# SunQM-5: Using the Interior {N,n//6} QM to Describe an Atom's Nucleus-Electron System, and to Scan from Sub-quark to Universe (Drafted in April 2018)

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### Abstract

In this paper, we proposed a brand new nuclear  $\{N,n/6\}$  QM. It showed that like the Sun-planet system, the atom's nucleus-electron system also follows the  $\{N,n//6\}$  QM. By using the nucleus-electron system's interior  $\{N,n//6\}$  QM with the ground state electron's orbit as  $e1\{0,1\}$ , the nuclides of hydrogen (Z=1), He (Z=2), Li (Z=3), Ne (Z=10), Fe (Z=26), Pd (Z=46), and Og (Z=118) atoms have the interior QM structure (in size) of e1{-3,1//6}, e1{-3,2//6}, e1{-3,3//6}, e1{-3,6//6} = 1{-3,6//6}  $e_{1}^{-2,1/6}$ ,  $e_{1}^{-2,2/6}$ ,  $e_{1}^{-2,3/6}$ , and  $e_{1}^{-2,6/6} = e_{1}^{-1,1/6}$ , respectively. Although all nucleons in those nuclides structured in  $\{N,n//6\}$  QM, this property actually is a hidden property (like a "recessive gene") that only showed up at the interior e1{N,n//6} QM using electron's n=1 shell as the reference point. This is because the strong EM-force shrank the orbital  $r_{1e}$  to  $r_{1e}/Z$ . The study revealed that Fe atom's nucleus is the only one that has the 100% nucleon occupancy in the n = 1 nucleon orbital shell space (among the size n=2 nuclides from Z = 11 to Z = 26), so that Fe element's nucleus has the most stable nuclear  $\{N,n/6\}$  QM structure in comparison to that of its adjacent elements. This result explains why Fe is more abundant than its adjacent elements, and why Fe atom is the heaviest nucleus that a nuclear fusion reaction can go without adding extra energy. This is exactly like why the inert element Ne is chemically most stable among the electron period 2 elements, because it has the outmost electron shell completely filled (or it has the 100% electron occupancy in the n=2 electron orbital shell space). This analysis revealed that the atom's nuclear  $\{N,n/6\}$  QM structure stability played an important role in determining the abundancy of each element in our universe. Therefore, the nuclear {N,n//6} QM structural analysis significantly improved our knowledge on the nuclear physics. Further analysis showed that although the {N,n//q} QM structures of both atom's nucleus shell and atom's electron shells are primarily in  $\{N,n/6\}$  QM at the low Z#, at the high Z#, both of them shifted to the  $\{N,n//7\}$  QM. Another analysis revealed that  $\{-10,1//6\}$  is the maximum (ever-possible) size of an atom. It also revealed that although  $\{N,n//6\}$  is the fundamental  $\{N,n//q\}$  QM structure in our universe, the disruption of either the primary EM-force or G-force formed  $\{N,n/6\}$  QM structure will form  $\{N,n/q\}$  QM structure with q quantum number other than 6.

#### Introduction

Based on Bohr model's QM and Schrodinger equation's QM, we developed a new version of QM, named {N,n} QM <sup>[1-9]</sup>. {N,n} QM has 4 bases: 1) interior {N,n} QM structure, 2) multiplier n' for the fine adjustment, 3) using Schrodinger equation's probability density  $r^2 * |R(n,l)|^2 * |Y(l,m)|^2$  and maximum mass occupancy theory to fill mass in space of {N,n} shell, 4) RF (rotation diffusion, or RotaFusion) of interior {N,n} QM to study our universe. It is much easier to set the universe as {0,1}, and then calculate everything else inward as the interior {N,n} QM structure. So in this paper, we will first explain how we come up the idea that using interior {N,n} QM to study our universe. Then we will present what we

have achieved in using the interior {N,n} QM to analyze our world (from micro- to macro-). Note: for {N,n} QM nomenclature as well as the general notes for {N,n} QM model, please see SunQM-1 section VII. Note: Microsoft Excel's number format is often used in this paper, for example:  $x^2 = x^2$ ,  $3.4E+12 = 3.4*10^{12}$ ,  $5.6E-9 = 5.6*10^{-9}$ . The reading sequence for SunQM series papers is: SunQM-1, 1s1, 1s2, 1s3, 2, 3, 3s1, 3s2, 3s3, 5, 5s1. For all SunQM series papers, reader should check "SunQM-9s1: Updates and Q/A for SunQM series papers" for the most recent updates and corrections.

Note: the original title of this paper is: SunQM-5: {N,n} QM (a new version of QM based on interior {N,n}, multiplier n',  $|R(n,l)|^2$  [Y(*l*,m)]<sup>2</sup> guided mass occupancy, and RF) and its application from sub-quark to universe (drafted in April 2018). The minimum revise has been made after April 2018.

#### I. Our Universe in view of interior QM

Table 1 (copied from paper SunQM-1s2's Table 1) showed a {N,n//6} QM structural scanning from size of subquark to universe. The Cold-G r-track (columns 3 to 5) used Sun-mass black hole's r at {-3,1} as the reference value for the calculation (which is equivalent to the current Sun that has no hydrogen fusion's heating caused size expansion). The Hot-G r-track (columns 6 to 8) used Sun surface's r at {0,2} as the reference value for the calculation (so it is with the Sun's hydrogen fusion heating caused size expansion effect). Columns 9 to 11 used proton's r at {-15,1} as the reference value for the calculation (so it is equivalent to a much higher heating and expansion effect due to it has much larger r). Columns 12 to 14 used electron's n=1 ground state's r at {-12,1} as the reference value for the calculation (and it is equivalent to an even higher heating and expansion effect due to it has the even larger r). Although it is better to use different r-track for different size range, for simplicity, we only use Hot-G r-track for all sizes (from macro- to micro-world) in this paper. Columns 15-16 showed the known objects that correlate to theses sizes, and the best fitted N super-shell set with  $\Delta N = 5$ . Column 17 showed the N super-shell set (with  $\Delta N = 5$ ) that do not belong to the regular N super-shell set in columns 15-16. Also see SunQM-1s2 for more explanations.

With the data in Table 1, now let us change the angle to view our universe by using the concept of interior {N,n} QM. The following discussion reviewed and extended the discussion in paper SunQM-2 section IV.

First let us study the Sun core's interior QM. Take Sun core as the most top structure  $\{0,1\}$ . Its gravitational interior QM generated a stable QM structure at size of  $\{-15,1\}$ , we call it proton (also can be neutron). There are total ~ 1.19E+57 of this  $\{-15,1\}$  structures (Sun's mass / proton's mass = 1.99E+30 kg / 1.67E-27 kg = 1.19E+57). The  $\{-15,1\}$  structures sit not only at the center of  $\{0,1\}$  structure, but also everywhere inside  $\{0,1\}$  structure (and lots of them sit inside  $\{0,2\}$  structure). The distribution of  $\{-15,1\}$  structures inside  $\{0,1\}$  structure follows the gravitational interior QM of  $\{0,1\}$ , so it has high density at the center of  $\{0,1\}$  and low density at the edge of  $\{0,1\}$  structure, and it forms the fine interior QM structures, like Sun's  $\{-1,n=1...5\}$ o,  $\{-2,n=1...5\}$ o, ... etc. Here the important thing is, we only have the total mass, and the space limitation (size of Sun core) for this mass, then the gravitational interior QM will determine the elementary particles, the mass distribution, etc. Note: for a matter wave view of Solar interior  $\{N,n\}$  QM, see SunQM-2 section IV-c.

Now let us repeat this analysis by using another system, say the whole Solar system (up to size of  $\{5,1\}$ ). For this, we need to reset Solar QM's  $\{5,1\}$  as the most top structure  $\{0,1\}$ , and name it as solarsys $\{0,1//6\}$ . Its gravitational interior QM generated a stable QM structure at size of solarsys $\{-20,1//6\}$ , we call it proton (or neutron). The distribution of many solarsys $\{-20,1//6\}$  structures inside solarsys $\{0,1//6\}$  structure follows the gravitational interior QM of solarsys $\{0,1//6\}$ , so some of these solarsys $\{-20,1//6\}$  structures form a stable solarsys $\{-5,2//6\}$  QM structure (we call it Sun) at the center of solarsys $\{0,1//6\}$  structure. Some other solarsys $\{-20,1//6\}$  structures (there are 3.57E+51 of them, calculated as Earth's mass / proton's mass = 5.97E+24 kg / 1.67E-27 kg = 3.57E+51) formed a stable QM structure in shell of solarsys $\{-4,5//6\}$ , we call it Earth. Again, here we only have the total mass, and the space limitation (size of Solar system) for this mass, then the gravitational interior QM will determine the elementary particles, the mass distribution, etc.

Table 1. {N,n//6} scanning from sub-quark to universe. Start from: 1) Sun black hole {-3,1} r = 2.95E+3 meters; 2) Sun core {0,1} r = 1.74E+8 meters; 3) proton {-15,1} r = 8.4E-16 meters; 4) electron n = 1 ground state {-12,1} r = 5.29E-11 meters. Using  $\Delta N = \pm 1$  (or 36×) to scan up and down.

											electron(						
		Cold-G r			Hot-G r			proton r			n=1) r			object			{K,N//6,n//
N=	total n=	track			track			track			track			correlated			6}
		m	AU	ly	m	AU	ly	m .	AU	ly	m	AU	ly				
-29	2.71E-23	1.01E-37			1.28E-37			1.37E-37			1.85E-37						
-28	1.63E-22	3.65E-36			4.61E-36			4.92E-36			6.65E-36			{-28,4} Plan	ck length 1.6	6E-35	
-27	9.77E-22	1.31E-34			1.66E-34			1.77E-34			2.39E-34						
-26	5.86E-21	4.73E-33			5.98E-33			6.38E-33			8.61E-33						
-25	3.52E-20	1.70E-31			2.15E-31			2.30E-31			3.10E-31			{-25,1}			{-5,0,1}
-24	2.11E-19	6.13E-30			7.75E-30			8.27E-30			1.12E-29						
-23	1.27E-18	2.21E-28			2.79E-28			2.98E-28			4.02E-28						
-22	7.60E-18	7.94E-27			1.00E-26			1.07E-26			1.45E-26						
-21	4.56E-17	2.86E-25			3.61E-25			3.86E-25			5.21E-25						
-20	2.74E-16	1.03E-23			1.30E-23			1.39E-23			1.88E-23			{-20,1}			{-4,0,1}
-19	1.64E-15	3.71E-22			4.68E-22			5.00E-22			6.75E-22						
-18	9.85E-15	1.33E-20			1.69E-20			1.80E-20			2.43E-20						
-17	5.91E-14	4.80E-19			6.07E-19			6.48E-19			8.75E-19			[-17,1} quar	k r=4.3E-19		
-16	3.54E-13	1.73E-17			2.19E-17			2.33E-17			3.15E-17						
-15	2.13E-12	6.23E-16			7.87E-16			8.40E-16			1.13E-15			{-15,1} prot	on r=8.4E-16	5	{-3,0,1}
-14	1.28E-11	2.24E-14			2.83E-14			3.02E-14			4.08E-14						
-13	7.66E-11	8.07E-13			1.02E-12			1.09E-12			1.47E-12						
-12	4.59E-10	2.90E-11			3.67E-11			3.92E-11			5.29E-11			{-12,1} H-ato	om r		
-11	2.76E-09	1.05E-09			1.32E-09			1.41E-09			1.90E-09			[-11,1} Pb at	om r=1.8E-9		
-10	1.65E-08	3.76E-08			4.76E-08			5.08E-08			6.86E-08			{-10,1} 50 n	m		{-2,0,1}
-9	9.92E-08	1.36E-06			1.71E-06			1.83E-06			2.47E-06						
-8	5.95E-07	4.88E-05			6.17E-05			6.58E-05			8.89E-05						
-7	3.57E-06	1.76E-03			2.22E-03			2.37E-03			3.20E-03						
-6	2.14E-05	6.32E-02			7.99E-02			8.53E-02			1.15E-01						
-5	1.29E-04	2.28E+00			2.88E+00			3.07E+00			4.15E+00			{-5,1} mete	r		{-1,0,1}
-4	7.72E-04	8.19E+01			1.04E+02			1.11E+02			1.49E+02						
-3	4.63E-03	2.95E+03			3.73E+03			3.98E+03			5.37E+03			{-3,1} Sun b	ack hole r=2	2.95E+3	
-2	2.78E-02	1.06E+05			1.34E+05			1.43E+05			1.93E+05						
-1	1.67E-01	3.82E+06			4.83E+06			5.16E+06			6.96E+06			{-1,1} Earth	size		
0	1	1.38E+08			1.74E+08			1.86E+08			2.51E+08			Sun core g{(	),1} gravity-ı	ŗ	{0,0,1}
1	6	4.95E+09			6.26E+09			6.69E+09			9.02E+09						
2	36	1.78E+11	1.19E+00		2.25E+11	1.51E+00		2.41E+11	1.61E+00		3.25E+11	2.17E+00		Mars {2,1} g	ravity-r		
3	216	6.42E+12	4.29E+01		8.11E+12	5.42E+01		8.66E+12	5.79E+01		1.17E+13	7.82E+01		Kuiper belt	{3,1}		
4	1296	2.31E+14	1.55E+03	2.44E-02	2.92E+14	1.95E+03	3.09E-02	3.12E+14	2.08E+03	3.30E-02	4.21E+14	2.81E+03	4.45E-02	Oort begin {	4,1}		
5	7776	8.32E+15	5.56E+04	8.80E-01	1.05E+16	7.03E+04	1.11E+00	1.12E+16	7.51E+04	1.19E+00	1.52E+16	1.01E+05	1.60E+00	Oort end {5,	1}		{1,0,1}
6	4.67E+04	3.00E+17	2.00E+06	3.17E+01	3.79E+17	2.53E+06	4.00E+01	4.04E+17	2.70E+06	4.27E+01	5.46E+17	3.65E+06	5.77E+01				
7	2.80E+05	1.08E+19	7.21E+07	1.14E+03	1.36E+19	9.11E+07	1.44E+03	1.46E+19	9.73E+07	1.54E+03	1.96E+19	1.31E+08	2.08E+03				
8	1.68E+06	3.88E+20	2.60E+09	4.10E+04	4.91E+20	3.28E+09	5.19E+04	5.24E+20	3.50E+09	5.54E+04	7.07E+20	4.73E+09	7.47E+04	{8,1} Milky \	vay, r=5~9E+	⊦4 ly	
9	1.01E+07	1.40E+22	9.34E+10	1.48E+06	1.77E+22	1.18E+11	1.87E+06	1.89E+22	1.26E+11	1.99E+06	2.55E+22	1.70E+11	2.69E+06	{9,1} to And	romeda Gal	axy, {9,2} lo	cal group
10	6.05E+07	5.03E+23	3.36E+12	5.32E+07	6.36E+23	4.25E+12	6.72E+07	6.79E+23	4.54E+12	7.18E+07	9.16E+23	6.13E+12	9.69E+07	{10,1} Virgo	SupClst r=5.	5E+7 ly	{2,0,1}
11	3.63E+08	1.81E+25	1.21E+14	1.91E+09	2.29E+25	1.53E+14	2.42E+09	2.44E+25	1.63E+14	2.58E+09	3.30E+25	2.21E+14	3.49E+09	{10,2} Lania	kea r=2.6E+	8ly	
12	2.18E+09	6.52E+26	4.36E+15	6.89E+10	8.24E+26	5.51E+15	8.71E+10	8.80E+26	5.88E+15	9.30E+10	1.19E+27	7.94E+15	1.26E+11	{11,5} obser	v Univ r=4.4	E+26	
13	1.31E+10	2.35E+28	1.57E+17	2.48E+12	2.97E+28	1.98E+17	3.14E+12	3.17E+28	2.12E+17	3.35E+12	4.28E+28	2.86E+17	4.52E+12	{13,1} our U	niverse?		
14	7.84E+10	8.45E+29	5.65E+18	8.93E+13	1.07E+30	7.14E+18	1.13E+14	1.14E+30	7.62E+18	1.21E+14	1.54E+30	1.03E+19	1.63E+14				
15	4.70E+11	3.04E+31	2.03E+20	3.22E+15	3.85E+31	2.57E+20	4.06E+15	4.11E+31	2.74E+20	4.34E+15	5.54E+31	3.70E+20	5.86E+15	{15,1} our u	niverse?		{3,0,1}
16	2.82E+12	1.10E+33	7.32E+21	1.16E+17	1.38E+33	9.25E+21	1.46E+17	1.48E+33	9.88E+21	1.56E+17	1.99E+33	1.33E+22	2.11E+17				
17	1.69E+13	3.94E+34	2.64E+23	4.17E+18	4.98E+34	3.33E+23	5.27E+18	5.32E+34	3.56E+23	5.62E+18	7.18E+34	4.80E+23	7.59E+18				
18	1.02E+14	1.42E+36	9.49E+24	1.50E+20	1.79E+36	1.20E+25	1.90E+20	1.92E+36	1.28E+25	2.02E+20	2.59E+36	1.73E+25	2.73E+20				
19	6.09E+14	5.11E+37	3.42E+26	5.40E+21	6.46E+37	4.32E+26	6.83E+21	6.90E+37	4.61E+26	7.29E+21	9.31E+37	6.22E+26	9.84E+21				
20	3.66E+15	1.84E+39	1.23E+28	1.94E+23	2.32E+39	1.55E+28	2.46E+23	2.48E+39	1.66E+28	2.62E+23	3.35E+39	2.24E+28	3.54E+23				{4,0,1}
21	2.19E+16	6.62E+40	4.43E+29	7.00E+24	8.37E+40	5.59E+29	8.85E+24	8.94E+40	5.97E+29	9.45E+24	1.21E+41	8.06E+29	1.27E+25				
22	1.32E+17	2.38E+42	1.59E+31	2.52E+26	3.01E+42	2.01E+31	3.18E+26	3.22E+42	2.15E+31	3.40E+26	4.34E+42	2.90E+31	4.59E+26				
23	7.90E+17	8.58E+43	5.74E+32	9.07E+27	1.08E+44	7.25E+32	1.15E+28	1.16E+44	7.74E+32	1.22E+28	1.56E+44	1.04E+33	1.65E+28				
24	4.74E+18	3.09E+45	2.07E+34	3.27E+29	3.91E+45	2.61E+34	4.13E+29	4.17E+45	2.79E+34	4.41E+29	5.63E+45	3.76E+34	5.95E+29				
25	2.84E+19	1.11E+47	7.44E+35	1.18E+31	1.41E+47	9.40E+35	1.49E+31	1.50E+47	1.00E+36	1.59E+31	2.03E+47	1.35E+36	2.14E+31	{25,1} our u	niverse?		{5,0,1}

Note: in this simple scale up/down calculation, the possible non-linearity of space at extreme larger end (and/or extreme small end) is not included.

