The periodic table form possible to coincide with a proton distribution

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Abstract

The periodic table form seems to coincide with a folding nuclear structure, a distribution of protons (nucleons) nearly the same as Lewis dot structure, which will grow vertical 4 A (representative), 4 B (transition) and 8 C (inner transition) α -clusters bound with valency neutron (excess neutron, ~ *A* - 2*Z*) to stand a core (1st period) of likely expanding in Co, Ni, Rh and Pd. In a typical fission of ²³⁵U+n→¹³⁷Ba+⁹⁷Kr+2n, the mass difference of ¹³⁷Ba - ⁹⁷Kr=40 is suggested to result from the 16 α -clusters of different length cleft into 9:7 (±1 A α -cluster, 2 clusters of 5 α) and 2n release from valency neutrons in its scission line; also, the 16 α -clusters cleft into 15:1 implies to point to that α (1 C) and cluster (1 B \geq ¹²C, 1 A \geq ²⁰Ne) decay. To cross nuclei, atoms and molecules, this work provides a way that may enable us closer to nature of the periodic table, the periodic law of dominating their behaviors.

Keywords: periodic table, proton distribution, folding nucleus, nuclear core, α -cluster, valency neutron

PACS:

01.30.Kj Handbook, dictionaries, tables, and data compilations
21.10.Dr Binding energies and masses
21.10.Ft Charge distribution
21.10.Gv Nucleon distributions and halo features
21.60.-n Nuclear structure models and methods
21.60.Gx Cluster models
23.35.+g Isomer decay
23.60.+e α decay
23.70.+j Heavy-particle decay
24.75.+I General properties of fission
32.10.-f Properties of atoms
33.15.-e Properties of molecules

Abstract Introduction Nuclear 4 steps and 16 α-clusters Composition of beta stable line Fission configuration Discussion References

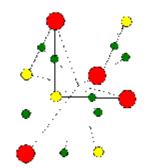
Introduction

The periodic table with the elements accumulated now is well-known and plays a guide role in physical and chemical fields. Its form, the periodic law, *Z*, the atomic number, is often explained by the electron since it is discovered (1) to establish atomic models (2-4), an atomic periodicity, though about in the same time it has been proven to result from the proton number (5, 6). This seems possible to attribute to that a clear cubic distributions of protons, the atomic number, in a nucleus (7) consisting of protons and neutrons (8) might has not been revealed (9-12), to the best of the author's knowledge. However, it may be a flaw to ignore Moseley's research (5, 6) that the proton number could convey a nuclear periodicity to a certain extent. As the shell model (12), its magic nuclei 2, 8, 20, 28, 50, 82 and 126 is apparently inconsistent with noble nuclei (gases) 2, 10, 18, 36, 54, 86 and 118, while noble nuclei appear to close naturally that could display a cubic *Z* in a tangible proton distributions. It was observed in analysis molecular structure, covering nuclear, atomic and molecular three levels, which is both quite simple and universal.

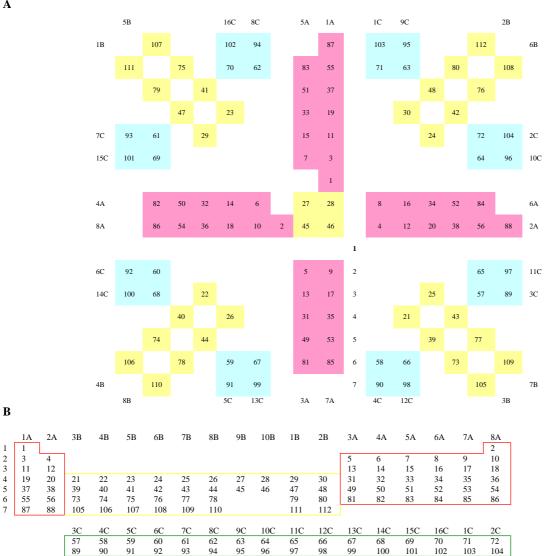
The observation is starting from a curiosity that whether an atomic mass has an influence upon molecular bond energy or not about in the summer of 1987. Because an element occurs some isotopes and then had no intention of taking their relative mass what want to see nucleons how to distribute in a molecule (atom), every element is represented by its maximal abundant isotope (Table 1) selected from a handbook. The simulation of nucleon distributions of atoms in a molecule is that according to their number of periods and groups. For instance, in ${}^{12}C^{1}H_{4}$, the isotopic masses marked in the molecular formula, ¹H (99.985%, natural abundance) is simple; ¹²C (98.89%) is in the 2nd period and group 4A, indicating that its 6 p should be separated into: 2 p in nuclear core and 4 p in nuclear skin, then its 6 n overlap this 6 p frame, that is, ${}^{12}C_{c-2+2}{}^{2222}$ (most notes in captions of Table 1, Figs. 1A-C and 2A), which is a tetrahedral distribution to connect 4 ¹H atoms, a ¹²C¹H₄ molecular-nuclear (M-N) structure (Supporting fig. S122), as it is well-known that a methane is a tetrahedral structure. Also, it seems to be straightforward that if alternating single and double bonds of a benzene ring (${}^{12,13,14}C_6{}^{1}H_6$, fig. S125), a buckyball (${}^{12,13,14}C_{60}$) and diamond (${}^{12,13,14}C_n$) are rooted in a tetrahedral nucleus of ${}^{12,13,14}C$. Here, black and white stones of Go (a Chinese game) were used to stand for protons and neutrons, respectively. It is useful in this early work; without it, it is difficult to simulate a M-N structure, nuclear structure and fission, since a nucleus is a system of numerous nucleons.

In order to test, this proton distribution hypothesis, which could be regarded as nucleon distributions for that neutrons by and large surround protons to fit in a nucleus, is used trying to account for nuclear fission phenomena (13): structures, mass difference and vanishing points of fragments, and emission angle of fission α -particles. Furthermore, it implies that both α and cluster (heavy-particle, mass A > 4) decay are similar to fission in principle that only mass ratios are different. Now first is to describe this nucleon distributions which seems to coincide with the periodic law that periods and groups of the periodic table are corresponding to horizontal 7 layers and vertical 16 α -clusters of nuclei, respectively (Figs. 1*A*-*C* and 2*A*). In the 1, 2-3, 4-5 and 6-7 periods/layers or nuclear 0, A, B and C 4 steps, respectively, basic, light, mid and heavy nuclei will be exhibited in nuclear growth (Table 1 and figs. S1-118) to pose different shapes that imply to be folding.

Fig. 1. Nuclear folded and unfolded frames.



А



С

(A) Nuclei seem to grow 4 A ($Z \ge 10$, red), 4 B ($Z \ge 26$, yellow) and 8 C α -clusters (axes) ($Z \ge 70$, green), where are ¹p (last proton, added to the prior element, Z - 1) locations of A (representative), B (transition) and C (inner transition) family elements, respectively. (B) Growing locations of ^{1}p of the atomic number 1-112. The period is in the middle (uncolored) and the group is in four sides. The groups 1A, 5A pair on A1 axis, 2A, 6A on A2 axis, 3A, 7A on A3 axis and 4A, 8A on A4 axis; it is the same way for the B and C family groups. Four ¹p of Z = 27, 28, 45 and 46 in old group 8B (American convention; groups 8-10, modern form), which was revised into groups 8-10B here, are sunk into the 1st period. (C) The A (red), B (yellow) and C (green) families correspond to 4 A, 4 B and 8 C α -clusters, respectively. As 8 C α -clusters occupy 16 ^lp, the number of C family elements is suggested to increase from 2×14 to 2×16 , then group 8B only leave ₇₈Pt and ₁₁₀Ds. The classification of C family is following previous B family that 71Lu and 72Hf of groups 1-2C into inner transition is analogous to ${}_{29}$ Cu and ${}_{30}$ Zn of groups 1-2B into transition, though C shell may have right been closed at ${}_{70}$ Yb (Table 1).

Table 1. A periodic distributions of nucleons for the maximal abundant isotopes.

