The spectra of hydrogen and deuterium interpreted with an alternative fine structure constant compared to the CODATA recommended value.

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#### **Abstract**

A modified Dirac expression for the electron binding energy in hydrogen-like atoms is presented, which allows a direct and unambiguous comparison among different fine structure constants without bound-state QED theory. The least squares analysis of the parameters, describing the spectra of hydrogen and deuterium, is grounded on two sets of the most accurately measured energy separations. The optimal spectroscopic fine structure constant is found to be equal to 0.00 72 84(1), disagreeing with the determinations ultimately based on renormalized QED but being in good agreement with the number constant  $2^{-6}\pi^{-\frac{2}{3}} \approx 0.00$  72 84 28 . The present work compiles experimental values of the Lamb shift of S, P, and D states with n = 1, 2, and 3 derived from those measurements. Accurate predictions for hyperfine splitting intervals with  $n > 1$  are given and compared to experimental values for  $n = 2$ .

**Keywords:** fine structure constant, bound-state QED, Dirac binding energy, hydrogen-like atoms, Lamb shift, hyperfine splitting.

## **Introduction**

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Today, the fine structure constant  $\alpha_{codata} \approx 0.00$  72 97 35 is derived using a Dyson power series in powers of the fine structure constant  $\alpha$  for the anomalous magnetic moment of the electron. The coefficients of this power series are not measurable and are calculated using many hundreds of complicated Feynman multiloop diagrams of quantum electrodynamics (free QED) that only a handful of theoretical physicists can master. Determinations of the fine structure constant by other means (quantum Hall effect, ac Josephson effect) also depend on QED and produce identical results agreeing with each other with a precision of better than 1 part per 10<sup>8</sup>. This is not surprising because various methods must produce the same result irrespective of the correctness of QED since they are based on the same theory [1a]. Unfavorably, extracting the fine structure constant from QED itself is not possible [1b]. The spectrum of hydrogen as the main historical source of the value for the fine structure constant no longer plays a role in the calculation of its value<sup>1</sup>, and the spectrum is used solely to calculate the auxiliary Rydberg constant  $R_{\infty}$  of infinitely heavy mass, with the unit of energy (J) defined as  $R_\infty \equiv \frac{1}{2}$  $\frac{1}{2}\alpha^2 m_e c^2$ , acting by its definition as a

<sup>&</sup>lt;sup>1</sup> Quotation from Kramida 2010, p. 608: It is not easy to compare the experimental energy levels and transition frequencies of H, D, and T with the QED calculations because the latter are in fact adjusted to fit the experimental transition frequencies by adjusting the fundamental constants entering into the QED equations.

universal scaling factor for all transitions and substituting for either electron mass *me* or fine structure constant  $\alpha$  as required. The author [2a] has derived the fine structure constant  $\alpha_{\text{geom}}$ represented by the unit invariant number constant  $\alpha_{geom}\equiv~2^{-6}\pi^{-\frac{2}{3}}\approx 0.00$  72 84 28 ( $\alpha^{-1}\approx$  137.28). This idea does not require a formula for an observable quantity from which the fine structure constant is evaluated.

### **The model: an empirical modification to the Dirac equation**

In the following, an algebraic scaling formula is presented that allows to decide, using the spectra of hydrogen (H) and deuterium (D), whether  $\alpha_{codata}$  or  $\alpha_{geom}$  describes its spectrum better. In an arbitrary hydrogen-like atom, the electron binding energy  $E(n, \ell, j)$  in the natural energy (J) unit  ${m_e c^2}$ Codata 2010 shall be represented for  $S(\ell = 0)$ ,  $P(\ell = 1)$ , and  $D(\ell = 2)$  states as

$$
E(n,\ell,j) = E_D(n,j) \cdot \gamma \cdot \{1 + A + \delta_{\ell 0} B/n + \delta_{\ell 1} \delta_{k1} C/n + \delta_{\ell 1} \delta_{k2} D/n + \delta_{\ell 2} \delta_{k2} E/n + \delta_{\ell 2} \delta_{k3} F/n\}
$$
 (1)

 $\delta$  is the Kronecker delta function, and  $\gamma$  is a scaling factor of all levels that creates a fictitious Dirac particle with mass *γ*{*me*}Codata 2010 moving in the field of a stationary point nucleus, and reducing the two-body problem to an equivalent one-body problem. This concept is not the correct treatment of the relativistic two-body quantum problem and is arbitrarily chosen as the starting point (gross structure, Δn  $\neq$  0) needing corrections. The dimensionless, relativistic Dirac binding energy  $E_D(n, j)$  for a fixed point nucleus Coulomb potential is analytically given by

$$
E_D(n,j)=f(n,j)-1
$$

where [CODATA 2018, eq. 25; Kramida, eq. 3]

$$
f(n,j) = \left[1 + \left(\frac{Z\alpha}{n-\delta}\right)^2\right]^{-\frac{1}{2}}
$$

$$
\delta = k - [k^2 - (Z\alpha)^2]^{\frac{1}{2}}
$$

$$
k = j + 1/2
$$

*Z* is the nuclear integer-charge, *n* is a positive integer called the principal quantum number,  $\ell =$ 0,1 to  $n - 1$  is the orbital angular momentum, and  $j = \ell \pm 1/2$  is the total angular momentum of the electron, which results from combining the orbital motion of the electron with its intrinsic angular momentum called spin. The Dirac binding energy fails to take into account the nuclear mass and recoil and provides the same energy levels for each atom. Also, effects caused by the difference in the nuclear charge distribution are not included. In all expressions, the Planck constant  $\{h\}$ Codata 2010 is suppressed because it is solely used as a conversion factor to convert energy (J) to frequency (Hz).

Each hydrogenic atom has its own unique parameter set *A* to *F*, which must experimentally be determined. They are corrections to the scaled Dirac energy levels  $E_{D}(n, j)\gamma$ , neglecting the hyperfine structure (ΔF ≠ 0), and depend on the nucleus and the charge *Z*. Accounting for perturbations of levels with values of orbital angular momentum  $\ell > 2$  is easily possible by analogy. With formula (1), a nucleus with one electron infinitely far away and no other electrons nearby has zero binding energy, so all bound state energy levels are negative. The  $\ell$  dependent terms in formula (1) remove the degeneracy in  $\ell$  and produce a splitting of levels with the same value of *j* but different values of ℓ, for example, between 2S1/2 and 2P1/2 convincingly proven to

exist by WE Lamb and RC Retherford who reported a splitting of 1062(5) MHz [3]. For this reason, the splitting of levels ( $\Delta \ell \neq 0$ ) is later called Lamb shift and since then theoretically explained by bound state QED, in contrast to ordinary QED for free leptons.<sup>2</sup>

The energy difference ΔE between two energy states is given by  $\Delta E \equiv E(n_2,\ell_2,j_2) - E(n_1,\ell_1,j_1)$ . For the ionization energy of the ground state, this results in  $I = \Delta E = E(n_2 \rightarrow \infty, 0, 1/2)$  –  $E(1,0,1/2) = -E(1,0,1/2)$ . The ionization energy is the negative of the ground state energy, which is the largest energy for each atom. A common, accurate method of estimating absolute ionization energies is based on a fit of the modified Ritz formula. Using a Ritz series formula does not depend on theoretical calculations of the binding energy of any level and is an independent test of the validity of the fine structure constant (see note 'a' in Table 5 and Table 7).

