

Challenging the Orbital Model



of
The Atom

by
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E2M considers that energy sources consist of concentrated energy in a toroidal form, flowing around the major radius (R or r_e in figure 1) with a flow component in the minor radius (r) direction to create a swirling, twisting flow pattern. Paralleling the energy flow within the torus is less dense field of electromagnetic energy field around the torus and spiralling through the central hole of the torus to generate what we identify as electric and magnetic fields.

The up/down component of the external electromagnetic field provides the field's North/South magnetic orientation, with the North Pole defined by the **divergent** energy flow and the South by the **convergent** flow. The circular flow component of the electromagnetic field around the North/South axis provides the electrical characteristics of the energy source: the convention assumed is **clockwise** for the **positive** pole and **anti-clockwise** for the **negative** when looking from outside towards the torus along its North-South axis.

The electromagnetic field orientation of the electromagnetic energy defines the chirality (or helicity) of the tori. The convention adopted is that a N- (North-Negative) pole or a corresponding S+ (South-Positive) pole defines an **electron**, and that a N+ or S- pole an **positron**, as shown in figure 2.

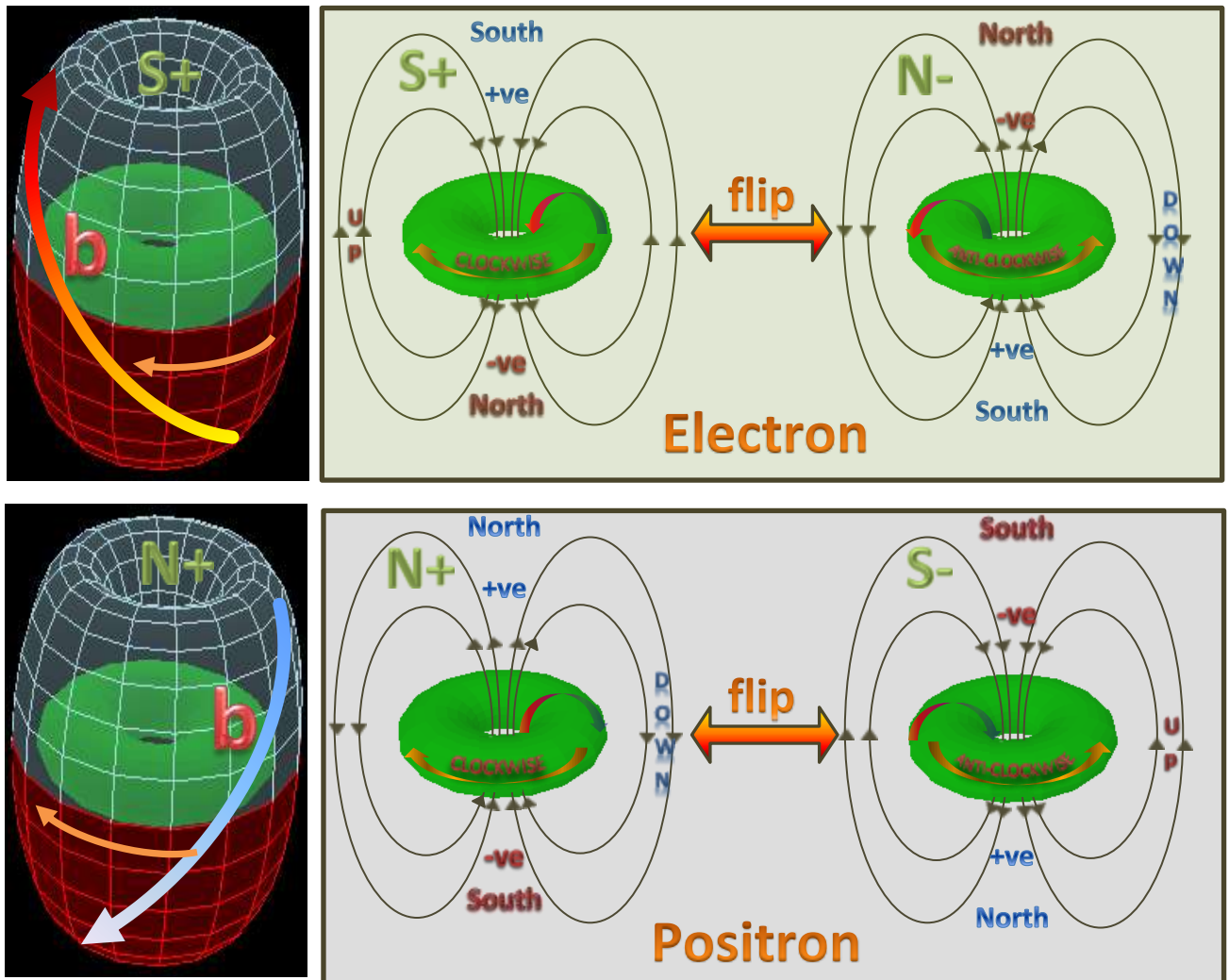


Figure 2: Electron and Positron Electromagnetic Field Patterns

As in figure 2, it is represented by a green torus, with the electromagnetic field shown as a nominal encompassing to torus, with its positive electric field side indicated by **blue** highlighting and the negative side by **red**.

Concentrated Energy Sources

Concentrated Energy Sources (CESs) are considered to be a primary energy source that represents the most fundamental energy source from which matter is derived: they are the E2M equivalent of [Preons](#). As for electrons and positrons, CESs are considered to consist of a central torus of concentrated energy surrounded by a swirling electromagnetic field. As for electrons and positrons, CESs can have different chirality. They are represented by a yellow torus (see figure 3) to distinguish them from electrons and positrons (green torus).

The rest mass of a CES is in the order of $52 \text{ MeV}/c^2$, which in energy terms means that it contains about 100 times more energy than an electron (or positron), which has a rest mass of $0.511 \text{ MeV}/c^2$.

CESs have the additional ability of being able to capture and absorb extra electromagnetic energy from other electromagnetic sources, thus increasing their net energy, or to lose some energy to their environment. Both processes are ongoing and in parallel, providing CESs with responsive energy transfer and balancing characteristics.

Unlike CESs, the electromagnetic energy level of electrons and photons remains relatively constant until it is absorbed by another structure or field. E2M also addresses the possibility that electrons and positrons are derived or secondary energy sources that are created by or derived from CESs.

Another important feature of CESs is their ability to group together to form **up** and **down quarks**, which in turn form into the **nucleons** (i.e. protons and neutrons), which build into normal matter.

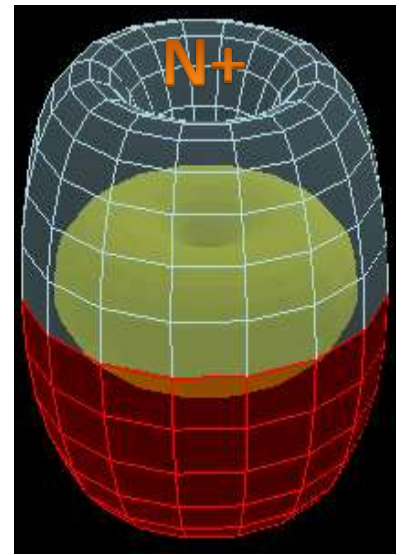


Figure 3: CES

Quarks

E2M considers that each up and down quark consists of a three-dimensional array of 6 CESs, held in place by their respective electromagnetic fields, as shown in the figure 4 Note that...

- The green cube is purely to highlight the spatial position of the CESs forming the quark structure.
- For a given electromagnetic polarity the rotational direction around the longitudinal axis can be clockwise or anti-clockwise. Where possible, it would be reasonable to assume that most pairs of CESs sharing the same longitudinal axis within a quark would both have the same rotational direction (e.g. a N- CES would most likely match with a S+ or a N+ CES).

An **up quark** consists of five outwardly pointing positive polarity CESs and one negative polarity CES. The electric charge corresponding to each pole can be considered to equate to a point charge of $1/6 e$ (elementary charge) located at the centre of a quark (i.e. at the centre of its green-cube geometry); thus the net electric charge of an up quark is $+5/6 - 1/6 = +2/3 e$.

Similarly a **down quark**, consisting of two outwardly pointing positive polarity CESs and four negative polarity CESs, resulting in a net electric charge of $-1/3 e$. Thus the electric charge estimates derived from the E2M model correspond with experimentally determined electric charge values of up and down quarks.

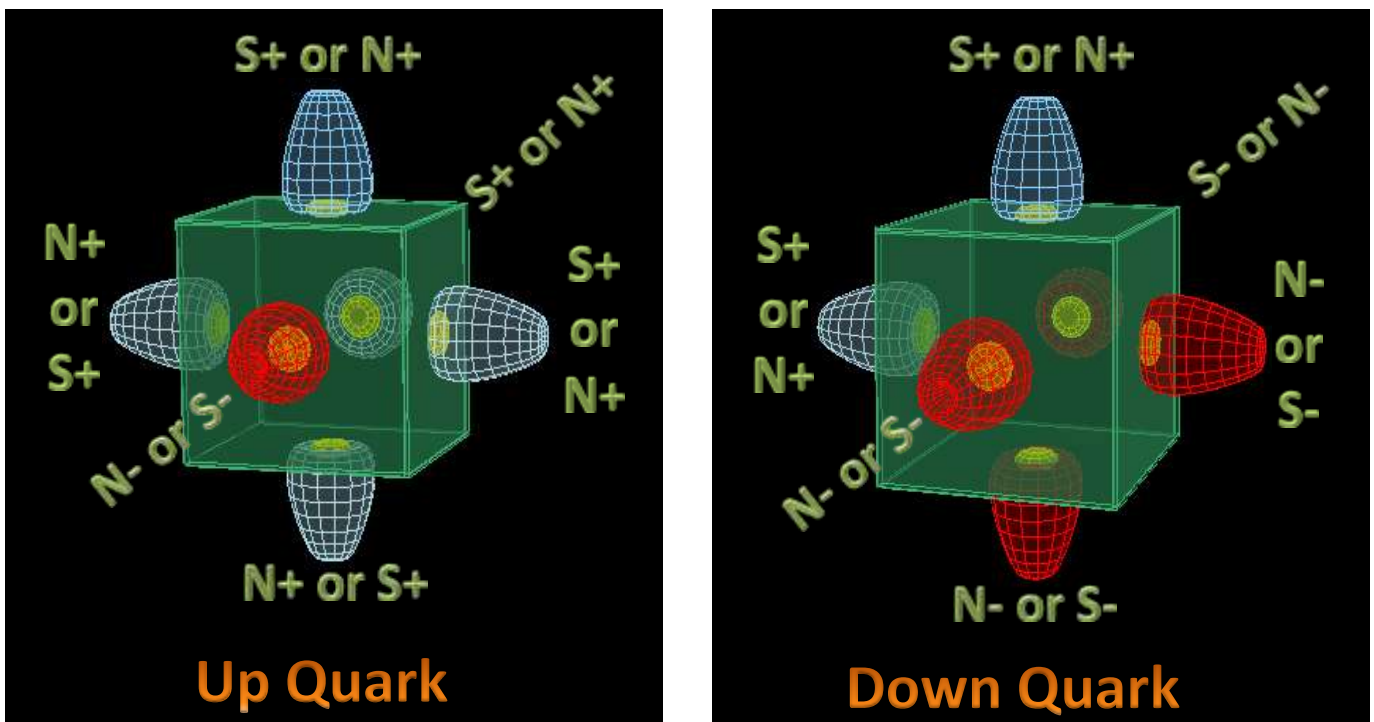


Figure 4: Up and Down Quark Structure and Electromagnetic Pole Patterns

The Nucleons

A **Proton** consists of two up quarks and one down quark, and a **Neutron** consisting of two down quarks and one up quark. Both nucleons have an 'L' shaped form, as shown in figure 5 ('l' form nucleons will be discussed shortly).

Note that the strong force joins between the up and down quarks provide for the 2-way movement of electromagnetic energy between quarks, thus providing an energy balancing mechanism between nucleon quarks.

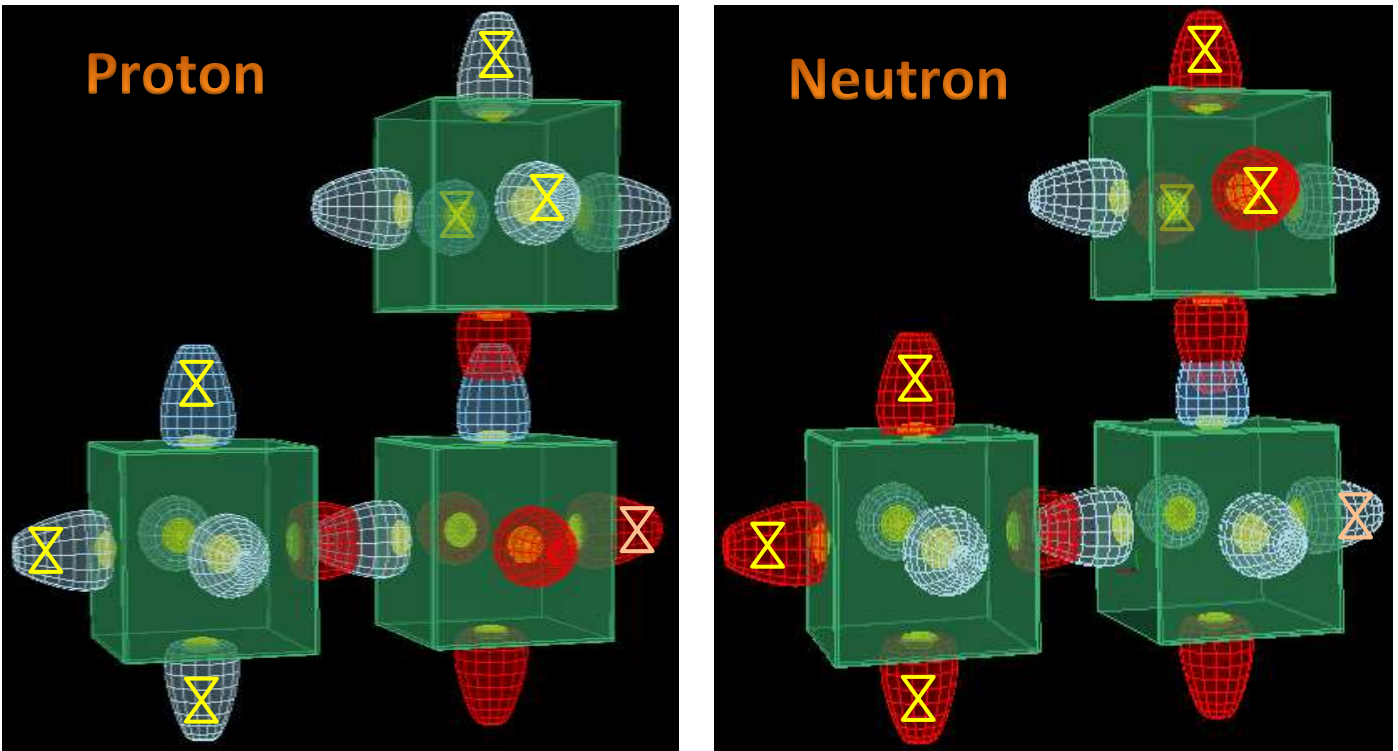


Figure 5: 'L' Form Proton and Neutron Structure

Protons consist of two up quarks ($+2/3 e$) and one down quark ($-1/3 e$), resulting in a net electric charge of $+1e$, whereas the Neutron, consisting of two down quarks and one up quark, is electrically neutral.

