



Simple and Validated Ultraviolet Spectrophotometric Method for the Estimation of Febuxostat in Bulk and Pharmaceutical Dosage Forms

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ABSTRACT

For estimation of the Febuxostat in bulk and pharmaceutical dosage forms, an ultra violet – spectrophotometric method has been developed and validated. The method employed 0.1 N NaOH as solvent. At 275 nm, the linear regression analysis data for the calibration plot showed good linear relationship with correlation coefficient value of 0.998 in the concentration range of 10 – 70 µg/ml. The limit of detection and limit of quantitation were found to be 0.239 µg/ml and 0.725 µg/ml respectively. The results demonstrated that the procedure is accurate, precise and reproducible (R.S.D. < 2 %).

Key words: Febuxostat, Validation, Estimation, UV spectrophotometry, 0.1 N NaOH, Bulk and Pharmaceutical forms.

INTRODUCTION

The Febuxostat chemically is 2-[3-cyano-4-(2-methylpropoxy) phenyl]-4-methylthiazole-5-carboxylic acid ¹ (Fig. 1) with a molecular weight of 316.38. The molecular formula is C₁₆H₁₆N₂O₃S. Tablets available in the local market for oral use contain the active ingredient, febuxostat in two dosage strengths; 40 mg and 80 mg ².

Febuxostat (TEI-6720; TMX-67) is a novel, orally administered, potent and non-purine analogue being developed by Teijin, with licensees

Ipsen and TAP Holdings, for the treatment of hyperuricemia in gout ³. Febuxostat was determined by HPLC ⁴ and UPLC MS/MS ⁵ and HPLC-FLU methods⁶. Febuxostat was estimated by UV spectrophotometric method in pure methanolic ^{7, 8} and methanol-alkali/acid mixture media⁹. According to Anjana *et al.*,². Febuxostat is slightly soluble in methanol and acetonitrile. In dosage form of Febuxostat, the inactive ingredients include lactose monohydrate, microcrystalline cellulose, hydroxypropyl cellulose, sodium croscarmellose, and silicon dioxide and magnesium stearate. The replacement of methanol

by pure sodium hydroxide solution as a solvent reduces the solubility of other ingredients. In addition, due to added advantages of sodium hydroxide solution as a solvent, different drugs were estimated using it as a solvent¹⁰⁻¹³. Hence, it was thought worthwhile to develop a rapid, sensitive, precise and accurate ultra violet – spectrophotometric method for validation of Febuxostat in bulk and pharmaceutical dosage forms by conducting systematic trials using pure sodium hydroxide solution as a solvent.

MATERIALS AND METHODS

The spectrophotometric measurements were carried out using a double beam LABINDIA UV – Visible spectrophotometer (UV – 3092) connected to computer and loaded with PMT detector UV – WIN 5 software. The instrument has an automatic wavelength accuracy of 0.1 nm and matched quartz cells of 10 mm (1.0 cm) cell path length. Shimadzu AUX-220 balance was used for weighing the samples. The absorption spectra of the reference and test solutions were carried out in a 1 cm quartz cells over the range of 200 – 400 nm. Febuxostat was tested for purity by measuring its melting point and IR spectra and no impurities were found. All the reagents used were of analytical grade.

Standard solution of Febuxostat

Standard stock solution (primary) was prepared by dissolving 10 mg of Febuxostat in 10 ml of 0.1 N NaOH to get concentration of 1mg/ml (1000µg/ml) and was stored at + 4°C during the study. Secondary stock solution was prepared daily by diluting 1ml of the primary stock solution to final volume of 10 ml using 0.1 N NaOH to get concentration of 0.1mg/ml (100µg/ml).

Preparation of calibration standard solutions

Suitable aliquots of the secondary standard solution of Febuxostat (10–70 ml) were transferred to a series of calibrated 100 ml standard volumetric flasks and the volume was made up to the mark with 0.1 N NaOH.

Scanning and determination of maximum wavelength (λ_{\max})

In order to ascertain the wavelength of

maximum absorption (λ_{\max}) of the drug, qualitative solution of the drug was prepared in 0.1 N NaOH and scanned by using UV spectrophotometer within the wavelength region of 200 – 400 nm against 0.1 N NaOH as blank. The calibration curve was constructed for absorbance versus concentration of Febuxostat. The resulting spectrum was shown in Figure 2, and the absorption curve showed characteristic absorption maxima at 275 nm for Febuxostat.

RESULTS AND DISCUSSION

The purpose of the present study is to develop a rapid, sensitive, precise and accurate ultra violet – spectrophotometric method for its validation in bulk and pharmaceutical dosage forms by conducting systematic trials.

Method Validation

Validation is one of the most important steps in method development for analytical determinations. The main validation parameters such as linearity and range, accuracy and precision, recovery, ruggedness, limit of detection (LOD) and limit of quantitation (LOQ) were evaluated in developed method¹⁴⁻¹⁷.