According to formula (1), the transition 2S1/2–4S1/2 energy difference is

$$
\Delta E = \left( E_{D(4,1/2)} - E_{D(2,1/2)} \right) \gamma + A \left( E_{D(4,1/2)} - E_{D(2,1/2)} \right) \gamma + B \left( \frac{E_{D(4,1/2)}}{4} - \frac{E_{D(2,1/2)}}{2} \right) \gamma = \Delta E^{meas}
$$

and for the classic Lamb shift 2P1/2–2S1/2, which is difficult to measure because of the very short lifetime (natural line width 100 MHz) of the 2P1/2 state, the energy difference is

$$
\Delta E = (E_{D(2,1/2)} - E_{D(2,1/2)})\gamma + A(E_{D(2,1/2)} - E_{D(2,1/2)})\gamma + B\frac{E_{D(2,1/2)}}{2}\gamma - C\frac{E_{D(2,1/2)}}{2}\gamma
$$
  
=  $\frac{1}{2}(B - C)E_{D(2,1/2)}\gamma = \Delta E^{meas}$ 

For each transition, analogous relations can be written representing, in most cases, an overdetermined linear system of equations for the parameters *A, B, C, D, E,* and *F*, which has a least squares solution dependent on  $\gamma$  if the equations are linearly independent. The least squares solution best reproduces the input data or the results of measurements by means of expression (1). The energy  $E(n, \ell, i)$  calculated with the solutions of the normal equations using formula (1) is independent of the scale factor  $\gamma$  and only depends on  $Z\alpha$  and quantum numbers.<sup>3</sup> This formalism allows, based on a set of measured energy separations, a direct and unambiguous comparison among different fine structure constants without theory (bound-state QED), except the relativistic Dirac equation, in the analysis.

### **The fine structure constant**

 $\overline{\phantom{a}}$ 

In Table C, Kramida [4] tabulated the most accurately measured fine structure intervals (differences between the corresponding energy levels) for hydrogen. Table C of this article is a copy of Kramida's Table C with two additional measurements at the end for the transitions 1S1/2– 3S1/2 and 2S1/2–8D5/2. In order to derive the fine-structure energy levels listed in Table C from the available experimental data, purely theoretical corrections using  $\alpha_{codata}$  were necessary due to the hyperfine splitting (hfs) of one or both fine-structure levels involved in the measured transition. Thus, most transitions listed in Table C cannot be considered purely experimental, and small systematic errors are most likely to exist in the input data (see Kramida 2010 section 4).

<sup>2</sup> Relation 15.3 in [2b] is an incorrect ansatz because it does not take into account the Lamb shift. All results related to relation 15.3 must be reconsidered.

<sup>3</sup> This is because the scaling relations  $A' = \frac{1-\gamma+A}{\gamma}$  $\frac{\gamma+A}{\gamma}$  ,  $B'=\frac{B}{\gamma}$  $\frac{B}{\gamma}$  ,  $C'=\frac{C}{\gamma}$  $\frac{c}{\gamma}$  ,  $D'=\frac{D}{\gamma}$  $\frac{D}{\gamma}$  ,  $E' = \frac{E}{\gamma}$  $\frac{E}{\gamma}$  ,  $F'=\frac{F}{\gamma}$  $\frac{p}{\gamma}$  apply.

The reported intervals detailed in Table C were utilized to adjust the six parameters *A\_hyd* to *F\_hyd* by solving the linear system of equations. Measurements marked "not used" were not used to determine the best compromise values of *A\_hyd* to *F\_hyd*, which only approximately satisfy all measured intervals. In total, there are six degrees of freedom to fit 29 input values, that is, transition frequency measurements. The reliability of the observations was not included in the calculation due to unknown systematic uncertainties previously addressed.

The scaling factor has no influence on the energy values and can arbitrarily be set to one. Table C lists, for the 29 measurements considered, the calculated differences  $Diff(\alpha) \equiv \Delta E^{meas} - \Delta E(\alpha)$ for  $\alpha = \alpha_{codata\ 2010}$  and  $\alpha = \alpha_{geom}$ . The mean absolute deviation (*MAD*) of the 29 input values (results of measurements) is 4.03 MHz for  $\alpha_{codata\ 2010}$  and 0.82 MHz for  $\alpha_{geom}$ , respectively. For deuterium, Kramida [4] displayed the most accurately measured fine structure intervals in Table G replicated in Table G of this article. In order to adjust the parameters *A\_deut* to *F\_deut*, 19 linear equations in six unknowns were utilized, resulting in a mean absolute deviation of 4.75 MHz for  $\alpha_{Codd}$  and 0.80 MHz for  $\alpha_{geom}$ , respectively. Minimizing the function  $MAD(\alpha)$  yields the optimal fine structure constant  $\alpha_{min}$  without knowing the scaling factor, that is, the Rydberg constant  $R_{atom}$ , meaning that the experimental values extracted are disentangled from the measurement of the absolute value of the Rydberg constant. The values obtained are displayed with estimated error bars in Table 1, revealing for both isotopes that  $\alpha_{min} = 0.00$  72 84(1) is very close to  $\alpha_{geom}$  and disagrees violently with  $\alpha_{codata\,2010}$ .



Historically, the optimal spectroscopic fine structure constant  $\alpha_{min}$  agrees with that found by Houston [5], calculating a value of  $\alpha \approx 0.00$  72 85 ( $\alpha^2 \approx 5.307 \times 10^{-5}$ ) from purely spectroscopic measurements of the Rydberg constants for hydrogen and helium. Houston 1927 assumed in his evaluation that the relativity equation of Sommerfeld is applicable. In the year 1930, Millikan [6] comments that the experimental situation clearly favors, due to his oil drop work, a value of  $\alpha \approx$ 0.00 72 84 ( $\alpha^{-1} \approx 137.29$ ) that Birge [7] confirmed with the value  $\alpha^{-1} = 137.31(5)$  in 1932 by the simultaneous evaluation of the electron charge and the Planck constant from several known functional relations between these two constants. It would be interesting to look at the history of the measurements after Millikan to extract why the physics community settled down to the fine structure constant  $\alpha \approx 0.00$  72 97 35, and today is aimlessly shifting the accuracy to ever higher decimal places.

Kramida 2010 derived optimized energy levels from the measured fine structure intervals listed in Table C using a least-squares optimization (LOPT) code. The level optimization procedure involved several iterations using various interpolation and extrapolation procedures and the

fitting of the Ritz series formulas for the nD5/2 and nS1/2 series. Table 5 is a copy of Kramida's Table 5 with 148 frequency intervals, which resulted from the semi-empirical least-squares level optimization procedure. Absolute ionization energies tabulated by Kramida 2010 and by NIST are additionally included in Table 5. The mean absolute deviation of the 148 high precision determinations (Ritz values) from calculated frequencies by formula (1) is 3.67 MHz for  $\alpha_{\text{codata 2010}}$  and 0.78 MHz for  $\alpha_{\text{geom}}$ , respectively. Ritz values do not depend on theoretical calculations of the binding energy of any level.

For deuterium, the measured fine-structure intervals from Table G were used to derive the energy levels using the LOPT code. Table 7 lists Ritz values of 116 intervals resulting from the least squares level optimization procedure of Kramida, similar to that employed for hydrogen. In Table 7, the ionization energies tabulated by Kramida 2010 and NIST are also included. The mean absolute deviation of the 116 Ritz values from calculated frequencies by formula (1) is 2.96 MHz for  $\alpha_{\text{codata 2010}}$  and 1.15 MHz for  $\alpha_{\text{geom}}$ , respectively.

The mean absolute deviations of the intervals listed in the four tables are summarized below.



In all cases,  $\alpha_{geom}$  gives significantly smaller mean absolute deviations and describes, without further assumptions, the fine-structure energy levels of H and D more accurately applying the simple expression (1). In summary, there is convincing spectroscopic evidence with a great deal of experimental truth that the fine structure constant  $\alpha$  based on renormalized QED determinations should be discarded.