An important characteristic of the nucleon structures relates to those CESs annotated with a yellow hourglass symbol in figure 5 act as **flip triggers**: should any one of these 6 flip trigger tori be flipped (from a positive outward electric field to a negative one or vice versa), they cause the other five, plus the one within the central quark annotated with an orange hourglass, to flip automatically. CES torus flipping provides nucleons with the ability to instantaneously convert from a proton into a neutron or vice versa: this capability is very important for atom-building and transformation processes.

Electron-Positron Annihilation

Electron–positron annihilation occurs when an electron (e^-) and a positron (e^+), the electron's antiparticle, collide. In a majority of cases, the result of the collision is the annihilation of the electron and positron, and the creation of gamma ray photons: $e^- + e^+ \rightarrow \gamma + \gamma$. The Wikipedia diagram for the annihilation is shown in the top of figure 6 (with a Feynman diagram insert). Each electron, positron and gamma ray photon represent an energy of $0.511 \text{ MeV}/c^2$.

The E2M explanation for electron–positron annihilation is shown in the bottom part of figure 6: as the electron and positron approach each other, their magnetic fields attract causing them to accelerate towards each other. As they get really close to each other, almost touching, their electric field repulsion intensity increases dramatically and the two particle explosively separate pushing each other away in opposite directions. Comparable with two solid incompressible objects in a head-on collision, the tori fly off in opposite directions (i.e. 180° to each other) at close to the speed of light and start free-spinning at a frequency corresponding to that of **Gamma (γ) radiation**.

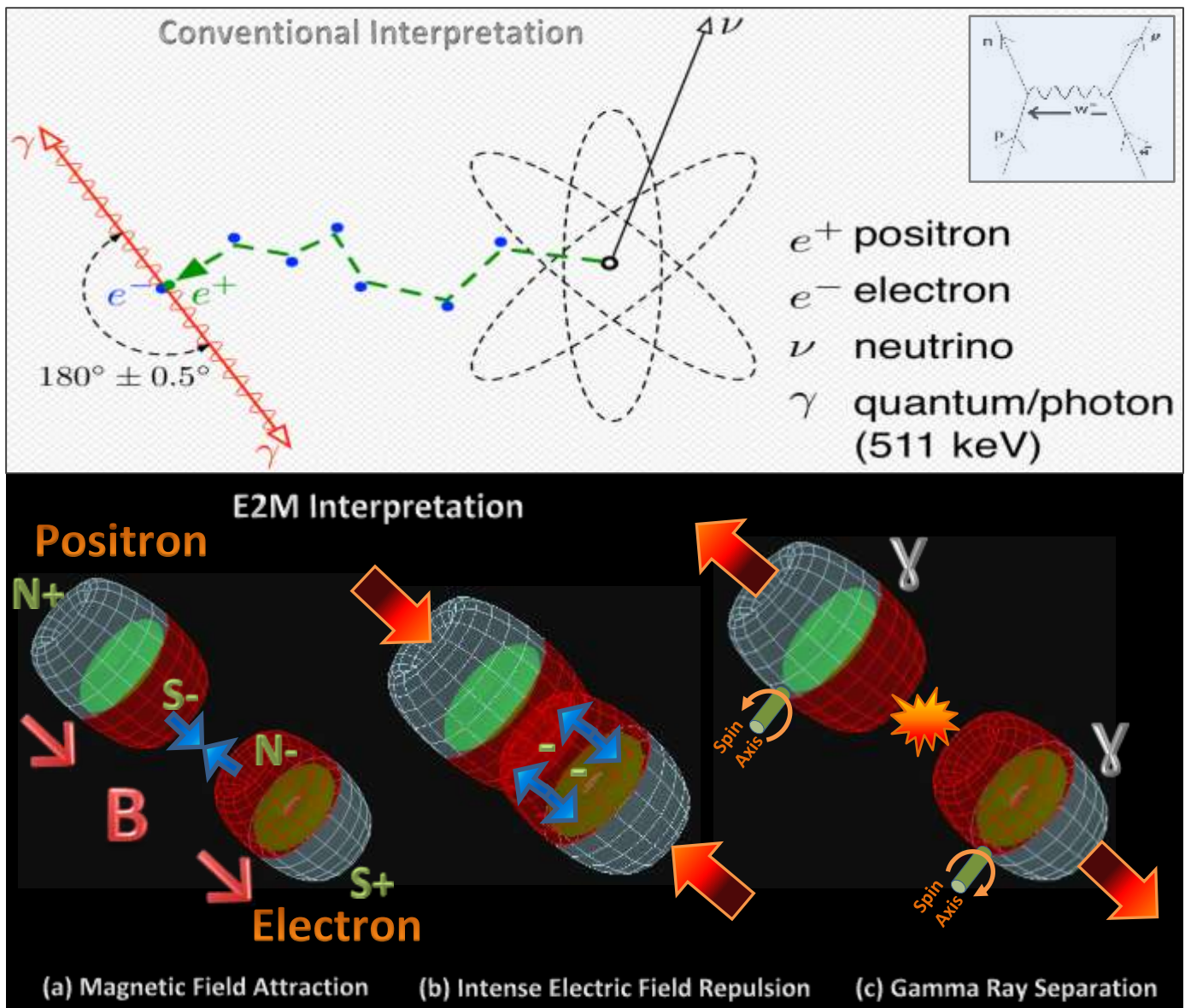


Figure 6: Electron-Positron Annihilation

Beta Decay and Electron Capture

Beta decay (β decay) is a type of 'weak reaction' radioactive decay in which a **beta ray** (a fast energetic electron or positron) and a neutrino are emitted from an atomic nucleus. It is quite prevalent across many elements of the Periodic Table as shown in figure 7. Stable isotopes are shown as the tenuous black line that roughly demarks the two types of decay.

Beta minus (or β^-) decay is when a neutron converts into a proton, increasing the atomic number of the atom concerned by 1; and **beta plus** (or β^+) decay is when a proton converts into a neutron, decreasing the atomic number by 1.

β^- decay is triggered when an electron internal to the atom becomes unstable and accelerates towards one of the neutron's negative flip-triggers causing it to flip. The flip causes the neutron's other 6 flip CESs to flip simultaneously, converting the neutron into a proton. The flipped CES causes the electron involved to be repelled at speed away from the atom as an **electron beta ray**.

A small amount of concentrated energy is released as an anti- neutrino (Note that a **neutrino** is considered to be a low-speed torus with a rest mass of about $0.12 \times 10^{-6} \text{ MeV}/c^2$ with the same chirality as an electron; an **anti-neutrino** has the same chirality as a positron).

Figure 8 (a to c) shows the mechanics of β^- decay: the strong attraction of opposite magnetic and electric poles causes the acceleration of the electron towards the trigger CES, compressing and condensing the electromagnetic energy between them, forming into a neutrino torus (brown torus of 8b). Pressure from the rapidly approaching electron causes the trigger CES to flip, thus reversing its magnetic field. Neither the neutrino nor the electron are flipped, but the magnetic field associated with the newly formed neutrino is reversed (changing from $+B+b$ to $-B+b$), transforming it instantly into an anti-neutrino energy pattern. The energy exchange associated with the torus flip causes the explosive and rapid ejection of the electron and anti-neutrino; the magnetic repulsion between the electron and anti-neutrino ensures that they take separate exit paths.

β^+ is essentially the reverse of β^- decay, being triggered when an positron internal to an atom gets energised and causes one of a proton's flip-trigger CESs to flip, turning the proton into a neutron, and the release of a positron beta ray and a neutrino. Note that 'a positron internal to an atom' is not a possibility for conventional theoretical Physics, but is fully feasible and expected for E2M, as will be explained shortly.

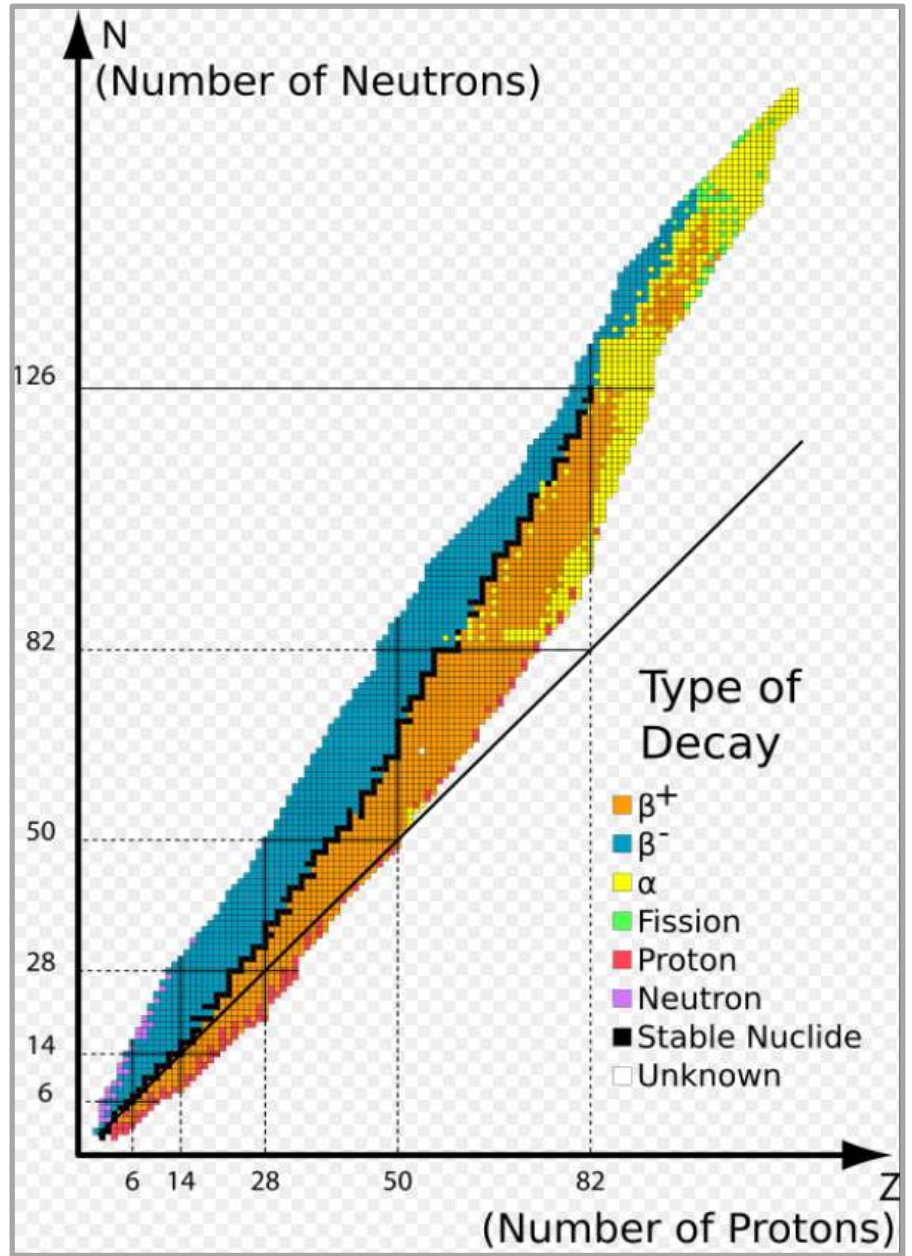
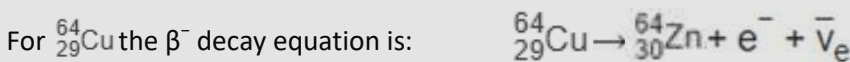


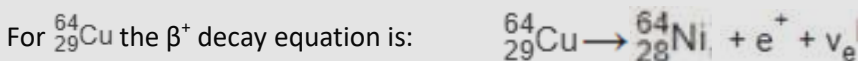
Figure 7: Plot of Beta Decay by Atomic Mass

Some examples of Beta decay are:

For β^- decay, a neutron is converted to a proton resulting in an electron and an electron anti-neutrino.



For β^+ decay, a proton is converted to a neutron resulting in a positron and an electron neutrino.



Note that ${}_{29}^{64}\text{Cu}$ has a half-life of 12.7 hours, undergoing proton decay to ${}_{28}^{64}\text{Ni}$, 61% of the time (18% by beta plus decay and 43% by electron capture decay), and neutron decay by beta minus decay to ${}_{30}^{64}\text{Zn}$, 39% of the time.