Note: the size of quark is obtained from [10].

We can expand this analysis to the galaxy, Virgo super cluster, or even to our universe. Now let us apply this analysis to our universe. We set the whole universe as the most top structure  $\{0,1\}$ , and we name it as **univ\{0,1//6\}**. Its gravitational interior QM generated a stable QM structure at size of univ $\{-NN,1//6\}$ , we call it proton (or neutron). We do not know exactly what NN number is, but we guess it is around 28~30, (estimated from Table 1, proton is  $\{-15,1\}$ , universe may be  $\{13,1\}$  to  $\{15,1\}$ ). The distribution of many univ $\{-NN,1//6\}$  structures inside univ $\{0,1//6\}$  structure follows the

gravitational interior QM of univ $\{0,1//6\}$ , so some of these univ $\{-NN,1//6\}$  structures form a stable univ $\{-NN+15+8,1//6\}$  QM structure (we call it Milky Way, in Table 8, at size of  $\{8,1\}$ ) inside the univ $\{0,1//6\}$  structure. Some other of these univ $\{-NN,1//6\}$  structures form a stable univ $\{-NN+15+5,1//6\}$  QM structure (we call it Solar system, in Table 1, at size of  $\{5,1\}$ ) inside the univ $\{0,1//6\}$  structure. Here the importance thing is, our universe can be treated as the only most top structure (with fixed total mass and space, the slow expansion of universe is ignored here), its gravitational interior QM will determine the elementary particles, the mass distribution, etc.

The traditional (exterior) QM provides us a bottom-up view of universe, a lot of details, but more difficult to get the whole picture of our universe. In contrast, the interior QM presents us a top-down view of the universe, which may have a profound impact on the current theories of our universe.

Now if somebody says the proton is not element enough as the elementary particles, and want to use quark ({-17,1} in Table 1) as the elementary particle, then in above analysis, we only need to replace -NN by -NN-2, that is all.

For a matter wave view of universe's interior  $\{N,n\}$  QM, see SunQM-2 section IV-c. For why the  $r_n = r_1 * n^2$  can be used for 3D  $\{N,n\}$  QM analysis, please see SunQM-3s1. For why the Solar QM  $\{N,n//6\}$  structure analysis can be extended to from sub-quark to universe, please see SunQM-1s2 section V. For the rest part of this paper, we will present how we use the interior  $\{N,n\}$  QM to study our real world (from micro to macro).

# II. Treating nucleus as the interior $\{N,n\}$ QM structure of the ground state electron orbit $\{0,1//6\}$ o revealed that the all nucleons in a nucleus also follow the $\{N,n//6\}$ QM structure

(Note: I am not a nuclear physicist. I am citizen scientist of QM. All I did here is to use  $\{N,n//6\}$  QM to study the nuclear physics).

#### II-a. Uncovering the hidden {N,n//6} QM structure for atom's nucleus-electron system

Since the Bohr model was developed for hydrogen atom's proton-electron system, for {N,n} QM, we also started from the atom's nucleus-electron system. In this section, we treated the nucleus as the interior {N,n} QM structure of the ground state electron {0,1} (named as **el{0,1}**) for atom's nucleus-electron system. The way to do this is, for each one of 118 chemical elements, we calculated the r for nucleus (named as nuc-r, or **r**<sub>nuc</sub>) and r for n = 1 state electron (named **r**<sub>el</sub>). So we can determine their interior {N,n} QM by calculating the n number = sqrt( $r_{nuc} / r_{el}$ ). Then we look for the interior {N,n//6} QM character. To do this, we constructed the Table 2.

In columns 1 through 6 of Table 2, the basic information of 118 elements, was copied from [11], and updated according to wiki "periodic table".

The main purpose of column 7 to 11 in Table 2 is to calculate out each element nucleus' r, and then the interior n (relative to hydrogen element's nucleus r and n). In column 7, all elements (except H) 's nucleus  $r_{e1}$  are adjusted to  $r_1/Z$  due to its Z protons shrink the  $r_{e1}$ . In column 8, all element's (except H's and He's)  $r_{nuc}$  were calculated as (see [12]):  $r_{nuc} = 1.25E-15 * (M#)^{(1/3)}$ , where M# is the Atomic mass number (the number of protons Z, plus the number of neutrons N). The Hydrogen element's  $r_{nuc} = 8.4E-16$  meters, was from wiki "proton", not from the calculation using the formula. The Helium element's  $r_{nuc} = 1.92E-15$  meters was obtained from experiment data [13], and the formula calculated result was  $r_{nuc} = 1.98E-15$  meters. In column 9, all elements'  $r_{nuc}$  were normalized to their own adjusted  $r_{e1}$ , named  $r_{nuc}/r_{e1}$ . In column 10, each element's  $r_{nuc}/r_{e1}$  was further normalized to hydrogen element's  $r_{nuc}/r_{e1}$  (=8.4E-16 / 5.29E-11). In column 11 (which took the square-root of column 10), it gave each element  $r_{nuc}$ 's (interior) quantum number  $n_{nuc}$  (against to the shrunk  $r_{e1}$  of this element) which was normalized to that of Hydrogen element. We can see that  $r_{nuc}$  (or size) of nucleus of H, He, Li, B, N, F/Ne has  $n_{nuc} = 1, 2, 3, 4, 5, 6$ , against its own shrunk  $r_{e1}$ , if we set H's  $r_{nuc}$  as  $n_{nuc} = 1$ .