In expression of nucleon distributions, subscript, left and right superscript of the periodic number are the numbers of ^vn, ^pn and ¹p respectively, and skins are axial configurations that 4 A, 4 B and 8 C axes in 7 codes (the Arabic numeral): 1, proton (p); 2, deuteron (d); 3, triton (t); 4, alpha particle (α); 0, neutron (n); <u>2</u>, dineutrons (^dn); <u>3</u>, ³He ion. Hyphen (-) is the same as upside. Chemical valency corresponding to axial configuration is listed in the 2nd period. In the 3rd period is grown mass between these nuclei. There will be a fluctuation only if ⁵⁸Ni (67.88%) and ¹⁰⁶Pd (27.33%) in group 10B are listed.

nucleus						nucleon distribution	chemical valency	grown mass	abundance% /half-lief
$^{1}\mathrm{H}$	1 ^{a-1}								99.9852
⁴ He	1 ^{a-1010}								100
⁷ Li	² 1 ²	2 ^{a-100}					1		92.5
⁹ Be	-	2 ^{a-2010}					2		100
11 B	-	2 ^{a-0222}					3		80.2
¹² C	-	2 ^{a-2222}					4		98.89
^{14}N	-	2 ^{a-4222}					- 3		99.63
¹⁶ O	-	2 ^{a-4242}					- 2		99.75
¹⁹ F	-	2 ^{a-4443}					- 1		100
²⁰ Ne	_	2^{a-4444}					0		90.5
²³ Na	_	⁸ 2 ⁸	3 ^{a-100}						100
²⁴ Mg	_	_	3 ^{a-1010}					1	78.99
²⁷ Al	-	-	3 ^{a-0222}					3	100
²⁸ Si	-	-	3 ^{a-2222}					1	92.2
³¹ P	-	-	3 ^{a-4232}					3	100
³² S	-	-	3 ^{a-4242}					1	95.0
³⁵ Cl	-	-	3 ^{a-4443}					3	75.77
⁴⁰ Ar	-	-	43 ^{a-4444}					5	99.59
³⁹ K	-	-	⁸ 3 ⁸	4	a-100				93.3
⁴⁰ Ca	-	-	-	4	a-1010				96.94
⁴⁵ Sc	-	-	⁸ ₄ 3 ⁸	4 ^{b-111} 2					100
⁴⁸ Ti	-	-	-	4 ^{b-2222}					73.7
⁵¹ V	-	-	-	4 ^{b-4232}					99.75
⁵² Cr	-	-	-	4 ^{b-4242}					83.79
⁵⁵ Mn	-	-	-	4 ^{b-4443}					100
⁵⁶ Fe	-	-	-	4 ^{b-4444}					91.7
⁵⁹ Co	⁴ 1 ³	-	-	4 -					100
⁶⁰ Ni	⁴ 1 ⁴	-	-	4 -					26.23
⁶³ Cu	-	-	-	4 -	a-100				69.1
⁶⁴ Zn	-	-	-	4 -	a-1010				48.9
⁶⁹ Ga	-	-	-	₄ 4 ⁻	a-111 <u>2</u>				60.0
⁷⁴ Ge	-	-	-	₄ 4 -	a-3232				36.4
⁷⁵ As	-	-	-	₄ 4 ⁻	a-4232				100
⁸⁰ Se	-	-	-	₈ 4 ⁻	a-4242				50.0
⁷⁹ Br	-	-	-	₄ 4 ⁻	a-4443				50.69
⁸⁴ Kr	-	-	-	₈ 4 ⁻	a-4444				57.0

25 m 1				16 116	_	- 1			
⁸⁵ Rb	-	-	-	¹⁶ ₈ 4 ¹⁶	5	a-1			72.17
⁸⁸ Sr	-	-	-	-	5	a-1010			82.6
⁸⁹ Y	-	-	838	-	5 ^{b-1112}				100
⁹⁰ Zr	-	-	-	-	5 ^{b-1212}	-			51.46
⁹³ Nb	-	-	-	-	5 ^{b-4333}				100
⁹⁸ Mo	-	-	-	-	5 ^{b-4343}	a-0000			23.78
⁹⁹ Tc	-	-	-	-	5 ^{b-4443}	-			$2.1 \times 10^5 \text{ yr}$
102 Ru	⁶ 1 ⁴	-	-	-	5 ^{b-4444}	-			31.64
103 Rh	⁶ 1 ⁵	-	-	-	5 -	-			100
¹⁰⁴ Pd	⁶ 1 ⁶	-	-	-	5 -	-			10.97
¹⁰⁷ Ag	-	-	-	-	5 -	a-1 <u>222</u>			51.82
¹¹⁴ Cd	-	⁸ 42 ⁸	⁸ 43 ⁸	-	5 -	a-11 <u>22</u>			28.86
¹¹⁵ In	-	-	-	-	5 -	a-0222			95.72
¹²⁰ Sn	_	-	_	-	5 -	a-3333			32.85
¹²¹ Sb	-	_	-	-	5 -	a-4333			57.25
¹³⁰ Te	-	-	-	-	₈ 5 -	a-4343			34.48
¹²⁷ I	_	_	-	-	45 ⁻	a-4443			100
¹³² Xe	_	_	-	_	₈ 5 -	a-4444			26.89
¹³³ Cs	_	_	_	_	¹⁶ 85 ¹⁶	6		a-1	100
¹³⁸ Ba		_	-	_	-	6		a-1122	71.66
¹³⁹ La	-	-			- ¹⁶ 5 ¹⁶	6 ^{c-1010}	-100 b-00	00 a-0000	99.91
¹⁴⁰ Ce	-	-	-	-		6 ^{c-1010}		-	
¹⁴¹ Pr	-	-	-	-	-	6 ^{c-1010}		-	88.48
	-	-	-	-	-	6 ^{c-1112}			100
¹⁴² Nd	-	-	-	-	-			-	27.11
¹⁴⁵ Pm	-	-	-	-	-	6 ^{c-2121}			17.7 yr
¹⁵² Sm	-	-	-	-	-	6 ^{c-3232}		-	26.7
¹⁵³ Eu	-	-	-	-	-	6 ^{c-4232}		-	52.2
¹⁵⁸ Gd	-	-	-	-	-	6 ^{c-4333}		-	24.7
¹⁵⁹ Tb	-	-	-	-	-	6 ^{c-4343}		-	100
¹⁶⁴ Dy	-	⁸ 2 ⁸	-	-	¹⁶ ₈ 5 ¹⁶	6 ^{c-4343-}		-	28.2
¹⁶⁵ Ho	-	-	-	-	-	6 ^{c-4443}		-	100
¹⁶⁶ Er	-	-	-	-	-	6 ^{c-4443}		-	33.4
¹⁶⁹ Tm	-	-	-	-	-		I-4443 b-00 <u>2</u>	22 -	100
¹⁷⁴ Yb	-	⁸ 42 ⁸	-	-	-	6 ^{c-4444}	-4444 -	-	31.84
¹⁷⁵ Lu	-	-	-	-	-	6	- b-1 <u>2</u>		97.4
$^{180}\mathrm{Hf}$	-	-	-	-	-	6	- b-22	2 <u>22</u> a- <u>2222</u>	35.1
¹⁸¹ Ta	-	-	-	-	-	6	- b-32	- 22	99.988
^{184}W	-	-	-	-	-	6	- b-33	- 33	30.7
¹⁸⁷ Re	-	-	-	-	-	26	- b-43	33 -	62.5
¹⁹² Os	-	-	-	-	-	66	- b-43	43 -	41.0
¹⁹³ Ir	-	-	-	-	-	66	- b-44	43 -	62.6
¹⁹⁴ Pt	-	-	-	-	-	66	- b-44	44 -	32.9
¹⁹⁷ Au	-	-	-	-	-	106		a-1 <u>222</u>	100
²⁰² Hg	-	-	-	-	-			a-11 <u>22</u>	29.7
²⁰⁵ Tl	-	-	-	-	-	166		a-223 <u>2</u>	70.5
²⁰⁸ Pb	-	-	-	-	-	166		a-3333	52.4
²⁰⁹ Bi	-	-	-	-	-	166		a-4333	100
²⁰⁹ Po	-	-	-	-	-	166		a-4342	103 yr
						-0-			5