Electron capture is very similar to β^+ decay except that the source of the positron that triggers the decay is a free positron rather than from an atom's structure. The E2M electron capture equation is: $P + e^+ \rightarrow N + \nu_e + e^+$

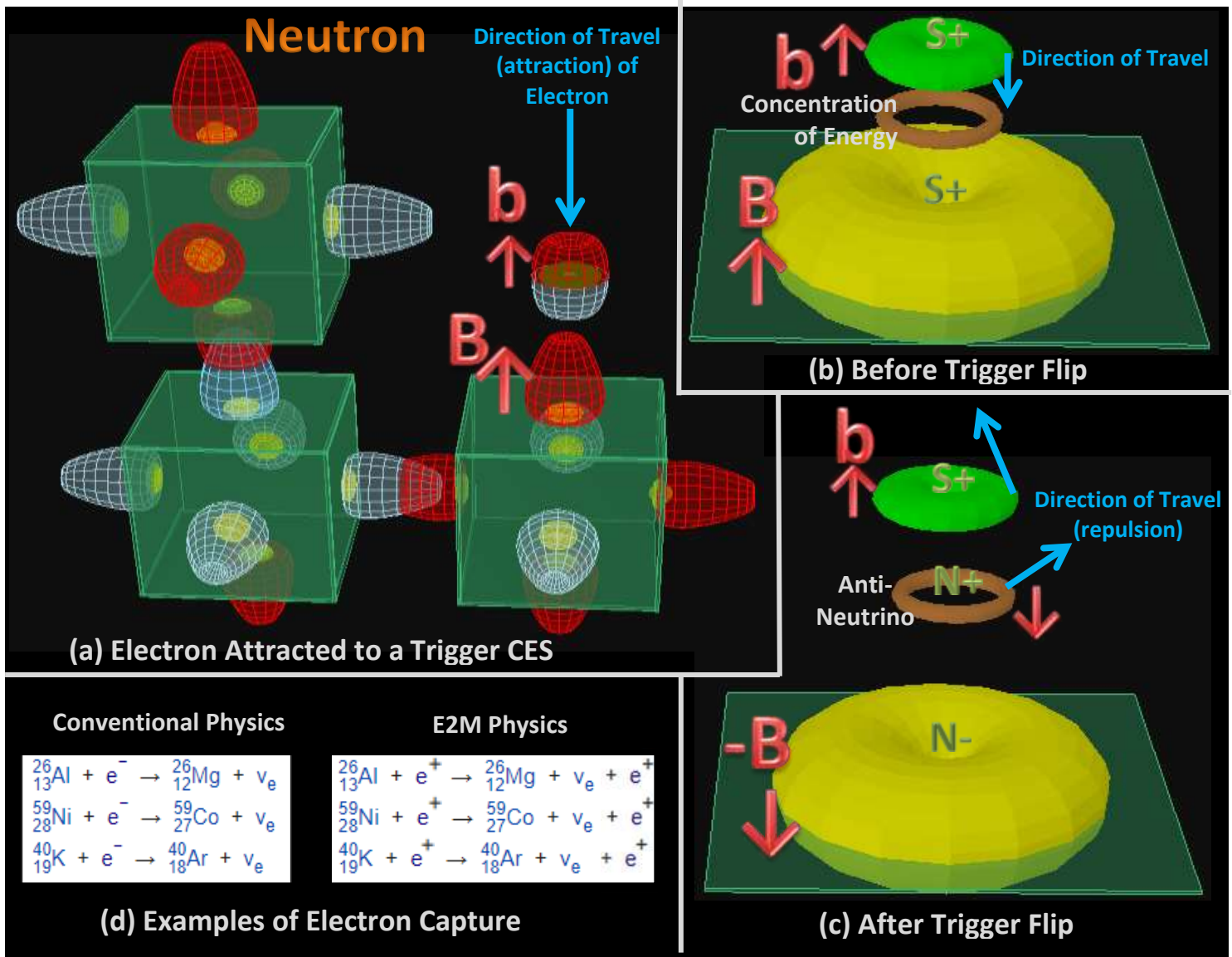
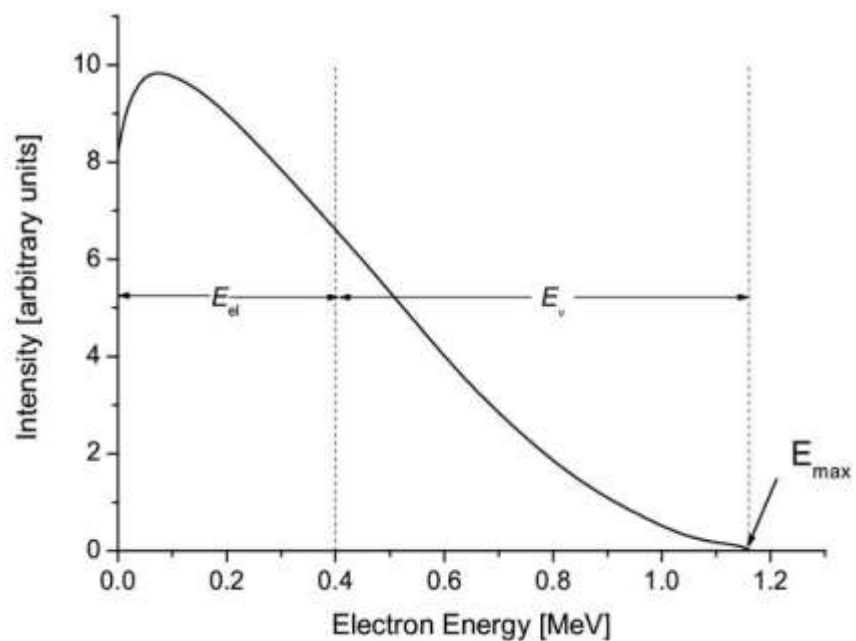


Figure 8: Beta Minus Decay and Examples of Electron Capture

For both β^+ and β^- decay the energy exchange associated with the tori flip results in the creation of an electron neutrino and anti-neutrino respectively, resulting in the energy distribution shown right. For electron capture, however, there is less momentum related energy redistribution which results in a more discrete energy profile.

The equations used to represent Beta decay and electron capture are quite deceptive because the electron being captured is shown on the left-hand side of the electron capture equation, but omitted in the beta decay equations – possibly because it is sourced from the atom itself. The conventional Physics approach involves the total exclusion of positrons from an atom's structure and from electric currents: E2M is not restricted by such constraints.



Electron and Positron Generation

As described in Appendix B, within a wire conductor electrically aligned electrons and positrons shuffle past each other, presenting as an electric current and creating a circular electromagnetic field around the wire. Although concentration of electrons (or, more accurately, the ratio of electrons to positrons) builds up on the positive side of a resistor and positrons on the negative, it is virtually impossible to distinguish between electrons and positrons within such a wire conductor, let alone separate or isolate them.

Electrons are most commonly sourced from thermionic cathode ray tubes (e.g. first generation TV tubes): those used in research settings are usually called **electron guns**. As for electric current, E2M contends that the electrons emitted from electron guns are a combination of electrons and positrons.

Low levels of positrons can be sourced from β^+ decay and Electron Capture. One practical application is **Positron Emission Tomography (PET)** which is a gamma imaging technique using radionuclide tracers that generate positrons which trigger electron-positron annihilation inside a patient's body: the resulting pair of gamma ray travel in opposite directions, and are detected allowing the tracer locations to be accurately mapped producing a high resolution image.

In order to generate larger groupings of positrons, the 'electron' beam from an 'electron' gun is concentrated and accelerated to increase the electrons' net energy; they are then used to bombard a metal film to release positrons. At CERN, the **LIL (Large Scale Electron-Positron Collider Injector Linac)** uses an 'electron' gun to assemble electrons with an energy of 80 keV; these are in turn accelerated to an energy of around 200 MeV which are then shot at a tungsten target to produce positrons that can be magnetically separated and further accelerated for particle collision purposes.

On a smaller scale, in their 2013 paper titled '[Table-Top Laser-Based Source of Femtosecond, Collimated, Ultra-relativistic Positron Beams](#)', by G. Sarri et. al. report the generation of a positron beam using a laser-driven particle acceleration setup (see figure 9). A petawatt (10^{15} W) laser was fired at a sample of inert helium gas, creating a stream of electrons (and, most likely, positrons) moving at very high speed, which were directed at a very thin sheet of metal foil: the resulting collisions produced a stream of electron and positron (and gamma ray) emissions which could be separated using magnets.

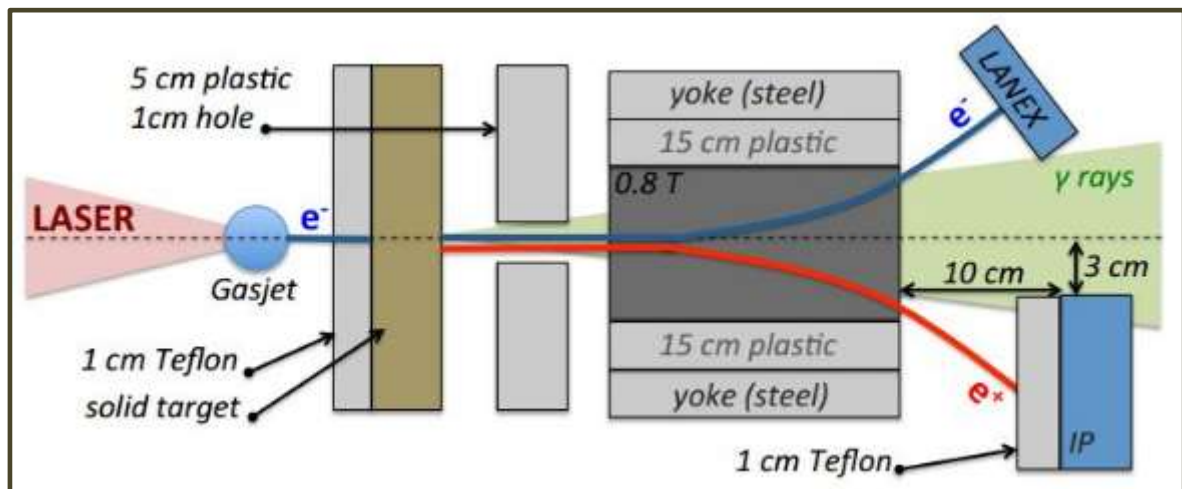


Figure 9: Laser Generated Electron and Positron Generation Setup

There would seem to be a great deal of commonality between both these examples of larger scale positron generation using energised 'electrons' and **electron-positron pair production** using high energy EMR (principally gamma radiation): both involve the bombardment of a metal target by highly energised particles. For the 'electron' gun approach (i.e. CERN and the table-top examples) the energy levels of the 'electrons' are significantly higher than that of the photons of the gamma rays, resulting in many more positrons being released. For EMR of even less energy, such as EMR in the visible light range, electron-positron pair production is called the **photo-electric** effect.

On the speculative side, it would seem that the mechanics of electron-positron pair production is analogous to that of electron-positron annihilation, except that resulting electrons and positrons are more restrained, resulting in particle velocities less than that of gamma radiation (i.e. the speed of light), and without the associated particle spin.

Atomic Structure of the Elements of the Periodic Table

As explained earlier, protons (U-D-U) consist of 2 up quarks and 1 down quark (a total of 18 CESs). Neutrons (D-U-D) consist of 2 down quarks and 1 up quark (figure 10).

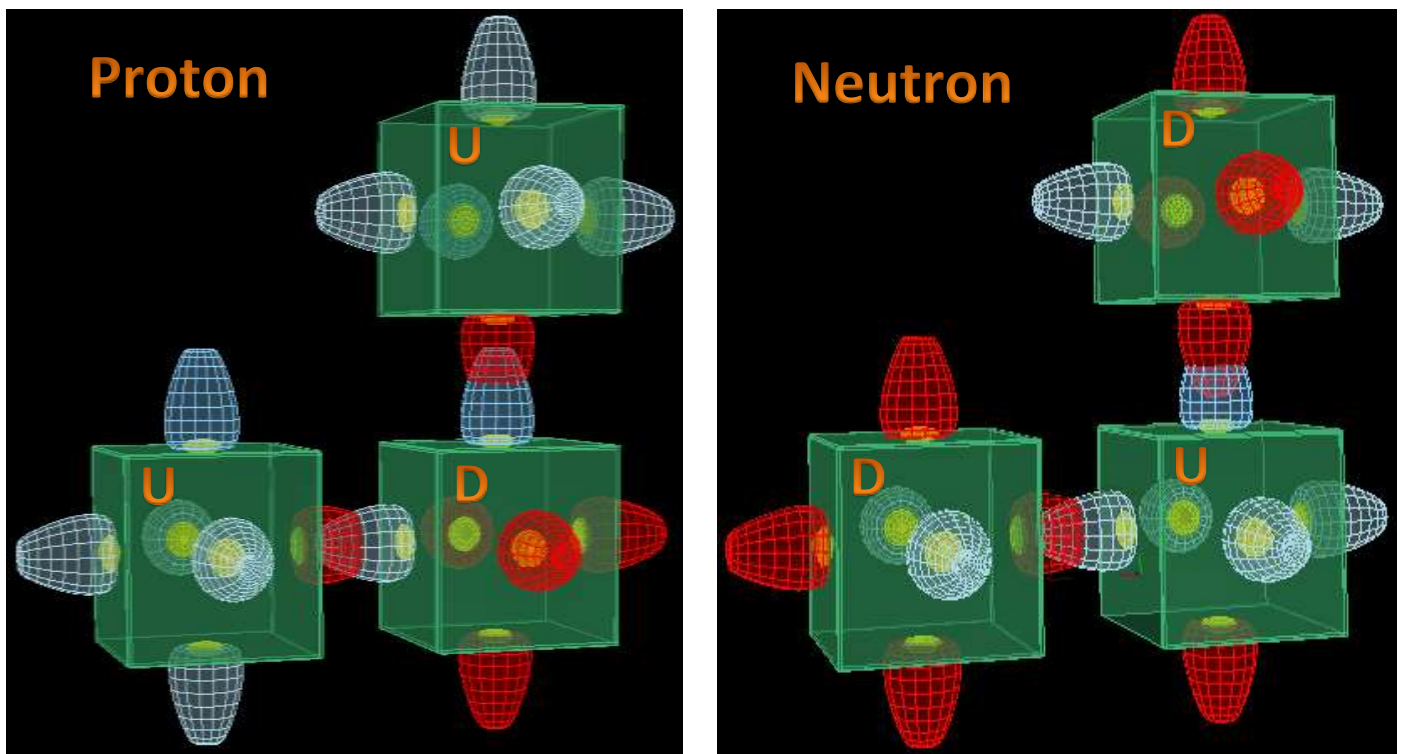


Figure 10: 'L' Form Proton and Neutron Structure

Hydrogen is the first element in the Periodic Table having an atomic number of 1. Its nucleus consists of a single proton. The **Standard Model** (SM) considers that, in its diatomic H_2 molecular form, it covalently shares two electrons.

The E2M view is that the H_2 molecule consists of two protons that are bonded internally by a pair of electron bonds. Figure 11 shows two configurations: one with the free up quarks on opposite side of the electron bonds, and the other with them both on the same side: both configurations are equally likely. Note also that the top and bottom up quarks are free to rotate around their vertical axis (as shown) and, as they are unrestrained, most likely do so.