Linearity and range

The absorbance values for different standard solutions (10, 20, 30, 40, 50, 60 and 70µg/ml) of Febuxostat were measured at λ_{\max} 275 nm, against 0.1 N NaOH as blank. Some workers^{7,8} reported that the λ_{\max} was in the range 306–324 for Febuxostat in methanol medium. When a methanolic solution of Febuxostat was diluted with sodium hydroxide solution, the λ_{\max} shifted to 260 nm from

Table 1: Calibration values of Febuxostat

Concentration (µg/ml)	Absorbance*
0	0.000
10	0.139
20	0.279
30	0.415
40	0.553
50	0.685
60	0.833
70	0.955

315nm, whereas, λ_{\max} was unaffected by diluting with hydrochloric acid solution⁹. Each point of the calibration graph corresponded to the mean value obtained from three independent measurements (Table 1). The calibration graph (Fig. 3) was constructed by plotting absorbance versus concentration of Febuxostat. The calibration graph

of the absorbance versus concentration was found to be linear over the range of 10-70 $\mu\text{g/ml}$ for the proposed method. The linear regression equation obtained was $y = 0.0138x - 0.001$ where y is the absorbance and x is the concentration of pure drug solution. The summary of optical and regression parameters was shown in Table 2.

Table 2: Statistical data of the regression equations and validation parameters for of Febuxostat

S. No.	Parameter	Observation
Optical characteristics		
1.	Apparent molar absorptivity (l/mol.cm)	4334
2.	Sandell's sensitivity ($\mu\text{g/cm}^2/\text{A}$)	0.07299
Regression analysis		
1.	Slope	0.0138
2.	Intercept	0.001
3.	Regression coefficient (r)	0.998
Validation parameters		
1.	λ_{\max} (nm)	275
2.	Beer's Law Limit (Linearity, $\mu\text{g/ml}$)	10 – 70
3.	Limit of detection ($\mu\text{g/ml}$)	0.239
4.	Limit of quantitation ($\mu\text{g/ml}$)	0.725

Table 3: Recovery of Febuxostat using the proposed UV method

Level of recovery (%)	Amount of drug added ($\mu\text{g/ml}$)	Amount of drug recovered ($\mu\text{g/ml}$)		% of recovery	
		Mean \pm SD *	%RSD	Mean \pm SD*	%RSD
50	30	30.00 \pm 0.11	0.37	100.01 \pm 0.37	0.37
100	40	40.14 \pm 0.06	0.15	100.35 \pm 0.15	0.15
150	50	50.14 \pm 0.06	0.13	100.27 \pm 0.13	0.13

* Average of three determinations

Table 4: Precision of the proposed method

Used Concentration ($\mu\text{g/ml}$)	Determined Concentration ($\mu\text{g/ml}$)*			
	Intra-day variation		Inter-day variation *	
	Mean \pm SD	% RSD	Mean \pm SD	% RSD
30	30.04 \pm 0.136	0.451	30.07 \pm 0.065	0.218
40	39.95 \pm 0.088	0.220	39.95 \pm 0.030	0.076

* Average of six determinations

Accuracy

To determine the accuracy of the proposed method, recovery studies were carried out by adding different amounts (50%, 100%, and 150%) of bulk samples of febuxostat to 20 µg/ml so that overall concentration will be within the linearity range. The accuracy was expressed in terms of percent recovery. The mean of percentage recovery values varied from 100.01 to 100.35 (Table 3), whereas in Paramdeep *et al.*, method⁷ the range was 99.23 – 100.58. The statistical analysis of data obtained for the estimation of febuxostat indicates a high level of accuracy for the proposed method as evidenced by the low values of standard deviation and relative standard deviation.

Precision

The precision of a method is defined as the closeness of agreement between independent test results obtained under optimum conditions. Two different concentrations of febuxostat in the linear range (30 and 40 µg/ml) were analyzed in six independent series in the same day (intra-day precision) and in six consecutive days (inter-day precision) and results were given in Table 4.

The % RSD values of intra-day and inter-day studied for the proposed method were in the

ranges 0.451 – 0.220 and 0.218 – 0.076 respectively compared to 0.08 – 2.05 and 0.174 – 1.862 respectively of method developed by Paramdeep *et al.*,⁷. This concludes that the precision of the current method was satisfactory.

Ruggedness

The ruggedness of the proposed method was evaluated by applying the developed procedure for assay of 30 µg/ml and 40 µg/ml of febuxostat using the same instrument by two different analysts under the same optimized conditions at different days. The obtained results were found to be reproducible, since there was no significant difference between analysts (Table 5). The % of RSD for two different analysts were in the range of 0.291 – 0.513 compared to 0.6513 – 1.0692 of Paramdeep *et al.*, method⁷. Thus, the proposed method was rugged.