# **The Lamb shift**

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Without knowing the exact scale factor *γ* and the fine structure constant, Lamb shifts cannot be extracted from the spectra. First, a clear definition is vital to avoid different interpretations of the term Lamb shift. Using formula (1), the definition for the Lamb shift  $\mathcal{L}(nS1/2)$  shall be

$$
\mathcal{L}(nS1/2) \equiv \mathcal{L}(n, 0, 1/2) \equiv E_D(n, 1/2) \cdot \gamma \cdot \{A + B/n\} = E_D(n, 1/2) \cdot \gamma_{zero} \cdot B/n
$$

which can easily be generalized to other Lamb shifts  $\mathcal{L}(n, \ell, j)$ . The choice of  $\gamma$  defines the values of the Lamb shifts, which are to be understood as effects in addition to what can be obtained from the scaled Dirac equation. In the literature, the reference point is not unique due to different corrections applied and, in most cases, even undefined. One possibility for an "experimental" reference point is to choose the scaling factor such that for  $\gamma = \gamma_{zero}$  the parameter *A* is zero, which implicitly defines  $\gamma_{zero}$  by  $A(\gamma_{zero}) = 0$  or explicitly<sup>4</sup> by  $\gamma_{zero} = 1 + A(\gamma = 1)$ . Corrections that affect all states are ascribed to the scaling factor  $\gamma_{zero}$  and the parameters *B* to *F*, which may

<sup>&</sup>lt;sup>4</sup> The explicit expression follows from the scaling relation  $A' = 0 = \frac{1 - \gamma_{zero} + A(\gamma = 1)}{\gamma_{zero}}$ .

be called dimensionless Lamb shift constants, include all corrections that cannot be described by  $\gamma_{zero}$ . The scaling factor  $\gamma_{zero}$  completely relies on an experimental input, to wit, Table C or Table G. The adjusted dimensionless parameters *B* to *F* and the experimentally determined absolute Lamb shifts for  $n = 1, 2$ , and 3 are detailed for  $\gamma = \gamma_{zero}$  in Table 1. No theoretical calculations except {*mec <sup>2</sup>/h*}Codata 2010 are involved to evaluate the experimental data for the absolute Lamb shifts from the input data.

For a nucleus of finite mass *M*, nuclear motion is accounted for nonrelativistically by replacing the electron mass *me* by the reduced mass, which multiplies all energy levels by the reduced mass correction factor  $\gamma_{red \; mass} \equiv \left(1 + \frac{m_e}{M}\right)$  $^{-1}$ . These factors are tabulated for hydrogen and deuterium in Table 1 using CODATA 2010 values for the electron-nuclear mass ratios based on high precision Penning trap mass spectrometry. Comparing the factors  $\gamma_{red \, mass \, Codata \, 2010}$  for both isotopes with the experimental scaling factors  $\gamma_{zero}$  ( $\alpha_{codata 2010}$ ) reveals that they are nearly equal. Interestingly, if the experimental scaling factors  $\gamma_{zero}(\alpha_{geom})$  are divided by the correction factor

$$
\gamma_{corr} \equiv \left(\frac{\alpha_{codata\ 2010}}{\alpha_{geom}}\right)^2 \approx 1.003\ 591\ 800
$$

there results 0.999 455 690 for H as compared with the CODATA 2010 reduced mass factor 0.999 455 679, and 0.999 727 643 for D as compared with 0.999 727 631, which means that the simple reduced mass correction factors for both isotopes are derived for  $\alpha_{geom}$  very accurately from essentially experimental spectroscopic data, implying a reduction of the relativistic two body problem to an equivalent one body equation in terms of a single effective mass. The deviations relative to the tabulated reduced mass factors  $\gamma_{red \, mass \, Codata \, 2010}$  are 1.0 parts per 10<sup>8</sup> for H and 1.2 parts per 108 for D, respectively. The close agreement

$$
\gamma_{zero}(atom) \approx \gamma_{red \, mass}(atom) \times \gamma_{corr}
$$

is quite remarkable, since the concept of reduced mass has no theoretical basis in relativistic quantum mechanics, and the assumption that the nucleus behaves inertly and plays no role other than its mass is an idea of the classic Bohr model. The factor  $\gamma_{corr}$  might be interpreted as a correction of the spectroscopic electron mass  ${m_e}$ Codata 2010 deduced by CODATA from experimental data through a least squares adjustment with the fine structure constant  $\alpha_{codata\,2010}$ determined by other independent measurements (non-spectroscopic), assuming for the energy level the expression [CODATA 2010, equ. 22]

$$
E(n,\ell,j) = -\frac{Z^2 R_\infty}{n^2} \{1 + \delta(n,\ell,j)\} \equiv -\frac{Z^2 R_\infty}{n^2} F_{n\ell j}^{rel}
$$

where  $\delta(n, \ell, j)$  is a dimensionless theoretical correction factor, small compared to one, that contains the details of the bound-state QED apparatus of each energy level, including the effect of the finite size of the nucleus as a function of the rms charge radius. Only the Rydberg constant  $R_{\infty}$ is an adjusted constant in the numerical evaluation carried out by CODATA. The method mentioned above consists of comparing measured transitions (primarily the 1S–2S frequency in H and the H–D isotope shift of the 1S–2S frequency) with intervals calculated from a complex theoretical equation for each energy level scaled with  $R_{\infty}$ , which indirectly assigns, using  $\alpha_{codd}$ a fitted value to the absolute electron mass { $m_e$ }<sub>Codata</sub> that can by no means be directly compared (i.e., without a theoretical contribution) to the artifact SI standard kilogram.

In order to clarify the significance of the theoretical corrections to the experimental data mentioned in the text, only the two gross structure transitions 1S1/2–2S1/2 and 2S1/2–8S1/2

from Table C (Kramida 2010, column largest theoretical correction) were fitted to the data points, from which "exact" values for [*γzero*] and the parameter [*B*], scaling the Lamb shifts of the nS1/2 levels in terms of the principal quantum number n, can be extracted. The results, reproducing the reduced mass factor of hydrogen without significant change, are displayed in brackets in Table 1, which give 2 466 061 413.1859 MHz for the 1S1/2–2S1/2 intervall and 770 649 350.012 MHz for the  $2S1/2-8S1/2$  intervall. The measured values are marked with an asterix (\*) in Table C.<sup>5</sup> The fitted parameter [*B*] is an accurate experimental value admitting a direct comparison with nS1/2 Lamb shift calculations.

# **Tritium**

Since tritium (T) is radioactive and difficult to handle, there is little information about its spectrum from which a reliable parameter set can be obtained. The Lamb shift parameters *B* to *F* of hydrogen and deuterium are similar in value, which suggests that the most accurate known Lamb shift parameters *B* to *F* of hydrogen could be used as a first approximation for tritium. The most accurate measured values of four fine structure transitions arranged by Kramida [4] in Table K are listed below and compared with the values calculated using formula (1),  $\alpha = \alpha_{a\text{eom}}$ ,  $A = 0$ , the parameters *B* to *F* of hydrogen (Table 1), and  $\gamma = \gamma_{red \, mass} (T) \times \gamma_{corr}$ .