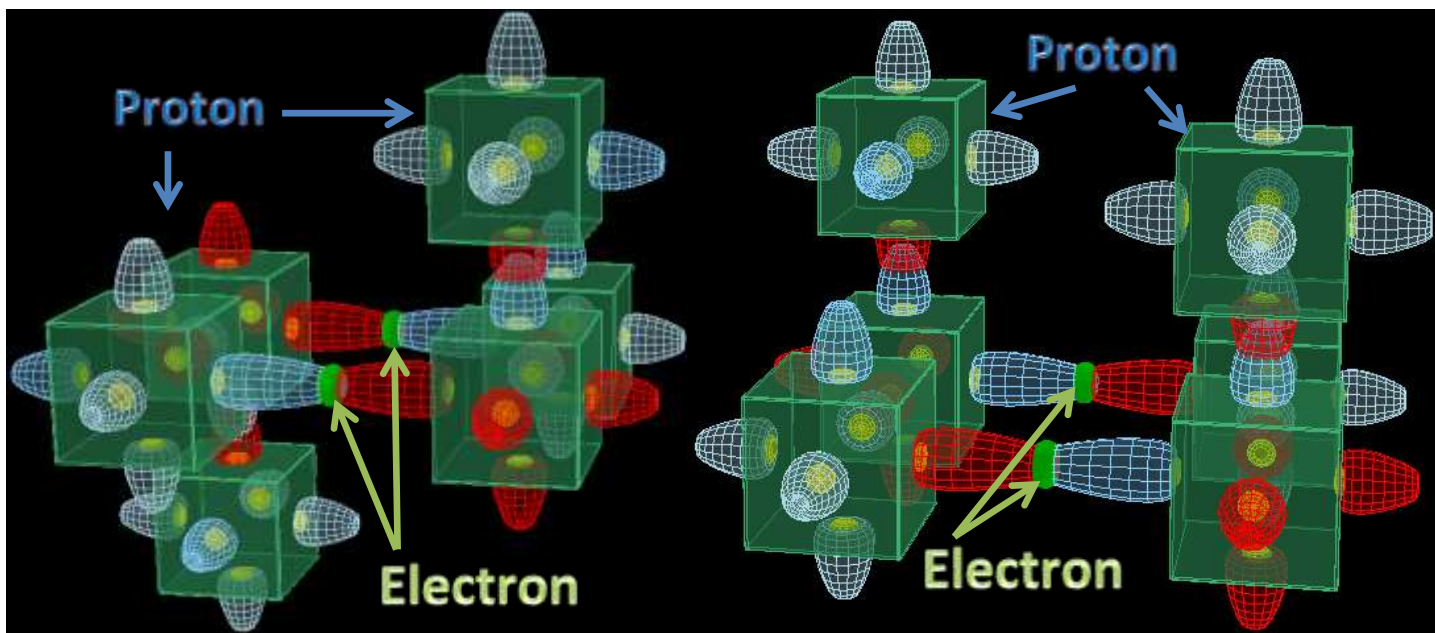


Figure 11: Models of Diatomic Hydrogen Molecule

The most striking feature of the E2M model for the H_2 molecule is that the electrons are part of the internal structure holding the protons together rather than being in a covalent 's' orbital around the protons.

The E2M view is that most of an atom's electrons are held internally, bonded between a pair of oppositely charged CESs, forming an integral part of the structure of the nucleus and adding strength to it. Some of an atom's electrons are external to the nucleus bonding atoms together to form molecules, ions and a wide range of chemical compounds. The internally bound electrons are considered to be an integral part of the nucleus's structure, whereas the outward facing bond-hook electrons can be gained, lost and/or exchanged dynamically and dictate valency options and bonding patterns.

The SM view is that electrons are external to the nucleus, orbiting at the speed of light around the nucleus in mathematically defined 'spdf' orbital patterns as point charge particles or a wave-like form. A major problem for this model is that no electrons have ever been located in the claimed orbital zones (not even for larger atoms supporting 100's of orbital electrons). Quantum Mechanics formalises the inability to confirm the physical presence of electrons in orbital zones by invoking **Heisenberg's Uncertainty Principle**, which essentially means that any attempt to observe them in a particular location (e.g. in an orbital) lessens the chance of locating them, let alone determining their momentum and related characteristics. This uncertainty principle is in effect a version of the statistical 'accuracy' versus 'precision' argument extended to the height of absurdity: even an imprecise observation of an orbital electron would be a plus to adding credibility to the stochastic model.

The other argument justifying the lack of physical evidence of orbital electrons is the **Observer Effect**, wherein the act of observing such electrons is considered to change their behaviour, presumably from particle to wave or vice versa.

The Observer Effect and Uncertainty Principle, in isolation or in combination, represent pretty unconvincing explanations for the lack of physical evidence relating to orbital electrons. For E2M electron orbitals around the nucleus are not needed, whereas the SM needs them to balance the charge of a positively charged nucleus.

As can be seen from the simplest of atomic structures such as for H₂ of figure 11, with only 2 protons (consisting of 36 CESs and their associated electromagnetic fields) 3D models quickly become quite complex: this worsens considerably with increasing atomic number. Fortunately, within the nucleus 'L' shaped nucleons combine, interlocking as shown in figure 12 to form layers of 'I' form protons and neutrons: the upper layer has the U-D-U (Up-Down-Up) quark pattern of protons, and the lower layer as shown has the D-U-D pattern of neutrons. Also note that whereas the nucleon layering is shown as a chain in figure 3, within the nucleus the layers form into closed polygonal forms (eg. figure 4).

Thus, rather than the complicated geometries of interlocking 'L' form nucleons, the structure within the nucleus can be represented more readily by proton and neutron layers consisting of overlapping bar-shaped 'I' form nucleons. Using a 'bar' notation to represent 'I' form nucleons the 3D models of the nucleus are simplified, making them more manageable and easier to understand.

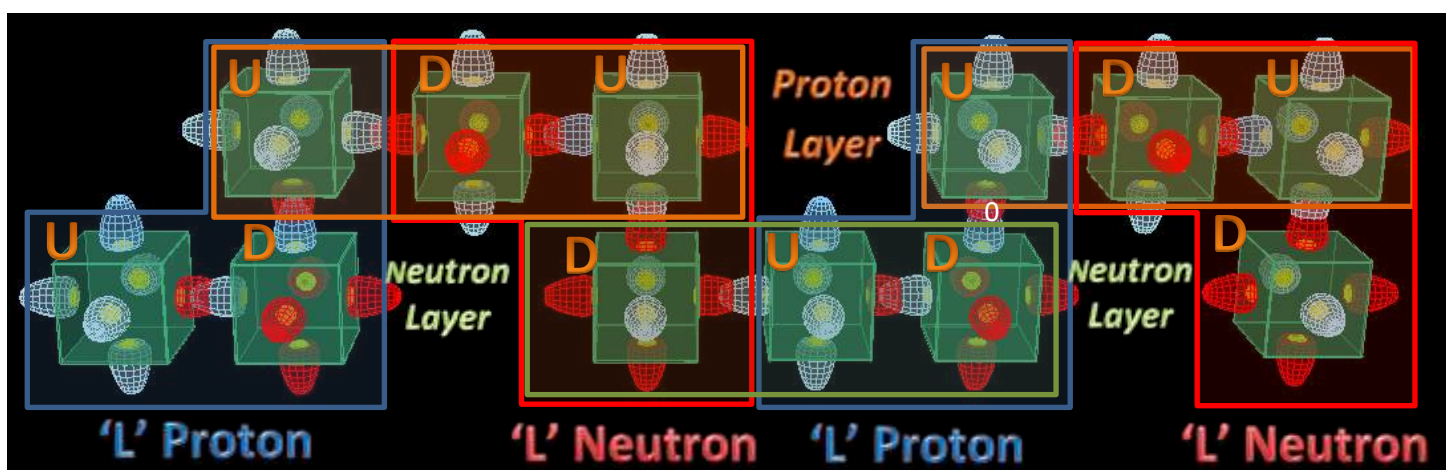


Figure 12: 'L' form Protons and Neutrons Interlock to create 'I' form Proton and Neutron Layers

The convention adopted is to represent 'I' protons by Golden-Yellow bars and 'I' neutrons by Mid-Green bars as shown in figure 4 for **Helium**. The 'bar' notation considerably reduces complexity due to excessive detail, allowing the nucleus geometry to be appreciated.

The second element in the Periodic Table is **Helium**, which is a colourless, odourless, tasteless, non-toxic, inert, monatomic gas, the first in the **noble gas** group in the periodic table.

He-4 consists of two 'L' form proton and neutron pairs inter-locking to form a 2 proton 'I' form and a 2 neutron 'I' form layer, and made stronger by 3 electron bonds as shown in figure 13.

The compact geometry of helium-4 results in a high binding energy (see graph right) compared with He-3, which consists of 2 'L' form protons, 1 'L' form neutron and only 1 electron bond (it has been generalised in terms of 'I' form nucleons in figure 4).

As **alpha radiation**, the nucleus of He-4 can be quite damaging to other matter over short distances.

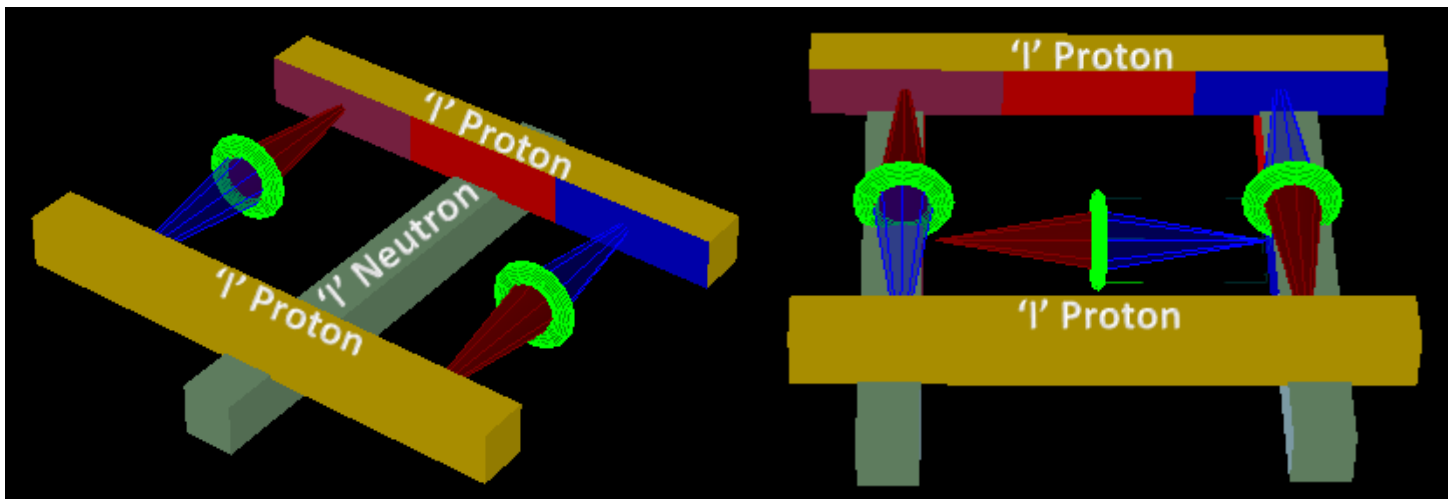
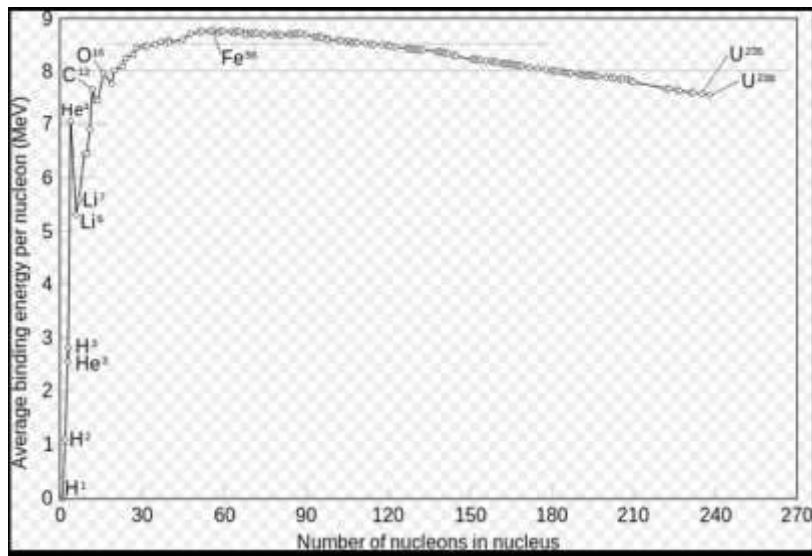


Figure 13: Bar Model of He-3 and He-4

E2M considers that He-4 contains 3 electrons that are internal to the nucleus rather than 2 electrons in the external $1s^2$ orbital as claimed by the 'spdf' model. He-3, on the other hand, is considered to contain 2 internal electrons.

The forth element in the Periodic Table is **Beryllium**, which is a divalent element only occurring naturally in combination with other elements within minerals. As a free element it is a steel-grey, strong, lightweight and brittle alkaline earth metal.

Primordial Beryllium predominantly consists of only one stable isotope, Be-9 (left in figure 14), and thus can be considered monoisotopic. On the other hand, radioactive cosmogenic Be-10 (right in figure 14) is produced in Earth's atmosphere by the cosmic ray spallation of oxygen to accumulate on surface soil. It has a half-life of 1.36 million years, with one of its two extra neutrons beta-decaying (as described earlier) into a proton, converting Be-10 into boron-10.

The extra neutrons of Be-9 and Be-10 are most likely 'L' form neutrons that attach in a variety of electromagnetically favourable positions around the nucleus: certainly a lot more work is required to determine their preferential positions, orientation and behaviour. In the meantime they are coloured coded (a brighter green than a neutron that is part of a neutron layer) to indicate that they are transitional or partially formed I-form neutrons.

Similarly, the relative length of electron bonds and the size of the electron tori are notional to highlight geometry rather than to indicate absolute or relative scale.

Be-9 and Be-10 are very interesting, as they display patterns repeated in elements throughout the Periodic Table. They demonstrate the way in which electron bonds can act as pillar and post supports holding nucleon layers together. For Be-9 there are more likely only 4 inter-layer bonds, with the other 2 neutron-to-neutron bonds being within the neutron layers. Thus the inter-layer bonding of Be-9 is weaker than that for Be-10 (that has 6 inter-layer bonds) and consequently, should it violently lose its extra neutron, the atom readily decays to Helium-4 in the form of alpha radiation.