²¹⁰ At	-	-	-	-	-	166 - a-44	42	8.3 h
²²² Rn	⁶ 61 ⁶	-	-	-	-	166 - a-44	¹⁴⁴ ⁴ 7	3.82 d
²²³ Fr	-	-	-	-	-	³² 166 ³²	7 b-0000 a-1	22 min
²²⁶ Ra	-	-	-	-	-	-	7 - a-1010	1602 yr
²²⁷ Ac	616	-	-	-	-	-	7 ^{c-1010-100} - a-0000	21.77 yr
²³² Th	-	-	-	-	-	-	7 ^{c-11<u>22</u>-11<u>22</u>}	100
²³¹ Pa	-	-	-	-	-	-	7 ^{c-11<u>22</u>-111<u>2</u>}	$3.3 \times 10^4 \text{ yr}$
²³⁸ U	-	-	-	-	-	-	7c-333 <u>2</u> -333 <u>2</u>	99.27
²³⁷ Np	-	-	-	-	-	-	7c-2223-222 <u>2</u>	2.1×10 ⁶ yr
²⁴⁴ Pu	-	-	-	-	-	-	70-3333-3333	8.3×10 ⁷ yr
²⁴³ Am	-	-	-	-	-	-	7°-4323-3323	7950 yr
²⁴⁷ Cm	-	-	-	-	-	-	7 ^{c-4323-4322} b- <u>2222</u> -	$1.6 imes 10^7$ yr
²⁴⁷ Bk	-	-	-	-	-	-	7°-4242-4322	1.4×10^3 yr
²⁵¹ Cf	-	-	-	-	-	-	7c-4343-4342	898 yr
²⁵² Es	-	-	-	-	-	-	7c-4443-4342	350 d
²⁵⁷ Fm	-	-	-	-	-	-	7c-4443-4442 - a- <u>2222</u>	101 d
²⁵⁸ Md	-	-	-	-	-	-	7 ^{c-4444-4442}	52 d
²⁵⁹ No	-	-	-	-	-	-	7 ^{c-4444-4444} b-0 <u>222</u> -	58 min
²⁶⁰ Lr	-	-	-	-	-	-	7 - b-2 <u>222</u> -	3 min
261 Rf	-	-	-	-	-	-	7 - b-32 <u>22</u> -	65 s
²⁶² Db	-	-	-	-	-	-	7 - b-233 <u>2</u> -	40 s
²⁶³ Sg	-	-	-	-	-	-	7 - b-2333 -	0.9 s
²⁷⁰ Bh	-	-	-	-	-	-	₆ 7 - b-4323 -	61 s
²⁷⁷ Hs	-	-	-	-	-	-	167 - b-4342 a-0000	
²⁷⁸ Mt	-	-	-	-	-	-	167 - b-4442 -	7.6 s
²⁸¹ Ds	-	-	-	-	-	-	167 - b-4444 a- <u>2</u> 000	11 s
²⁸¹ Rg	-	-	-	-	-	-	167 a-2000	26 s
²⁸⁵ Cn	-	-	-	-	-	-	167 - a-32 <u>22</u>	29 s
²⁸⁷ 113	-	-	-	-	-	-	167 - a-333 <u>2</u>	
²⁸⁸ 114	-	-	-	-	-	-	167 a-3333	
²⁸⁹ 115	-	-	-	-	-	-	167 - a-4333	
²⁹⁰ 116	-	-	-	-	-	-	167 - a-4343	
²⁹¹ 117	-	-	-	-	-	-	167 - a-4443	
²⁹² 118	616	42 ^{a-4444}	43 ^{a-4444}	84 ^{b-4444} a-4444	85 ^{b-4444 a-4444}	166 ^{c-4444-4444} b-4444 a-4444	167 ^{c-4444-4444 b-4444} a-4444	

Nuclear 4 steps and 16 α-clusters

The 1st period: nuclear core. Folding nuclei appear to be woven of 7 kind of particles: ^{1,2,3}H, ^{3,4}He, valency neutron and di-neutrons (unclear who termed), according to the number of periods and groups, which can be separated into core, middle and skin three parts, core+middle = ^mc ($Z \ge 11$); skin mass, ^ms, its particle structures and distributions is called axial configuration in ground state. Generally, ^mc is a noble nucleus and core is a ⁴He of the closed 1st period in a range Z = 3-26, for it will expand in Z = 27, where Z = 21, 27 and 57 are three key points in proton distributions, implying to differ from prevailing electron distributions (configurations). On the other hand, in nuclear growth a nucleon behavior seems to loom up a tetrahedral shape having some "nucleon valency" (~ 4) to bind other nucleons or basic nuclei (Fig. 2*A*) with an explicit direction, suggesting that a molecular bond may result essentially from this character (Fig. 2*B*).

Figures. $\angle A - B$ also imply that a nucleus and a molecule able to be a three dimensional structure depend largely on that an α -particle is tetrahedral, a most concise solid structure in nature.

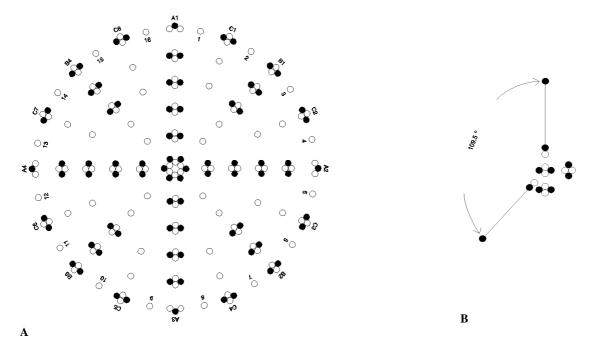


Fig. 2. An images of 208 Pb (unfolded) and $^{1}H_{2}^{16}$ O.

(*A*) Closed and open circles are protons and neutrons, respectively. In ²⁰⁸₈₂₊₈₆₊₄₀Pb_{c-6+6,v-40}^{c-4444.4444,b-4444,a-3333}, right superscript is axial configuration, subscript c-6+6 is core (n+p), v-40 is ^vn number and 82+86+40 is $Z+^{p}n+^{v}n$ three frames (shells), where ^pn is pair neutron accompanying proton to form ^{2,3}H and ^{3,4}He, and ^vn is valency neutron (single open circles) to fill gaps between axes and shells. In order to describe fission, a nuclear coordinate was introduced that the serial numbers of axes and layers are after A, B and C letters; for valency neutron is N and it is clockwise rotating from A1 axis to the origin point: $1 \rightarrow 4$, $1 \rightarrow 8$ and $1 \rightarrow 16$ in A, B and C steps, respectively. This figure suggests that in 1-1/4-8 fission (from N1-6 through 1/4 core to N8-6), 1-8 and 8-1 sectors generally are light (A_L) and heavy (A_H) fragments, respectively (21-23), some of 6 ^vn (N1-6, 1-4, 1-2 and N8-6, 4-4, 2-2) are become unbound neutrons (24, 25) in the scission line, and angular distributions of α -particle (26, 27) are 90 - 22.5° for A_L and 90+22.5° for A_H coming from C1-6 or C4-6. (*B*) In the center of ¹H₂¹⁶O is a ¹⁶O_{c-2+2}⁴²⁴² and the farthest are 2 ¹H, suggesting that its chemical valency and their angle are rooted in 2 d on the nuclear skin. If the 2 d of ¹⁶O⁴²⁴² (99.756%) were replaced by 1 or 2 t, it is ¹⁷O⁴³⁴² (0.039%) or ¹⁸O⁴³⁴³ (0.205%).