Detection of LOD and LOQ

For determination of sensitivity of the proposed method, LOD and LOQ were calculated. Based on the signal to noise ratio they were quantified. The lowest detectable concentration of the analyte by the method is LOD where as the minimum quantifiable concentration is LOQ. LOD and LOQ for febuxostat were calculated according

Table 5: Ruggedness data of Febuxostat by two analysts at different days

Used Concentration (µg/ml)	Determined Concentration (µg/ml)*			
	Analyst 1		Analyst 2	
	Mean ± SD	% RSD	Mean ± SD	% RSD
30	30.16 ± 0.155	0.513	30.17 ± 0.088	0.291
40	39.99 ± 0.148	0.370	40.05 ± 0.099	0.247

* Average of six determinations

Table 6: Assay of Pharmaceutical Formulation

Formulation	Label Claimed (mg)	Amount Recovered* (mg)	% Drug Recovered	%RSD
FEBUGET® Tablets	40	39.5653±0.2898	98.91	0.7326

* Average of three determinations

to the ICH guidelines by using S (relative standard deviation of the response) and δ (slope of the calibration curve).

$$\text{LOD} = 3.3 \times \sigma / S = 0.239 \mu\text{g/ml and}$$

$$\text{LOQ} = 10 \times \sigma / S = 0.725 \mu\text{g/ml}$$

In the method developed by Sheth *et al.*,⁹ LOD and LOQ values were 0.5871 and 1.7793 respectively. These results indicate that the present proposed method was sensitive to detect and quantify.

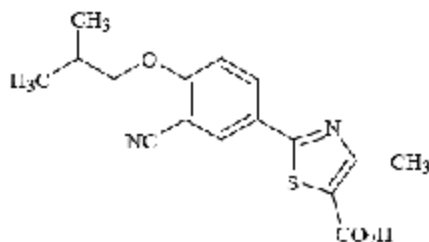


Fig. 1: Chemical Structure of Febuxostat

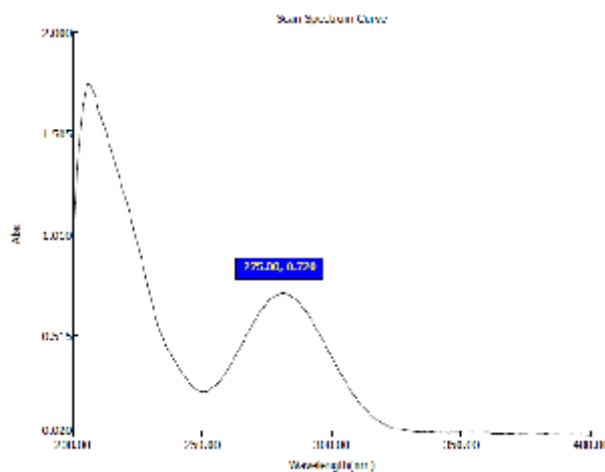


Fig. 2: UV Spectrum of Febuxostat

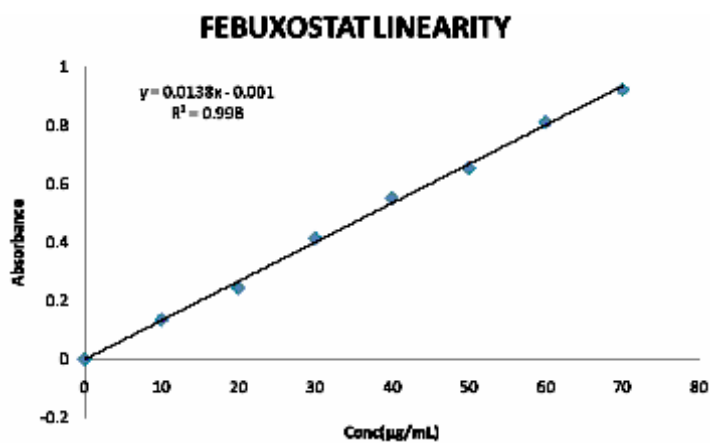


Fig. 3: Calibration graph of Febuxostat

Analysis of pharmaceutical formulations

Ten Febuxostat tablets were taken and the average weight was determined. The tablets were ground and mixed well. The powder of the sample equivalent to 10 mg of Febuxostat was accurately weighed and transferred into a 10 ml volumetric flask. About 7 ml of diluents was added, sonicated to dissolve it completely and made the volume up to the mark with diluent. Mixed well and filtered through Whatmann filter paper. An aliquot equivalent to 20 mg of the sample was pipetted into a 10 ml volumetric flask and made up to the mark after filtration. From the absorbance value, the drug content per tablet (on an average weight basis) was calculated and the results were tabulated (Table 6). Good recovery values of drugs shows that the proposed method can be successfully applied to the determination of febuxostat in pharmaceutical formulations without any interference from common excipients.

CONCLUSIONS

In the proposed UV spectrophotometric method for the estimation of febuxostat, low cost sodium hydroxide solution replaced the usage of organic solvents. This method has the lowest LOD value and is more sensitive method. From the results obtained, we concluded that the suggested method showed high sensitivity, accuracy and precision. Therefore the developed method can be used for routine analysis for estimation of febuxostat in bulk and pharmaceutical dosage forms.

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