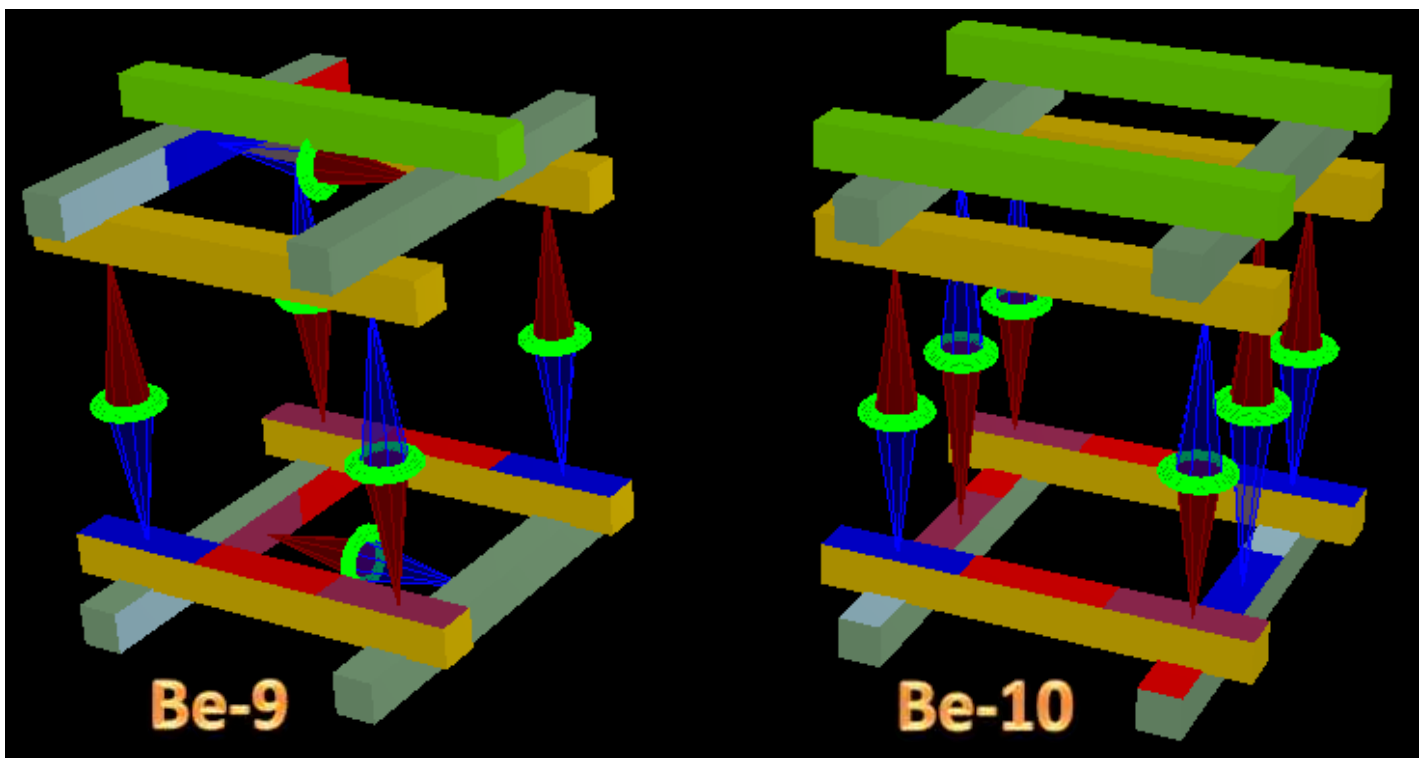


Figure 14: Bar Model of Beryllium-9 and Beryllium-10

For larger atoms further up the Periodic Table, the use of pairs of proton-to-proton (PP) and neutron-to-neutron (NN) bonding to hold nucleon layers together and to hold one polygonal form embedded within another become common features of nucleus structure. Polygonal form embedding will be discussed in more detail shortly.

Also, as the Atomic number of elements increase, so does the size and number of polygonal nucleon layer shapes possible and thus the number of forms an element can take: Carbon 12 is a good example of such **allotropy**.

Carbon is the sixth element in the Periodic Table, and Carbon-12 has two allotropic forms: one hexagonal presenting as **Graphite**, and the other tetragonal form presenting as **Diamond** (figures 15 to 17).

Graphite's nucleus consists of two hexagonal layers each made from 6 overlapping 'I' form nucleons: three neutrons and three protons. The two layers are joined (vertically as shown in figure 15 by nine internal electron bonds: three connecting matching neutrons and six connecting matching protons).

Graphite commonly occurs in sheets wherein external electron bonds (horizontal inter-atom bonding of figure 16) to create a larger-scale hexagonal sheet pattern. Graphite is soft and layered as dictated by its nuclear structure.

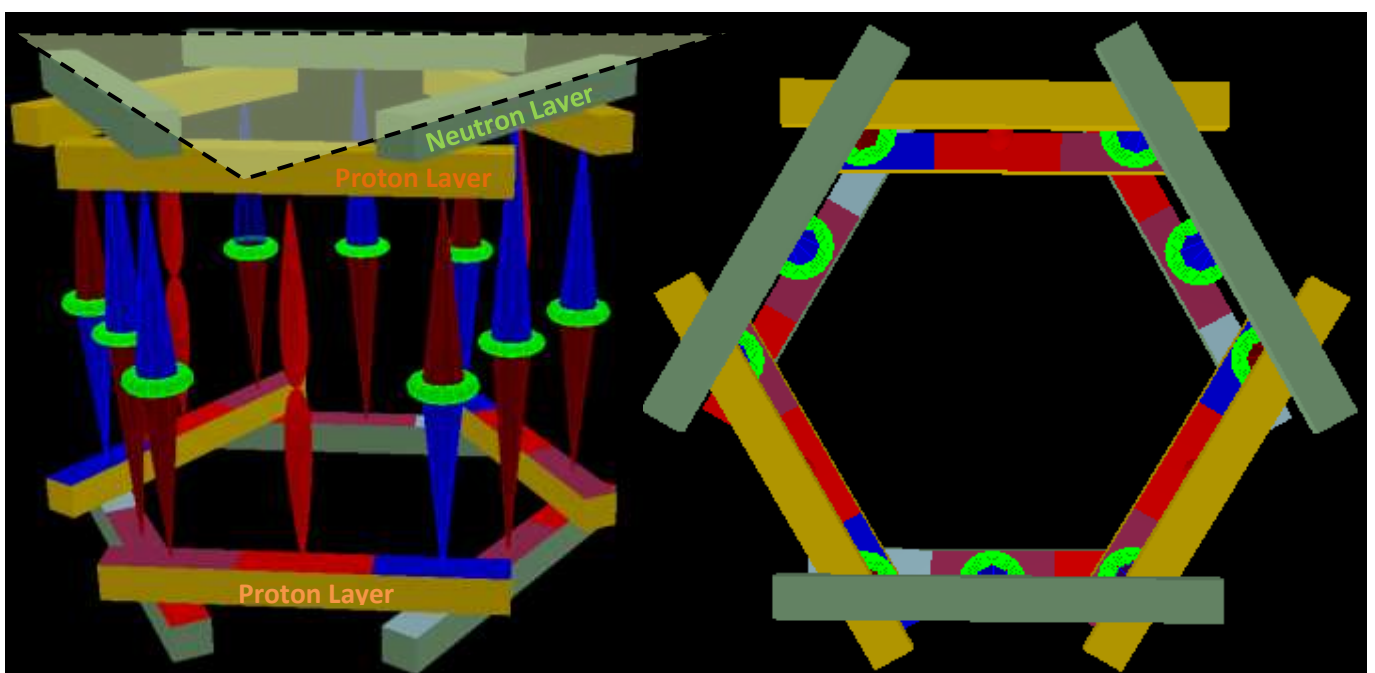


Figure 15: Bar Model of C-12 Graphite: Side and Plan Orthographic Views

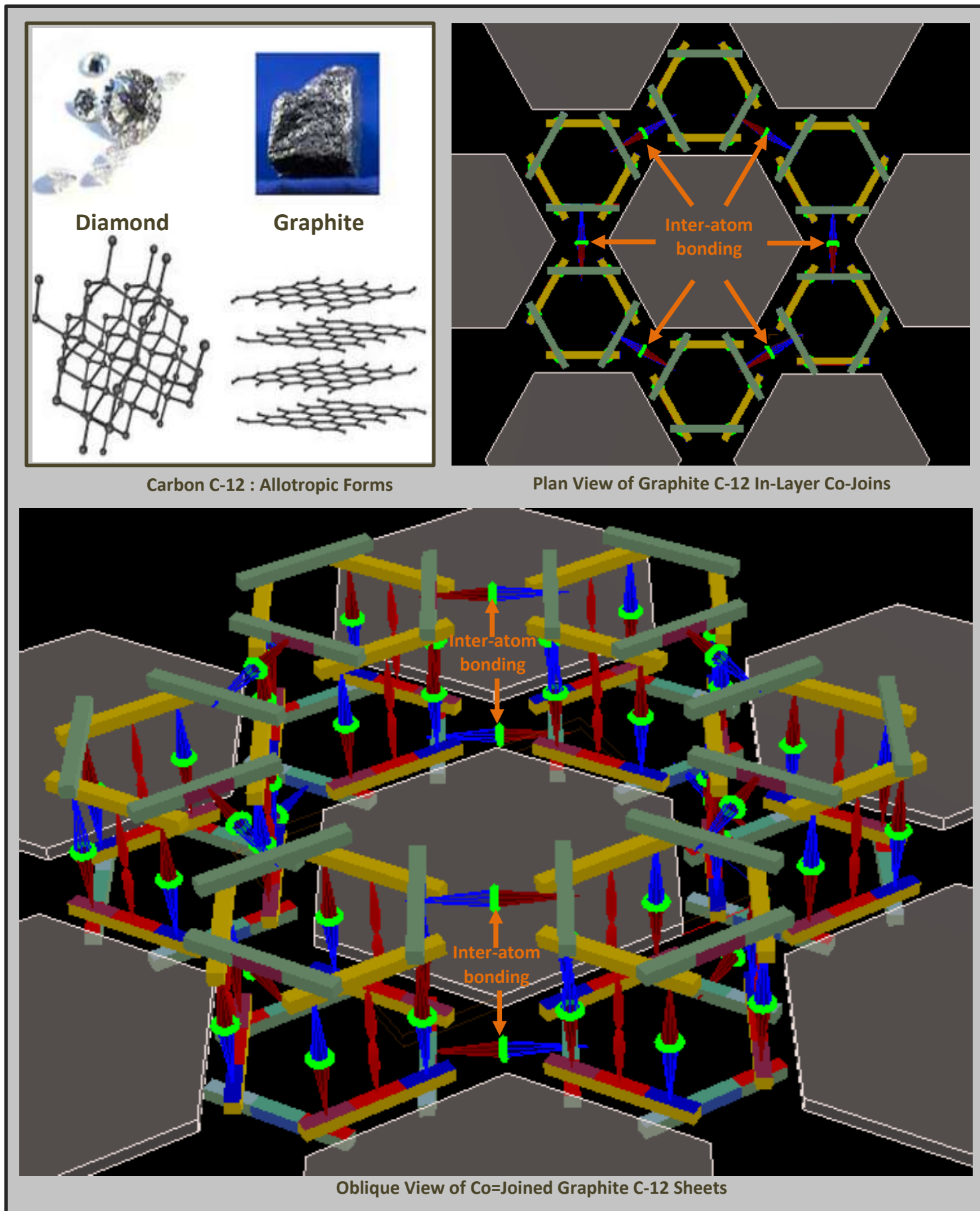


Figure 16: C-12 Allotropic Forms and Bar Models of Graphite Sheets

The Diamond nucleus, on the other hand, consists of a tight, strong tetragonal structure with nine cross-lashing internal electron bonds, as shown in figure 6. It forms into a cubic-form crystal system resulting in one of the hardest elements in the Periodic Table.

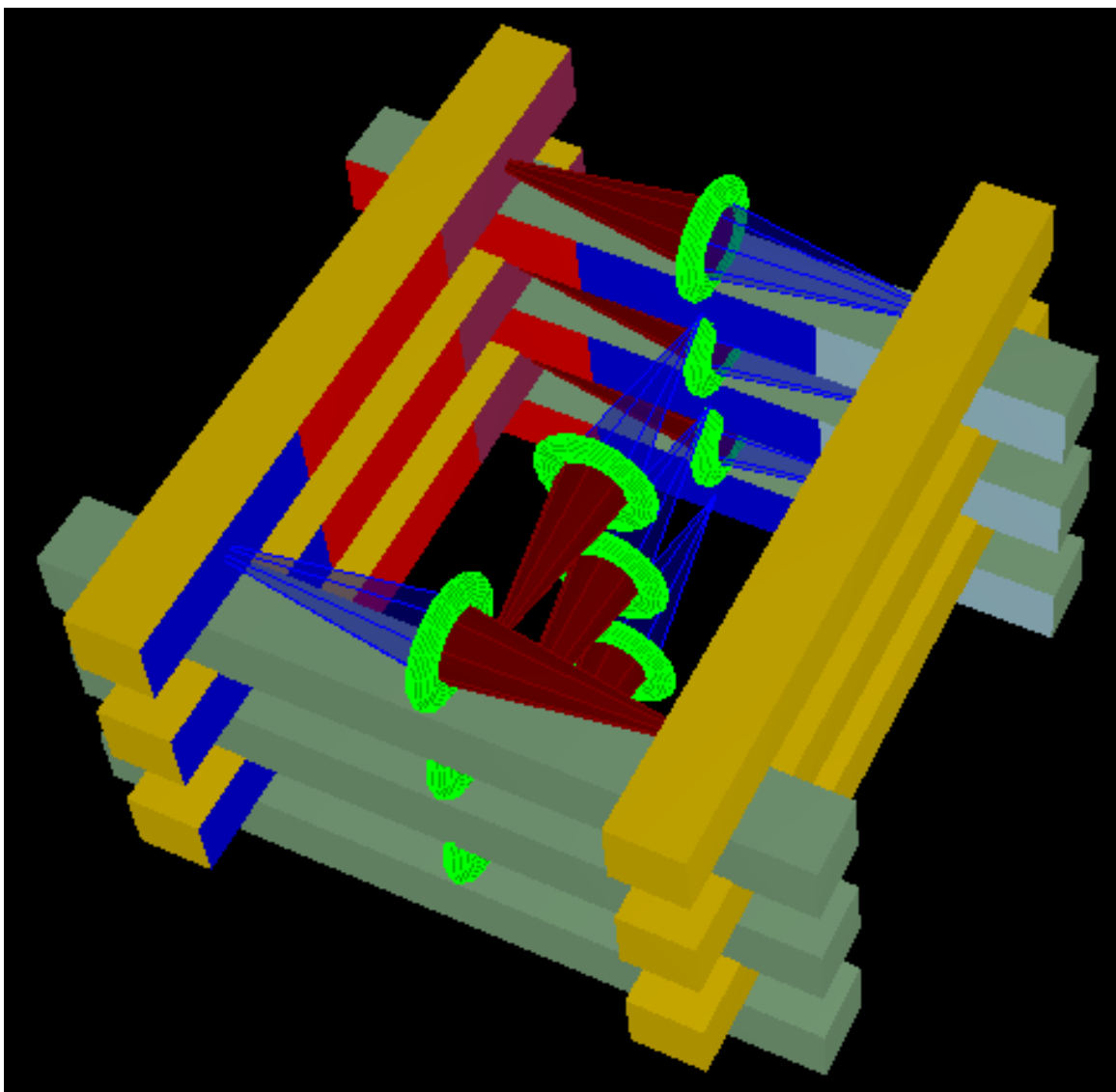


Figure 17: Bar Model of C-12 Diamond

The contrast between the physical characteristics of these 2 allotropic forms of C-12, which have the same nucleon count and internal electron count, is extreme: Graphite is a layered, grey material that is soft enough to be used for pencil 'lead', whereas Diamond is clear and hard enough to cut glass.