The 2-3 periods: Lewis dot structure of nuclear skin. In F₂, O₂ and N₂ molecules, a single, a double and a triple bond coincided with a pair of t in two nuclear skins of (¹⁹F⁴⁴⁴³)₂, two pairs of d in (¹⁶O⁴²⁴²)₂ and three pairs of d in (¹⁴N⁴²²²)₂ suggest that molecular different shapes may be rooted in axial configurations (see also figs. S119-124: ²³Na³⁵Cl, ⁹Be³⁵Cl₂, ¹¹B¹⁹F₃, ¹²C¹H₄, ¹⁴N₂ and ¹⁶O₃, respectively). In the 2nd layer, nuclei start to grow up to 4 α-particles (8 elements) standing on 4 nucleons of a basal tetrahedral α-particle. In ⁷Li_{c-2+2}¹⁰⁰, axial configuration is a-100 that its 1¹p with 2 ^pn are dispersed on 3 axes of 4 hole axes, thus its skin has merely a proton in that molecular structure is reflected 1 of chemical valency. Generally, the number is equal between chemical main valency and ^{1,2,3}H (p, d and t) at top of nuclear 4 A axes (Table 1), which is nearly the same as Lewis dot structure (3) that an α-particle corresponds to a lone pair of electrons. For example, in ¹⁶O, 2 dots and 2 lone pairs of electrons will identify with skin 2 d and 2 α of ¹⁶O_{c-2+2}⁴²⁴²; if 2 ¹H atoms descended on the 2 d extended lines, geometrical angle being 109.5°, it is a M-N structure of ¹H₂¹⁶O (Fig. 2*B*). At near the 3rd layer end, nuclei begin to grow valency neutrons that first clearly to emerge 4 ^vn will be in ⁴⁰Ar_{v-4} in stable nuclei (~ 300), where the 4 ^vn position is to grow out 4 B α-clusters of next step.

The 4-5 periods: nuclear expansible core. Along with nuclear growing to the 4th layer, nuclear skin area will increase enough to hold another 4 B α -clusters (transition metals) in between 4 A α -clusters; further, in order to support increasing mass its core will be intensified in old group 8B. In

terms of electron distributions, proton distribution of ${}_{21}Sc$ is 2 'p of ${}_{19}K$ and ${}_{20}Ca$ to float on 2 A axes, only the ${}^{1}p$ of ${}_{21}Sc$ fit on B axis, but it is not very appropriate from nucleon arrangements of subsequent elements to speculate; i.e., the 3 ${}^{1}p$ of ${}_{19}K$, ${}_{20}Ca$ and ${}_{21}Sc$ should simultaneously glide upon B axes. However, it is hardly possible that protons like electrons is a probability distribution. In comparison, a pair distributions of electrons and protons are ${}_{21}Sc$:[e-(18)ds²/p-(18)d³] (in proton distributions, A = s+p that in the range Z = 3-26, basal 2 p and 2 n which 4 A axes stand on have not been distinguished, B = d and C = f, respectively). Thus, from ${}_{19}K$ to ${}_{26}Fe$ in a total of 8 ${}^{1}p$ is piled on 4 B axes. The 2 ${}^{1}p$ of ${}_{27}Co$ and ${}_{28}Ni$ begin to sink into the core (Fig. 1*B*). The 2 ${}^{1}p$ of ${}_{29}Cu$ and ${}_{30}Zn$ will slip back upon A axes, though they are B family elements, then until to ${}_{36}Kr$, the 8 ${}^{1}p$ are piled on 4 A axes. The 5th layer is the same as the 4th layer, and, the 2 ${}^{1}p$ of ${}_{45}Rh$ and ${}_{46}Pd$ in groups 9-10B (Fig. 1*C* and figs. S45-46) have also been sunk into the core, which may be finished that a total of 6 ${}^{1}p$ (Z = 1, 2, 27, 28, 45 and 46) with 6 n are to form an innermost close-packed core (Fig. 2*A*).

Originally, here it also felt puzzling that where the ¹p of ₂₇Co should put on. Going through a period of time fiddling with its disorderly nucleon distribution, it seems to indicate that the ¹p of ₂₇Co to put on the center is an optimal option due to ⁵⁶Fe skin closure, i.e., ⁵⁶Fe_{c-2+2}⁴⁴⁴⁴ (91.7%) \rightarrow ⁵⁹Co_{c-4+3}⁴⁴⁴⁴ (100%) (figs. S26-27), ₂₇Co:[e-(18)d⁷s²/p-(19)d⁸]. First, it can explain a problem that why exist old group Fe-Co-Ni 8B in the periodic table. Second, this performance will enable a nucleus to possess a definite hub, a Coulomb repulsion center, otherwise its shape can not be opened up, however, like a tiny liquid drop (11). Corresponding to this is in that a chart (14), per nucleon binding energy ~ 8.7 MeV is maximal, as nuclear mass increase to *A* ~ 60, i.e. Fe-Co-Ni region, which would imply that though at Fe-Co-Ni region the nuclear core has been intensified immediately, a sharp change of nucleon distributions, this curve remains to fall. In addition, ferromagnetic only occurs in Fe, Co and Ni at room temperature which possibly has a link to the nuclear core expansion, i.e. a nuclear growth show to be rather suitable, such as c-2+2 in ¹⁶O, ⁴⁰Ca, c-4+4 in ⁶⁰Ni, ⁸⁸Sr and c-6+6 in ¹²⁰Sn, ²⁰⁸Pb these magic nuclei (Table 1).

The 6-7 periods: folding nucleus. One of extraordinary feature in the periodic table is old group 8B existence, implying that nuclei contain a expansible core, the other is the number of inner transition metals, from where a particular place, $57La:[e-(54)ds^2/p-(54)f^3]$, start to grow out between 4 A and 4 B α -clusters, implying that nuclei are folding. The extrapolation is that, if C family ^lp were fitting on 6 faces or 12 sides of Fig. 1A, it needs 12 ^lp or 24 ^lp, what is both impractical. Therefore, it should averagely vacate 4 sides in the 12 sides that only occupy 8 sides to grow 8 C α -clusters. On the other hand, whether C family contains 14 elements? If so, while assume that α is one of particles to construct a nucleus, as such an odd number of 7 α will be asymmetric in a nuclear shape (coordinate). So far, it is thought that nuclear shapes might have not been so easy to recognize, whereas in here are tangible that nuclear shapes almost are tetrahedral in A step (¹²C, ²⁸Si) and cubic in B step (⁷⁴Ge, ¹²⁰Sn), but in C step (²⁰⁸Pb), the shape would be kept in a phase that between cubic and flat, which may result from a nuclear vibratory pattern that only allows to occur 8 C α -clusters. In fact, this nuclear pattern first is two dimensional using Go game stones to put on the floor (see also figs. S1-125), whereas it is so coincidental that when it is folded into three dimensional, such as ${}^{12}C$, ${}^{74}Ge$ and ${}^{208}Pb$ (Fig. 2A). However, from the number of inner transition metals being 16 or 14, not other numbers, and valency neutron number (see later), it strongly suggests that a nuclear shape is folding; further, the extrapolation of C family 8 α -particles appears helpful to explain angle distribution of fission α particle (see later).