14010 2.	Cu	Toura		1 mic	101 [1	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	2111 591				cetton	l byble	III Ous		1	· J•		1	-	
						calc. nuc-	-n relative to	H's nuc-n					calc atom-	r use nuc-r	atom size	n=2 effect		relativisti	c n <sub>nuc</sub>	
						adj. r <sub>e1</sub>	r <sub>nuc</sub>						atom-r	atom-r'						
												interior							relativisti	i
											interior	{N,n} use				atom-n	atom-n		с	relativ.
							r <sub>nuc</sub> =1.25E-		$(r_{nuc}/r_{e1})/$	n <sub>nuc</sub>	r <sub>nuc</sub>	adj		r'=r1*n^2	theory	=sqrt	=sqrt	v <sub>1</sub> =sqrt[	$r_{e1r}=r_1/Z^*$	n <sub>nuc</sub>
		proton #					15*M^(1/3		(H-r <sub>nur</sub> /H-	=sqrt(r <sub>n</sub> /r	=(n <sub>nuc</sub> /6^	r <sub>1</sub> =e1{0,1	r=r <sub>e1</sub> *n <sub>nuc</sub>	r <sub>1</sub> =r <sub>1</sub> /Z,	calc.	(atom-	(atom-	(Ze²/4πε <sub>0</sub>	sqrt(1-	=sqrt(r <sub>n</sub> /r
Name	Symbol	Z	Mass #	atomic#	neutron #	$r_{e1}=r_1/Z$	)	r <sub>nuc</sub> /r <sub>e1</sub>	r <sub>e1</sub> )	1)	3)^2*r <sub>e1</sub>	}	^2	use n <sub>nuc</sub>	atom-r	r_/H-r_1)	r <sub>n</sub> /r <sub>e1</sub> )	)/r <sub>e1</sub> /m]	$v_1^2/c^2$ )	e1r)
						m	m		m	-	m		m	m	E-12. m			m/s	m	
Hvdrogen	н	1	. 1	1	. 0	5.29E-11	1 8.4E-16	1.59E-05	5 1.00E+00	1.0	1.13E-15	e1{-3.1}	5.29E-11	5.29E-11	53	1.00	1.00	2.19E+06	5.29E-11	1.0
Helium	He	2	4	2	2	2.65E-11	1 1.92E-15	7.26E-05	4.57E+00	2.1	2.59E-15	e1{-3.2}	2.42E-10	1.21E-10	31	0.76	0.77	4.38E+06	2.64E-11	2.1
Lithium	Li	3	5 7	' 3	4	1.76E-11	1 2.39E-15	1.36E-04	8.54E+00	2.9	3.23E-15	e1{-3.3}	4.52E-10	1.51E-10	167	1.78	1.78	6.56E+06	1.76E-11	2.9
Bervllium	Be	4	9	4	5	1.32E-11	1 2.60E-15	1.97E-04	1.24E+01	L 3.5	3.51E-15		6.55E-10	1.64E-10	112	1.45	1.46	8.75E+06	1.32E-11	3.5
Boron	В	5	11		6	1.06F-11	1 2.78E-15	2.63E-04	1.65E+01	4.1	3.75E-15	e1{-3.4}	8.75E-10	1.75E-10	87	1.28	1.28	1.09F+07	1.06E-11	4.1
Carbon	c	6	12	e	5 6	8.82E-12	2 2.86E-15	3.25E-04	2.04E+01	4.5	3.86E-15	;	1.08E-09	1.80E-10	67	1.12	1.13	1.31E+07	8.81E-12	2 4.5
Nitrogen	N	7	14	7	7	7.56E-12	2 3.01E-15	3.99E-04	2.51E+01	5.0	4.07E-15	e1{-3.5}	1.33E-09	1.90E-10	56	1.03	1.03	1.53E+07	7.55E-12	2 5.0
Oxygen	0	8	16	ε ε	8 8	6.61E-12	2 3.15E-15	4.76E-04	3.00E+01	L 5.5	4.25E-15		1.59E-09	1.98E-10	48	0.95	0.95	1.75E+07	6.60E-12	2 5.5
Fluorine	F	g	19	9	10	5.88E-12	2 3.34E-15	5.67E-04	3.57E+01	6.0	4.50E-15	e1{-3,6}	1.89E-09	2.10E-10	42	0.89	0.89	1.97E+07	5.87E-12	2 6.0
Neon	Ne	10	20	10	10	5.29E-12	2 3.39E-15	6.41E-04	4.04E+01	6.4	4.58E-15	e1{-3,6}	2.14E-09	2.14E-10	38	0.85	0.85	2.19E+07	5.28E-12	2 6.4
Sodium	Na	11	23	11	. 12	4.81E-12	2 3.55E-15	7.39E-04	4.66E+01	L 6.8	4.80E-15		2.46E-09	2.24E-10	190	1.89	1.90	2.41E+07	4.79E-12	2 6.8
Magnesium	Mg	12	24	12	12	4.41E-12	2 3.61E-15	8.18E-04	5.15E+01	L 7.2	4.87E-15		2.72E-09	2.27E-10	145	1.65	1.66	2.63E+07	4.39E-12	2 7.2
Aluminium	Al	13	27	13	14	4.07E-12	2 3.75E-15	9.22E-04	5.80E+01	L 7.6	5.06E-15		3.07E-09	2.36E-10	118	1.49	1.49	2.84E+07	4.05E-12	2 7.6
Silicon	Si	14	28	14	14	3.78E-12	2 3.80E-15	1.00E-03	6.33E+01	L 8.0	5.12E-15		3.35E-09	2.39E-10	111	1.45	1.45	3.06E+07	3.76E-12	2 8.0
Phosphorus	Р	15	31	. 15	5 16	3.53E-12	2 3.93E-15	1.11E-03	7.01E+01	L 8.4	5.30E-15		3.71E-09	2.47E-10	98	1.36	1.36	3.28E+07	3.51E-12	2 8.4
Sulfur	S	16	32	16	5 16	3.31E-12	2 3.97E-15	1.20E-03	7.56E+01	L 8.7	5.36E-15		4.00E-09	2.50E-10	88	1.29	1.29	3.50E+07	3.28E-12	2 8.7
Chlorine	Cl	17	35	17	18	3.11E-12	2 4.09E-15	1.31E-03	8.28E+01	l 9.1	5.52E-15		4.38E-09	2.57E-10	79	1.22	1.22	3.72E+07	3.09E-12	9.1
Argon	Ar	18	40	18	3 22	2.94E-12	2 4.27E-15	1.45E-03	9.16E+01	L 9.6	5.77E-15		4.85E-09	2.69E-10	71	1.16	i 1.16	3.94E+07	2.91E-12	9.6
Potassium	к	19	30	19	21	2.78E-12	2 3.88E-15	1.40E-03	8.79E+01	L 9.4	5.24E-15		4.65E-09	2.45E-10	243	2.14	2.14	4.16E+07	2.76E-12	9.4
Calcium	Ca	20	40	20	20	2.65E-12	2 4.27E-15	1.62E-03	1.02E+02	2 10.1	5.77E-15		5.38E-09	2.69E-10	194	1.91	1.92	4.38E+07	2.62E-12	2 10.1
Scandium	Sc	21	. 45	21	. 24	2.52E-12	2 4.45E-15	1.76E-03	1.11E+02	10.5	6.00E-15		5.88E-09	2.80E-10	184	1.86	1.87	4.59E+07	2.49E-12	2 10.6
Titanium	Ti	22	48	22	26	2.40E-12	2 4.54E-15	1.89E-03	1.19E+02	2 10.9	6.13E-15		6.29E-09	2.86E-10	176	1.82	1.82	4.81E+07	2.37E-12	2 11.0
Vanadium	v	23	51	. 23	28	2.30E-12	2 4.64E-15	2.02E-03	1.27E+02	11.3	6.26E-15		6.71E-09	2.92E-10	171	1.80	1.80	5.03E+07	2.27E-12	2 11.3
Chromium	Cr	24	52	24	28	2.20E-12	2 4.67E-15	2.12E-03	1.33E+02	11.5	6.30E-15		7.05E-09	2.94E-10	166	1.77	1.77	5.25E+07	2.17E-12	11.6
Manganese	Mn	25	55	25	30	2.12E-12	2 4.75E-15	2.25E-03	1.41E+02	11.9	6.42E-15		7.48E-09	2.99E-10	161	1.74	1.74	5.47E+07	2.08E-12	2 12.0
Iron	Fe	26	56	i 26	30	2.03E-12	2 4.78E-15	2.35E-03	1.48E+02	12.2	6.46E-15	e1{-2,2}	7.83E-09	3.01E-10	156	1.72	1.72	5.69E+07	2.00E-12	12.3
Cobalt	Co	27	58	27	31	1.96E-12	2 4.84E-15	2.47E-03	1.56E+02	2 12.5	6.53E-15		8.23E-09	3.05E-10	152	1.69	1.70	5.91E+07	1.92E-12	2 12.6
Nickel	Ni	28	58	28	30	1.89E-12	2 4.84E-15	2.56E-03	1.61E+02	2 12.7	6.53E-15		8.53E-09	3.05E-10	149	1.68	1.68	6.13E+07	1.85E-12	2 12.8
Copper	Cu	29	64	29	35	1.82E-12	2 5.00E-15	2.74E-03	1.73E+02	13.1	6.75E-15		9.13E-09	3.15E-10	145	1.65	1.66	6.35E+07	1.78E-12	13.3
Zinc	Zn	30	65	30	35	1.76E-12	2 5.03E-15	2.85E-03	1.79E+02	13.4	6.78E-15		9.50E-09	3.17E-10	142	1.64	1.64	6.56E+07	1.72E-12	2 13.6
Gallium	Ga	31	. 70	31	. 39	1.71E-12	2 5.15E-15	3.02E-03	1.90E+02	13.8	6.95E-15		1.01E-08	3.24E-10	136	1.60	1.60	6.78E+07	1.66E-12	2 14.0
Germanium	Ge	32	73	32	41	1.65E-12	2 5.22E-15	3.16E-03	1.99E+02	14.1	7.05E-15		1.05E-08	3.29E-10	125	1.54	1.54	7.00E+07	1.61E-12	2 14.3
Arsenic	As	33	75	33	42	1.60E-12	2 5.27E-15	3.29E-03	2.07E+02	14.4	7.12E-15		1.10E-08	3.32E-10	114	1.47	1.47	7.22E+07	1.56E-12	14.6
Selenium	Se	34	79	34	45	1.56E-12	2 5.36E-15	3.45E-03	2.17E+02	14.7	7.24E-15		1.15E-08	3.38E-10	103	1.39	1.40	7.44E+07	1.51E-12	2 15.0
Bromine	Br	35	80	35	6 45	1.51E-12	2 5.39E-15	3.56E-03	2.24E+02	15.0	7.27E-15		1.19E-08	3.39E-10	94	1.33	1.33	7.66E+07	1.46E-12	2 15.2
Krypton	Kr	36	84	36	6 48	1.47E-12	2 5.47E-15	3.73E-03	2.35E+02	15.3	7.39E-15		1.24E-08	3.45E-10	88	1.29	1.29	7.88E+07	1.42E-12	15.6
Rubidium	Rb	37	85	37	48	1.43E-12	2 5.50E-15	3.84E-03	2.42E+02	15.6	7.42E-15		1.28E-08	3.46E-10	265	2.24	2.24	8.10E+07	1.38E-12	2 15.9
Strontium	Sr	38	88	38	3 50	1.39E-12	2 5.56E-15	3.99E-03	2.52E+02	15.9	7.50E-15		1.33E-08	3.50E-10	219	2.03	2.03	8.31E+07	1.34E-12	2 16.2
Yttrium	Y	39	89	39	50	1.36E-12	2 5.58E-15	4.11E-03	2.59E+02	2 16.1	7.53E-15		1.37E-08	3.51E-10	212	2.00	2.00	8.53E+07	1.30E-12	! 16.4
Zirconium	Zr	40	91	. 40	51	1.32E-12	2 5.62E-15	4.25E-03	2.68E+02	2 16.4	7.59E-15		1.42E-08	3.54E-10	206	1.97	1.97	8.75E+07	1.26E-12	! 16.7
Niobium	Nb	41	. 93	41	52	1.29E-12	2 5.66E-15	4.39E-03	2.76E+02	16.6	7.64E-15		1.46E-08	3.57E-10	198	1.93	1.93	8.97E+07	1.23E-12	2 17.0
Molybdenum	Mo	42	96	i 42	54	1.26E-12	2 5.72E-15	4.54E-03	2.86E+02	2 16.9	7.73E-15		1.51E-08	3.60E-10	190	1.89	1.90	9.19E+07	1.20E-12	! 17.3
Technetium	Тс	43	98	43	55	1.23E-12	2 5.76E-15	4.68E-03	2.95E+02	2 17.2	7.78E-15		1.56E-08	3.63E-10	183	1.86	1.86	9.41E+07	1.17E-12	2 17.6
Ruthenium	Ru	44	101	. 44	57	1.20E-12	2 5.82E-15	4.84E-03	3.05E+02	2 17.5	7.86E-15		1.61E-08	3.67E-10	178	1.83	1.83	9.63E+07	1.14E-12	2 17.9
Rhodium	Rh	45	103	45	5 58	1.18E-12	2 5.86E-15	4.98E-03	3.14E+02	2 17.7	7.91E-15		1.66E-08	3.69E-10	173	1.81	1.81	9.85E+07	1.11E-12	2 18.2
Palladium	Pd	46	106	46	60	1.15E-12	2 5.92E-15	5.14E-03	3.24E+02	18.0	7.99E-15	e1{-2,3}	1.71E-08	3.73E-10	169	1.79	1.79	1.01E+08	1.08E-12	2 18.5
Silver	Ag	47	108	47	61	1.13E-12	2 5.95E-15	5.29E-03	3.33E+02	18.3	8.04E-15	e1{-2,3}	1.76E-08	3.75E-10	165	1.76	i 1.77	1.03E+08	1.06E-12	! 18.8
Cadmium	Cd	48	112	48	8 64	1.10E-12	2 6.03E-15	5.47E-03	3.44E+02	18.6	8.13E-15		1.82E-08	3.79E-10	161	1.74	1.74	1.05E+08	1.03E-12	! 19.2
Indium	In	49	115	49	66	1.08E-12	2 6.08E-15	5.63E-03	3.55E+02	18.8	8.20E-15		1.88E-08	3.83E-10	156	1.72	1.72	1.07E+08	1.01E-12	! 19.5
Tin	Sn	50	119	50	69	1.06E-12	2 6.15E-15	5.81E-03	3.66E+02	19.1	8.30E-15		1.94E-08	3.87E-10	145	1.65	1.66	1.09E+08	9.85E-13	19.8
Antimony	Sb	51	. 122	51	. 71	1.04E-12	2 6.20E-15	5.98E-03	3.76E+02	19.4	8.37E-15	i	1.99E-08	3.90E-10	133	1.58	1.59	1.12E+08	9.63E-13	3 20.1
Tellurium	Те	52	128	52	2 76	1.02E-12	2 6.30E-15	6.19E-03	3.90E+02	2 19.7	8.50E-15		2.06E-08	3.97E-10	123	1.52	1.52	1.14E+08	9.41E-13	20.5
Iodine	I	53	127	53	8 74	9.98E-13	6.28E-15	6.30E-03	3.96E+02	19.9	8.48E-15	i	2.10E-08	3.96E-10	115	1.47	1.47	1.16E+08	9.20E-13	3 20.7
Xenon	Xe	54	131	. 54	1 77	9.80E-13	6.35E-15	6.48E-03	4.08E+02	2 20.2	8.57E-15	i	2.16E-08	4.00E-10	108	1.43	1.43	1.18E+08	9.00E-13	3 21.1
Cesium	Cs	55	133	55	5 78	9.62E-13	6.38E-15	6.63E-03	4.18E+02	2 20.4	8.61E-15	i	2.21E-08	4.02E-10	298	2.37	2.37	1.20E+08	8.81E-13	3 21.4
Barium	Ва	56	137	56	5 81	9.45E-13	3 6.44E-15	6.82E-03	4.30E+02	2 20.7	8.70E-15		2.27E-08	4.06E-10	253	2.18	2.19	1.23E+08	8.62E-13	\$ 21.7
Lanthanum	La	57	139	57	82	9.28E-13	6.48E-15	6.98E-03	4.39E+02	2 21.0	8.74E-15		2.32E-08	4.08E-10	)			1.25E+08	8.44E-13	\$ 22.0
Cerium	Ce	58	140	58	8 82	9.12E-13	6.49E-15	7.12E-03	4.48E+02	2 21.2	8.76E-15		2.37E-08	4.09E-10	)			1.27E+08	8.26E-13	\$ 22.2
Praseodymiur	r Pr	59	141	. 59	82	8.97E-13	6.51E-15	7.26E-03	4.57E+02	2 21.4	8.78E-15		2.42E-08	4.10E-10	247	2.16	5 2.16	1.29E+08	8.09E-13	\$ 22.5
Neodymium	Nd	60	144	60	84	8.82E-13	6.55E-15	7.43E-03	4.68E+02	2 21.6	8.84E-15		2.48E-08	4.13E-10	206	1.97	1.97	1.31E+08	7.93E-13	\$ 22.8
Promethium	Pm	61	. 145	61	. 84	8.67E-13	6.57E-15	7.57E-03	4.77E+02	2 21.8	8.86E-15		2.52E-08	4.14E-10	205	1.97	1.97	1.33E+08	7.77E-13	3 23.1
Samarium	Sm	62	150	62	88	8.53E-13	6.64E-15	7.78E-03	4.90E+02	2 22.1	8.96E-15		2.59E-08	4.18E-10	238	2.12	2.12	1.36E+08	7.61E-13	\$ 23.4
Europium	Eu	63	152	63	8 89	8.40E-13	6.67E-15	7.94E-03	5.00E+02	2 22.4	9.00E-15		2.65E-08	4.20E-10	231	2.09	2.09	1.38E+08	7.46E-13	3 23.7
Gadolinium	Gd	64	157	64	93	8.27E-13	6.74E-15	8.16E-03	5.14E+02	2 22.7	9.10E-15		2.72E-08	4.25E-10	233	2.10	2.10	1.40E+08	7.31E-13	\$ 24.1
Terbium	Tb	65	159	65	94	8.14E-13	6.77E-15	8.32E-03	5.24E+02	2 22.9	9.14E-15		2.77E-08	4.26E-10	225	2.06	2.06	1.42E+08	7.16E-13	3 24.4
Dysprosium	Dy	66	163	66	5 97	8.02E-13	6.83E-15	8.52E-03	5.37E+02	2 23.2	9.22E-15		2.84E-08	4.30E-10	228	2.07	2.08	1.44E+08	7.02E-13	\$ 24.7
Holmium	Но	67	165	67	98	7.90E-13	6.86E-15	8.68E-03	5.47E+02	2 23.4	9.25E-15		2.89E-08	4.32E-10	)			1.47E+08	6.89E-13	\$ 25.0
Erbium	Er	68	167	68	99	7.78E-13	6.88E-15	8.85E-03	5.57E+02	2 23.6	9.29E-15		2.95E-08	4.34E-10	226	2.06	2.07	1.49E+08	6.75E-13	3 25.3
Thulium	Tm	69	169	69	100	7.67E-13	6.91E-15	9.01E-03	5.68E+02	2 23.8	9.33E-15		3.00E-08	4.35E-10	222	2.05	2.05	1.51E+08	6.62E-13	3 25.6
Ytterbium	Yb	70	173	70	103	7.56E-13	6.97E-15	9.22E-03	5.80E+02	2 24.1	9.40E-15	e1{-2,4}	3.07E-08	4.39E-10	222	2.05	2.05	1.53E+08	6.50E-13	3 26.0
Lutetium	Lu	71	. 175	71	. 104	7.45E-13	6.99E-15	9.38E-03	5.91E+02	2 24.3	9.44E-15	e1{-2,4}	3.13E-08	4.40E-10	217	2.02	2.03	1.55E+08	6.37E-13	3 26.3
Hafnium	Hf	72	178	72	106	7.35E-13	3 7.03E-15	9.57E-03	6.03E+02	2 24.6	9.49E-15		3.19E-08	4.43E-10	208	1.98	1.98	1.58E+08	6.25E-13	3 26.6
Tantalum	Та	73	181	73	108	7.25E-13	3 7.07E-15	9.76E-03	6.14E+02	24.8	9.54E-15		3.25E-08	4.45E-10	200	1.94	1.94	1.60E+08	6.13E-13	26.9
Tungsten	W	74	184	74	110	7.15E-13	3 7.11E-15	9.95E-03	6.26E+02	2 25.0	9.60E-15		3.31E-08	4.48E-10	193	1.91	1.91	1.62E+08	6.02E-13	3 27.3
Rhenium	Re	75	186	75	5 111	7.05E-13	3 7.14E-15	1.01E-02	6.37E+02	25.2	9.63E-15		3.37E-08	4.49E-10	188	1.88	1.89	1.64E+08	5.90E-13	3 27.6

Table 2. Calculation of interior  $\{N,n\}$  QM system for nuclear-electron system based on  $e1\{0,1\}$ .

Osmium	Os	76	190	76	114	6.96E-13	7.19E-15	1.03E-02	6.50E+02	25.5	9.70E-15		3.44E-08	4.53E-10	185	1.87	1.87	1.66E+08	5.79E-13	28.0
Iridium	Ir	77	192	77	115	6.87E-13	7.21E-15	1.05E-02	6.61E+02	25.7	9.73E-15		3.50E-08	4.54E-10	180	1.84	1.84	1.68E+08	5.68E-13	28.3
Platinum	Pt	78	195	78	117	6.78E-13	7.25E-15	1.07E-02	6.73E+02	25.9	9.78E-15		3.56E-08	4.56E-10	177	1.83	1.83	1.71E+08	5.58E-13	28.6
Gold	Au	79	197	79	118	6.70E-13	7.27E-15	1.09E-02	6.84E+02	26.2	9.82E-15		3.62E-08	4.58E-10	174	1.81	1.81	1.73E+08	5.47E-13	28.9
Mercury	Hg	80	201	80	121	6.61E-13	7.32E-15	1.11E-02	6.97E+02	26.4	9.88E-15		3.69E-08	4.61E-10	171	1.80	1.80	1.75E+08	5.37E-13	29.3
Thallium	TI	81	204	81	123	6.53E-13	7.36E-15	1.13E-02	7.10E+02	26.6	9.93E-15		3.75E-08	4.63E-10	156	1.72	1.72	1.77E+08	5.27E-13	29.7
Lead	Pb	82	207	82	125	6.45E-13	7.39E-15	1.15E-02	7.22E+02	26.9	9.98E-15		3.82E-08	4.66E-10	154	1.70	1.71	1.79E+08	5.17E-13	30.0
Bismuth	Bi	83	209	83	126	6.37F-13	7.42F-15	1.16E-02	7.33E+02	27.1	1.00F-14		3.88F-08	4.67E-10	143	1.64	1.64	1.82F+08	5.07F-13	30.4
Polonium	Po	84	209	84	125	6 30E-13	7 42E-15	1 18E-02	7 42E+02	27.2	1 00F-14		3 92F-08	4 67E-10	135	1.60	1.60	1 84F+08	4 98F-13	30.6
Astatine	At	85	210	85	125	6 22F-13	7 43E-15	1 19E-02	7 52E+02	27.4	1.00E-14		3 98F-08	4 68E-10	100	1.00	1.00	1.86E+08	4 88F-13	31.0
Radon	Rn	86	220	86	136	6 15E-13	7.57E-15	1 23E-02	7 75E+02	27.8	1.00E 11		4 10E-08	4 77E-10	120	1 50	1 51	1 88F+08	4 79F-13	31.6
Francium	Fr	87	222	87	136	6.08E-13	7 58E-15	1.25E-02	7.85E+02	27.0	1.02E 14		4.15E-08	4.77E-10	120	1.50	1.51	1 90E+08	4 70E-13	31.0
Radium	Ra	88	225	88	130	6.01F-13	7.50E 15	1.25E 02	7.05E+02	20.0	1.02E 14		4.13E 00	4.97E 10				1.93E+08	4.61E-13	32.3
Actinium	Ac	90	220	90	120	5 0/E-12	7.62E-15	1.270 02	9 09E±02	20.2	1.025-14		4.275-09	4.00E 10				1.055±00	4.526-12	22.5
Thorium	Th	00	227	00	1/2	5.94L-13	7.69E-15	1.200-02	8.00L+02	20.4	1.031-14		4.271-08	4.000-10				1.07E±00	4.321-13	22.0
Drotactinium	Do	50	232	30	142	5.00L-13	7.000-10	1.310-02	0.2315+02	20.7	1.046-14		4.331-08	4.041-10				1.005+00	4.451-13	33.0
Uranium	rd II	91	231	91	140	5.010-15	7.0/2-15	1.32E-02	0.510702	20.0	1.046-14		4.40E-08	4.05E-10				2.01E+00	4.55E-15	22.2
Nenturi	U Nor	92	230	92	140	5.75E-15	7.735-15	1.35E-02	0.400702	29.1	1.035-14		4.49E-00	4.00E-10				2.010+00	4.202-13	24.2
Neptunium	пр	93	237	93	144	5.09E-13	7.74E-15	1.30E-02	8.50E+U2	29.3	1.04E-14		4.53E-08	4.8/E-10				2.03E+08	4.18E-13	34.2
Plutonium	Pu	94	244	94	150	5.63E-13	7.81E-15	1.39E-02	8.74E+02	29.6	1.05E-14		4.62E-08	4.92E-10				2.06E+08	4.09E-13	34.7
Americium	Am	95	243	95	148	5.5/E-13	7.80E-15	1.40E-02	8.82E+02	29.7	1.05E-14		4.6/E-08	4.91E-10				2.08E+08	4.01E-13	35.0
Curium	Cm	96	247	96	151	5.51E-13	7.84E-15	1.42E-02	8.96E+02	29.9	1.06E-14	e1{-2,5}	4.74E-08	4.94E-10				2.10E+08	3.93E-13	35.4
Berkelium	Bk	97	247	97	150	5.45E-13	7.84E-15	1.44E-02	9.06E+02	30.1	1.06E-14	e1{-2,5}	4.79E-08	4.94E-10				2.12E+08	3.85E-13	35.8
Californium	Ct	98	251	98	153	5.40E-13	7.88E-15	1.46E-02	9.20E+02	30.3	1.06E-14		4.87E-08	4.97E-10				2.14E+08	3.77E-13	36.3
Einsteinium	Es	99	252	99	153	5.34E-13	7.90E-15	1.48E-02	9.31E+02	30.5	1.07E-14		4.92E-08	4.97E-10				2.17E+08	3.69E-13	36.7
Fermium	Fm	100	257	100	157	5.29E-13	7.95E-15	1.50E-02	9.46E+02	30.8	1.07E-14		5.00E-08	5.00E-10				2.19E+08	3.62E-13	37.2
Mendelevium	Md	101	258	101	157	5.24E-13	7.96E-15	1.52E-02	9.57E+02	30.9	1.07E-14		5.06E-08	5.01E-10				2.21E+08	3.54E-13	37.6
Nobelium	No	102	259	102	157	5.19E-13	7.97E-15	1.54E-02	9.68E+02	31.1	1.08E-14		5.12E-08	5.02E-10				2.23E+08	3.46E-13	38.1
Lawrencium	Lr	103	286	103	183	5.14E-13	8.24E-15	1.60E-02	1.01E+03	31.8	1.11E-14		5.34E-08	5.19E-10				2.25E+08	3.39E-13	39.1
Rutherfordium	Rf	104	267	104	163	5.09E-13	8.05E-15	1.58E-02	9.97E+02	31.6	1.09E-14		5.27E-08	5.07E-10				2.28E+08	3.31E-13	39.1
Dubnium	Db	105	268	105	163	5.04E-13	8.06E-15	1.60E-02	1.01E+03	31.7	1.09E-14		5.33E-08	5.08E-10				2.30E+08	3.24E-13	39.6
Seaborgium	Sg	106	289	106	183	4.99E-13	8.26E-15	1.66E-02	1.04E+03	32.3	1.12E-14		5.52E-08	5.20E-10				2.32E+08	3.16E-13	40.6
Bohrium	Bh	107	270	107	163	4.94E-13	8.08E-15	1.63E-02	1.03E+03	32.1	1.09E-14		5.44E-08	5.09E-10				2.34E+08	3.09E-13	40.6
Hassium	Hs	108	269	108	161	4.90E-13	8.07E-15	1.65E-02	1.04E+03	32.2	1.09E-14		5.49E-08	5.08E-10				2.36E+08	3.01E-13	41.1
Meitnerium	Mt	109	278	109	169	4.85E-13	8.16E-15	1.68E-02	1.06E+03	32.5	1.10E-14		5.60E-08	5.14E-10				2.38E+08	2.94E-13	41.8
Darmstadtium	Ds	110	281	110	171	4.81E-13	8.19E-15	1.70E-02	1.07E+03	32.7	1.11E-14		5.67E-08	5.16E-10				2.41E+08	2.87E-13	42.4
Roentgenium	Rg	111	282	111	171	4.77E-13	8.20E-15	1.72E-02	1.08E+03	32.9	1.11E-14		5.73E-08	5.16E-10				2.43E+08	2.79E-13	43.0
Copernicium	Cn	112	285	112	173	4.72E-13	8.23E-15	1.74E-02	1.10E+03	33.1	1.11E-14		5.80E-08	5.18E-10				2.45E+08	2.72E-13	43.6
Nihonium	Nh	113	286	113	173	4.68E-13	8.24E-15	1.76E-02	1.11E+03	33.3	1.11E-14		5.86E-08	5.19E-10				2.47E+08	2.65E-13	44.3
Flerovium	Fl	114	289	114	175	4.64E-13	8.26E-15	1.78E-02	1.12E+03	33.5	1.12E-14		5.93E-08	5.20E-10				2.49E+08	2.57E-13	45.0
Moscovium	Mc	115	290	115	175	4.60E-13	8.27E-15	1.80E-02	1.13E+03	33.7	1.12E-14		5.99E-08	5.21E-10				2.52E+08	2.50E-13	45.6
Livermorium	Lv	116	293	116	177	4.56E-13	8.30E-15	1.82E-02	1.15E+03	33.9	1.12E-14		6.07E-08	5.23E-10				2.54E+08	2.43E-13	46.4
Tennessine	Ts	117	294	117	177	4.52E-13	8.31E-15	1.84E-02	1.16E+03	34.0	1.12E-14		6.12E-08	5.23E-10				2.56E+08	2.35E-13	47.2
Oganesson	Og	118	294	118	176	4.48E-13	8.31E-15	1.85E-02	1.17E+03	34.2	1.12E-14	e1{-2.6}=e	6.18E-08	5.23E-10				2.58E+08	2.28E-13	47.9
Table 2h The e	estimater	virtual 7>	137 ator	n that onl	v exist in th	e high G-n	ressured wo	rld like in {	(-3 n=2 6) c	elestial ho	dy but no	t in the no	rmal world							{0.48//6}=
Tuble 20. The C	cstimatee	viittuui 2>	2 56		y existin ti	ie nigit o p		ind like in p	[ 3,11-20] 0	crestian be	Juy, but no	e in the no								[0,40,70]-
			2.30																	
ratio of		ncoude 7	true	true	true						ov po et e -									
		µseudo Z	uue	uue	u de						expected									
iviass/Atom#		#	mass #	acomic#	neutron #		0.555.15	2.077.0	4 207 2		n • -		C 007 5	5 ac= · ·						
Predicted e1{-1	1,1} nucle	128	320	128	192	4.13E-13	8.55E-15	2.07E-02	1.30E+03	36.1	36	e1{-1,1}	6.89E-08	5.38E-10						
Predicted e{-1,	,2} nucleu	360	900	360	540	1.47E-13	1.21E-14	8.21E-02	5.17E+03	71.9	72	e1{-1,2}	2.74E-07	7.60E-10						
Predicted e1{-1	1,3} nucle	670	1675	670	1005	7.90E-14	1.48E-14	1.88E-01	1.18E+04	108.8	108	e1{-1,3}	6.26E-07	9.35E-10						
Predicted e1{-1	1,4} nucle	1020	2550	1020	1530	5.19E-14	1.71E-14	3.29E-01	2.07E+04	144.0	144	e1{-1,4}	1.10E-06	1.08E-09						
Predicted e1{-1	1,5} nucle	1420	3550	1420	2130	3.73E-14	1.91E-14	5.12E-01	3.22E+04	179.5	180	e1{-1,5}	1.71E-06	1.20E-09						
Predicted e1{-1	1,6} nucle	1870	4675	1870	2805	2.83E-14	2.09E-14	7.39E-01	4.65E+04	215.7	216	e1{-1,6}=e	2.46E-06	1.32E-09						