E2M contends that the nucleus structure of all of elements in the Periodic Table defines their physical and chemical characteristics.

Thus the key differences in the physical properties of the C-12 allotropes diamond and graphite are considered to be directly attributable to differences in nucleus structure.

Whereas nucleon layers for Carbon-12 can have a tetrahedral or hexagonal form, those for the Oxygen atom have an octagonal form. As atomic number increases, the maximum number of sides that a polygonal nucleon layers can adopt increases, but the polygonal forms for nucleus layers can only be in even side number increments, as they are formed by linked I-form nucleons (or inter-locked L-formms) of adjacent proton and a neutron layers. Polygons with an odd number of sides can only appear as a subset of a larger polygon with an even number of sides (e.g. pentagon geometries can only be a subset of a decagon geometry). The range of polygonal possibilities and their layer count (per pattern per element) can be seen in the arithmetic progressions of the figure 10 tabulation.

The 8th element of the Periodic Table is **Oxygen**, which is a highly reactive non-metal and oxidizing agent that, by mass, is the third-most abundant element in the universe after hydrogen and helium. At standard temperature and pressure, two atoms of the element oxygen bind to form dioxygen, a colourless and odourless diatomic gas with the formula O₂, which constitutes 20.8% of the Earth's atmosphere. As compounds, including oxides, the element makes up almost half of the Earth's crust.

The **Oxygen** atom has 12 electrons as 8 proton and 4 neutron inter-layer bonds. **Dioxygen** (O₂) has 24 electrons (22 well protected inter-layer electrons and 2 covalent bond electrons as shown in figure 18.

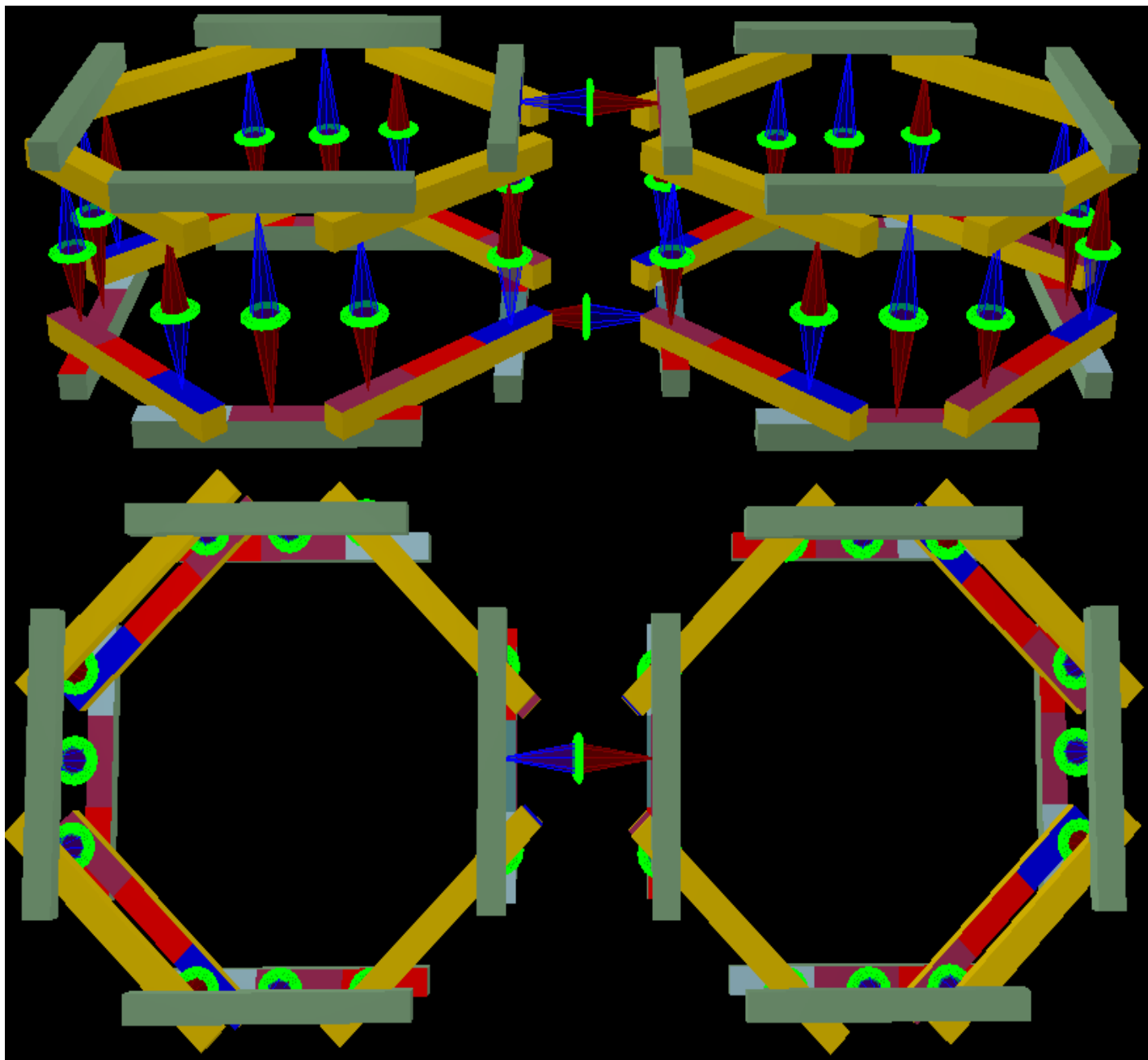


Figure 18: Bar Model of Dioxygen (O₂)

As well as the development of larger polygonal forms, the increase in atomic number of elements within the Periodic Table is also accommodated by the embedding of one polygonal form within another. Such embedding of one atomic structure within another starts at the lower end of the Periodic Table with Beryllium-9, which contains two embedded He-4 molecules and continues with larger polygonal forms further up the Periodic Table.

Elements of the 3rd period are transitional embedding tetragon forms within octagon forms (figure 11 and ↓ in table right).

By the 4th and 5th period the hexadecagon (16-gon) form is added (lower figure 24 graphic), with the triacontadigon (32-gon) by the 6th period.

Similar embedding of hexagon forms within the dododecagon (12-gon) form is possible, even extending to the tetracosagon (24-gon) form.

Framework Geometry	Period Number						
	1	2	3	4	5	6	7
Tetragon			↓	↓	↓	↓	↓
Octagon			↓	↓	↓	↓	↓
Hexadecagon				↓	↓	↓	↓
Triacontadigon						↓	↓

Periods 1 to 3 Potential Nucleus Structures																		
----- Polygon Names/No. Sides of Polygon/External Angles/Level Adjacent Face Angles -----																		
names	T e t r a	H e x a	O c t a	D e c a	D o d e c a		H e x a d e c a				T e t r a c o s a				T r i c o n t a d i			
# polygon sides	4	6	8	10	12	14	16	18	20	22	24	26	28	30	32	34	36	
# protons	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
Extremal Angle Degs	90	60	45	36	30	25.71	22.5	20	18	16.36	15	13.85	12.86	12	11.25	10.59	10	
Level Adjacent Angle	180	120	90	72	60	51.43	45	40	36	32.73	30	27.69	25.71	24	22.5	21.18	20	
AN Element	----- Possible Nuclide Geometries and Layer Counts -----																	
H																		
2 He	1																	
3 Li		1																
4 Be	2	1+1	1															
5 B		1+2	1+1	1	Magnesium Diborate (MgB2) = Hexa												Decaborane (B10H14) = Hexa	
6 C	3	2			1													
7 N		2+1		1+2		1	Racetams (drug class) and Pyrrolidine/tetrahydropyrrole (CH2)4NH, NH3 = Deca											
8 O	4		2	1+3			1	Tetrahydrofuran (THF)										
9 F		3					1											
10 Ne	5				2						1							
11 Na																		
12 Mg	6	4	3			2					1							
13 Al																		
14 Si	7			2+4			2				1	Siloles (silacyclopentadienes) = Deca						
15 P		5			3													
16 S	8			4	3+1			2								1		
17 Cl																		
18 Ar	9	6				3					2							

Figure 19: Nuclei Polygonal Forms for Periods 1 to 3 of the Periodic Table

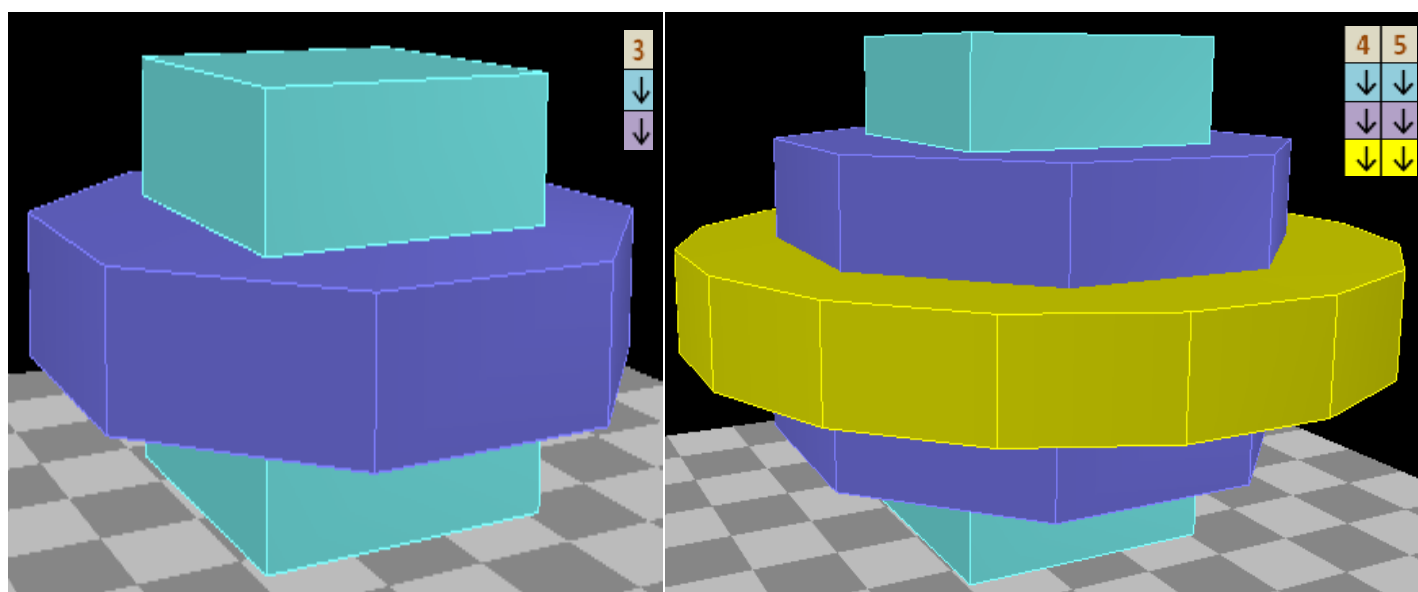


Figure 20: 4-8-16 Polygonal Embedding of Nuclei (Period 3 and Onwards)

The embedding of polygonal forms is well demonstrated by comparing the atomic structure of Copper, Silver and Gold in group 1B spanning periods 4 to 6 of the periodic table and having atomic numbers 29, 47 and 79 respectively.

Figure 21 shows nested tetragon, octagon and hexagonal proton/neutron layers for Cu-63 and the SM equivalent (insert of figure 21). The larger outer ring structure consists of 2 hexagonal nucleon layers separated and supported by a combination of proton and neutron electron bonds (vertical as shown). The octagonal layers are attached to the enclosing hexagonal layers by 16 proton electron bonds (4 sets of 4 shown as horizontal offset by 90° to each other). The nested tetragon is more rigidly attached to the enclosing octagon by neutron electron bonds.