Vertical 16 *a*-clusters. As in Table 1, configurations of nuclear vertical 16 *a*-clusters show to correspond to the groups that is almost the same in a big group, regardless of group A, B and C. For example, in big group 7 (7A, 7B, 13-14C), axial configuration 4443 is always found in ¹⁹F, ³⁵Cl, ⁷⁹Br, ¹²⁷I of 7A, ⁵⁵Mn, ⁹⁹Tc, ¹⁹³Ir of 7B and ¹⁶⁵Ho^{c-4443-4343}, ¹⁶⁶Er^{c-4443-4443} of 13-14C. In big group 3, a di-neutrons may often serve as a proton in skins of mid-heavy nuclei; e.g., ⁴⁵Sc^{b-1112} and ⁸⁹Y^{b-1112} in 3B, each of them has 3 ¹p to stay on 3 axes of 4 B axes, then a ^dn will substitute for a proton to occupy a surplus axis, thus, the 4 particles (3¹p+1^dn) are formed a stable b-tetrahedron out of their ^mc, i.e., ⁴⁰Ar+b-1112 (⁴⁵Sc, 100%) and ⁸⁴Kr+b-1112 (⁸⁹Y, 100%), similar to a b-tetrahedron of ⁴⁰Ar+b-2222 (⁴⁸Ti, 73.7%) that is different from an a-tetrahedron of such as

In stable nuclei, ⁴⁵Sc is likely emerging di-neutrons for the first time, which seems a unique structure. In group 1A, a ⁷Li (92.5%) may prefer a-100 to a-3 (a single triton, alternative) in its skin, including below ²³Na¹⁰⁰ (100%) and ³⁹K¹⁰⁰ (93.3%), because skin particle masses will smoothly increase from 1 to 4 along with sweeping big groups from 1 (1A, 1B, 1C) to 8 (8A, 8-10B, 15-16C). In other words, nuclear skin may be thicker and thicker from left to right in the periodic table.

Evidently, there is a correspondence between the periodicity of atoms and nuclei that a nuclear axial configuration is repeated in every layer, such as the a-4443 in ¹⁹F (100%), ³⁵Cl (75.77%), ⁷⁹Br (50.69%) and ¹²⁷I (100%) in group 7A that all of their chemical main valency is -1. Perhaps, the different physical and chemical properties of A, B and C three family elements are principal rooting in that A long, B mid and C short α -clusters extend a different depths in nuclei (Fig. 2A). Such as lanthanide contraction, it may be relevant to C family 16¹p trapped in 8 C axes where is low lying between 4 A and 4 B axes, implying that nuclear radii (15) might have a link to atomic radii (16). For example, in ²³Na³⁵Cl M-N structure (fig. S119), thin a-100 in ²³Na may be looser to the ^mc than thick a-4443 in ³⁵Cl, somewhat similar to that nucleon halo, if involved nuclear force; i.e., the interstice between the ^mc and ^ms in ²⁰Ne+a-100 (²³Na) is larger than in ²⁰Ne+a-4443 (³⁵Cl), a factor possible influence on their atomic radii (0.15 and 0.09 nm, respectively). However, it seems reasonable to conclude that the short 2-3, long 4-5 and very long 6-7 periods are to originate from nuclear 4 (A), 4+4 (A+B) and 4+4+8 (A+B+C) α -clusters standing on the expanding core (the very short 1 period), respectively. The vertical 16 α -clusters in Fig. 2A show to be bound with valency neutrons that are excess neutrons in nuclei in all probability.

Composition of beta stable line

Stable nuclei. A distinction between electron and proton distributions is proton distributions accompanied with neutrons. Generally in a nucleus neutrons exceed protons in number, as the slope of beta stable line (neutron-proton ratios) plotting in a chart (17), suggesting to derive from

$$A = Z + {}^{\mathrm{p}}\mathrm{n} + {}^{\mathrm{v}}\mathrm{n}, \qquad [1]$$

where ^pn is nearly the same as proton in number and distribution, and ^vn = 2(2², 2³, 2⁴) are valency neutron holes in A, B and C steps of nuclei, respectively (Fig. 2*A*), e.g. ¹³²Xe = 54 + 54 + 2(2²+2³) in Table 1. As shown in Fig. 2*A*, a natural element (nucleus) at most comprise 40 ^vn derived from the 2-6 layers. In addition, in a nucleus ^vn approximates to α in number, e.g., ¹³²Xe = 24(^vn+ α) + 612, where 612 is its core in *zA* (*Z* = 6, *A* = 12, its structure is very different from a ¹²C); the 612 might be replaced by a 618 in ²²²Rn (Table 1), even in ³⁰²118 (*Z* = 118, fig. S118*B*), a 6n-6p-6n sandwich core.

Table 1 shows an interesting phenomenon that grown masses are an odd number in most cases, in particular, 1, 3, 1, 3, 1, 3 and 5 in the 3rd period that when grow from odd to even Z, only fill 1 ^{lp} (${}^{23}Na^{100} \rightarrow {}^{24}Mg^{1010}$) or with 4 ^{vn} (${}^{35}Cl^{4443} \rightarrow {}^{40}Ar_{v-4}{}^{4444}$), but 1 ^{lp} is often accompanied by 2 ^{pn} (${}^{28}Si^{2222} \rightarrow {}^{31}P^{4232}$) from even to odd Z. Commonly, isotopic mass change for light nuclei was ^{pn} as ${}^{16,17,18}O$ in Fig. 2*B*. For mid-heavy nuclei, e.g., ${}^{108}Sn_{v-8}{}^{2222}$ (10.5 min) and ${}^{124}Sn_{v-20}{}^{3333}$ (5.94%), both the ^{pn} and ^{vn} were varied (Table 2).

In some cases this primary neutron fit is alternative. For example, a ³⁶S (0.014%) is ³⁶S_{v-4}⁴²⁴² (fig. S16*D*) or ³⁶S_{v-2}⁴³⁴³ that how to balance ^{pn} and ^{vn}, and 20 ^{vn} of ¹²⁷I is v-0,4,8,8 (fig. S53) or v-4,4,8,4 (Table 1) in the 2-5 layers. Perhaps, in a nucleus neutron different distributions are corresponding to that nuclear isomers (18) (marked m1, m2, m3...), while its proton frame is unconcerned; e.g., ⁷⁹Br_{v-4,4} (50.69%) and ^{79m}Br_{v-0,0,8} (4.86 s), its inner 8 ^{vn} would be excited to outer (figs. S35A-B), or other distributions in the 2-4 layers. For nuclei, however, figs. S3A-18*C* indicate that: the higher is the abundance (half-life), the more concise is the structure, such as ¹²C²²²² (98.89%), ¹³C³²²² (1.11%), ¹⁴C³²³² and ²⁸Si²²²² (92.2%), ²⁹Si³²²² (4.7%), ³⁰Si³²³² (3.1%).

1 able 2. A tentative arrangements of ¹⁰⁰ ¹⁰⁰

Commonly, in these 32 neutrons (132 - 100), ^vn ~ 20, ^pn ~ 8, and 4 n in the 6th layer is neutron skin (halo), meaning that Sn isotopes might only grow ^vn from $A > {}^{120}$ Sn. All of valency neutrons in the 2-6 layers is compiled in a column.

Sn isotope	core+middle	valency neutron	axial conf	figuration	abundance% / half-life
100	76	0, 4	b-4444	a-1111	1.1 s
108	-	0, 4, 4	-	2222	10.5 min
109	-	0, 4, 4	-	3222	18.0 min
110	-	0, 4, 4	-	3232	4.0 h
111	-	0, 4, 4	-	3332	35.3 min
112	-	0, 4, 4	-	3333	0.96
113	-	0, 4, 8	-	3222	115.2 d
114	-	0, 4, 8	-	3232	0.66
115	-	0, 4, 8	-	3332	0.35
116	-	0, 4, 8	-	3333	14.3
117	-	4, 4, 8	-	3222	7.61
118	-	4, 4, 8	-	3232	24.03
119	-	4, 4, 8	-	3332	8.58
120	-	4, 4, 8	-	3333	32.85
121	-	4, 4, 8, 1	-	-	~50 yr
122	-	4, 4, 8, 2	-	-	4.7 2
123	-	4, 4, 8, 3	-	-	129 d
124	-	4, 4, 8, 4	-	-	5.94
125	-	4, 4, 8, 5	-	-	9.65 d
126	-	4, 4, 8, 6	-	-	2.3×10^5 yr
127	-	4, 4, 8, 7	-	-	2.12 h
128	-	4, 4, 8, 8	-	-	60.0 min
129	-	4, 4, 8, 8, 1	-	-	7.5 min
130	-	4, 4, 8, 8, 2	-	-	3.72 min
131	-	4, 4, 8, 8, 3	-	-	61 s
132	-	4, 4, 8, 8, 4	-	-	40.6 s

Unstable nuclei. Away from the beta stable line, ³He was estimated to generate in skin of neutron-deficient light nuclei, e.g. ¹⁹Ne_{c-2+2}⁴⁴⁴³ (17.4 s) and ¹⁷Ne_{c-2+2}⁴³³³ (0.108 s). Furthermore, a heavy nucleus lack of neutrons, ^vn, to bind its vertical A-5 α , B-3 α and C-1 α clusters may more easily cause decay (emission) (19, 20), corresponding to the fact that α -decay only happen in neutron-deficient side in a range *Z* ~ 62-82 (in mid nuclei almost never) that begin to grow C-1 α clusters, where also likely is a point of fission α -particle (Fig. 2*A*). To some extent, both α and cluster decay seem to result from the 16 α -clusters cleft into 15:1 similar to nuclear fission that only cleft ratios are different (Table 3).