# **The hyperfine splitting**

Formula (1) considers the Lamb shifts as a perturbation to the scaled Dirac energy levels  $E_D(n, j)$ y. In the following, the hyperfine splitting manifested as a small splitting of the fine-structure energy levels is regarded, in analogy to the Lamb shift, as a perturbation to  $E(n, \ell, j)$  of the form

$$
E(n,\ell,j,F) = E(n,\ell,j)\{1 - Q(\ell,j,I,F) \cdot Z \cdot \widetilde{B}/n\}
$$
\n(2)

The number *I* is the spin of the nucleus, and  $F$  is the total angular momentum for the whole atom with the possible values being  $j + I$ ,  $j + I - 1$ , ...,  $|j - I|$ . Formula (2) can be deduced from the rephrased equations 5 and 6 (without off-diagonal terms) given by Kramida 2010 by setting

$$
\frac{Z^2 R_{\infty}}{n^2} F_{n\ell j}^{rel} \equiv -E(n, \ell, j)
$$

$$
\frac{[F(F + 1) - I(I + 1) - j(j + 1)]}{j(j + 1)(2\ell + 1)I} \equiv Q
$$

 $\overline{\phantom{a}}$ 

<sup>&</sup>lt;sup>5</sup> The exact measurements are not obtained because the numbers in Table 1 are rounded to 12 decimal places.

$$
\alpha^2 \frac{\mu_{nucl}}{\mu_B} \equiv \tilde{B}
$$

The minus sign in relation (2) ensures that the energy of the F level is lifted and that of the F-1 is lowered. From formula (2), the hfs interval between two adjacent levels in a hyperfine multiplet can be written as

$$
(\Delta E)^{hfs}_{n\ell j} = E(n,\ell,j,F) - E(n,\ell,j,F-1) = -E(n,\ell,j) \cdot \frac{2F}{j(j+1)(2\ell+1)I} \cdot Z \cdot \tilde{B}/n \tag{3}
$$

wich takes into accout Lamb shifts of energy levels. Because the relativistic Dirac theory naturally implies the electron's magnetic dipole moment to be exactly  $\mu_e = \mu_B$ , the ratio  $\frac{\mu_e}{\mu_B}$  in equation (6) given by Kramida 2010 was set to one. From simplicity requirements on the approximating function, the reduced mass prefactor  $\gamma_{red~mass}^{3}$  has been omitted, assuming that a correction in the form of an overall reduced mass factor is adequate. These assumptions make relation (3) compatible with the uncorrected expression (22.13) of Bethe and Salpeter [10], and equation (41) of Grifffiths [11] setting  $g_e$  = 2.

The value of  $\frac{\mu_{nucl}}{\mu_B}$  is not directly accessible experimentally, but it can be traced back to maser experiments by Winkler and coworkers, who determined the bound particle ratio of the magnetic moments of the nucleus and electron  $\frac{\mu_{nucl}(atom)}{\mu_e(atom)}$ , implying that the bound electron moment in Bohr magnetons  $\frac{\mu_e (atom)}{n}$  $\frac{atom}{\mu_B}$  must also be known. This ratio can be approximated by  $\frac{\mu_e(free)}{\mu_B}$  $\frac{f(\theta)}{\mu_B} \equiv 1 + a_e$ , where  $a_e$  [Codata 2010] is the very accurately measured electron magnetic moment anomaly. Replacing the fine structure constant  $\alpha$  by  $\alpha_{geom}$  entails that, in addition to 1+ $a_e$ , the multiplying factor  $\gamma_{corr}$  must be taken into account, since the electron mass is given by  $\gamma_{corr}\{m_e\}_{codata}$ . Combining all this gives

$$
\tilde{B} \equiv \alpha^2 \frac{\mu_{nucl}}{\mu_B} = \alpha^2 \cdot \frac{\mu_{nucl}}{\mu_e} \cdot \frac{\mu_e}{\mu_B} \approx \alpha^2 \cdot \frac{\mu_{nucl}}{\mu_e} \cdot (1 + a_e) \approx \alpha_{geom}^2 \cdot \frac{\mu_{nucl}}{\mu_e} \cdot (1 + a_e) \gamma_{corr}
$$
(4)

The following table lists the hyperfine splittings of hydrogen and deuterium, which were calculated using relations (3) and hypothesis (4). The scaling factor  $\gamma_{zero}$  and the parameter *B* used to compute  $E(n, \ell, j)$  correspond to the values listed in Table 1.



The comparison of theoretical results with experimental values shows a remarkable agreement (or coincidence?) for hydrogen, and reflects an unclear discrepancy although small with a substantial correction factor for deuterium. This might be due to the fact that the ratio of the magnetic moment of the deuteron to the magnetic moment of the electron in the 1S state of deuterium has never been published. This magnetic ratio should be reevaluated.

Due to the lack of knowledge on the number of multiplying factors that control hypothesis (4), the dimensionless parameter  $\tilde{B}$ , responsible for the splitting, is difficult to calculate. But the existence of high-precision experimental data on the 1S hfs splitting makes it possible to derive an empirical,

simple scaling law for other hfs intervals, if the dimensionless parameter  $\tilde{B}$  is the same for all levels. For hydrogen, the scaling law from the ground-state hyperfine splitting is then

$$
(\Delta E)^{hfs}_{n\ell j}(F-1;F) = (\Delta E)^{hfs}_{1S1/2}(0;1) \cdot \frac{E(n,\ell,j)}{E(1,0,1/2)} \cdot \frac{F}{j(j+1)(2\ell+1)} \cdot \frac{3}{4} \cdot \frac{1}{n}
$$
(5)

For the scaling law from the ground-state hyperfine splitting of deuterium, the factor 3/4 in formula (5) must be replaced by 1/2. In Table 2, theoretical predictions using the scaling relation (5) are given and compared to experimental results for 2S1/2 states. For both isotopes, the agreement between theory and experiment is impressive with, surprisingly, a relative deviation of 30 ppm that is independent of the nucleus. Unfortunately, besides the 2P1/2 hfs splitting in hydrogen of 59.22(14) MHz [Kramida 2010, Table A], no reliable experimental data are available for other excited states. Theoretical values, calculated by different authors using QED with  $\alpha_{codata}$ , can be found in Kramida's work, allowing a comparison with the values calculated by means of expression (5).

## **References**

- [1] The fine structure constant. Toichiro Kinoshita Rep Prog Phys **59** 1459 (1996) a) section 2.10, section 3 b) section 4
- [2] On the Origin of Natural Constants. Hans Peter Good, De Gruyter Berlin (2018) a) p. 42 b) p. 184
- [3] Fine Structure of the Hydrogen Atom. Part II W E Lamb and R C Retherford Phys Rev **81**, 222 (1951)
- [4] A critical compilation of experimental data on spectral lines and energy levels of hydrogen, deuterium, and tritium. A E Kramida Atomic Data and Nuclear Data Tables **96** 586 (2010)
- [5] A spectroscopic Determination of e/m. W V Houston Phys Rev **30** 608 (1927), p. 613
- [6] The most probable 1930 values of the electron and related constants. R A Millikan Phys Rev **35** 1231 (1930), p. 1235
- [7] Probable Values of  $e$ ,  $h$ ,  $e/m$ , and  $\alpha$ . R T Birge Phys Rev **40** 228 (1932)
- [8] CODATA 2018, Table X
- [9] Measurement of the 2S1/2–8D5/2 transition in hydrogen. A D Brandt et al Phys Rev Lett **128** 023001 (2022)
- [10] Quantum Mechanics of One- and Two-Electron Atoms. H A Bethe, E E Salpeter, Springer-Verlag Berlin (1957), p. 110
- [11] Hyperfine splitting in the ground state of hydrogen. D J Griffiths Am J Phys **50** 698 (1982), p. 701
- [12] CODATA 1998 a) Table XIV.B.1, B9 : Winkler et al Phys Rev A **5** 83 (1972) b) Table XIV.B.1, B10 : Phillips et al (1984), private communication, unpublished result



**Table 1:** The adjusted dimensionless parameters *A* to *F* and the absolute Lamb shifts for *n* = 1, 2, and 3.

### **Notes:**

The value  $\alpha_{min}$  is the result of minimizing the mean absolute deviation based on Table C or Table G using formula (1) and the solutions of the linear system of equations. The minima were determined by quadratic regression of 13 values of the discretized function *MAD(α;Table C)* or *MAD(α;Table G)* with *α* = [0.00 72 78, 0.00 72 90] and *Δα* = 0.00 00 01.

The grey-shaded fields present the adjusted dimensionless parameters in formula (1) to compute the essentially experimental binding energy  $E(n, \ell, j)$  for  $\alpha_{geom}$ . All numbers are rounded to 12 decimal places.