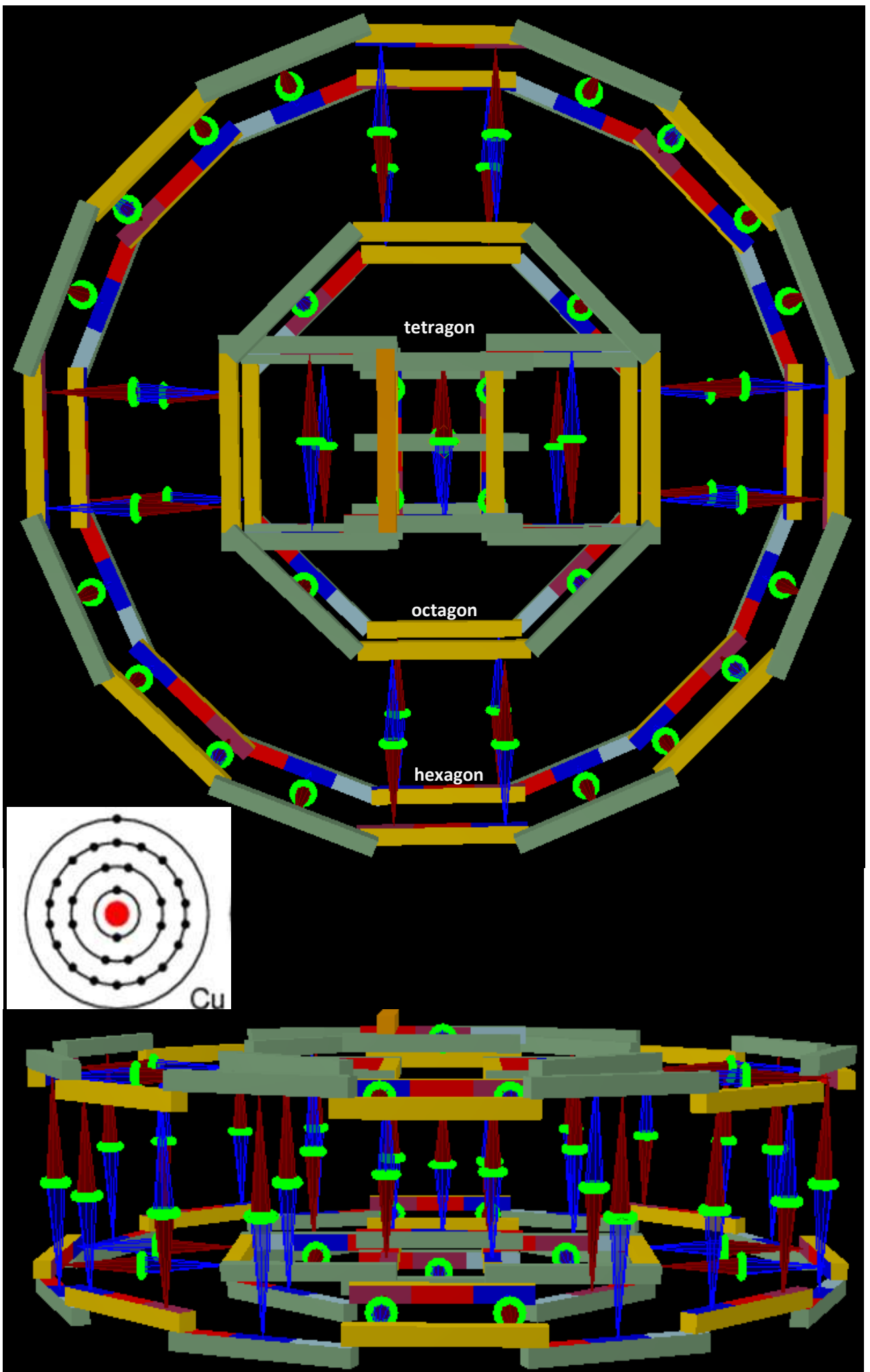


Figure 21: Bar Model of Copper-63

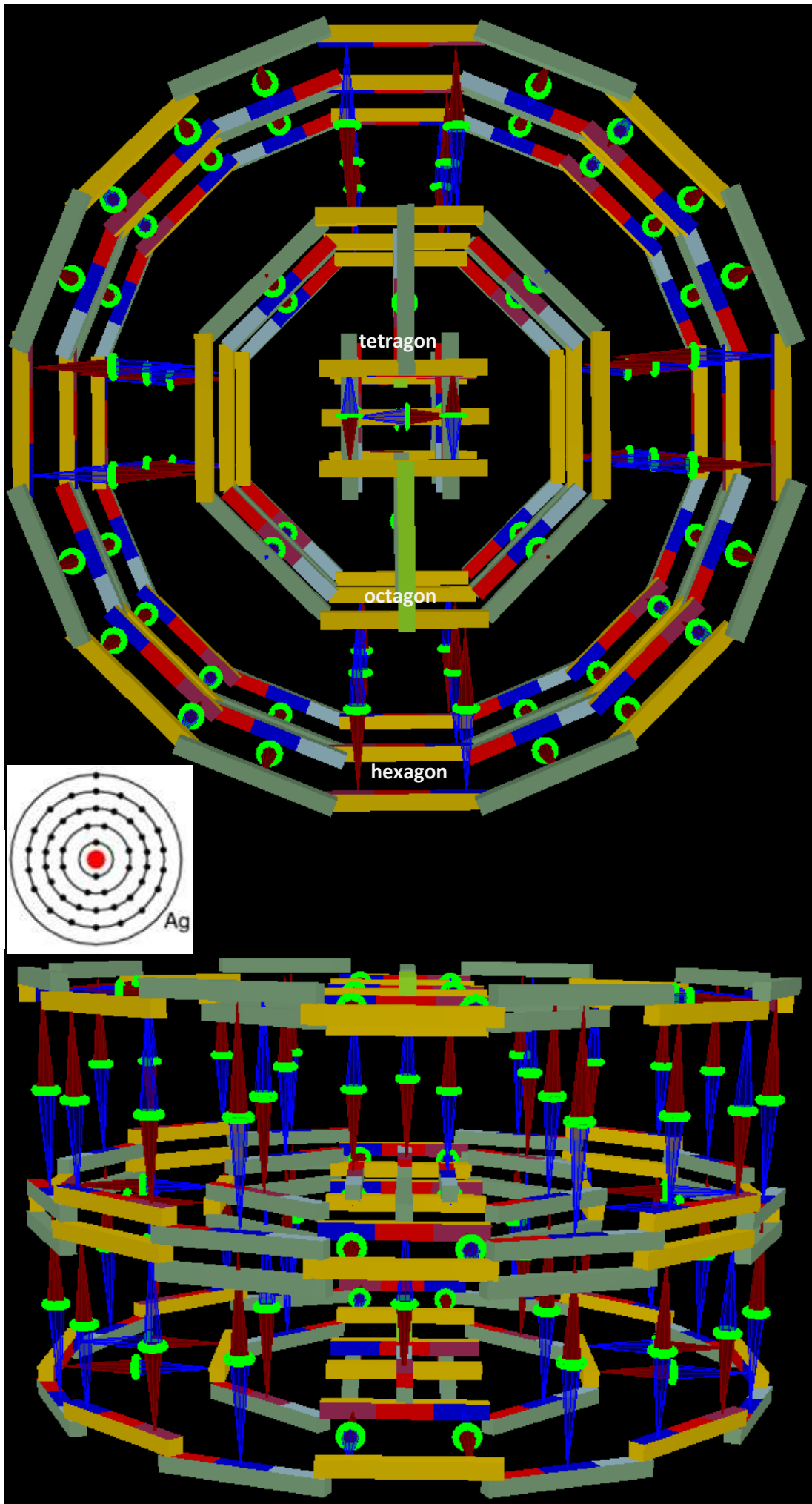


Figure 22: Bar Model of Silver-107

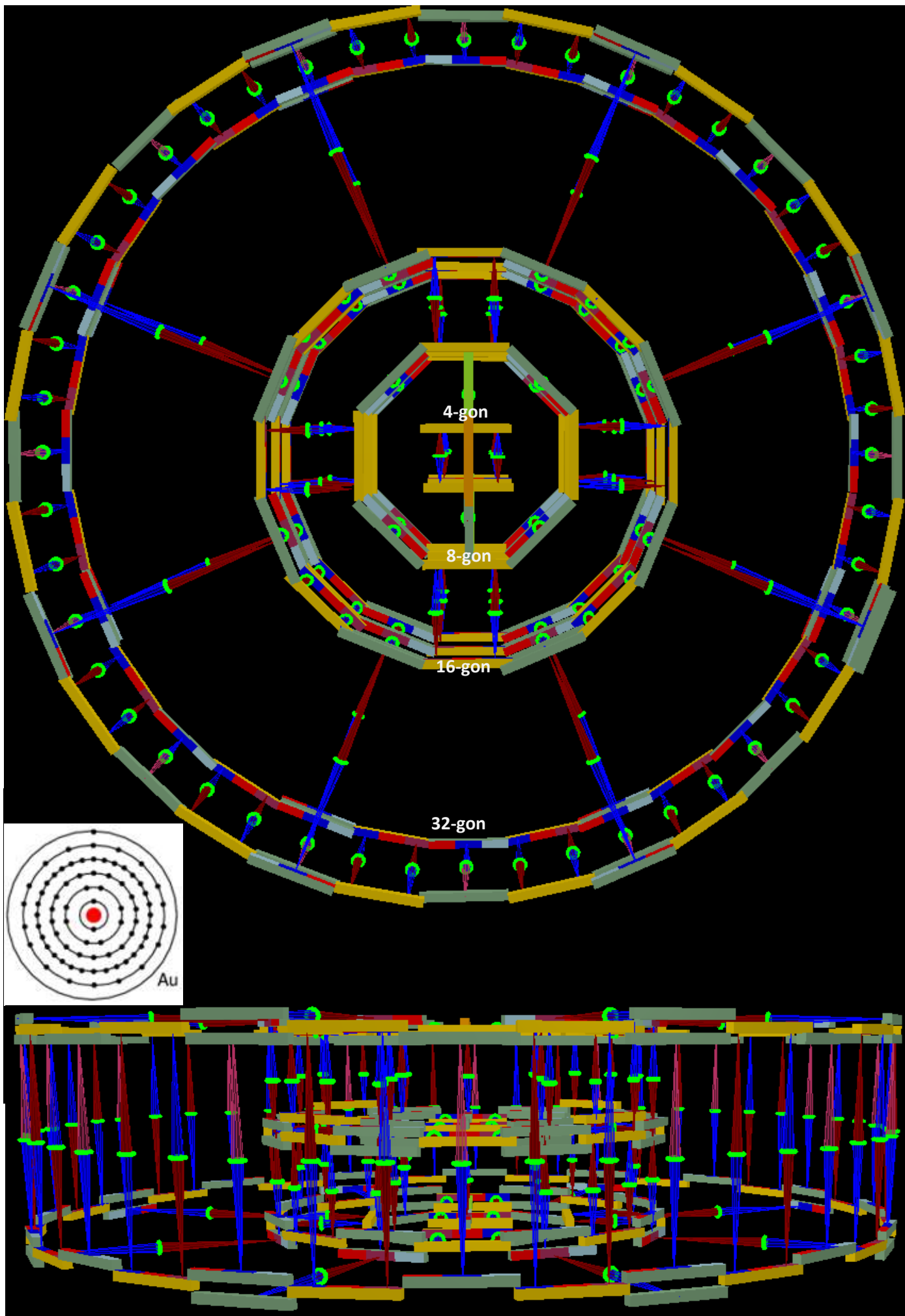
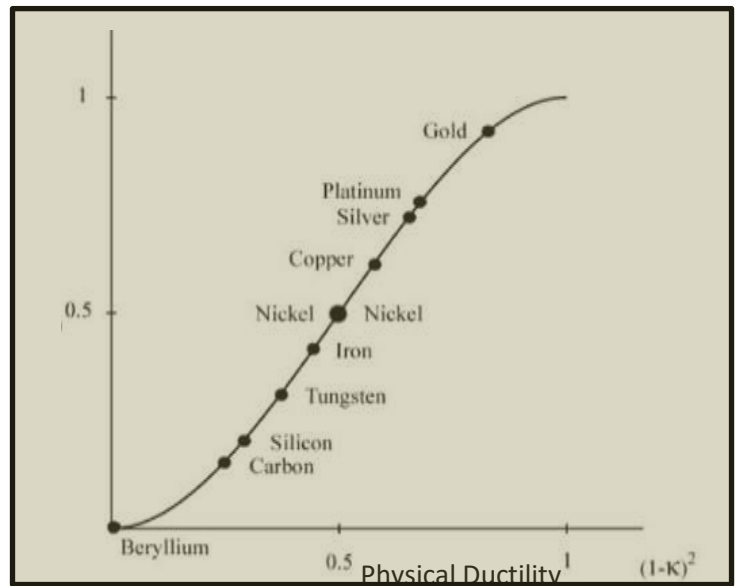


Figure 23: Bar Model of Gold-197

The physical characteristics of Copper, Silver and Gold match quite well to the E2M atomic model for them. Gold's two triacontadigon (32-gon) layers (figure 23) overlap its central hexagon layer contributing to its malleability - Gold is the most malleable of metals, followed by Silver, Aluminium and Copper.

Gold and Silver are also the 2 most ductile of metals (see graph right) with Copper being the 6th behind Platinum, Iron and Nickel. Also, the Shear Modulus of Copper is 48 GPa, reflecting its more compact form, compared with 30 and 27 GPa for Ag and Au respectively.

The SM represents the nucleus as an amorphous conglomerate of nucleons with no particular structure of note apart from size. E2M, on the other hand, claims that the unique structure each atom's nucleus contributes towards its physical form and characteristics. Even for large atoms such as Copper, Silver and Gold, each proton and neutron is needed and accounted for within their structure, as shown in their 3D bar models and summarised in the table below.



Metal Isotope	Level 1 (Top)		Level 2 (Middle)		Level 3 (Bottom)		Totals	
	Protons	Neutrons	Protons	Neutrons	Protons	Neutrons	Protons	Neutrons
²⁹ Cu-63	14+1	17	-	-	14	17	29	34
⁴⁷ Ag-107	15	15+1	18	28	14	16	47	60
⁷⁹ Au-197	30+1	46	18	26	30	46	79	118

Nucleons cannot just be added to the nucleus arbitrarily here and there to build up nucleon numbers and to transform one element's structure into that of another that might be structurally similar (e.g. from a Copper atom to Silver to Gold): there are changes in structure that have proven to be quite subtle and non-intuitive. For example:

- an additional neutron layer to support the central hexagonal layer in Silver and Gold;
- an additional neutron layer to support Gold's two triacontadigon (32-gon) layers;
- a weakening of the tetragon structure for Gold from that for Copper and Silver; and
- the staggered neutron bond supports between the tetragon layers within Gold.

Importantly, there is little room to vary the nucleon geometry for a particular isotope of an element. Apart from Carbon allotropes, the nucleon count corresponds to only one geometry, making it unique. However, the geometry may include several polygonal forms that resemble the nesting of other elements of a smaller atomic number. Should one or more nucleons be different or out of place, then you have a different element or isotope that may or may not be stable.

Bonding and Free Electrons

Bonding between CESs within up and down quarks can only occur when the facing poles have the opposite electric charge, thus having the same circular electromagnetic energy flow direction. Should the pole charge be opposite then should the magnetic poles also be opposite, then a strong bond, such as those holding the up and down quarks together to form the nucleons, is formed; otherwise an electron or positron bond is formed should the facing magnetic poles be the same, as shown in the table right. No bond is formed for any other combination.

Bond Type	CES1	CES2
Inter-quark bond (Strong Force)	N+	S-
	S+	N-
Bitron (Electron/ Positron) bond	N+	N-
	S+	S-

The bonds holding up (U) and down (D) quarks together to form 'L' form protons (U-D-U) and neutrons (D-U-D), and binding 'L' form nucleons together to create 'l' form nucleon layers are strong inter-quark bonds. As well as being very strong, inter-quark bonds act as two-way conduits (blue or orange flow directions in figure 24(a)) to distribute energy evenly throughout nucleons and nucleon layers.

When two CESs with opposite chirality are held in close proximity, a balance between the forces generated by their opposite electric charges and like magnetic poles is established. With their North poles (N+ and N-) or their South poles (S+ and S-) facing each other, their swirling electromagnetic fields overlap so as to concentrate energy between them as a torus-shaped structure called a **bitron**. As a torus shaped concentration of electromagnetic energy, a bitron contains approximately the same energy level and thus resembles an electron or a positron, except that the contained energy flows in only one direction, as shown by the blue and orange arrows of figure 24(b), with no magnetic component (i.e. there is no swirling energy flow around the small radius of the torus) as for electrons or positrons.

It is only as it is physically removed from the bond field that a bitron's energy is induced to swirl (i.e. small radius spin) like that of an electron or positron, as dictated by though which side of the bond field (i.e. the CES1 or CES2 side) that it exits. Thus, statistically, a bitron has equal probability of becoming either an electron or positron upon release from the parent bond. (*Note that so far bitron bonds have been loosely referred to as electron bonds*).