In neutron-rich nuclei, neutron halo might often perch on top of axial extended lines; e.g., a structure of ${}^{31}F^{-2}1^{2}_{,4}2^{a-4443}_{,3^{a-0000,b-0000}}$ was 4 vn with other 8 n (neutron halo) to cage ${}^{19}F^{-2}1^{2}_{,2^{a-4443}}$ (figs. S9A-B). In addition, a neutron skin (halo) may happen in a stable nucleus; e.g., ${}^{136}Xe^{-616,8428,8438,168416,168516,6^{b-0000}}$ (8.9%), this 4 n (6^{b-0000}) skin will float on ${}^{132}Xe$, because its ^pn and ^vn shells are closed.

Note that, unexpectedly, all of Z, ^pn and ^vn three shells closed ($A = 2Z + ^vn$, ideal nucleus) shows not a most stable structure from two abundance (half-life) lines of ideal nuclei and their maximal abundant isotopes (Table 1) intersecting at ¹³²Xe_{v-4,4,8,8} (26.89%), i.e., for ideal nuclei, the ^vn too many in ²⁴Ne_{v-4} (3.38 min), ⁴⁴Ar_{v-4,4} (11.87 min) and ⁸⁸Kr_{v-4,4,8} (2.84 h), but too few in ²¹²Rn_{v-} 4,4,8,8,16 (23.9 min). Nevertheless, here was unable to find a better way to fit pair and valency neutrons with the line of beta stability. In the followings, it seems that released neutrons may generally come from ^vn, not ^pn in fission.

Fission configuration

As fission can direct reflect details of a nuclear structure, there is an attempt to depict it to test this nucleon distributions; at this stage that fission configuration has not been very clear a brief depiction may be more effective. Generally, nuclei can occur asymmetric and symmetric fission (21-23, 28-35). Asymmetric fission shows peaking to Z = (86+102) / 2 = 94 in thermal neutron (n,f) and spontaneous (sf) fission, where 86 and 102 are A and C closed shells, implying that it may only occur within the two shells. Its process is possible that: when a neutron land on a nuclear skin and induce the skin particles to glide, more or less, to the one side, or from side to side repeatedly, then the nucleus would be cleft, likely in that when its shape is just unfolded and thus its cleft place is a line, not a point, somewhat similar to a crystal cleft. If increase excitation energy, the particles on the skin has insufficient time to glide and the core, a relatively firm six neutron center, might be broken up, thus a symmetric fission will happen. Accordingly, the core, middle and skin of a folding nucleus may play different parts in fission.

fission	cleft ratio	Z-A distri	bution
type and path	of 16 α -clusters	A_L	A_{H}
1-1/2-9	$2^{a}2^{b}4^{c}: 2^{a}2^{b}4^{c}$ (8:8)	¹¹⁸ Pd	¹¹⁸ Pd
2-1/2-9	$2^{a}2^{b}3^{c}: 2^{a}2^{b}5^{c}(7:9)$	¹¹¹ Tc	¹²⁵ In
3-1/2-9	$2^{a}1^{b}3^{c}: 2^{a}3^{b}5^{c}$ (6:10)	⁹⁵ Rb	^{141}Cs
4-1/2-9	$2^{a}1^{b}2^{c}: 2^{a}3^{b}6^{c}(5:11)$	⁸⁸ Se	¹⁴⁸ Ce
1-1/4-8	$1^{a}2^{b}4^{c}: 3^{a}2^{b}4^{c}$ (7:9)	⁹⁴ Kr	¹⁴² Ba
2-1/4-8	$1^{a}2^{b}3^{c}: 3^{a}2^{b}5^{c}$ (6:10)	⁸⁵ As	¹⁵¹ Pr
3-1/4-8	$1^{a}1^{b}3^{c}: 3^{a}3^{b}5^{c}(5:11)$	⁶⁹ Co	¹⁶⁷ Tb
4-1/4-8	$1^{a}1^{b}2^{c}: 3^{a}3^{b}6^{c}$ (4 : 12)	⁶⁰ Cr	¹⁷⁶ Er
1-0-9	$2^{a}2^{b}4^{c}: 2^{a}2^{b}4^{c}$ (8:8)	¹¹² Tc	¹²⁴ In
2-0-9	$2^{a}2^{b}3^{c}: 2^{a}2^{b}5^{c}$ (7:9)	¹⁰⁵ Zr	¹³¹ Te
3-0-9	2ª1 ^b 3 ^c : 2ª3 ^b 5 ^c (6:10)	⁸⁹ Se	¹⁴⁷ Ce
4-0-9	$2^{a}1^{b}2^{c}: 2^{a}3^{b}6^{c}(5:11)$	⁸² Ga	¹⁵⁴ Pm
Cluster decay			
1-0-2	1^{c} : $4^{a}4^{b}7^{c}$ (1 : 15)	⁴ He	³²³ Th
2-0-3	1^{b} : $4^{a}3^{b}8^{c}$ (1 : 15)	^{12}C	²²⁴ Rn
4-0-5	1^{a} : $3^{a}4^{b}8^{c}$ (1 : 15)	²⁰ Ne	²¹⁶ Pb

Table 3. Some fission types and paths of ²³⁶U.

Fission type (different divide of core $_{6}12$ nucleons) was mainly classified into: 1/2, $_{3}6:_{3}6$; 1/4, $_{3}3:_{3}9$; 0, 0: $_{6}12$. The ^vn number error is about ± 3 in mass division, which could be related to isotopic products. Skin nucleons and valency neutrons in the scission lines have not been allotted to cluster decay.

Fragment structures. On the whole, fragment charge and mass is a statistical distribution. In Fig. 2*A* it appears to come from 16 α -clusters of a nucleus cleft into different ratios (colonies, Table 3), in which asymmetric fission is a double peak yield curve (21-23) that a contribution of skin particle glide might be involved. As a shift ratio of ^ms is a random distribution at present, while ^mc (²¹²Rn⁴⁴⁴⁴, fig. S864, ²⁰⁸Pb³³³³+4^lp, Fig. 2*A*) has several fixed divisions, a fission system (*zA* > ²¹²Rn) is resolved: X = ^mc + ^ms. Thus, a fragment often is a mixture of ^mc and ^ms. For example, in 1-1/4-8 fission of ²³⁵U+n, a structure of ²³⁵U being ⁶¹⁶, ⁸42⁸, ⁸43⁸, ¹⁶84¹⁶, ¹⁶85¹⁶, ³²₁₆6³², 7^{a-0000,b-0000,c-2222-²²²⁰, the ^mc is lined off: ²¹²Rn _{1-1/4-8} \rightarrow ₅₃129 + ₃₃83, [2] where subscript is fission path that its image is somewhat analogous to a tetrahedron cutting off an angle (1 A axis, 3:1, 3^a2^b4^c:1^a2^b4^c</sub>) nearby center, and the ^ms falls into a-000,b-00,c-2222 + a-0,b-00,c-2222 = ₃13 + ₃11, [3] a total of 24 nucleons on the skin including a thermal neutron; then add Eq. 3 to Eq. 2,}

$$^{235}\text{U}+n_{1-1/4-8} \rightarrow (_{53}129+_{3}13) + (_{33}83+_{3}11) = {}^{142}\text{Ba} + {}^{94}\text{Kr},$$
 [4]

[5] [6]

 $^{235}\text{U}+\text{n}_{1-1/4-8} \rightarrow _{53}129 + (_{33}83+_{6}24) = ^{129}\text{I} + ^{107}\text{Y},$

$$^{235}\text{U}+n_{1-1/4-8} \rightarrow (_{53}129+_{6}24) + _{33}83 = ^{153}\text{Pr} + ^{83}\text{As},$$

that is, when the $_{3}13$ glide into 1-8 sector [5] or the $_{3}11$ into 8-1 sector [6], they are 107 Y (94+13) or 153 Pr (142+11), suggesting that a fragment peak yield (28-31) may be partly contributed from the $_{3}13$ and $_{3}11$ glide. In Eq. 4 the ^ms has not glided that two fragment peak yields are near 6%.