For the classic Lamb shift 2P1/2–2S1/2 of hydrogen the energy difference  $E(2,1,1/2) - E(2,0,1/2)$  is 1058.86 MHz using {*mec <sup>2</sup>/h*}Codata 2010.

			hydrogen		deuterium	
$(\Delta E)_{hfs}$	1S1/2	(MHz)	1420.405 751 768(1)	a)	327.384 352 5222(17)	a)
$\tilde{B}$		(ppm)	0.080 997 276	b)	0.024 884 934	b)
$(\Delta E)_{hfs}$	2S1/2	(MHz)	177.55 15		40.92 32	
			177.55 69	a)	40.9245	a)
	3S1/2		52.6077		12.12 54	
	4S1/2		22.1938		5.1154	
	5S1/2		11.3632		2.6191	
	6S1/2		6.5759		1.51 57	
	7S1/2		4.14 11		0.95 45	
	8S1/2		2.7742		0.6394	
	9S1/2		1.94 84		0.4491	
	10S1/2		1.42 04		0.3274	
	11S1/2		1.0672		0.24 60	
	12S1/2	(MHz)	0.82 20		0.1895	
	$2P1/2(F = 1)$	(MHz)	59.1839			
	$2P3/2(F = 2)$		23.6733			
	$3P1/2(F = 1)$		17.53 59			
	$3P3/2(F = 2)$		7.0143			
	$3D3/2(F = 2)$		4.2086			
	$3D5/2(F = 3)$		2.70 55			
	$4D5/2(F = 3)$		1.14 14			
	$4F7/2(F = 4)$		0.6039			
	$7F5/2(F = 3)$		0.15 21			
	$8D3/2(F = 2)$		0.22 19			
	$8D5/2(F = 3)$		0.14 27			
	$8F5/2(F = 3)$	(MHz)	0.10 19			
	$4D5/2(F = 7/2)$	(MHz)			0.2046	
					0.14 62	
	$4D5/2(F = 5/2)$					
	$8D5/2(F = 7/2)$				0.02 56	
	$8D5/2(F = 5/2)$	(MHz)			0.0183	

**Table 2:** Absolute values of the hyperfine splitting frequencies in H and D calculated by scaling from the hfs of 1S1/2.

#### **Notes:**

Hfs values without a reference are obtained with the help of formula (5) and can be compared with theoretical values of Kramida 2010 [hydrogen: Table A and Table 1; deuterium: Table B and Table 2].

a) Experimental value [Kramida 2010, p. 591].

b) Calculated from the 1S hfs using formula (3).

interval		Measured value (MHz)	Unc. $(MHz)$		Diff (MHz)
				$\alpha_{geom}$	$\alpha_{\text{codata}}$
$1S1/2 - 2S1/2$	$\ast$	2466061413.187074	0.000034	$-0.1$	2.1
2P1/2-2S1/2		1057.847	0.09	$-1.0$	$-2.3$
2P1/2-2P3/2		10969.13	0.1	$-0.1$	$-0.1$
$2S1/2 - 2P3/2$		9911.201	0.012	0.8	2.1
$2P1/2 - 3D3/2$	not used	456685852.8	1.7		
2S1/2-3P1/2		456681549.9	0.3	$-1.5$	$-5.1$
2S1/2-3P3/2		456684800.1	0.3	$-1.5$	$-5.0$
3P1/2-3D3/2	not used	456675968.3	3.4		
$2S1/2-4P1/2$		616520017.568	0.015	$-0.3$	$-3.0$
$2S1/2 - 4S1/2$		616520150.636	0.01	0.4	$-2.4$
2S1/2-4P3/2		616521388.672	0.01	$-0.3$	$-3.0$
2S1/2-4D5/2		616521843.441	0.024	1.3	$-9.6$
$2S1/2 - 6S1/2$		730690017.097	0.021	0.2	$-0.1$
$2S1/2 - 6D5/2$		730690518.592	0.011	0.4	$-2.3$
2S1/2-8S1/2	$\ast$	770649350.012	0.09	0.3	1.5
$2S1/2 - 8D3/2$		770649504.45	0.08	0.0	0.4
$2S1/2 - 8D5/2$		770649561.584	0.06	0.3	0.5
2S1/2-10D5/2		789144886.411	0.039	0.2	1.7
2S1/2-12D3/2		799191710.473	0.09	0.1	2.4
2S1/2-12D5/2		799191727.404	0.07	0.2	2.4
3P1/2-3S1/2		314.818	0.048	1.1	0.7
3P1/2-3D3/2	not used	3244.9	3.1		
$3S1/2 - 3P3/2$	not used	2933.5	1.2		
$3S1/2 - 3D3/2$		2929.9	0.8	$-3.8$	$-18.9$
$3S1/2 - 3D5/2$		4013.155	0.048	2.6	$-16.5$
3D3/2-3P3/2		5.5	0.9	2.7	18.3
3D3/2-3D5/2	not used	1083	0.29		
$3P3/2 - 3D5/2$	not used	1078	1.1		
$4P1/2 - 4S1/2$		133.2	0.6	0.8	0.7
			0.22	$-0.3$	$-0.3$
$4P1/2 - 4P3/2$		1370.85			$-5.5$
$4P1/2 - 4D3/2$		1371.1	1.2	1.1	
$4S1/2 - 4D3/2$	not used	1235	2.1		
$4S1/2 - 4P3/2$		1237.79	0.29	$-1.0$	$-0.8$
$4S1/2 - 4D5/2$		1693	0.4	1.0	$-7.0$
$4D3/2 - 4F5/2$	not used	456.8	1.6		
$4D3/2 - 4D5/2$	not used	458	2.2		
$4P3/2 - 4D5/2$	not used	455.7	1.6		
$4D5/2 - 4F7/2$	not used	227.96	0.41		
$5P1/2 - 5S1/2$	not used	64.6	5		
$5P1/2 - 5D3/2$	not used	704	7		
$5S1/2 - 5P3/2$	not used	622	10		
5P3/2-5D5/2	not used	232.2	2.9		
5D5/2-5F7/2	not used	117	1.5		
$1S1/2 - 3S1/2$	[8]	2922743278.678	0.013	0.2	$-1.5$
$2S1/2 - 8D5/2$	$[9]$	770649561.5709	0.02	0.3	0.5