Note-1: e1{N,n} (a {N,n} QM structure with the prefix "e1") always has a shrunk (or compressed)  $r_{e1}$  unless it is specified as not shrunk. A {N,n} QM structure without any prefix (e.g., {-12,2}) is always  $r_1$  uncompressed since it is under Sun{0,1//6}. Note-2: Yellow cells in column 2 are inert elements (stable states for  $n_{electron}$ ). Green cells in column 11 are expected stable states (for  $n_{nucleus}$ ) for e1{N,n//6} QM structure. Blue cells in column 21 are expected stable states (for  $n_{nucleus}$ ) for e1{N,n//7} QM structure.

Note-3: For Table 2b, see result & discussion 2.

In column 12, start with the shrunk  $r_{e1}$  (in column 7), we calculated back the nucleus  $r_{nuc}$  through the interior {N,n} calculation:  $r_{nuc} = (n_{nuc}/6^3)^{2*}r_{e1}$ , where  $n_{nuc}$  is in column 11, and  $r_{e1}$  is in column 7. Here we assumed that the EM-force caused (interior) {N,n} QM structure also has base- $n = 5*6^{\circ}$ . The back-calculated  $r_{nuc}$  in column 12 matched the  $r_{nuc}$  in column 8 quite well. This confirms that the atom's nucleus-electron (EM-force) system can be described with a {N,n//6} QM structure. Notice that back-calculated  $r_{nuc}$  (in column 12) is  $1.35 \times$  larger than the original one (in column 8). This is because the formula  $r_{nuc} = (n_{nuc}/6^{\circ}3)^{\circ}2*r_{e1}$  is an approximation. If we use the accurate formula  $r_{nuc} = (n_{nuc}/251)^{\circ}2*r_{e1}$ , where 251 come from SQRT(1/(8.4E-16/5.29E-11)), then we will get exact same  $r_{nuc}$  as in column 8. However we like to use  $r_{nuc} = (n_{nuc}/6^{\circ}3)^{\circ}2*r_{e1}$  here because it is purely based on {N,n//6} QM model calculation.

In column 13, the interior {N,n//6} was assigned for each element, with its own shrunk electron ground state  $r_{e1}$  (in column 7) as  $e1\{0,1//6\}$ . We can see that while the hydrogen atom's nucleus (proton, r = 8.4E-16 meters) is the  $e1\{-3,1//6\}$ 

e1{0,1/6^3} interior QM structure of Bohr radius ( $r_{e1} = 5.29E-11$  meters), Ne element's nucleus is about the e1{-2,1/6} interior QM structure of its electron's n = 1 shell e1{0,1//6}, and Og element's nucleus is about the e1{-1,1//6} interior QM structure of its electron's n = 1 shell. In this way, we established the interior e1{N,n} QM relationship between element's nucleus and its electron n = 1 shell (see Figure 1 for illustration). If we use exterior {N,n//6} QM, and take Hydrogen atom's  $r_{nuc}$  (relative to electron n = 1 shell) as {0,1//6} in size, then He's  $r_{nuc}$  is {0,2//6} in size, Ne's nucleus (approximately) finishes the first super-shell of nucleons {0,n=2..6} = {1,1//6} in size, Og's nucleus (approximately) finishes the second super-shell of nucleons {1,n=2..6} = {2,1//6} in size. If we use interior {N,n//6} QM, and take Og atom's nucleus r (relative to electron n=1 shell) as {0,1//6} in size, then Ne's nucleus is (approximately) at {-1,1//6} in size, Hydrogen atom's nucleus is (approximately) at {-2,1//6} in size.

The shrinking of  $r_{e1}$  as Z increasing can be explained as that the (strong) EM-force shrinks the space (in comparison with the G-force). In column 14, a pretend atom's r (not a nucleus' r) was calculated by using  $n_{nuc}$  and (fixed)  $r_1 = 5.29E-11$  meters, with  $r_n = r_1*n_{nuc}^2$ . So this is the idealized  $r_n$  (or size of atom) if there is no EM-force caused  $r_{e1}$  shrinking (when the nucleus increases its Z from 1 to 118). In column 15, as a comparison to column 14,  $r_n$  is calculated by using  $n_{nuc}$ , but with  $r_{e1} = 5.29E-11 / Z$ . The result is useless because no electron shielding effect is considered (Note: as a citizen scientist, I don't know how to calculate the electron shielding effect).



Figure 1. Illustration of the nuclear {N,n//6} QM structure by using the ground state electron's orbit as  $e1\{0,1//6\}$ . To illustrate that  $e1\{-3,1//6\}$  is the size of a single proton, a  $e1\{-2,1//6\}$  sized nucleus contains up to 10 protons, and a  $e1\{-1,1//6\}$  sized nucleus contains up to 118 protons. Note: the r-dimension is not on scale.

#### More results and discussions on Table 2:

1) For columns 7 to 11, we set  $e1\{0,1//6\}$  at Bohr radius  $a_0 = r_1 = 5.29E-11$  meters. Hydrogen atom's  $r_{nuc}$  is  $(1/6^3)^2 of r_{e1}$ , so its  $r_{nuc}$  is at  $e1\{0,1/6^3//6\} = e1\{-3,1//6\}$  QM state (in size). Helium atom's  $r_{nuc}$  is  $(2/6^3)^2 of$  its shrunk  $r_{e1}$ , so its  $r_{nuc}$  is at  $e1\{0,2/6^3//6\} = e1\{-3,2//6\}$  QM state (in size). Li atom's  $r_{nuc}$  is  $(3/6^3)^2 of$  its shrunk  $r_{e1}$ , so its  $r_{nuc}$  is at  $e1\{0,3/6^3//6\} = e1\{-3,3//6\}$  QM state (in size). Ne (and/or F) atom's  $r_{nuc}$  is about  $e1\{0,1/6^2//6\} = e1\{-2,1//6\}$  interior QM state (in size) of its electron n = 1 shell, Og atom's  $r_{nuc}$  is about  $e1\{0,1/6//6\} = e1\{-1,1//6\}$  interior QM state (in size) of its electron n = 1 shell. Og atom's  $r_{nuc}$  is about  $e1\{0,1/6//6\} = e1\{-1,1//6\}$ , those  $\{N,n=2...5//6\}$  sized nuclei with integer n # should be more stable than those non-integer n's nuclei. These integer n(s) are: 1 (for H), 2 (for He), 3, 4, 5, 6, 12, 18, 24, 30, 36, etc. However, this higher stability is for the nucleus, not for electron shells, so they do not necessary to have higher stability in the (chemical element's) periodic table. See section II-b for more detailed explanation. We also hope that the (isotopic) nuclear physics experimental data will support our prediction. This work uncovered that the  $\{N,n//6\}$  QM structure of nucleus is a "recessive gene", it is not a "dominant gene". You will not be able to see it unless you use the interior  $\{N,n\}$  QM analysis based on the ground state electron  $\{0,1\}$  for atom's nucleus-electron system. If using the normal  $r_1 = r_{nuc=1}$ , then you don't see any  $\{N,n//6\}$  QM structure!

2) For (the bottom) Table 2b, from point view of {N,n} QM structure, the e1{-1,n=2..6//6} sized super large nucleus (or atoms with Z > 118) is OK to exist (although in the traditional Bohr QM, the orbital velocity of n=1 electron has to be smaller than the speed of light, so Z cannot be larger than 137). Those n(s) are 2, 3, 4, 5, 6 (multiply by 6^2 to get the total

 $n_{nuc} = 72$ , 108, 144, 180, 216 with  $r_1$  at e1{-3,1//6}, see column 12 of Table 2b). To add the predicted nuclides of e1{-1,n=2..6//6} sizes in Table 2b, we first manually adjusted their Z numbers (in column 3, Z > 118) to get the mass number M# = Z# \* 2.5. (Note: this M# / Z# =2.5 was assumed to be a constant for all super large atoms and this is the best "citizen scientist leveled" approximation I can do now). We then got their  $r_{nuc} = 1.25 * (M#)^{(1/3)}$ , then calculated the  $n_{nuc} = sqrt(r/r_1)$  (in column 11) to match the expected  $n_{nuc}$  (in column 12). So the predicted atom's (pseudo) Z# are about 360, 670, 1020, 1420, 1870. The analysis in Table 2b gave the following hypothesis: it is possible to form atom with Z >137. When Z >137, the Z become pseudo-Z, any proton beyond 137 will be merged with electron, so it will keep the true proton number < 137 in a nucleus (even its pseudo-Z > 137, see SunQM-5s1 for more explanation, this may cause M# / Z# ratio increase as Z# increasing). Thus, it will also keep electron's  $v_1 < c$  (probably by sharing electrons between nuclei, or forming electron sea). Note: other scientists may already have many discussions about this topic before. Here we repeated this discussion only under the framework of the {N,n//6} QM.

3) Now we try to calculate out the maximum  $r_{nuc}$  (which is caused by the maximum  $n_{nuc}$  but before the  $r_{nuc}$  crash with the electron n = 1 shell, and remember this el shell shrinks as Z# increases). At the nucleus size of el {0,1//6} = el {-1,6//6}, we obtained  $r_{nuc}/r_1 = 0.739$  (in column 9). It is very close to  $(n=5^2)/(n=6^2) = 0.694$ . We believed that this is the maximum  $r_{nuc}$  (without considering the relativity) before crash into its own shrunk el {0,1//6} electron shell. So (without considering the relativity), el {0,1//6} = el {-1,6//6} sized nucleus is the maximum-ever-possible nucleus (without crash into the shrunk  $r_1$  electron shell). At the max  $r_{nuc} = 2.09E$ -14 meters (see Table 2b last row, and column 8), electron's  $r_1 = 2.83E$ -14 meters,  $Z \approx 1870$ , so the number of electrons  $\approx 1870$ . (Note: it corresponds to the maximum electron shell of  $n \le 22$ , the detailed calculation of the electron configuration is not shown here, but we can imagine that a chemical element periodical table with 22 electron periods, although most of them may have to merge with the protons in the nucleus?). However, we cannot use  $r_n = r_1/Z * n^22$  to calculate the maximum electron shell's  $r_n$  (or max atom-r), due to that it does NOT count-in the inner electron shield effect that will increase outer electron's  $r_1$ .

4) (Note: here we rename the period in Mendeleev's periodic table as "electron period", to distinguish it from the N period in {N,n} QM). For column 14, if use  $n_{nuc}$ , and use unshrunk  $r_1 = 5.29E-11$  and fix this value for all elements, then calculate the maximum electron's  $r_n = r_1 * n_{nuc}^2$  (which may relate to the size of an unshrunk atom, using  $n_{nuc}$  in column 11). Then we see that for hydrogen atom,  $r_{unshrunk} = 5.29E-11$  meters,  $n_{nuc} = e1\{-3,1//6\}$ ; for Ne atom,  $r_{unshrunk} = 2.14E-9$  meters,  $n_{nuc} \approx e1\{-2,1//6\}$ ; for Og atom,  $r_{unshrunk} = 6.17E-8$  meters,  $n_{nuc} \approx e1\{-1,1//6\}$ . When comparing these  $r_{unshrunk}$  values to the {N,n} of sub-quark to universe in Table 1, we immediately realized that Hydrogen atom's  $r_{unshrunk} = Z*r' = 5.29E-11$  meters matches to {-12,1//6} 's r = 3.67E-11, Ne atom (Z = 10, the end of electron period 2) 's  $r_{unshrunk} = 2.14E-9$  meters closely matches to {-11,1//6} 's r = 1.32E-9, Og atom (Z = 118, the end of electron period 7) 's  $r_{unshrunk} = 6.16E-8$  meters closely matches to {-10,1//6} 's r = 4.76E-8. So {-10,1//6} super stable {N,n} QM structure is the largest atom Og (Z=118) if there is no Z induced (electron n = 1 shell)  $r_1$  shrink. For column 15, after including Z induced  $r_1$  contraction, then we are not able to calculate the right electron's  $r_n$  due to the inner electron shielding effect, so no useful result is obtained in this column.