1 ne valley is a symmetric fission: $^{235}U+n_{1-1/2-9} \rightarrow _{46}118 + _{46}118$, near 0.1% yield.

Fragment mass difference. Between two fragments, a mass difference can be setting up by a single C, B or A axis (Table 3). To an A axis, it is, e.g.,

 235 U+n $_{1-1/2-8} \rightarrow (_{53}126+_{3}13) + (_{33}86+_{3}11) = {}^{139}$ Ba + 97 Kr = ($_{46}118+_{10}21$) + ($_{46}118-_{10}21$), [8] where $_{10}21 = _{10}20 + 1$, a cluster of 5 α (A3 axis) plus 1 n at A3-7 position, when the scission line sweep anticlockwise from 1-1/2-9 [7] to 1-1/2-8.

Fragment vanishing points. Fragment yields, including of A_H, A_L (22), neutron (24, 25, 36) and α -particle (21), show to vanish in the same two points at leftmost sides of double mass peaks, ~ 130 and ~ 80, which both suggest to come from an α -cluster colony where its 2 edges are enclosed by 2 A long axes, i.e., $3^a 2^b 4^c$ (A_H, e.g. 8-1 sector) and $2^a 1^b 2^c$ (A_L, e.g. 4-9 sector) α -cluster colonies. Take neutron yields for example, in

 252 Cf $_{1-1/4-8} \rightarrow _{53}129 + (_{33}83 + _{12}40) = _{53}129 + _{45}123$,

[9]

where ${}_{53}129$ is a minimal ${}^{32b}4^{c}$ colony, and ${}_{45}123$ is a complementary colony $(1^{a}2^{b}4^{c})$ to yield maximal neutrons (~3) (24), suggesting that maximal neutron yield is from a sector of 2 C axis edges. In 254 Fm (sf) that both 252 Cf and 254 Fm neutron shells may be ${}^{p}n-114+{}^{v}n-40$ shows a similar result of neutron yield: minimum at 129-130 of A_H and maximum at 123-124 of A_L (36). In 4-0-9 fission, there is

 $^{252}Cf_{4-0-9} \rightarrow (_{56}142 + _{9}28) + (_{30}70 + _{3}12) = {}^{170}Tb + {}^{82}As,$ [10]

which is an extreme event that $2^{a}1^{b}2^{c}+_{3}12$ (⁸²As) has not held any ^vn in the scission line. In addition, it seems that a neutron deficient fragment is also nearby an enclosed $3^{a}2^{b}4^{c}$ or $2^{a}1^{b}2^{c}$ colony. For instance, in $^{238}U+^{12}C$ (35), a $^{73}{}_{33}As$ likely come from $_{30}70+_{3}3$ in 4-1/4-9 fission, where $_{30}70 = 5\alpha \times 2+3\alpha+1\alpha \times 2+10^{v}n$, a minimal $2^{a}1^{b}2^{c}$ colony, and $_{3}3$ is from core. For ^{72}As and ^{69}Zn in $^{238}U+p$ (37), they might come from other ways, such as a $1^{a}2^{b}2^{c}$ colony, because the 10 ^vn normally cannot be released inside a minimal $2^{a}1^{b}2^{c}$ colony ($^{70}{}_{30}Zn$) (35), unless one of them has become a delayed neutron (38), different from prompt neutrons (25) which might come from N1-6 and N8-6 that are ^vn not ^pn in 1-1/4-8 fission (Fig. 2A), when the fragment was reconstituted to turn into a daughter nucleus.

Skin particle glide. Comparing Eq. 5 and Eq. 6 suggests that the ^ms prefer to glide upon top of A_L ($3^{a}2^{b}4^{c}$: $1^{a}2^{b}4^{c}$ + ^ms), for the yield of Eq. 5 is usually higher than Eq. 6. This view is also illustrated by that A_H masses are nearly constant, while A_L masses increase in ²²⁹Th, ^{233,235}U, ²³⁹Pu, ²⁴⁵Cm, ²⁴⁹Cf and ²⁵⁴Es (n,f) (22). However, particles at skin may glide limited in the two closed shells that transition from asymmetry to symmetry is in two sides of Z 94±6 ($_{88}$ Ra- $_{100}$ Fm) (39), implying that its mass number is neither too many, nor too few. For example, in the light side of Z 94±6 there is ²²⁶Ra (³He, df) that simultaneously reveal asymmetric and symmetric fission (40). In the heavy side of Z 94±6, transition is in Fm isotopes (36, 39, 41-43), which from ²⁵⁴ $_{100+114+40}$ Fm^{a-0000,b-2222,c-4443-4443} (sf, asymmetric) to ²⁵⁸ $_{100+118+40}$ Fm^{a-2222,b-2222,c-4443-4443} (²⁵⁷Fm+n, symmetric), the a-0000 was replaced by the a-2222 at the top of 4 A axes (see superscript and subscript). This is likely that the closed ^pn-118 shell (see also fig. S100) fence against the skin particles to glide and drive a ²⁵⁸Fm bisected. In the case of ²⁵² $_{102+110+40}$ No^{a-0000,b-0000,c-4444-4444}, though its Z-102 shell is closed (see also fig. S102), it is an asymmetric fission (44), possible that it lack 8 ^pn than ²⁵⁸Fm in the skin to resist the ^ms glide.

Possible to result in even-Z fragment that its energy release is greater than odd (fine structure of fragment masses, interval $A \sim 5$, ^vn+ α) (28-31), part of skin particles might be fused in glide, since in ²³⁵U (n,f), its skin having no an inherent α -particle in ground state, has occurring polar α -particle emission, about 0° or 180° with respect to the fission axis (45-47); further, it is over 3 times for A_L to A_H flight directions (46), which is in favor of the ^ms to glide upon A_L also. In order to clarify whether polar emission come from the ^ms glide, if possible, could try to detect that symmetric fission; e.g. ²⁰⁸Pb (Fig. 2A) or ²⁵⁷Fm (n,f), their ^ms are almost unable to glide and estimate that polar emission is less than ²³⁵U (n,f).

1 ernary fission, a and cluster decay. Binary fission in Fig. 2A could draw a line from one side through the core to the opposite side. When draw three lines, e.g., 1-c (from N1-6 to core), 2-c and 9-c. it is a ternary 1-2-9 fission. Usually, 1-2 sector (one of 8 C axes) is a place of light charged particle (LCP) emission (21, 26, 27, 48, 49) and can partition it into three points: C1-7 (p, d, t, α), C1-6 (α) and C1-6+C1-7 (^{7,8,9}Li, ^{9,10}Be). Among them most probable emission is α and its angle to differ from polar emission is perpendicular to the fission axis, nearly 90±22.5° to A_H and A_L, respectively, where 22.5° = 360° /16. Since a fission nucleus is almost impossibly complete unfolded, its α emitting angle is within 67.5° -112.5°, which is in good agreement with results reported (26, 27) that is difficult to explain up to now, to the author's knowledge. In addition, it is possible that LCP, just in the two ends of a scission line, could serve as a probe to identify axial configurations on the condition that the ^ms has not glided or in glide zA of some LCP have not varied. LCP emission probabilities in per 10³ sf of ²⁵²Cf^{a-0000,b-2222,c-4343-4343} are: α -3.3, t-0.2, d-0.02 and p-0.06, respectively (48), slightly less than ideal that its axial configuration has 4 α and 4 t, no d and p.