**Table C**: The best available measurements of fine-structure intervals in *hydrogen* [4].





interval	$\Delta E_{\text{Ritz}}$ (MHz)	Unc. (MHz)		$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)
	(Ritz values)		$\alpha_{geom}$	$\alpha_{\text{codata}}$
Ionization <sup>a)</sup>	3288086856.8	0.7	$-0.5$	5.8
Ionization b)	3288086857.128	0.003	$-0.1$	6.1
$1S1/2-2P1/2$	2466060355.339	0.009	0.9	4.4
1S1/2-2S1/2	2466061413.18707	0.00003	$-0.1$	2.1
$1S1/2-2P3/2$	2466071324.389	0.012	0.7	4.2
$1S1/2-3P1/2$	2922742963.15	0.21	$-1.6$	$-2.9$
$1S1/2 - 3S1/2$	2922743277.97	0.22	-0.5	-2.2
1S1/2-3P3/2	2922746213.24	0.21	$-1.7$	$-2.9$
	3082581430.756	0.015	$-0.4$	$-0.8$
$1S1/2-4P1/2$			0.3	$-0.3$
$1S1/2 - 4S1/2$	3082581563.823	0.01		
$1S1/2 - 4P3/2$	3082582801.858	0.01	$-0.5$	$-0.9$
$1S1/2 - 5P1/2$	3156563616.6	1.1	0.0	0.9
$1S1/2 - 5S1/2$	3156563684.8	1.1	0.4	1.3
$1S1/2 - 5P3/2$	3156564318.6	1.1	$-0.1$	0.9
$1S1/2 - 5D5/2$	3156564549.7	0.7	-1.0	$-4.2$
$1S1/2-6P1/2$	3196751390.79	0.3	$-0.2$	1.8
$1S1/2 - 6S1/2$	3196751430.284	0.021	0.1	2.0
1S1/2-6P3/2	3196751797.05	0.03	$-0.2$	1.8
$1S1/2-7P1/2$	3220983314.5	1.2	$-0.2$	2.7
$1S1/2 - 7S1/2$	3220983339.4	1.2	0.0	2.8
$1S1/2 - 7P3/2$	3220983570.4	1.2	$-0.2$	2.7
$1S1/2 - 7D5/2$	3220983655.4	0.7	0.3	1.6
$1S1/2 - 8P1/2$	3236710746.525	0.018	0.0	3.5
$1S1/2 - 8S1/2$	3236710763.199	0.009	0.1	3.6
1S1/2-8P3/2	3236710917.916	0.019	0.0	3.5
1S1/2-9P1/2	3247493411.9	1.2	0.2	4.1
$1S1/2 - 9S1/2$	3247493423.6	1.2	0.2	4.2
1S1/2-9P3/2	3247493532.3	1.2	$0.2\,$	4.1
1S1/2-9D5/2 3247493572		0.7	0.1	3.3
1S1/2-10P1/2	3255206183.1	1.2	0.3	4.7
1S1/2-10S1/2	3255206191.6	1.2	0.4	4.7
1S1/2-10P3/2	3255206270.8	$1.2$	$0.3\,$	4.6
1S1/2-11P1/2	3260912757.7	$\frac{1}{1.2}$	0.3	4.9
1S1/2-11S1/2	3260912764.1	1.2	$0.4\,$	5.0
1S1/2-11P3/2	3260912823.6	1.2	0.3	4.9
1S1/2-11D5/2	3260912845.1	0.7	$0.0\,$	4.2
1S1/2-12P1/2	3265253073.3	$\frac{1}{1.2}$	0.3	5.1
1S1/2-12S1/2	3265253078.2	1.2	0.3	5.1
		1.2	$0.2\,$	5.1
1S1/2-12P3/2	3265253124			$-2.3$
2P1/2-2S1/2	1057.848	0.09	$-1.0$ $-0.2$	$-0.2$
2P1/2-2P3/2	10969.05	0.015		
2P1/2-3S1/2	456682922.63	0.22	-1.4	-6.6

**Table 5:** Frequencies of fine-structure transitions in *hydrogen* derived from the level-optimization procedure [Kramida 2010, Table 5].

interval	$\Delta E_{\text{Ritz}}$ (MHz)	Unc. (MHz)		$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)
	(Ritz values)		$\alpha_{geom}$	$\alpha_{codata}$
$2P1/2 - 3D3/2$	456685852.6	0.6	$-5.0$	$-25.5$
2P1/2-4S1/2	616521208.484	0.013	$-0.6$	$-4.7$
2P1/2-4D3/2	616522444.5	0.23	$-2.2$	$-12.7$
2P1/2-5S1/2	690503329.4	1.1	$-0.5$	$-3.2$
2P1/2-5D3/2	690503960.5	0.7	$-3.1$	$-9.0$
2P1/2-6S1/2	730691074.945	0.023	$-0.8$	$-2.4$
2P1/2-6D3/2	730691441.05	0.04	$-1.4$	$-4.9$
2P1/2-7S1/2	754922984.1	1.2	$-0.9$	$-1.5$
2P1/2-7D3/2	754923214.8	0.7	$-1.1$	$-3.0$
2P1/2-8S1/2	770650407.86	0.012	$-0.8$	$-0.8$
2P1/2-8D3/2	770650562.298	0.012	$-1.0$	$-1.9$
2P1/2-9S1/2	781433068.3	1.2	$-0.6$	$-0.2$
2P1/2-9D3/2	781433176.5	0.7	$-1.0$	$-1.2$
2P1/2-10D3/2	789145915	0.05	$-1.0$	$-0.6$
2P1/2-11S1/2	794852408.8	1.2	$-0.5$	0.6
2P1/2-11D3/2	794852467.8	0.7	$-1.0$	$-0.2$
2P1/2-12S1/2	799192722.9	1.2	$-0.5$	0.8
2P1/2-12D3/2	799192768.321	0.013	$-1.0$	0.1
$2S1/2 - 2P3/2$	9911.202	0.012	0.8	2.1
$2S1/2-3P1/2$	456681549.96	0.21	$-1.4$	$-5.0$
$2S1/2 - 3S1/2$	456681864.78	0.22	$-0.4$	$-4.3$
$2S1/2 - 3P3/2$	456684800.05	0.21	-1.5	$-5.1$
$2S1/2 - 4P1/2$	616520017.569	0.015	$-0.3$	$-2.9$
2S1/2-4S1/2	616520150.636	0.01	0.4	-2.4
$2S1/2 - 4P3/2$	616521388.671	0.01	$-0.3$	$-3.0$
2S1/2-4D5/2	616521843.443	0.024	1.3	-9.6
2S1/2-5P1/2	690502203.4	1.1	0.1	$-1.2$
$2S1/2 - 5S1/2$	690502271.6	1.1	0.5	$-0.9$
$2S1/2 - 5P3/2$	690502905.4	1.1	0.1	$-1.3$
2S1/2-6P1/2	730689977.6	0.03	$-0.1$	$-0.3$
$2S1/2 - 6S1/2$	730690017.097	0.021	0.2	$-0.1$
2S1/2-6P3/2	730690383.86	0.04	$-0.1$	$-0.3$
$2S1/2 - 6D5/2$	730690518.592	0.011	0.4	$-2.3$
2S1/2-7P1/2	754921901.3	1.2	$-0.1$	0.5
$2S1/2 - 7S1/2$	754921926.2	1.2	0.1	0.7
2S1/2-7P3/2	754922157.2	1.2	0.0	0.6
2S1/2-8P1/2	770649333.338	0.018	0.1	1.4
$2S1/2 - 8S1/2$	770649350.012	0.009	0.3	1.5
$2S1/2 - 8D3/2$	770649504.45	0.08	0.0	0.4
2S1/2-8P3/2	770649504.729	0.019	0.1	1.4
$2S1/2 - 8D5/2$	770649561.584	0.007	0.3	0.5
$2S1/2 - 9P1/2$	781431998.7	1.2	0.3	$2.0\,$
2S1/2-9S1/2	781432010.4	$1.2\phantom{0}$	$0.4\,$	2.0

**Table 5:** Frequencies of fine-structure transitions in *hydrogen* derived from the level-optimization procedure [Kramida 2010, Table 5].