There is an even more important result that can be extracted from this analysis of  $n_{nuc}$  (in column 13): If taking Hydrogen atom's  $n_{nuc} = 1$  as  $\{0,1//6\}$ , named as proton $\{0,1//6\}$ , or **prot\{0,1//6\}**, then elements from He to Ne (with  $n_{nuc}$  from 2 to ~6, or from Z = 2 to Z = 10, or up to the end of electron period 2 in the chemical element periodic table) will fill the  $n_{nuc}$ 's super-shell space prot $\{0,n=1..5//6\}$  o fully (with the final size at prot $\{0,6//6\}$  = prot $\{1,1//6\}$ ); and elements from Na to Og (with  $n_{nuc}$  from > 6 to ~36, or from Z = 11 to Z = 118, or from electron period 3 to electron period 7 in the electron period table) will fill the  $n_{nuc}$ 's super-shell space prot $\{1,n=1..5//6\}$  o fully (with the final size at prot $\{1,6//6\}$  = prot $\{2,1//6\}$ , see Table 3's period N = -16, -15, and -14). If there are electron period 8 elements (Z ≥ 119), they most likely need to start a new  $n_{nuc}$ 's super-shell prot $\{2,n=1..5//6\}$  o and to fill it. But to start a new  $n_{nuc}$  super-shell is much more difficult than to finish the rest space in the same super-shell, and this means that the Og may be the last element we can find. Or we may never be able to produce element in electron period 8, not due to it start a new (electron shell) period, but due to it starts a new  $n_{nuc}$  (nucleon) super-shell.

5) For columns 16-18. In column 16, atomic radii calculated by Enrico Clementi, et al. <sup>[14]</sup> (from wiki "Atomic Radius") are listed. This data is very close to the experimental measured atom's r. In column 17, the n of the outmost electron shell is

calculated (relative to Hydrogen atom's n=1, or r = 53E-12 meters) as =sqrt[(column-11) / 53]. In column 18, the n of the out-most electron shell is calculated (relative to  $r_{e1} = 5.29E-11$  meters, or = sqrt[(column-11) \*E+12 / 5.29E-11 ]), and the result is the same as that in Column 17. The result revealed that almost all atoms have theirs n within 2, or their size of r  $\leq 4 \times 5.29E-11$  meters = 2.1E-10 meters. Here we call it "**atom size n=2 effect**", meaning although atom's out most electron shell may be up to n=7, but its real size is no more than n=2. Obviously, this effect is due to the strong EM-force induced  $r_n$  QM space shrinking.

6) For columns 19-21, the orbital electron's high speed caused relativity effect is included for the calculation (at the citizen scientist level). Once add relativity, for large Z, the relativistic

$$r_{e1r} = r_{e1} / Z^* \operatorname{sqrt}(1 - v_1^2 / c^2)$$

eq-1

become even smaller <sup>[15]</sup>, then  $n_{nuc} = \operatorname{sqrt}(r_{n-nuc} / r_{e1r})$  will become even bigger. (Note: For  $r_{n-nuc}$ , the relativity is not calculated because I don't how to calculate it. Also, as a citizen scientist, I don't know how correct the eq-1 is. Therefore, the whole Discussion-6 is a citizen scientist leveled discussion). For example, Og-118's  $r_{e1} = a_0 / 118 = 4.48E-13$  meters, but due to its  $v_1 = 2.58E+8$  m/s close to light speed c, its  $r_{e1r}$  shrunk to 50.8% (= 2.28E-13 meters). So the relativistic  $n_{nuc} = \operatorname{sqrt}(r_n / r_{e1r}) = 47.9$  for Og-118 element (see column 21). If it is {N,n//6} QM structure, then {0,48//6}={1,8//6} ≈{2,1.3//6}. If it is {N,n//7} QM structure, then {0,48} ≈ {0,49} = {1,7//7} = {2,1//7}, which 49 = 7^2, and Z = 118 is the end of electron period 7.

At beginning, we assumed that the EM-force caused (interior) {N,n} QM structure also has the base-pFactor-n =  $5*6^{\circ}$ . (Note: we should **change the name of "pFactor" into "qFactor"**, because it is same as the quantum number q in {N,n//q} or |qnlm>). The reason is that G-force has {N,n//6} QM system, and this n//6 "gene" should come from its ancestor of either the F4-force (the degenerated G-, EM-, S-, W-forces, see SunQM-2), or the unified three pair (G/RFg, E/RFe, S/RFs) forces (see SunQM-6). So it is expected that F4-force's another descendant EM-force also carries the same n//6 "gene". It is also expected that F4-force's 3rd descendant, the S-force (including its residue force, i.e., the nuclear force), also carries the same n//6 "gene". This expectation is supported by the resulted n<sub>nuc</sub> in column 11 (without relativity) that matches the nuclear {N,n//6} QM structure (due to H's n<sub>nuc</sub> = 1 at e1{-3,1//6}, Ne's n<sub>nuc</sub>  $\approx 6$  at e1{-2,1//6}, and Og's n<sub>nuc</sub>  $\approx 36$  at e1{-1,1//6}, and this nuclear structure is caused by the nuclear force plus the EM-force).

But after adding relativity, it seems that the {N,n/7} QM structure matches even better at high Z (due to H's  $n_{nuc} = 1$  at e1{-3,1//7}, Ne's  $n_{nuc} = 6.4 \approx 7$  at e1{-2,1//7}, and Og's  $n_{nuc} \approx 48 \approx 7^{2}$  at e1{-1,1//7}, see column 21). So it seems that due to the shrink of the electron ground state  $r_{e1}$ , the electron's  $r_n$  is also shrunk, and it frees up the original electron's  $r_n = 5$  (or even  $r_n = 4$ ) orbital shell space so that the higher n (n = 6, 7, or even higher electron orbit) can fill in. Due to that we do not know how to calculate the shrunk electron's  $r_n$ , we are not able to estimate how many extra n (=7, 8, etc.) shells of electron can be added to fill-in the left-over electron shell space (for the shrunk electron's  $r_n$  orbital space). According to the "atom size n=2 effect" in discussion 5, it can be  $\geq 7$ .

General relativity tells us that when Gravity-force getting stronger, the space getting shrink (in force direction). It can be expanded to as F4-force (or its descendants S-force and EM-force) getting stronger, the space getting shrink. So can we attribute the shrinking of  $r_{e1}$  (=  $r_1 / Z$ , in column 7) and the shrinking of  $r_n$  (=  $r_{e1} * n^2$ ) as Z increasing (equivalent to EM-force increasing) to the general relativity effect? The form of eq-1 may suggest that these are two different effect. The relativity effect of the {N,n} QM calculation is a too big topic for this paper. It will be studied in our future paper SunQM-7s1 (hopefully). Note: I am not a relativity physicist, Discussion 6 contained "citizen scientist leveled" calculations, and I am not sure they are correct or not. I want to present these discussions because they are very novel (at least to me), and they might inspire other scientists (even my calculations might be not correct).

# II-b. The $\{N,n//6\}$ (and $\{N,n//7\}$ ) QM structural stability of the nuclides played an important role in determining the abundancy of each element in our universe

15,1//6}, then, nuclides of He through Ne belong to the  $\{-15,n=1..5//6\}$  o orbital super shell space, and nuclides of Na through Og belong to the  $\{-14,n=1..5//6\}$  o orbital super shell space (see Table 3). Similarly, because H atom has its outmost electron orbital shell at  $\{-12,1//6\}$  o, then, electron period 1 through electron period 7 elements belong to the  $\{-12,n=1..7//6\}$  o orbital super shell space.

By comparing the {N,n//6} QM structural information of nuclides (in Table 2) to the abundancy of these elements in our galaxy (in Figure 2), it provided us some new knowledge on the nuclear physics. The size n = 2 nuclides at {-14,2//6} (with  $n_{nuc} \le 12 = 2 * 6^{1}$ , or all nucleons in the orbital shell space of {-14,1//6}o, see Table 2 and Table 3) ended at Fe element's nucleus. It means that Fe nucleus is the only completely filled nucleus (or it has 100 % nucleon occupancy in the {-14,1//6}o nucleon orbital shell space) among the nucleon orbital n = 1 nuclides (Z = 11 ... 26,  $n_{nuc} = 7$  ... 12, or among Na, Mg, Al, Si, P, S, Cl, Ar, K, Ca, Sc, Ti, V, Cr, Mn, and Fe), and even among the orbital n = 2 nuclides (Z = 27 ... 45,  $n_{nuc} = 13$  ... 17), so it is the most stable {N,n//6} QM structure among the (orbital) n = 1 (and even orbital n = 2) nuclides. This is exactly like why the inert element Ne is chemically most stable atom among the electron period 2 elements because it has the outmost electron shell completely filled (or it has the 100% electron occupancy in the n = 2 electron orbital shell space). This explained why Fe element is relatively abundant in the universe (in comparison to its adjacent elements, like Cr, Mn, Co, Ni, etc., see Figure 2), simply because its nucleus (and not its electron shell) is more stable than that of its adjacent elements. This also explained why Fe atom is the heaviest element that a nuclear fusion reaction can form without adding extra energy.

Next, for the size n = 3 nuclides (or all nucleons in the nucleon orbital shell space of n = 2), if still using {N,n//6} QM structure, the nuclides will end at {-14,3//6}, between Pd (Z= 46,  $n_{nuc} = 18.0 = 3 * 6^{1}$ ) and Ag (Z= 47,  $n_{nuc} = 18.1 \approx 3 * 6^{1}$ ). However, Figure 2 showed that while none of them has a high abundancy, Xe (Z= 54 and the adjacent) nuclides showed the abnormal high abundancy. Since Xe has the relativistic  $n_{nuc-rel} = 21.1$  which fits to {-14,3//7} sized QM structure (see Table 2), it revealed that from Z = 26 (Fe) to Z = 54 (Xe), the nuclides {N,n//q} structure transformed from {-14,2//6} to {-14,3//7} structurally, so that Xe (and the adjacent) nuclides have relative high stability (and therefore, the relative higher abundancy). Again next, for size n = 4 nuclides (or all nucleons in the nucleon orbital shell space of n = 3), they should follow the {N,n//7} QM structure, and we are looking for  $n_{nuc-rel} = 28$  in Table 2. Figure 2 showed that a group of elements from Os (Z= 76,  $n_{nuc-rel} = 28$ ), to Pb (Z= 82,  $n_{nuc-rel} = 30$ ) have relative high abundancy, and they fitted to the {-14,4//7} sized QM structure quite well. Thus, this analysis revealed that the nucleus' {N,n} QM structure stability played an important role in determining the abundancy of each element in our universe.

More interestingly, Xe is also an inert element that ends the electron period 5 in the chemical element periodic table. This may implies that when adding electron to the 4d orbit of the electron period 5 elements, the electron orbital  $\{N,n//q\}$  QM structure starts to transform from  $\{N,n//6\}$  QM to  $\{N,n//7\}$  QM, so that at the end of electron period 5 (and for the rest element with Z > 54), not only their nuclides become the  $\{N,n//7\}$  QM, but also the electron orbit structure become the  $\{N,n//7\}$  QM.



Figure 2. Abundances of the chemical elements in the Solar System. Copied from wiki "chemical element". Author: Ken Croswell. Copyright: CCBY-SA 3.0.

Table 3. {N,n//6} QM structural periodic table for atom's nuclide structures (N = -15..-14), and for atom's electron structures (N = -12).

		n= "n state"	or "n shell" or "	n orbit space'	•								
{N,n//6	5}	1	2	3	4	5	6 ={N+1,1//6}	7	8	9	10	11	12 = {N+1,2//6}
N=	-16					H nucleus n <sub>nuc</sub> =1, orbit {-16,5}o size {-15,1}							
nuclide	-15	He nucleus n <sub>nuc</sub> =2, orbit {-15,1}o size {-15,2}	Li nucleus n <sub>nuc</sub> =3, orbit {-15,2}o size {-15,3}	Be, B nuclides n <sub>nuc</sub> <=4, orbit {-15,3}o size {-15,4}	C, N, nuclides n <sub>nuc</sub> <=5, orbit {-15,4}o size {-15,5}	O, F,Ne nuclides n <sub>nuc</sub> <=6, orbit {-15,5}o size {-15,6//6} ={-14,1}	Na, Mg, nuclides n <sub>nuc</sub> <=7, orbit {-15,6}o size {-15,7//6}	Al, Si, nuclides n <sub>nuc</sub> <=8, orbit {-15,7}o size {-15,8//6}	P, S, Cl, nuclides n <sub>nuc</sub> <=9, orbit {-15,8}o size {-15,9//6}	Ar, K, Ca, nuclides n <sub>nuc</sub> <=10, orbit {-15,9}o size {-15,10//6}	Sc, Ti, V, nuclides n <sub>nuc</sub> <=11, orbit {-15,10}o size {-15,11//6}	Cr,Mn,Fe, nuclides n <sub>nuc</sub> <=12, orbit {-15,11}o size {-15,12//6} ={-14,2}	
nuclide	-14	orbit {-14,1}o, size {-14,2}, n <sub>nuc</sub> =712, Z=1126 nuclides Na,Mg,AI,Si,P,S,CI ,Ar,K,Ca,Sc,Ti,V,Cr ,Mn,Fe	orbit {-14,2}o, size {-14,3}, n <sub>nuc</sub> =1318, Z=2747 nuclides Co,Ni,Cu,Zn,Ga, Ge,As,Se,Br,Kr, Rb,Sr,Y,Zr,Nb,Mo ,Tc,Ru,Rh,Pd,Ag	orbit {-14,3}o, size {-14,4}, n <sub>nuc</sub> =1924, Z=4870 nuclides	orbit {-14,4}o, size {-14,5}, n <sub>nuc</sub> =2530, Z=7196 nuclides	orbit {-14,5}o, size {-14,6} ={-13,1}, n <sub>nuc</sub> =3136, Z=97118 nuclides	Og118 nucleus size {-13,1}						
	-13						{-12,1} H-atom size						
atom	-12	H, He, electron shell orbit	period 2 atom's electron outer shell (unshrunk) orbit, max actual size of atom {-12,3}	period 3 atom's electron outer shell unshrunk orbit	period 4 atom's electron outer shell unshrunk orbit	period 5 atom's electron outer shell unshrunk orbit	period 6 atom's electron outer shell unshrunk orbit	period 7 atom's electron outer shell unshrunk orbit					
	-11					{-10,1} max atom theoretical size							

Note: For the complete {N,n//6} QM master periodic table (N = -24 ... 15, and n = 1 ... 12), see SunQM-7's Table 1. Note:  $n_{nuc}$  is defined by the size of nucleus. According to the rule of "all matter between  $r_n$  and  $r_{n+1}$  belongs to orbit n", its matter's nucleon orbital shell space is at  $n_{nuc}$  -1. For example, Helium's nucleus has  $n_{nuc} = 2$  for its size, and its nucleon orbital shell space should be at n =  $n_{nuc}$  -1 = 1, or {-15,1//6}o.

More discussions on Table 3 and Figure 2:

1) In Table 3, the (Mendeleev) electron periodic table of elements (the electron configuration) is compressed into N = -12 (as a single period) with n = 1 to 7. The information of the electron configuration and the chemical property of these elements are buried after this compression.

2) It is interesting to see that a hydrogen nucleus (with  $n_{nuc} = 1$ , see Table 2) has size of  $\{-15,1//6\}$ , it is "accompanied" by a helium nucleus (with  $n_{nuc} = 2$ ) at  $\{-15,1//6\}$  o nucleon orbital shell space with a size of  $\{-15,2//6\}$ . Also, an oxygen nucleus (with  $n_{nuc} = 5.5 \approx 1 * 6^{1}$ ) has size of  $\{-14,1//6\}$ , it is "accompanied" by a Fe nucleus (with  $n_{nuc} = 12.2 \approx 2 * 6^{1}$ ) at  $\{-14,1//6\}$  o nucleon orbital shell space with a size of  $\{-14,2//6\}$ . Notice that in Figure 2, among  $n = 1 \dots 6$  of  $n_{nuc} = n * 6^{0}$  nuclides, or from H to Ne nuclides, hydrogen nucleus ( $n_{nuc} = 1$ ) has the highest abundancy (= stability), and the "accompanied" helium nucleus ( $n_{nuc} = 2$ ) has the second highest abundancy (= stability). Also among  $n = 1 \dots 6$  of  $n_{nuc} = n * 6^{1}$  nuclides, or from O to Og nuclides, Oxygen nucleus ( $n_{nuc} \approx 1 * 6^{1}$ ) has the highest abundancy/stability, and the "accompanied" Fe nucleus ( $n_{nuc} \approx 2 * 6^{1}$ ) has the second highest abundancy/stability. A more detailed discussion will be given in SunQM-7's section I-e, and SunQM-5s2's section III.