If the track varied from 1-2-9 to 1-4-8 randomly, it is a three large fragment fission (50, 51), in which 235 U (n,f), three sectors are about 1-4 (35 Mg), 4-8 (56 Sc) and 8-1 (145 Pr). It thus suggests that lighter fragments, including α and cluster decay, may come from a different combination of A, B and C axes, e.g., 1^c, $A \sim 10$, LCP; 1^b2^c, $A \sim 30$, ²⁸Mg (51); 1^a1^b1^c, $A \sim 50$, ⁴⁷Ca, ⁴⁸Sc (37). From here, it is tempting to expect that a nucleus might deeply be fragmented into four large fragments; e.g., in a quaternary 1-3-9-14 fission of ²³⁵U+n, four sectors are 1-3 (²⁵F), 3-9 (⁹³Rb), 9-14 (⁶⁹Co) and 14-1 (⁴⁹K). However, four coincident fragment angular, energy and mass as a function of excitation energy in a quaternary fission would shed interesting light upon that whether a nuclear shape is from folded (tetrahedral 4 A axes) to unfolded (plane 4 A axes) in fission.

Discussion

In this paper, it has described that folding nuclei can grow vertical 16 α -clusters bound with valency neutrons to stand the core. In fission, some of vn in a scission line will convert into unbound neutrons and lead to the 16 α -clusters of different length cleft, like a molecular bond broken, into symmetric 8:8, or asymmetric 9:7 (Fig. 2A) that its a pair fragment mass difference is about 5 $\alpha \times 2 = {}_{20}40 \ (\pm 1 \text{ A axis}, 2 \text{ clusters of 5 } \alpha)$, which is well consistent with data obtained (21-23), for a typical example, ${}^{235}\text{U}+n \rightarrow {}^{137}\text{Ba}+{}^{97}\text{Kr}+2n$.

To explain why two coincident fragments have a characteristic mass difference is a central problem in nuclear fission testing nuclear models. Sometimes, a heavy fragment was explained near Z-50, N-82 doubly magic shells (41) and a light fragment near N-50 shell (21), which might be less straightforward than this explanation that the fragment mass difference results from ± 1 A axis. However, It seems that there has no a clear distinction between proton and neutron shells. Likely, their shells are the same only in magic number 2 (⁴He), 8 (¹⁶O) and 20 (⁴⁰Ca) (Table 1); for the 28, 50 and 82, it is to differ because of valence neutron emergence, suggesting that proton shells are constant, whereas neutron shells are not in a number. For example, Z-50 shell is all along fixed, when isotope mass increasing from ${}^{100}_{50+46+4}$ Sn to ${}^{132}_{50+54+28}$ Sn (Table 2), while N-50 shell has largely been changed that the ^vn are gradually turned into the ^pn to pair increased ¹p, when isotone charge increasing from ${}^{82}_{32+36+14}$ Ge to ${}^{100}_{50+46+4}$ Sn. However, it is likely different that *N*-50 shell in ⁸⁸Sr (⁸⁴Kr+a-1010) from ⁸⁹Y (⁸⁴Kr+b-111<u>2</u>), *N*-82 shell in ¹³⁸Ba from ¹³⁹La, ¹⁴⁰Ce, ¹⁴¹Pr and ¹⁴²Nd (Table 1). For that *N*-126 shell, in Fig. 2A shows to derive from ^pn-86+^vn-40, a frame to grow ${}^{208}_{82+86+40}$ Pb 3333 , 209 Bi 4333 , 210 Po 4343 , 211 At 4443 and ${}^{212}_{86+86+40}$ Rn 4444 . Though only Z-2, N-2, Z-28 and N-126 shells are closed here, magic nuclei are encouragingly displayed a definite images (Figs. 2A-B), and, it is undeniable that all the magic nuclei together with their neighbor nuclei able to grow is so smooth in Table 1, which would be helpful to account for magic number phenomena in future.

On the other hand, it is possible that a nucleus might emerge different structures in ground and excited states. For example, a ¹⁶O in ground state is ¹⁶O_{c-2+2}⁴²⁴² and in excited state is 4 α -structure that skin 2 d of ¹⁶O_{c-2+2}⁴²⁴² were combined 1 α ; otherwise in ground state how a 4 α -structure of ¹⁶O can carry two hydrogen atoms to build a water molecule? Whereas a ¹H₂¹⁶O M-N structure

will nave understood at a glance in Fig. 2*B*, suggesting that the action of a nucleus in a molecule might have to take into account (52). However, a compelling relation between light nuclei and simple molecules may has been carried out in this work, which is an integrative result of Lewis dot structure (3) (Table 1 and figs. S119-125) and alpha-particle model (9) plus valency neutrons to some extent. Moreover, the scope of atoms was broadened to proton distributions in addition to electron distributions, though not to incorporate them. As the proton is predominant in an atom in a sense, the element properties may essentially result from the proton, *Z*, the atomic number, which is shown to occupy constant spatial positions not only in nuclei, but also in molecules as well (Fig. 2*B*), implying that the periodic law interpreted by proton might be less complicated, that is, the proton arrangements in nuclei are likely corresponding to the element arrangements in the periodic table (Figs. 1*A*-*C*).

Accordingly, the periodic law seems to dominate structures of not only molecules and atoms, but nuclei. Further, this folding nuclear structure appears to be feasible in depicted typical fission phenomena. Namely, a nuclear fission is likely that its 16 α -clusters are cleft into different ratios from 15:1 (1^c, α decay; 1^b or 1^a, cluster decay, essentially similar to three large fragments in a fission) to 8:8 that a most probable ratio is 9:7 $(3^{a}2^{b}4^{c}:1^{a}2^{b}4^{c})$ and its charge and mass difference is about $_{ZA} \ge _{20}40$ in two coincident fragments, released prompt neutrons are only valency neutrons from a scission line (Fig. 2A), α -particle angle within 90 \pm 22.5° to its fission axis is emitting from one of C family 8 α -particles, and, all fragment, neutron and α -particle yields suggests to vanish in the same two points: $3^{a}2^{b}4^{c}$ (A_H ~ 130) and $2^{a}1^{b}2^{c}$ (A_L ~ 80) α -cluster colonies (21-27). Taken together, however, there is a possibility that the periodic table form is rooting in nuclei that can only grow 1 α-particle in the 1st layer (core) intensified by 4^lp of ₂₇Co, ₂₈Ni, ₄₅Rh and ₄₆Pd, and then grow 2^3 , 2^4 and $2^5 \alpha$ -particles together with nearly same number valency neutrons in the 2-3, 4-5 and 6-7 layers, respectively (Fig. 2A), which is in general agreement with the line of beta stability. Consequently, the atomic number of the 1-7 periods is derived from Z 118 = $2(1+2^3+2^4+2^5) + 4$ that noble nuclei are able to demonstrate perfect nucleon distributions (Table 1) in three dimensions. Furthermore, the number of the elements $[2(2^3+2^4+2^5))$ besides groups 9-10B] and valency neutrons $(2^3+2^4+2^5)$ in the A, B and C steps being a square relation therefore indicates that an element nucleus unusually is a two dimensional structure in a nuclear phase, a folding nucleon plate; so that gives rise to it, a crude nucleon aspect was suggested from macrocosm. However, though here is an empirical distribution of nucleons, it provides a promising primary way that may be beneficial to further clarify and/or integrate nuclear, atomic and molecular structures.

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Supporting figs. S1-125 (35 MB)

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