interval	$\Delta E_{\text{Ritz}}$ (MHz)	Unc. (MHz)		$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)
	(Ritz values)		$\alpha_{geom}$	$\alpha_{\text{codata}}$
2S1/2-9P3/2	781432119.1	1.2	0.3	2.0
2S1/2-10P1/2	789144769.9	1.2	0.5	2.5
2S1/2-10S1/2	789144778.4	1.2	0.5	2.5
2S1/2-10P3/2	789144857.6	1.2	0.4	2.5
2S1/2-10D5/2	789144886.41	0.4	0.2	1.7
2S1/2-11P1/2	794851344.5	1.2	0.4	2.8
2S1/2-11S1/2	794851350.9	1.2	0.5	2.8
2S1/2-11P3/2	794851410.4	1.2	0.4	2.8
2S1/2-12P1/2	799191660.1	1.2	0.4	3.0
2S1/2-12S1/2	799191665	1.2	0.4	3.0
2S1/2-12D3/2	799191710.473	0.01	0.1	2.4
2S1/2-12P3/2	799191710.9	1.2	0.4	3.0
2S1/2-12D5/2	799191727.404	0.007	0.2	2.4
2P3/2-3S1/2	456671953.58	0.22	$-1.1$	$-6.4$
2P3/2-3D3/2	456674883.5	0.6	$-4.9$	$-25.3$
2P3/2-3D5/2	456675966.74	0.22	1.4	$-22.9$
2P3/2-4S1/2	616510239.434	0.016	$-0.4$	$-4.5$
2P3/2-4D3/2	616511475.45	0.23	$-2.0$	$-12.5$
2P3/2-4D5/2	616511932.24	0.03	0.5	$-11.7$
2P3/2-5S1/2	690492360.4	1.1	$-0.3$	$-2.9$
2P3/2-5D3/2	690492991.4	0.7	$-2.9$	$-8.9$
2P3/2-5D5/2	690493225.3	0.7	$-1.7$	$-8.4$
2P3/2-6S1/2	730680105.895	0.024	$-0.6$	$-2.2$
2P3/2-6D3/2	730680472	0.04	$-1.2$	-4.7
2P3/2-6D5/2	730680607.39	0.016	$-0.4$	$-4.4$
2P3/2-7S1/2	754912015	1.2	$-0.7$	-1.4
2P3/2-7D3/2	754912245.7	0.7	$-0.9$	$-2.8$
$2P3/2 - 7D5/2$	754912331	$0.7\,$	$-0.4$	$-2.6$
2P3/2-8S1/2	770639438.81	0.015	$-0.5$	$-0.6$
2P3/2-8D3/2	770639593.248	0.015	$-0.8$	$-1.7$
$2P3/2 - 8D5/2$	770639650.382	0.014	$-0.5$	$-1.5$
2P3/2-9S1/2	781422099.2	1.2	$-0.4$	$0.0\,$
2P3/2-9D3/2	781422207.5	0.7	$-0.8$	$-0.9$
2P3/2-9D5/2	781422247.6	0.7	$-0.6$	$-0.9$
2P3/2-10S1/2	789134867.2	1.2	$-0.3$	0.5
			$-0.8$	$-0.4$
2P3/2-10D3/2 2P3/2-10D5/2	789134945.95	0.05 0.04	$-0.6$	$-0.3$
	789134975.21	1.2	$-0.3$	0.8
2P3/2-11S1/2	794841439.7			
2P3/2-11D3/2	794841498.7	0.7	$-0.8$	$-0.1$
2P3/2-11D5/2	794841520.7	$0.7\,$	$-0.7$	0.0
2P3/2-12S1/2	799181753.8	1.2	$-0.4$	0.9
2P3/2-12D3/2	799181799.271	0.015	$-0.7$	0.3
2P3/2-12D5/2	799181816.202	0.014	$-0.6$	0.4

**Table 5:** Frequencies of fine-structure transitions in *hydrogen* derived from the level-optimization procedure [Kramida 2010, Table 5].

interval	$\Delta E_{\text{Ritz}}$ (MHz)	Unc. $(MHz)$		$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)
	(Ritz values)		$\alpha_{geom}$	$\alpha_{\text{codata}}$
3P1/2-3S1/2	314.82	0.05	1.1	0.7
$3P1/2 - 3D3/2$	3244.8	0.6	$-2.6$	$-18.2$
3P1/2-3P3/2	3250.09	0.03	$-0.1$	0.0
$3S1/2 - 3D3/2$	2929.9	0.6	$-3.8$	$-18.9$
$3S1/2 - 3P3/2$	2935.27	0.06	$-1.2$	$-0.8$
$3S1/2 - 3D5/2$	4013.16	0.05	2.6	$-16.5$
3D3/2-3P3/2	5.3	0.6	2.5	18.1
$3D3/2 - 3D5/2$	1083.2	0.6	6.3	2.4
3P3/2-3D5/2	1077.89	0.07	3.7	$-15.8$
$4P1/2 - 4S1/2$	133.067	0.018	0.7	0.6
4P1/2-4P3/2	1371.102	0.018	$-0.1$	0.0
$4P1/2 - 4D3/2$	1369.08	0.23	$-0.9$	$-7.5$
$4S1/2 - 4D3/2$	1236.02	0.23	$-1.6$	$-8.0$
4P3/2-4D5/2	454.77	0.03	1.6	$-6.6$
$5P1/2 - 5S1/2$	68.201	0.019	0.4	0.4
5P1/2-5D3/2	699.2	1.3	$-2.2$	$-5.6$
5P1/2-5P3/2	702.019	0.012	0.0	0.0
5S1/2-5P3/2	633.818	0.022	$-0.4$	$-0.4$
$5D3/2 - 5D5/2$	233.92	0.08	1.3	0.5
5P3/2-5D5/2	231.1	1.3	$-0.9$	$-5.1$
8D3/2-8D5/2	57.134	0.01	0.3	0.1
12D3/2-12D5/2	16.931	0.012	0.1	0.1

**Table 5:** Frequencies of fine-structure transitions in *hydrogen* derived from the level-optimization procedure [Kramida 2010, Table 5].

### **Notes:**

a) Ritz series limit [Kramida 2010, Table D].

b) NIST: Atomic Spectra Database 78 [version 5.11]: Ionization Energies Form. Energy has been determined from bound-state QED ab initio calculations.

interval	$\Delta E_{\text{Ritz}}$ (MHz)	Unc. (MHz)		$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)
	(Ritz values)		$\alpha_{geom}$	$\alpha_{codata}$
Ionization <sup>a)</sup>	3288981521.1	2.3	$-1.1$	3.1
Ionization b)	3288981522.062	0.003	$-0.09$	4.1
$1S1/2 - 2P1/2$	2466731348.24	0.06	0.7	2.5
$1S1/2 - 2S1/2$	2466732407.52171	0.00015	0.1	0.5
$1S1/2-2P3/2$	2466742320.1	0.3	0.0	2.0
$1S1/2-3P1/2$	2923538219.2	0.3	$-0.6$	$-3.9$
$1S1/2 - 3S1/2$	2923538534.6	0.5	0.7	$-3.0$
$1S1/2 - 3P3/2$	2923541470.1	0.3	$-0.9$	$-4.1$
$1S1/2-4P1/2$	3083420177.53	0.17	$-0.6$	$-3.0$
$1S1/2 - 4S1/2$	3083420311.095	0.02	$0.5\,$	$-2.1$
$1S1/2 - 4P3/2$	3083421549.28	0.15	$-0.4$	$-2.8$
$1S1/2 - 5P1/2$	3157422491	7	$-3.0$	$-4.1$
$1S1/2 - 5S1/2$	3157422559	6	$-2.9$	$-4.0$
$1S1/2 - 5P3/2$	3157423193	7	$-3.3$	$-4.3$
$1S1/2 - 5D5/2$	3157423433.2	2.3	5.0	$-0.3$
$1S1/2-6P1/2$	3197621201	7	$-2.5$	$-2.5$
$1S1/2 - 6S1/2$	3197621241	6	$-1.8$	$-1.8$
$1S1/2-6P3/2$	3197621608	7	$-1.9$	$-1.9$
$1S1/2-6D5/2$	3197621746.3	2.3	2.2	$-0.2$
$1S1/2-7P1/2$	3221859720	6	$-0.7$	0.1
$1S1/2 - 7S1/2$	3221859745	6	$-0.4$	0.4
$1S1/2 - 7P3/2$	3221859976	6	$-0.6$	0.2
$1S1/2 - 7D5/2$	3221860061.9	2.3	0.8	0.1
$1S1/2 - 8P1/2$	3237591432.01	0.3	0.1	1.5
1S1/2-8S1/2	3237591448.768	0.07	0.3	1.7
1S1/2-8P3/2	3237591603.48	0.04	0.1	1.6
1S1/2-9P1/2	3248377032	7	0.9	2.8
1S1/2-9S1/2 3248377044		$6\,$	1.2	3.1
1S1/2-9P3/2	3248377153	$\frac{7}{1}$ .	1.5	3.4
1S1/2-9D5/2 3248377191.5		2.3	$0.2\,$	1.4
1S1/2-10P1/2	3256091901		$0.2\,$	2.5
1S1/2-10S1/2 3256091910		$\frac{7}{6}$	0.7	3.0
1S1/2-10P3/2	3256091989	7	0.4	2.7
1S1/2-11P1/2	3261800029	$\frac{7}{2}$	$0.8\,$	3.4
$S1/2 - 11S1/2$	3261800035	6	$0.5\,$	3.0
1S1/2-11P3/2 3261800095		$\overline{a}$	0.9	3.5
1S1/2-11D5/2	3261800116.1	2.3	$0.2\,$	2.4
1S1/2-12P1/2	3266141526	6	1.2	4.0
1S1/2-12S1/2	3266141531	6	1.3	4.1
1S1/2-12P3/2 3266141577		$\overline{7}$	1.4	4.2
2P1/2-2S1/2	1059.28	0.06	$-0.7$	$-1.9$
2P1/2-2P3/2	10971.9	$0.4\phantom{0}$	$-0.7$	$-0.4$
2P1/2-3S1/2	456807186.4	0.5	$0.0\,$	-5.4