3) If we apply (a kind of) Fourier analysis to Figure 2, then it is expected to see that the  $\Delta n_{nuc} = 1$  is the primary base "frequency", the  $\Delta n_{nuc} = 2$  is the secondary base "frequency". Or, if we apply (a kind of) Principal Component Analysis (PCA, or similarly the SVD), then the first principal component vector must point to  $\Delta n_{nuc} = 1$ , and the second principal component vector must point to  $\Delta n_{nuc} = 2$ . (Note: as a citizen scientist, I don't know how to do those calculations right now, but I can guess how the result looks like).

### II-c. As the force (or mass) increased, the quantum number q also increased in {N,n//q} QM structure

The above analysis revealed that although the  $\{N,n//q\}$  QM structures of both atom's nucleus shells and atom's electron shells are primarily in  $\{N,n//6\}$  QM at the low Z#, at the high Z#, both of them shifted to the  $\{N,n//7\}$  QM. This result suggested that as the force (or electron, or mass) increases, the original  $\{N,n//q\}$  QM structure may increase the

quantum number q from low value to high value. Actually, this is not the first time we see this kind phenomenon. Before (in SunQM-3s7) we have showed that the {N,n//q} QM (which can also be expressed as |qnlm>) can have superposition of different QM states, in the current case, between quantum number q = 6 and q = 7. We may can explain the above result as: in the low Z range, the superposition (equilibrium) shifted to {N,n//6} side; and in the high Z range, the superposition (equilibrium) shifted to {N,n//6} side; and in the high Z range, the superposition (equilibrium) shifted to {N,n//7} side. Og (Z=118, n<sub>nuc</sub> = 34.2  $\approx$  36 under q=6) fits to {-14,5//6}o, which ends the N= -14 super shell, and the same Og (Z=118, n<sub>nuc</sub> = 47.9  $\approx$  49 under q=7) fits to {-14,6//7}o, which also ends the N= -14 super shell. Thus, the superposition of q = 6, and q = 7 gives the self-consistent explanation. Especially a {N,n//7} QM structure fits to both the atom's nucleus {N,n//q} QM (see Table 2 last column) and the atom's electron shell {N,n//q} QM (see Table 3, N = -12 period), this really makes us to believe that {N,n//7} QM is also correct for (at least the high Z numbered) atoms (probably due to the increased relativity effect).

In SunQM-1s3 and SunQM-3s7, we have shown that the original Saturn (when it was just formed ~4.5 billion years ago) had only  $\sim 20\%$  of current mass, and had a {N,n/2} QM structure. After it captured large amount of mass (of the evaporated H/He/H<sub>2</sub>O/CH<sub>4</sub>, etc. from inside the {1,6//6} pre-Sun ball), its mass increased to 100% of the current mass, and its  $p\{N,n/2\}$  QM structure changed to the superposition of  $p\{N,n/3\}$  and  $p\{N,n/2\}$ . Similarly, the original Jupiter (when it was just formed ~4.5 billion years ago) had only ~10% of current mass, and had a  $p\{N,n/2\}$  QM structure. After it captured large amount of mass (of the evaporated  $H/He/H_2O/CH_4$ , etc. from inside the {1,6//6} pre-Sun ball), its mass increased to 100% of the current mass, and its  $p\{N,n/2\}$  QM structure changed to the superposition of  $p\{N,n/5\}$  and  $p\{N,n/3\}$ . On the other hand, the original Earth (when it was just formed ~4.5 billion years ago) had ~25x of current mass, and had a  $p\{N,n/2\}$ QM structure. After it lost 96% of mass (24x of Earth mass, the whole original atmosphere shell with ~2000 km thickness of  $H/He/H_2O/CH_4$ , etc.), its mass decreased to 1x of the current mass, and its p{N,n/2} QM structure changed to the  $p\{N,n/2/2\}$ , meaning its original q=2 decreased to  $\frac{1}{2}$  of q=2, and the leftover  $p\{N,n/2\}$  QM structure re-structured to be a new  $p\{N,n/4\}$  QM structure (see SunQM-3s7's section I-e and section V). Therefore, we do have the evidences that not only the increasing of force/mass will increase the q value of  $\{N,n/q\}$ , but also the decreasing of force/mass will decrease the q value of  $\{N,n//q\}$ . Furthermore, this analysis showed us several true examples that the quantum number q can be in values of 7 (the {N,n//7} QM structure of nuclides at high Z#), 6 (the standard {N,n//6} QM structure from N = -17 to N = 10), 5 (Jupiter's internal and surface  $p\{N,n/5\}$  OM structure), 4 (equivalent to  $\{N,n/2\}$  OM naturally), 3 (Saturn's internal and outer ring  $p\{N,n/3\}$  QM structure, and Jupiter's outer ring  $p\{N,n/3\}$  QM structure), 2 (the initial  $p\{N,n/2\}$  QM structure for all original planets, the current Neptune's  $p\{N,n/2\}$  QM structure), 1 (Earth's initial  $p\{N,n/2\}$  QM structure is now become  $p\{N,n/2/2\} \rightarrow p\{N,n/4\}$  QM structure with q = 2 \* 1/2 = 1, after stripping off 24/25 = 96% of mass). So far, we have not seen any obstacle for q quantum number to take the even higher values of 8, 9, etc., or the even lower values of  $2*\frac{1}{2}*\frac{1}{2}=1/2$ , or even  $2*\frac{1}{2}*\frac{1}{2}=\frac{1}{4}$ , etc.

# II-d. The definition of an electron's ground state vs. excited state in an atom is different for {N,n} QM, and for Bohr-QM.

See SunQM-7's section I-f.

### II-e. Using {N,n} QM to analyze the 3D distribution of protons inside a atom's nucleus

 $\{N,n\}$  QM can also be used to analyze the 3D distribution of protons and neutrons inside an atom's nucleus. Here let's use Fe element's nucleus as the example. A Fe element's nucleus contains total 56 nucleons, including 26 protons and 30 neutrons. All of them are doing orbital movement (or at least their matter waves are doing orbital movement, while the nucleon particles may be apparently doing random collision) inside the size of  $\{-14,2//6\}$ . According to Table 3, the size of  $\{-14,2//6\}$  contains  $\{-15,n=1..11//6\}$  orbital shells, and the higher the n, the higher the QM state energy it will have. Because of the Coulomb repulsive potential energy, all (positively charged) protons should have higher QM state energy (than that of

neutrons). So protons mostly take the high n orbital shells (e.g.,  $\{-15,11//6\}$  orbital shells, close to the outer edge of the nucleus, see Figure 1), while neutrons are pushed inward to take the low n orbits. Without the outside force, the distribution of protons is isotropic. Thus, the 3D distribution of protons (or neutrons) can be described by Born probability (because it is in equilibrium state) of r^2 \*  $|R(n,l)|^2$  \*  $|Y(l,m)|^2$ , (where n = 1 ... 11, l = 0 ... (n-1), and m = -l, ... +l), and can be further simplified to r^2 \*  $|R(n)|^2$  (because of the isotropic,  $|Y(l,m)|^2 = 1$ ). When added the outside magnetic force, the nucleus has the orientation with the z-axis overlap to the outside magnetic force-line. Those nucleons (proton? neutron?) that can interact with the outside magnetic force to strongly decrease their QM state energy will take (the relatively low QM state energy of) nLL orbits, while the rest nucleons (neutron? proton?) will be pushed to the two poles (with the relative high QM energy orbit of n*l*0). Thus, the distribution of protons (which is always mostly at high n orbit, or near the out-edge of nucleus) become anisotropic. (Note: because I am only a citizen scientist of QM, not a nuclear physicist, I don't know whether it is proton or neutron that will be pushed to the two poles of the nucleus under the outside magnetic field. If any reader knows, please teach me).

Alternatively, the size of  $\{-14,2//6\}$  can be roughly divided into two (spherical or shell) spaces: the  $\{-14,1//6\}$  sized core (which has relatively low QM state energy, or it can be defined as the (relative) ground state), and the  $\{-14,1//6\}$  o orbital shell (which has relatively high QM state energy, or it can be defined as the 1<sup>st</sup> excited state). Protons will (mostly) take the  $\{-14,1//6\}$  o orbital shell (due to they have higher QM state energy). The  $\{-14,1//6\}$  o orbital shell has n=1, so the  $|1,0,0\rangle$ , or  $|R(1,0)|^2 * |Y(0,0)|^2$ , perfectly describes the isotropic distribution of protons inside Fe element nucleus (when without outside force). Under the outside magnetic field, the outer-edge nucleons (of Fe nucleus) has size of n  $\approx 2$ , and can be roughly treated as partially in  $|2,1,m\rangle$  QM state. Then those nucleons (proton? neutron?) that can interact with the outside magnetic force to strongly decrease their QM state energy will take the  $|2,1,1\rangle$  orbit (or be pushed to the equator, because  $|2,1,1\rangle$  orbit has the lower QM state energy), while the rest nucleons (neutron? proton?) will be forced to take  $|2,1,0\rangle$  orbit (or be pushed to the two poles, because  $|2,1,0\rangle$  orbit has the higher QM state energy).

The similar analysis can be used for any element's nucleon 3D distribution inside the nucleus (see the illustration in Figure 3). We also believed that the proton's 3D re-distribution information inside a nuclear  $\{N,n//6\}$  QM structure will soon be added in the explanation a number of nuclear physical processes, including nuclear fission, alpha-decay, beta-decay, even the difference between ferromagnetism vs. ferrimagnetism, etc.



Figure 3a (left). To illustrate that if protons interact with the outside force will decrease their QM state energy more than that of neutrons, then more protons in each nuclear super-shell will move to the approximate  $|2,1,1\rangle$  QM state (near equator). Figure 3b (right). To illustrate that if neutrons interact with the outside force will decrease their QM state energy more than that of protons, then more protons in each nuclear super-shell will be pushed to the approximate  $|2,1,0\rangle$  QM state (near two poles). Note: the r-dimension is not on scale.

# **II-f.** Using Schrodinger equation and probability r-distribution to describe an atom's nucleus-electron system (and *N*-body problem)

(Note: In this paragraph, the italic N means N-body's N, not  $\{N,n\}$  's N). The interior  $\{N,n\}$  QM of atom's nucleuselectron system reveals that the core of this  $\{N,n\}$  QM structure is N-body (rather than the one-body core in the Solar system's interior  $\{N,n\}$  QM structure). Likewise, the (one-body) Sun core  $\{0,1\}$  could be treated as that there were 4 of quarter-massed small-Sun doing RF movement (exactly like the nucleus of Helium with 4 nucleons), or could be treated as there were ~20 of small-Sun doing RF movement (exactly like the nucleus of Ne with 20 nucleons), or, could be treated as there were ~300 of Jupiter-sized small-Sun doing RF movement (like Og-118 nucleus with ~294 nucleons). Or, it even could be treated as there were (Sun's mass / proton's mass =) 1.19E+57 of nucleons doing RF movement. Since Sun's interior {N,n} QM is described by using Schrodinger equation's probability density  $r^2 * |R(n,l)|^2 * |Y(l,m)|^2$  to fill in mass, (He, Ne, and all other atoms') nuclides can also be treated by using Schrödinger equation's probability density  $r^2 * |R(n,l)|^2$ \* $|Y(l,m)|^2$  to fill in one (fundamental) mass to orbital shell spaces of e1{-1,n=1..5//6}o, e1{-2,n=1..5//6}o, and e1{-3,n=1..5//6, and this (fundamental) mass is a e1{-3,1//6} sized RF QM structure which we call it proton (and/or neutron). All this interior  $e_1\{N,n\}$  is relative to the electron ground state  $e_1\{0,1//6\}$  as  $r_1$ . For  $e_1\{-2,1//6\}$  sized orbital spherical space it has the maximum occupancy number of  $\sim 20$  of e1{-3,1//6} RF ball structures, and for e1{-1,1//6} sized orbital shell space it has the maximum occupancy number of  $\sim 300$  of e1{-3,1//6} RF ball structures. Note: it is easy to understand that using Schrodinger equation's probability density  $r^2 * |R(n,l)|^2 * |Y(l,m)|^2$  to fill in mass (N point mass, with large N number, billions, like gas molecule) to orbit space. It is not easy to think N-body problem (with  $N=3 \sim 10$ ) can also be treated in the same way as N in billions. Just think for a 3-body, each point center has some probability to stay in orbital shell space. (See SunQM-7's section II-c for more discussion on this topic).

Therefore, the nucleus can be described as the interior {N,n} QM structure of the ground state electron's orbital shell  $e_{1}\{0,1//6\}$ , i.e.,  $e_{1}\{-1,n=1..5//6\}$ ,  $e_{1}\{-2,n=1..5//6\}$ , and  $e_{1}\{-3,n=1..5//6\}$ , with the 1st, 2nd, and 3rd level of RF^ (note: see SunQM-2 's section IV-c for the definition of RF^), with matter wave  $\lambda = 1 \times, 6 \times, \text{ or } 36 \times \text{ of } 2\pi r_{1N}$ , where  $r_{1N}$  is r of  $e_{1}\{N,1//6\}$ . We can use Schrodinger equation's probability density r^2 \* |R(n,l)|^2 \* |Y(l,m)|^2 to fill in a number of  $e_{1}\{-3,1//6\}$  RF QM structures (we called them as protons/neutrons, see Figure 4). and the maximum occupancy for  $e_{1}\{-3,1//6\}$  spherical space,  $e_{1}\{-2,1//6\}$  orbital spherical space, and  $e_{1}\{-1,1//6\}$  orbital shell space is 1, ~20, ~300 of proton/neutron respectively. According to this understanding, we plotted the Figure 4 to show that nucleus as the interior  $e_{1}\{N,n//6\}$  QM structure of its own shrunk  $e_{1}\{0,1//6\}$  electron ground state, and all of them are governed by Schrodinger equation's probability density r^2 \* |R(n,l)|^2 \* |Y(l,m)|^2 = 0



Figure 4. Plot of the radial probability density  $r^2 * |R(n,l)|^2$  vs.  $r/r_1$  for the interior {N,n//6} QM of an atom's nucleuselectron system from e1{-4,n//6} to e1{1,n//6}. Note: the y-axis is not on scale. Note: The thick red curve of  $r^2 * |R(n=1,l)|^2$  has the size of n=2 and it fits the "atom size n=2 effect" of the shrunk electron shells (of the electron period 1 through 7). Notice that in this representation, only the Z=1 element is well represented; all Z > 1 elements should have  $r_{e1} = r_1/Z$  and those  $r_{e1}$  are ignored in the red curve representation in Figure 1. For example, the Z=36 element should have  $r_{e1} = r_1/Z$  and those  $r_{e1}$  are ignored in the red curve representation in Figure 1.

 $r_1/Z = r_1/36$  and it is ignored in the Figure 1's red curve representation. Part of reasons for the ignorance is that Figure 1 needs to represent Z = 1..118 at one time.

														r/r1 log compress at 4
a <sub>0</sub> =	0.0529	nm												4
												0.1X		
r/a0 =	r <sup>2</sup>  R5,0  <sup>2</sup>	r <sup>2</sup>  R5,1  <sup>2</sup>	r <sup>2</sup>  R5,2  <sup>2</sup>	r² R5,3 ²	r² R5,4 ²	n=1,I=0	n=2,ΣI=0,1	n=3,Σl=02	n=4,Σl=03	n=5,ΣI=04	Σn=15,Σl	Σn=15	Σn=1.2	log(r/r1)
0.2	0.015998	8.44E-05	3.03E-08	1.82E-12	2.07E-17	2.027433	0.25175842	0.07450383	0.03141784	0.01608275	2.4011959	0.24012	2.279191	0.2
0.4	0.040984	0.001099	1.7E-06	4.22E-10	1.95E-14	5.436116	0.66229479	0.19530913	0.08225982	0.0420848	6.4180646	0.641806	6.098411	0.4
0.6	0.056701	0.004516	1.69E-05	9.78E-09	1.04E-12	8.198859	0.97105424	0.28490991	0.11978578	0.06123341	9.6358428	0.963584	9.169914	0.6
0.8	0.058719	0.011529	8.28E-05	8.83E-08	1.7E-11	9.770418	1.12346398	0.32803181	0.13768822	0.0703318	11.429933	1.142993	10.89388	0.8
1	0.049544	0.022638	0.000275	4.76E-07	1.46E-10	10.23329	1.15904046	0.33806031	0.14186273	0.07245751	11.944713	1.194471	11.39233	1
2	0.001911	0.112167	0.008734	7.33E-05	1.01E-07	5.539702	1.70554862	0.54844411	0.23732733	0.12288537	8.1539072	0.815391	7.24525	2
3	0.087288	0.143192	0.047879	0.001123	3.89E-06	1.686863	4.23519485	1.28863674	0.54548692	0.27948548	8.0356675	0.803567	5.922058	3
4	0.140406	0.074903	0.124776	0.006661	4.63E-05	0.405853	6.46298537	1.70577778	0.69034111	0.34679285	9.6117498	0.961175	6.868838	4
5	0.081866	0.006064	0.210513	0.02339	0.000289	0.085822	6.8992841	1.60685839	0.64161136	0.32212309	9.5556991	0.95557	6.985106	4.30103
6	0.009006	0.019887	0.260754	0.058728	0.001199	0.016725	5.9040222	1.51828863	0.66553062	0.34957334	8.4541401	0.845414	5.920747	4.477121
7	0.015773	0.107725	0.248857	0.116501	0.003753	0.003081	4.36404759	1.87526001	0.91112496	0.49261016	7.6461236	0.764612	4.367128	4.60206
8	0.100597	0.202397	0.181739	0.193653	0.009563	0.000545	2.90861106	2.69618312	1.29699366	0.68794882	7.5902813	0.759028	2.909156	4.69897
9	0.197087	0.242533	0.092774	0.279865	0.020816	9.33E-05	1.79515774	3.71109967	1.63977488	0.8330755	7.9792011	0.79792	1.795251	4.778151
10	0.241285	0.210098	0.022511	0.360169	0.040019	1.56E-05	1.04416978	4.60002243	1.80695536	0.87408091	8.3252441	0.832524	1.044185	4.90309
12	0.133588	0.047385	0.035228	0.445349	0.111337	4.11E-07	0.30941742	5.28209782	1.65071641	0.77288663	8.0151187	0.801512	0.309418	5
14	0.003249	0.01506	0.217312	0.386333	0.233707	1.02E-08	0.08061689	4.55040985	1.5250704	0.85566032	7.0117575	0.701176	0.080617	5.079181
16	0.077737	0.178673	0.377637	0.224532	0.399169	2.45E-10	0.01915154	3.21879457	2.01605722	1.25774797	6.5117513	0.651175	0.019152	5.146128
18	0.27589	0.356738	0.377952	0.064715	0.582433	5.68E-12	0.00424424	1.97692646	2.96268514	1.65772775	6.6015836	0.660158	0.004244	5.20412
20	0.392865	0.384808	0.237482	0	0.75056	1.28E-13	0.00089096	1.09228161	3.83918001	1.76571553	6.6980681	0.669807	0.000891	5.255273
22	0.337164	0.258853	0.074238	0.065063	0.874735	2.85E-15	0.00017908	0.55590065	4.25450181	1.61005302	6.4206346	0.642063	0.000179	5.30103
24	0.174224	0.091491	0.000229	0.234567	0.938267	6.21E-17	3.4734E-05	0.2649428	4.12576476	1.43877838	5.8295207	0.582952	3.47E-05	5.361728
27	0.002603	0.007221	0.134416	0.555149	0.917695	1.95E-19	2.8103E-06	0.07905793	3.24298461	1.61708413	4.9391295	0.493913	2.81E-06	5.414973
30	0.112528	0.20552	0.445905	0.79272	0.79272	5.96E-22	2.1572E-07	0.02150134	2.13598278	2.34939336	4.5068777	0.450688	2.16E-07	5.462398
33	0.417615	0.527699	0.725881	0.864958	0.619289	1.79E-24	1.5873E-08	0.00543514	1.23167877	3.15544177	4.3925557	0.439256	1.59E-08	5.50515
36	0.706667	0.773864	0.850368	0.791595	0.445272	5.27E-27	1.128E-09	0.0012949	0.64042136	3.56776632	4.2094826	0.420948	1.13E-09	5.556303
40	0.852436	0.842775	0.77556	0.580112	0.257828	2.18E-30	3.176E-11	0.00017749	0.23604795	3.3087107	3.5449361	0.354494	3.18E-11	5.612784
45	0.686303	0.62964	0.504996	0.314748	0.113309	1.25E-34	3.4571E-13	1.3448E-05	0.0579471	2.24899649	2.306957	0.230696	3.46E-13	5.662758
50	0.406218	0.355021	0.259695	0.142494	0.04398	7.03E-39	3.5746E-15	9.3656E-07	0.01244192	1.2074081	1.219851	0.121985	3.57E-15	5.70757
55	0.195242	0.164811	0.112749	0.056266	0.015438	3.86E-43	3.546E-17	6.098E-08	0.00240651	0.5445057	0.5469123	0.054691	3.55E-17	5.748188
60	0.080542	0.06622	0.043037	0.01995	0.004988	2.09E-47	3.3996E-19	3.7582E-09	0.00042806	0.21473734	0.2151654	0.021517	3.4E-19	5.78533
65	0.029529	0.023778	0.014837	0.006483	0.001503	1.11E-51	3.1674E-21	2.2126E-10	7.1083E-05	0.0761291	0.0762002	0.00762	3.17E-21	5.819544
70	0.009852	0.0078	0.004708	0.00196	0.000427	5.86E-56	2.8802E-23	1.2532E-11	1.1145E-05	0.02474613	0.0247573	0.002476	2.88E-23	5.880814

Table 4.  $r^2 |R(n,l)|^2$  vs.  $r/r_1$  for n=1 to 5, (for Figure 4 through Figure 6).

Note (for Table 4 and Figure 4):

1) The purpose of this plot is to demonstrate that the nucleus is an interior  $\{N,n\}$  QM structure of  $e1\{0,1//6\}$  electron shell.

The probability density scale between different N super-shells is not on scale.

2) For super-shells from  $e_{-4,n/6}$  to  $e_{-1,n/6}$ , orbit n = 1..5, size n = 2..6.

3) Proton size is at  $e_{1}^{-3,1/6}$ , but its orbit covers  $e_{1}^{-4,n=1..5/6}$  o. n=1 electron at orbit  $e_{1}^{0,1/6}$  o, it covers space from  $e_{1}^{0,1/6}$  to  $e_{1}^{0,2/6}$ , not  $\{-1,n=2..5/6\}$  o.

4) For e1{-1,n//6} super-shell in the Figure 4, the probability density is scaled down to  $0.1\times$ , to illustrate that: a) there is no (Z > 118) nucleus has been found in this super-shell yet; b) for the Z  $\leq$  118 nuclides, nucleons never go to this super-shell space.

5) Due to that I am not able to obtain R(n,l) for n=6, or n=7, so the probability density distribution of e1{0,n=1..7//6}o is actually  $\Sigma n=1..5$ .

6) The e1{1,n//6} super-shell (uncompressed space) of electron is empty for (electron orbital) period 6 to 35, so its amplitude is scaled down to  $0.1 \times$ .

### III. Treat atom as the interior $\{N,n//6\}$ QM structure of a theoretical $e_{0,1//6}$ electron shell

There are two questions that we are going to discuss here.

First question, how to explain n=7 (or electron period 7) electron shell in a {N,n//6} QM system? As shown in Figure 4, when we take ground state electron as  $e1\{0,1//6\}$ , then  $e1\{0,6//6\} = e1\{1,1//6\}$  size (with  $e1\{0,5//6\}$  o orbital space completely filled in) is expected to be a super stable state in {N,n//6} QM. If so, then the element (i.e., electron

configuration) period table should end at Xe (Z=54) at the end of electron period 5. According to {N,n//6} QM structure, if more electrons need to be added, then the next electron shell should be  $e1\{1,1//6\}o=e1\{0,n=6..11//6\}o$  orbital space, or n = 6..11, and then  $e1\{1,2//6\}o = e1\{0,n=12..17//6\}$ , or n = 12..17 (similar as Jupiter's n=12 which covered n=12..17 orbital space relative to four rocky planets' n=3, 4, 5, 6 orbital space in Solar system). Although we do not know how to calculate the shrunk electron's  $r_n$ , but wiki "atomic radius" <sup>[14]</sup> gives the empirically measured atomic radius for electron period 7, and it is in the range between 216E-12 to 175E-12 meters. Comparing it to the un-shrunk n = 6 's r = 36\*5.29E-11 = 1904E-12 meters, it seems that besides n=7, more extra n(s) can be added in to this gap.

One possible (and the less favorite) answer is, like Asteroid belt orbit at  $\{1,8//6\}$ o, between shell  $\{1,6//6\} = \{2,1//6\}$ and shell  $\{1,12//6\} = \{2,2//6\}$ , there is enough space to put (orbital) n=6, 7, 8, ..., 11, for  $\{N,n\}$  super shell, if needed. If so, then according to the bottom part of Table 2, the maximum Z for nucleus could be ~1870 before the  $r_{nuc}$  crashes to the shrunk  $r_{e1}$ , and electron shell could be  $\approx 22$  (according to Table 2b, Z=1870). Another possible (and our favorable) answer is, the shrinking of the electron's ground state  $r_{e1}$  causes the  $r_n$  shrinks coordinately, so that the new electron periods (n=7, 8, ...) can be directly add in. That means, the original  $\{N,n//6\}$  structure is now changed to  $\{N,n//x\}$ , where x=7, 8, etc.

The second question is, how to apply the interior  $\{N,n//6\}$  QM to electron's n shells? In Figure 4, the electron n shell is an exterior  $\{N,n\}$ , with ground state at  $e_1\{0,1\}$ . Now we try to use the maximum n of electron shell as  $\{0,1\}$  (named as  $e6\{0,1//6\}$  for the interior {N,n//6} calculation. But the problem is (same as before), we do not know how to calculate the shrunk electron's  $r_n$  (due to that we do not know how to calculate the electron shielding effect). Result/Discussion 4 on Table 2 (in section II-a) provides some clue. After a second thought, we suddenly realized that this is exactly an externalized  $r_1$  (or  $r_1 > r_{surface}$ ) effect, because that the too strong EM-force shrinks the  $r_n$  space. For element Ag (in electron period 5), although its real  $r_{n=5}$  shrunk to 165E-12 meters (see Table 2 column 16), its original  $r_{n=6}$  (at 36\*5.29E-11=1.9E-9 meters) still functions as  $e6\{0,1/6\}$  for the interior {N,n/6} purpose (for electron shells of n=1...5). For element Pb (in electron period 6), although its real  $r_{n=6}$  shrunk to 180E-12 meters, its original  $r_{n=6}$  (at 36^2\*5.29E-11=6.86E-8 meters) is still function as  $e6\{0,6//6\}$  $e6\{1,1/6\}$  for the interior {N,n} purpose. This externalization effect is exactly the opposite of the internalization effect that was discussed in the section XI of the paper SunQM-1s3, where the meter-sized body has too small G-force so that its gravity  $r_1$  is << of its  $r_{surface}$ , or  $r_1$  is internalized. So for the interior {N,n} QM, the e6{0,1//6} for electron period 1 to 5 elements is at 1.9E-9 meters (which is  $\{-11, 1/6\}$  size under Sun $\{0, 1\}$ , see Table 1, notice that if use Sun $\{0, 1\}$ , the 1.9E-9 meters becomes 1.32E-9 meters). The e6{0,1//6} for electron period 6, 7 (and up) elements is at  $r \approx 6.86E-8$  meter (which is {-10,1} size under Sun{0,1} at 4.76E-8 meters). Since  $\{-10,1\}$  is the unidentified super-super stable QM structure in  $\{N//6, n//6\}$  QM theory, so now we can assign  $\{-10,1\}$  as the maximum atom's externalized  $r_1$  (which is hidden by the strong EM force induced  $r_n$  QM space shrink). This maximum atom's externalized  $r_1$  at {-10,1} size is just like the "recessive gene", it is there, but it does not show up (or it is hidden by the strong EM force induced  $r_n$  OM space shrink).

Figure 5 showed the interior {N,n} QM structure for electron period 1 through 7 shells using  $r_{e6} = 1.9E-9$  meters as  $e6{0,1}$ . The result is similar as that in Figure 4, except the  $r_1$  is shifted up by  $\Delta N = +1$  (from  $r_{e1}$  to  $r_{e6}$ ).



Figure 5. The interior {N,n//6} QM structure for electron period 1 to 7 shells using  $r_{e6} = 1.9E-9$  meters as  $e6{0,1//6}$ . Note: the y-axis is not on scale.

#### IV. Interior {N,n} QM analysis for a proton (or neutron) in a proton-quark system

Particle physics told us that each proton (or neutron) is made of three (different) quarks. Table 1 showed that quark is an interior  $\Delta N = -2$  of {N,1//6} QM structure of a nucleon. So if we treat proton as {0,1//6} (named **prot{0,1//6}**), then quark will be its interior prot{-2,1//6} QM structure. We can use Schrodinger equation's probability density r^2 \*|R(n,1)|^2 \*|Y(1,m)|^2 to fill in a number of prot{-2,1//6} sized ball structures (we call it quark) into the size space of prot{-2,1//6}, or prot{-1,1//6}, or prot{0,1//6}, and the maximum (and the minimum) occupancy for prot{0,1//6} size space is 3.

Figure 6 plotted the radial probability density  $r^2 * |R(n,l)|^2 vs. r/r_1$  for interior {N,n} QM for a proton made of quarks. The purpose of this plot is to demonstrate that the quark prot{-2,1//6} is an interior {N,n} QM structure of prot{0,1//6} proton shell. Note: 1) The probability density between different N super-shells is not on scale. Does the inner one has higher probability than the outer one? (actually we don't know the answer); 2) This is a typical 3-body problem: each one quark has size of prot{-2,1//6}, three of them doing RF movement in the orbital space of {-2,n=1..5} o and {-1,n=1..5} o to form a proton with size of prot{0,1//6}. So a 3-body doing RF movement problem can be described under {N,n} QM structure with Schrodinger equation determined probability radial distribution. (For more discussions, also see SunQM-7's Table 1, or SunQM-5s2).



Figure 6. Plot of  $r^2 * |R(n,l)|^2$  vs.  $r/r_1$  for interior {N,n} QM for a proton-quark system from prot{-2,n//6} to prot{0,n//6}. Note: the y-axis is not on scale.

# V. Exploring the interior {N,n//6} QM structure of a quark, made of $\Delta N$ = -3 and $\Delta N$ = -8 super stable {N,n//6} QM structure (similar to different levels of strings)

The analysis of Figure 4 through Figure 6 demonstrated that we can use this kind of figure as a general tool to explore the internal structure of a fundamental particle (like an atom, a nucleus, etc.) if we know its structure is based on  $\{N,n//6\}$  QM. From Table 1, we had hypothesized that at the size range much below quark, it also follows the  $\{N//6,n//6\}$  QM. So in this section, we will use this tool to further explore the possible sub-structure of quark.

Table 1 showed that there are (at least) two levels of super stable {N,n//6} QM structures inside quark, we call them **string**{-20,1//6} and **string**{-25,1//6}, one at  $\Delta N = -3$ , another one at  $\Delta N = -8$  if quark is {0,1//6} (named **qk**{0,1//6}). So these super stable QM structures can be treated as the interior {N,n} QM structure of quark. Figure 7 plotted the radial probability density r^2 \*|R(n,l)|^2 vs. r/r\_1 for interior {N,n//6} QM of a quark made of string{-20,1//6} using qk{0,1//6}. So a (unknown) number (that equivalent to the mass occupancy, or equivalent to 3 quarks per nucleus) of super stable string{-20,1//6} moving in 3 levels of super-shell spaces from qk{-3,n=1..5//6} o to qk{-2,n=1..5//6} o to qk{-1,n=1..5//6} o, under qk{0,1//6}.

Figure 8 plotted the radial probability density  $r^2 * |R(n,l)|^2$  vs.  $r/r_1$  for the interior {N,n//6} QM using string{-20,1//6} as {0,1//6} (named as **str{0,1//6}**). So a (unknown) number (that also equivalent to the mass occupancy, or

equivalent to 3 quarks per nucleus) of super stable string $\{-25,1//6\}$  moving in 5 levels of super-shell spaces from str $\{-5,n=1..5//6\}$  o through str $\{-1,n=1..5//6\}$  o, under str $\{0,1//6\}$ .



Figure 7. Plot of  $r^2 * |R(n,l)|^2$  vs.  $r/r_1$  for interior {N,n//6} QM of a quark made of string{-20,1//6} under qk{0,1//6}. Note: the y-axis is not on scale.



Figure 8. Plot of  $r^2 * |R(n,l)|^2$  vs.  $r/r_1$  for the interior {N,n//6} QM using string{-20,1//6} as {0,1//6}. Note: the y-axis is not on scale.

## VI. White dwarf, neutron star, and black hole re-analyzed by using the interior {N,n} QM structure of nucleuselectron system

Note: due to the size limitation of this paper, this section is moved into the next paper SunQM-5s1.

#### VII. Interior {N,n} QM for molecules and solids, using carbon (diamond) or H<sub>2</sub>O (ice) as example

In previous sections, we used the interior  $\{N,n//6\}$  QM to describe structures of atom, nucleus, quark, string $\{-20,1\}$ , and string $\{-25,1\}$ . Now we try to use the same interior  $\{N,n//6\}$  QM to describe the structures of molecule, daily-life-world mass body, celestial body, and up to our universe.

First let us use the interior  $\{N,n//6\}$  QM to describe the structure of a molecule, using carbon (diamond) molecules as the example. we should admit that the molecule is the most difficult part to use interior  $\{N,n\}$  QM to describe, and we still have a lot of confusion for our selfies.