**Table 7:** Frequencies of fine-structure transitions in *deuterium* derived from the level-optimization procedure [Kramida 2010, Table 7].

interval	$\Delta E_{\text{Ritz}}$ (MHz)	Unc. (MHz)	$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)	
	(Ritz values)		$\alpha_{geom}$	$\alpha_{\text{codata}}$
	456810114.2	1.8	$-5.6$	$-26.3$
$2P1/2 - 3D3/2$				
2P1/2-4S1/2	616688962.86	0.07 0.24	$-0.2$ $-1.8$	-4.5 $-12.5$
2P1/2-4D3/2	616690198.83 690691211		$-3.3$	$-6.3$
2P1/2-5S1/2		7 2.3	2.8	$-3.4$
2P1/2-5D3/2	690691850.8			
2P1/2-6D3/2	730890262.7	2.3	0.8	$-2.9$
2P1/2-7S1/2	755128397	7	$-0.9$	$-1.9$
2P1/2-7D3/2	755128628.2	2.3	$-0.6$	-2.8
2P1/2-8S1/2	770860100.53	0.6	$-0.5$	$-0.8$
2P1/2-8D3/2	770860254.98	0.06	$-0.7$	$-1.8$
2P1/2-9S1/2	781645696	7	0.8	0.9
2P1/2-9D3/2	781645803.1	2.3	$-0.8$	$-1.2$
2P1/2-10S1/2	789360562	7	0.3	0.8
2P1/2-10D3/2	789360640.26	0.07	$-0.7$	$-0.6$
2P1/2-11S1/2	795068687	7	0.0	0.8
2P1/2-11D3/2	795068745.9	2.3	$-0.6$	$-0.1$
2P1/2-12S1/2	799410183	7	0.8	1.9
2P1/2-12D3/2	799410227.32	0.06	$-0.7$	0.1
$2S1/2 - 2P3/2$	9912.6	0.3	$-0.1$	1.5
2S1/2-3P1/2	456805811.7	0.3	$-0.7$	-4.4
$2S1/2 - 3S1/2$	456806127.1	0.5	0.7	$-3.5$
$2S1/2 - 3P3/2$	456809062.6	0.3	$-0.9$	-4.6
$2S1/2 - 4P1/2$	616687770.01	0.17	$-0.6$	$-3.5$
$2S1/2 - 4S1/2$	616687903.573	0.02	0.5	$-2.6$
$2S1/2 - 4P3/2$	616689141.76	0.16	$-0.4$	$-3.3$
2S1/2-4D5/2	616689596.72	0.04	1.6	$-9.5$
2S1/2-5P1/2	690690083	6	-3.6	-5.1
$2S1/2 - 5P3/2$	690690785	7	$-3.8$	$-5.4$
$2S1/2-6P1/2$	730888794	7	$-2.0$	$-2.5$
$2S1/2 - 6P3/2$	730889200	$\overline{7}$	$-2.4$	$-2.9$
$2S1/2-7P1/2$	755127312	7	$-1.2$	$-0.9$
2S1/2-7P3/2	755127568	7	$-1.2$	$-0.8$
$2S1/2 - 8P1/2$	770859024.49	0.03	0.0	1.0
$2S1/2 - 8P3/2$	770859195.96	0.04	0.0	1.0
$2S1/2-9P1/2$	781644625	7	1.3	2.8
2S1/2-9P3/2	781644745	7	0.9	2.4
2S1/2-10P1/2	789359494	7	0.7	2.5
2S1/2-10P3/2	789359582	7	0.9	2.7
2S1/2-11P1/2	795067621	6	0.3	2.4
2S1/2-11P3/2	795067687	6	0.3	2.4
2S1/2-12P1/2	799409119	7	1.7	4.0
2S1/2-12P3/2	799409169	7	0.9	3.2
2P3/2-3S1/2	456796214.5	0.6	0.7	$-5.0$
2P3/2-3D3/2	456799142.3	1.8	$-4.9$	$-25.8$

**Table 7:** Frequencies of fine-structure transitions in *deuterium* derived from the level-optimization procedure [Kramida 2010, Table 7].

interval	Unc. (MHz) $\Delta E_{\text{Ritz}}$ (MHz)			$\Delta E_{\text{Ritz}} - \Delta E_{\text{calc}}$ (MHz)
	(Ritz values)		$\alpha_{geom}$	$\alpha_{codata}$
2P3/2-4S1/2	616677991	0.3	0.5	$-4.1$
2P3/2-4D3/2	616679227	0.4	-1.0	-12.0
2P3/2-4D5/2	616679684.1	0.3	1.7	$-11.0$
2P3/2-5S1/2	690680239	7	$-2.7$	$-5.9$
2P3/2-5D3/2	690680878.9	2.3	3.6	$-2.9$
2P3/2-5D5/2	690681113	2.3	5.0	$-2.4$
2P3/2-6S1/2	730878921	7	$-1.6$	$-3.7$
2P3/2-6D3/2	730879290.9	2.3	1.6	-2.4
2P3/2-6D5/2	730879426.3	2.3	2.3	$-2.1$
2P3/2-7S1/2	755117425	7	$-0.3$	$-1.5$
2P3/2-7D3/2	755117656.4	2.3	0.2	$-2.2$
2P3/2-7D5/2	755117741.7	2.3	0.7	$-2.0$
2P3/2-8S1/2	770849128.7	0.3	0.3	-0.3
2P3/2-8D3/2	770849283.1	0.3	0.0	-1.4
2P3/2-8D5/2	770849340.3	0.3	0.4	-1.2
2P3/2-9S1/2	781634724	7	1.4	1.2
2P3/2-9D3/2	781634831.3	2.4	0.0	$-0.7$
2P3/2-9D5/2	781634871.4	2.3	0.2	$-0.6$
2P3/2-10S1/2	789349590	7	0.9	1.1
2P3/2-10D3/2	789349668.4	0.3	0.1	-0.1
2P3/2-10D5/2	789349697.7	0.3	0.3	0.0
2P3/2-11S1/2	795057715	7	0.6	1.1
2P3/2-11D3/2	795057774	2.3	0.1	0.3
2P3/2-11D5/2	795057796	2.3	0.2	0.4
2P3/2-12S1/2	799399211	$\overline{7}$	1.4	2.2
2P3/2-12D3/2	799399255.5	0.3	0.1	0.6
2P3/2-12D5/2	799399272.4	0.3	0.2	0.6
3P1/2-3S1/2	315.4	0.4	1.3	1.0
3P1/2-3P3/2	3250.9	0.4	$-0.3$	$-0.2$
$3D3/2 - 3D5/2$	1083.6	2.4	6.4	2.5
4P1/2-4S1/2	133.57	0.17	1.1	0.9

**Table 7:** Frequencies of fine-structure transitions in *deuterium* derived from the level-optimization procedure [Kramida 2010, Table 7].

#### **Notes:**

a) Ritz series limit [Kramida 2010, Table H].

b) NIST: Atomic Spectra Database 78 [version 5.11]: Ionization Energies Form. Energy has been determined from bound-state QED ab initio calculations.