For a molecule (that is made of many atoms)'s interior {N,n} QM, we have nucleus-electron's EM-force caused QM {N,n//6}, and G-force caused QM {N,n//6}. For nucleus-electron's EM-force caused QM {N,n//6}, we believe that it is most reasonable to choose (the compressed)  $e6{0,1//6}$  's  $r_{e6} = 36 * r_{e1}$  as the molecule's  $r_1$ , and use molecule's (ball-like) body surface r as the interior {0,1} (and we named it as **bs{0,1//6}**, meaning this {0,1} is using (ball-like) **b**ody **s**urface r as {0,1}). For a single carbon molecule (or atom), Z = 6, its measured  $r_{surface} \approx 7E-11$  meters (from wiki "atomic radius", also see Table 2 column 16), or, its  $bs{0,1//6}$  at  $r_{surface} \approx 7E-11$  meters. From the calculation, its  $r_1 = r_{e6} = 36 * 5.29E-11 / (Z=6) = 3.17E-10$  meters. Therefore, its  $r_1$  is externalized relative to its real  $r_{surface}$  (or it has the externalized  $e6{0,1}$  effect, or  $r_1 > r_{surface}$ ).

For a (diamond carbon) molecule made of five carbon atoms (named C5, see Figure 9) using chemical bond (which is a residue force of the EM-force between the neighboring nucleus-electron), the C-C (sp3-sp3) bond length = 1.54E-10meters (from wiki "Bond length"). This C5 molecule is always rotating around, making an effective  $r_{surface} \approx 1.89E-10$  meters (see Table 5 column 2) for its bs $\{0,1\}$ . Its r<sub>1</sub> increases extremely slow (comparing to the single C-atom), so it is still r<sub>1</sub> = 3.17E-10 meters. Thus, this C5 molecule still has externalized  $r_1$  (> $r_{surface}$ ). For this C5 molecule, electrons in the (n=1) inner shell of carbon are completely localized to each C atom. Electrons in the (n=2) outer shell are shared with immediate neighboring carbon atoms, so they are partly delocalized. Although practically there is no electron orbiting the outer surface of (the ball-like) C5 molecule, but by constructing the contour line of C5 molecule's electron density 3D map, we can imagine that each C atom may contribute < 0.01% of one electron to (spherically) orbit directly at the C5-molecule's out surface with  $r \approx 1.89E-10$  meters. Here the point is, for a molecule, the  $r_1$  for interior {N,n} QM is no longer associated with the molecule's outer most electron shell (like atom did in Figure 5) which is related to r<sub>surface</sub>. It is only associated with each (or averaged) atom's  $e1\{1,1//6\} = e6\{0,1//6\}$  's  $r_1$ , and this  $r_1$  increases extreme slowly with the increasing number of bound atoms (associated with rsurface). Why? Because the r1 is only related to the nucleus-electron system's (primary) EM-force caused QM effect. Notice that here the crystal lattice QM effect inside the carbon molecule (or inside the diamond ball) is not within our spec (because it does not belong to the (primary) EM-force caused QM effect). We only study the  $\{N,n\}$  QM effect exerted by a carbon molecule to its (molecular) surface and the outside space. So when adding more C atoms to a carbon (diamond) molecule, we can practically ignore the extreme small increases of  $b_{s}\{0,1//6\}$  's r<sub>1</sub>.



Figure 9. A diamond molecule made of five carbon atoms (the top un-shield part). Copied from: Dr Stan Fowler, "The difference between diamond and graphite, giant covalent structures", https://www.youtube.com/watch?v=FeZIIR50XoY. 0:48 / 4:19.

Table 5. Calculation of a carbon ball's r<sub>surface</sub> from its C atom numbers. Note: Diamond carbon mass density D≈3500 kg/m<sup>3</sup> obtained from wiki "Diamond".

D=3500 kg/m^3, 1 carbon =1.				
# of C atoms	5	100	1.00E+06	7.50E+29
kg	9.95E-26	1.99E-24	1.99E-20	1.49E+04
Vol = Mass / Density, m^3	2.84E-29	5.69E-28	5.69E-24	4.26E+00
r <sub>surface</sub> =, m	1.89E-10	5.14E-10	1.11E-08	1.01E+00
r <sub>1</sub> =36*5.29E-11/(Z=6), m	3.17E-10	3.17E-10	3.17E-10	3.17E-10

For a carbon (diamond) molecule with ~ 100 of C atoms, its  $r_{surface}$  is estimated to be almost same as the bs{0,1//6} 's  $r_1$  (see Table 5 column 3). For a carbon (diamond) molecule with 1E+6 of C atoms, its  $r_{surface}$  ( $\approx 1.1E$ -8 meters, see Table 5

column 4) is > bs{0,1//6} 's  $r_1 \approx 3.17E-10$  meters). For a carbon (diamond) ball with  $r_{surface} = 1$  meter (about 7.5E+29 of C atoms), its  $r_{surface} >> bs{0,1//6}$  's  $r_1$ .

Remember this is (primary) EM-force related  $r_1$ . What it means is that for a carbon diamond ball (with 1E+6 of C atoms), there is practically no (primary) EM-force caused {N,n} QM effect on its surface (at the gigantic molecule level). If you want to study the (primary) EM-force caused interior {N,n} QM of the meter-sized carbon ball, it is basically meaningless because it is within the range that comparable to (the interior QM's)  $r_1$ , which is  $<< r_{surface}$ . So for a molecule composed of  $\geq 1E+6$  atoms, the nucleus-electron system's (primary) EM-force caused bs{N,n} interior QM can be completely ignored, because the true structure of these atoms is formed under the (residue) EM-force (e.g., chemical bond force, etc.). It can be explained as that in a residue EM-force formed molecule, the residue EM-force (the chemical bond force) expanded the size of this molecule so that it is no longer follow the primary EM-force formed {N,n//6} QM structure.

This explanation is also valid for the G-force caused {N,n//6} QM analysis. For a molecule composed of many atoms, the G-force caused bs{N,n//6} interior QM can be completely ignored (because the true structure of these atoms is formed under the (residue) EM-force, e.g., chemical bond force, etc.), and the chemical bond increase the  $r_{surface}$  dramatically. This causes G-forced QM's  $r_1 << r_{surface}$ . Because G-force produces QM effect only near  $r_1$ , so at  $r_{surface}$  (>>  $r_1$ ), there is practically no G-force produced QM effect. So for the same carbon (diamond) ball with  $r_{surface} = 1$  meter, its  $r_{surface}$  is much greater than not only the EM-force caused bs{0,1//6} 's  $r_1$ , but also the G-force caused bs{0,1//6} 's  $r_1$ . As the mass of the carbon ball further increases to equal or above the mass of Earth, then its  $r_{surface}$  will be dominated by G-force (not the chemical bond force anymore), then its  $r_{surface}$  will be comparable to the G-forced  $r_1$  (notice that G-forced  $r_1$  has nothing to do with the primary EM-forced  $r_1$ , and apparently the G-forced  $r_1$  increases as  $r_{surface}$  increasing). So then we can use the interior {N,n} QM to study the G-force caused interior {N,n} QM structure of Earth, Jupiter, Sun, Solar system, ... etc. (as shown in paper SunQM-1s3).

We can also use the water molecules (or ice ball) to explain the interior {N,n} at molecular level. For a single H<sub>2</sub>O molecule, its effective  $r_{surface}$  is smaller than the (primary) EM-force caused effective bs{0,1//6} 's r<sub>1</sub>. So the r<sub>1</sub> is externalized. For multiple H<sub>2</sub>O molecules that form a tiny ice ball, they are bound by hydrogen-bond force (which is a residue force of the EM-force between atoms). Around 100 water H<sub>2</sub>O molecules is expected to have its  $r_{surface}$  to be the same as (or larger than) the (primary) EM-force caused effective bs{0,1//6} 's r<sub>1</sub>. For an ice ball with r = 0.25 meter, its r<sub>surface</sub> >> not only the (primary) EM-force caused bs{0,1//6} 's r<sub>1</sub>, but also the G-force caused bs{0,1//6} 's r<sub>1</sub>. So if you want to study the (either EM-force or G-force caused) {N,n} QM of the meter-sized ice ball, it is basically meaningless because it is in range of r<sub>1</sub>, which is << r<sub>surface</sub>. As the r of the ice ball further increases to the size of the ice-planet Neptune, then its r<sub>surface</sub> is dominated by G-force (not by H-bond anymore) so that it is become comparable to G-forced QM's r<sub>1</sub>. So then we can use the interior {N,n} QM to study the G-force caused interior {N,n} QM structure of the icy-planet Neptune (as shown in paper SunQM-1s3 and SunQM-3s7).

For the daily-life-world objects, they are made of molecules (with chemical bond), or a collection of molecules like biological cells, or even human body (formed with chemical bond, salt bridge, hydrogen bond, van der Waals bond, etc., all belong to the residue EM-force). Their interior {N,n} QM structures should be analyzed in the same way as that for the water/ice ball. We spend so much effect to study the meter-sized carbon (or H<sub>2</sub>O) giant molecule's (or ball's) {N,n} QM, simply because I was a biophysicist. So I interested in human body's {N,n} QM more than the celestial body's {N,n} QM. Unfortunately, the result showed that the chemical bond force expanded the size of an (C-C and C-H composed) mass to meter-size, and it makes the  $r_{surface}$  (which is controlled by chemical bond force, not by the primary EM- or G-force) >>  $r_1$  of (either the (primary) EM-force caused or G-force caused) {N,n} QM, so that the (EM- or G-forced) {N,n} QM analysis is not suitable for (meter-sized) human body. Alternatively, **does this analysis implies that the {N,n} QM structure is not able to produce any life structure, because {N,n} QM is a too simple physical process, lack of the diversity (that is required for the life structure)?** 

In the size range of N= -10 to N=0 super shells, because both the (primary) EM-force caused {N,n//6} QM structure and the G-force caused {N,n//6} QM structure is disrupted by the strong (residue) EM-force (such as the chemical bond, etc.), the fine physical structures in this size range are no longer follow the{N,n//6} QM's structure. For example, Solar system's eight planets have the size in the range of {-1,n=1..4//6}, and all of them have the initial fine {N,n//q} QM structure of q=2, not q=6. The current fine structure of these planets are even more diversified: Saturn is currently in the superposition of {N,n//3} and {N,n//2}QM structure, and Jupiter is currently in the superposition of {N,n//5} and {N,n//3}QM structure

(see section II-c). This analysis revealed that  $\{N,n//6\}$  is the fundamental  $\{N,n//q\}$  QM in our universe, and the disruption of either the primary EM-force or G-force formed  $\{N,n//6\}$  QM will causes a  $\{N,n//q\}$  QM structure with q other than 6. For this reason, although the  $\{N=-10..0,n//6\}$  sized range still follows the  $\{N,n//6\}$  QM as a whole, we are not able (or it is meaningless) to plot a general figure based on  $\{N,n//6\}$  QM structure (like the Figure 4 through Figure 8) to explore the fine structures in this size range.

### VIII. Interior {N,n//6} QM analysis for Virgo super cluster, Milky way galaxy, and Solar system

(Continued from the section I of this paper). Figure 10 plots the radial probability density  $r^2 * |R(n,l)|^2 vs. r/r_1$  for the interior {N,n//6} QM structural analysis using Virgo Super Cluster {10,1//6} as {0,1} (named as **VSC{0,1//6}**). So a (unknown) number (that also equivalent to the mass occupancy, or equivalent to 3 quarks per nucleus) of galaxies at size of VSC{-2,1} that moving in 2 levels of super-shell spaces from VSC{-2,n=1..5//6}o through {-1,n=1..5//6}o, under VSC{0,1//6}. Also there are unknown number (that also equivalent to the mass occupancy, or equivalent to 3 quarks per nucleus) of stars at size of VSC{-10,2//6} that moving in 10 levels of super-shell spaces from VSC{-10,n=2..5//6}o through {-1,n=1..5//6}o, under VSC{0,1//6}.

Notice that in Figure 10 (and also in Figures 2 through 6), the y-axis is not on scale. In the future, with more research added, we may be able to generate the (closed to the true) probability values for each N super shell and n-shell. All this work suggested that the structure of our universe (from the whole universe down to the sub-quark) can be described by Schrodinger equation and solution (in terms of probability).



Figure 10. Plot of  $r^2 * |R(n,l)|^2$  vs.  $r/r_1$  for the interior {N,n} QM using Virgo Super Cluster (VSC) {10,1} as VSC{0,1}. Note: the y-axis is not on scale.

### IX. Using interior {N,n} QM to describe three-body (or N-body) problem

Moved to SunQM-7 (due to the size limitation of this paper).

#### X. Using interior {N,n} to explore the size of our universe and the smallest string through the number guessing

Moved to SunQM-7 (due to the size limitation of this paper).

### Conclusion

The interior {N,n} QM structural analysis is a novel method to study a atom's nucleus-electron system. The hidden property of {N,n//6} QM structure of nucleons inside a nucleus is uncovered only by using the n=1 ground state electron as  $e1\{0,1//6\}$ . The result reveals that nuclides of hydrogen (Z=1), He (Z=2), Li (Z=3), Ne (Z=10), Og (Z=118) atoms have the interior QM structure of  $e1\{-3,1\}$ ,  $e1\{-3,2\}$ ,  $e1\{-3,3\}$ ,  $e1\{-3,6\} = e1\{-2,1//6\}$ , and  $e1\{-1,1//6\}$ , respectively. This result revealed that because Fe element is the only element that has ~100% nucleon occupancy in the n =1 nucleon orbital shell space (among the n = 1 nuclides of Z = 11 ... 26, and also among its adjacent nuclides in n = 2 nuclides of Z = 27 ... 45), so it has the most stable {N,n//6} QM nucleus structure among its adjacent nuclides.

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[10] https://www.theguardian.com/science/life-and-physics/2016/apr/07/how-big-is-a-quark How big is a quark? Jon Butterworth, 4/7/2016, "the ZEUS experiment has just released a new limit on the size of the quark. 0.43E-18 meters, That's 2000 times smaller than a proton radius,

ZEUS Collaboration: H. Abramowicz, et al., Limits on the effective quark radius from inclusive ep scattering at HERA, arXiv:1604.01280, 4/5/2016.

[11] "Numerical list of Atomic Numbers of Elements" at "http://www.elementalmatter.info/number-protons-neutrons.htm". Updated according to wiki "periodic table".

[12] wiki "Atomic nucleus", "The diameter of the nucleus is in the range of  $1.75 \text{ fm}(1.75 \times 10-15 \text{ meters})$  for hydrogen (the diameter of a single proton) to about 15 fm for the heaviest atoms, such as uranium... The stable nucleus has approximately

a constant density and therefore the nuclear radius R can be approximated by the following formula,  $R = r_0 * A^{(1/3)}$ , where A = Atomic mass number (the number of protons Z, plus the number of neutrons N) and  $r_0 = 1.25$  fm =  $1.25 \times 10-15$  m. In this equation, the "constant"  $r_0$  varies by 0.2 fm, depending on the nucleus in question, but this is less than 20% change from a constant. In other words, packing protons and neutrons in the nucleus gives approximately the same total size result as packing hard spheres of a constant size (like marbles) into a tight spherical or almost spherical bag (some stable nuclei are not quite spherical, but are known to be prolate"

[13] Helium nuclear r = 19E-16 meters, from "http://iopscience.iop.org/article/10.1088/0031-8949/1995/T58/002", "compared with the theoretical prediction which has similar accuracy and depends also on the rms nuclear radius of 3He. For this we determine a preliminary value of 1.923 (37) fm".

[14] From wiki "Atomic Radius", Clementi, E.; Raimond, D. L.; Reinhardt, W. P. (1967). "Atomic Screening Constants from SCF Functions. II. Atoms with 37 to 86 Electrons". Journal of Chemical Physics. 47 (4): 1300–1307.

[15] John Dirk Walecka, Introduction to General Relativity. 2007. p153, equation 7.113.

[16] A series of my papers that to be published (together with current paper):
SunQM-5s1: White dwarf, neutron star, and black hole re-analyzed by using {N,n} QM.
SunQM-7: Using {N,n} QM and Simultaneous-Multi-Eigen-Description (SMED) to describe our universe.
SunQM-7s1: Relativity and {N,n} QM.
SunQM-9s1: Addendums, Updates and Q/A for SunQM series papers.

[17] Major QM books, data sources, software I used for this study are:
Douglas C. Giancoli, Physics for Scientists & Engineers with Modern Physics, 4th ed. 2009.
David J. Griffiths, Introduction to Quantum Mechanics, 2nd ed., 2015.
John S. Townsed, A Modern Approach to Quantum Mechanics, 2nd ed., 2012.
Stephen T. Thornton & Andrew Rex, Modern Physics for scientists and engineers, 3rd ed. 2006.
James Binney & David Skinner, The Physics of Quantum Mechanics, 1st ed. 2014.
Wikipedia at: https://en.wikipedia.org/wiki/
(Free) online math calculation software: WolframAlpha (https://www.wolframalpha.com/)
(Free) online spherical 3D plot software: MathStudio (http://mathstud.io/)
(Free) offline math calculation software: R
Microsoft Excel, Power Point, Word.
Public TV's space science related programs: PBS-NOVA, BBC-documentary, National Geographic-documentary, etc.

Journal: Scientific American.