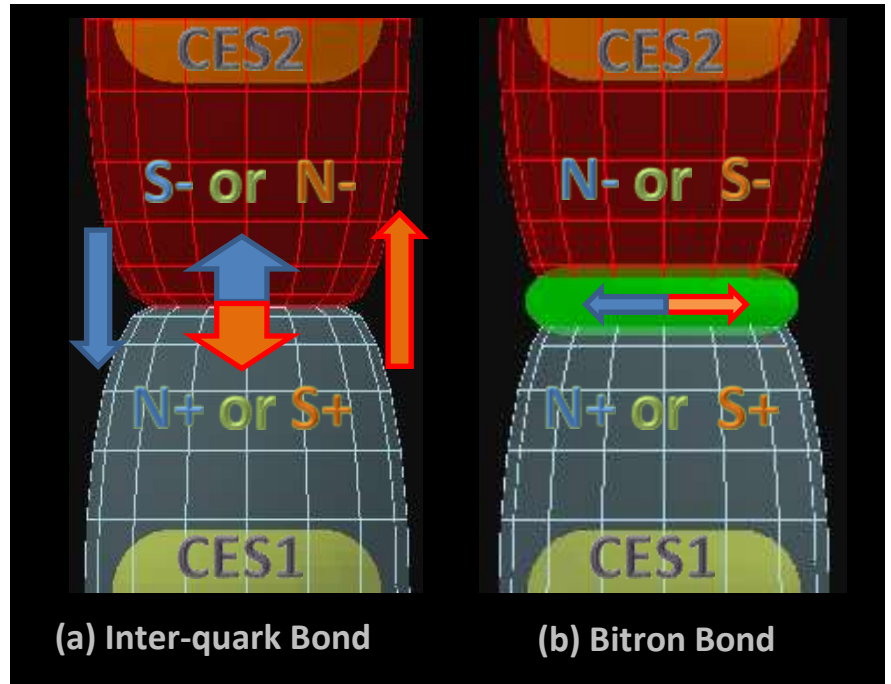


Figure 24: Inter-Quark and Bitron Bonding

The central postulate of E2M is that there is only one fundamental source of electromagnetic energy from which normal matter is formed, and that source is the CES. Thus all other energy manifestations, including electrons, positrons, neutrinos, anti-neutrinos, quarks, nucleons and photons, are derived directly or indirectly from CESs.

Free electrons and free positrons can be created by the electromagnetic bump-release of the bitron from a bonded pair of CESs; or by radioactive decay; or by chemical reactions (wherein bitron bonds are formed and/or re-assigned).

When an electromagnetic force (emf) is applied across a conductor (most metals), free electrons and positrons are aligned and induced to move as a current. When the emf is removed, the free electrons and positrons are randomly aligned by the electromagnetic fields of nearby atoms, with many becoming loosely attached to nearby un-bonded outward facing CES tori.

Free electrons and positrons can be captured (not to be confused with β decay electron capture) by a nucleon by becoming attached to the outward-facing side of an un-bonded CES, extending the strength and reach of that CES's electromagnetic field and thus its potential to attract and bond with other atoms and partake in chemical reactions.

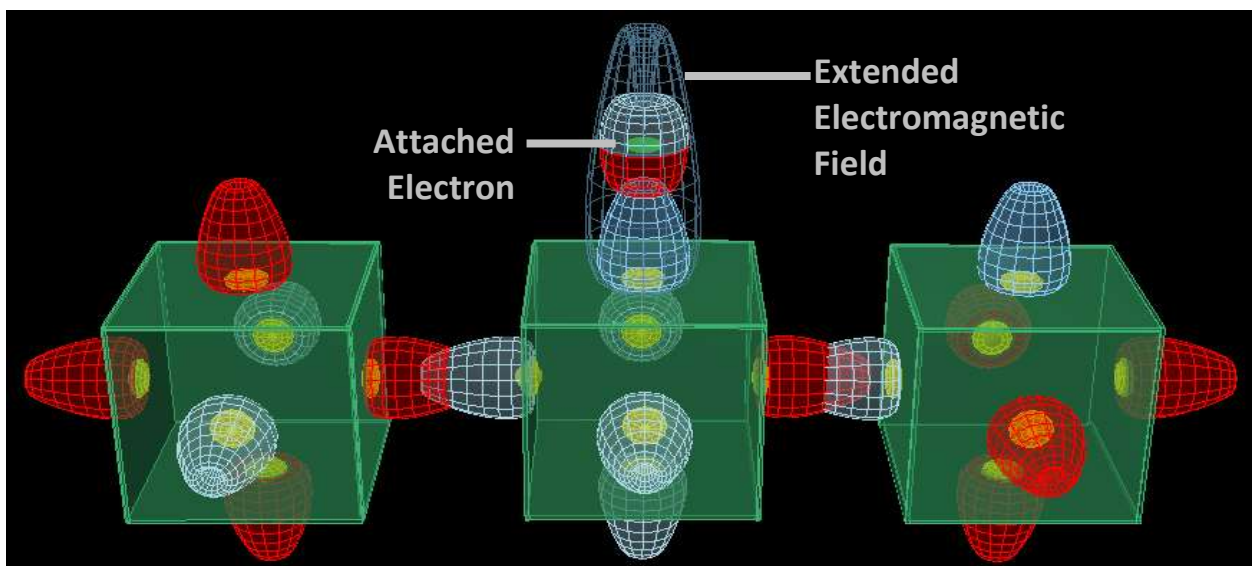


Figure 25: Free Electron Attachment to an 'I' form Neutron

Complex Chemical Compounds

A chemical compound is composed of atoms from more than one element held together by chemical bonds. For E2M chemical bonds result from bitron bonds or electromagnetic adsorption (e.g. oxygen adsorption). Conventional and E2M descriptions for specific types of chemical compound formation are summarised in the table:

Compound Type	Conventional Description	E2M Description
Molecules	An electrically neutral group of two or more atoms held together by chemical bonds (electron transfer or sharing).	An electrically neutral group of two or more atoms held together by bitron bonds.
Ionic Compounds	An electrically neutral chemical compound composed of ions held together by electrostatic forces between positively charged ions (cations) and negatively charged ions (anions).	An electrically neutral group of two or more atoms held together by bitron bonds. As separate ions (e.g. in solution) cations have a majority of their out-facing CES poles being a positive (N+ or S+), and a negative (N- or S-) majority for anions.
Intermetallic Compounds	Composed of definite proportions of two or more elemental metals wherein electrons form what may be termed an "electron sea" in which valence electrons are free to move from one atom to another to create a bonding effect.	Composed of definite proportions of two or more elemental metals, which pack due to the size and geometry compatibility of their nuclei that allows their atoms to inter-lock with possible electromagnetic adsorption between nuclei.
Coordination Complexes	Consists of a central atom or ion (often a metal) called the coordination centre, and a surrounding group of ligands (bound molecules or ions).	Consists of a central atom or ion (often a metal) called the coordination centre, and a surrounding group of ligands (bound molecules or ions) held by bitron bonds.

Electromagnetic **adsorption** occurs when atoms are aligned so that the bitron bonds supporting their nucleon layers interact, bending each other to create a bond, which, for oxygen adsorption, is only slightly less strong than bitron bonding. With bitron bonds under stress, many bitron tori are released as free electrons and positrons, resulting in high electrical and thermal conductivity.

Interstitial alloys provide examples of where adsorption is the main type of bonding. Figure 26 shows Cu-Au adsorption for Rose (or Russian) Gold, which is an interstitial alloy (e.g. 18K rose gold is 75% gold, 22.25% copper, 2.75% silver). For **substitution alloys** (e.g. bronze and brass, where some Cu atoms are substituted with either Sn or Zn atoms respectively) the size and electromagnetic structure of the base metal and alloying agents must be compatible.

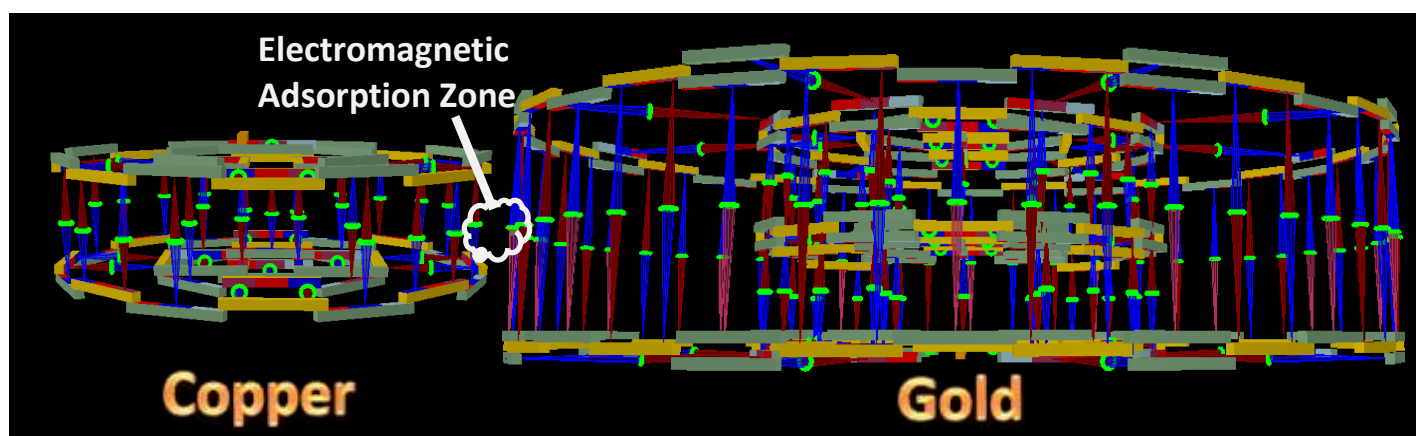


Figure 26: Electromagnetic Adsorption between Copper and Gold Atoms

As a noble metal, gold does not electron bond with oxygen. The top part of figure 27 shows two alternative models of oxygen (within water molecules) adsorption patterns for gold nanoparticles. Both patterns are compatible with the E2M model for gold, with water molecule potential bonding locations for each adsorption pattern shown by the symbols \otimes and \oplus . For oxygen such adsorption is only slightly less strong than that of bitron bonding.

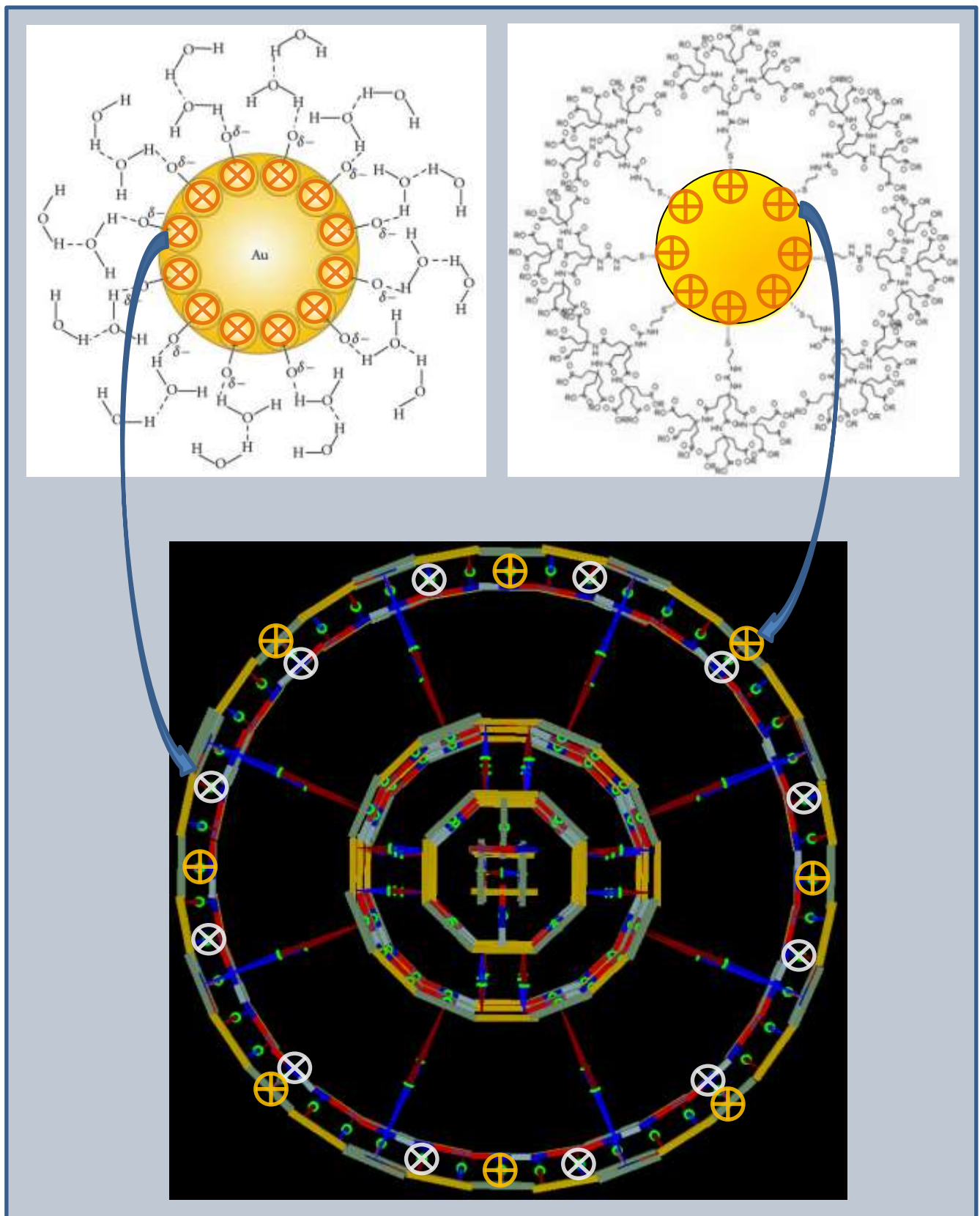


Figure 27: Models for Gold Nanoparticles in Water

For **catlysis** (a pre-cursor to many chemical reactions), some **enablers** (chemicals involved in a starter chemical reaction that is reversed by the end of the main reaction) induce the inter-layer bonds of reactant molecules to bend and stress to breaking point. Thus such enablers are active in the formation of reactant ions, which chemically interact to create molecular and ionic compounds via the formation of the appropriate bitron bonds.

Adsorption can provide a bond-like binding for elements and compounds together, or it can be a pre-cursor leading to a breaking of structural bitron bonds in one or more elements or compounds, freeing electrons and/or positrons, and facilitating the formation of new bitron bonds. Whilst not the only effective adsorption mechanism, water based oxygen adsorption is quite strong and widespread, enabling many chemical reactions, particularly those involving ions in aqueous solutions.

Bitron bonding also commonly occurs on the top and bottom nucleon layers of an atom. Nucleons not locked into a complete nucleon layers (i.e. are attached as an 'L' form nucleon), and excess neutrons in particular, are excellent candidates for bitron bonding with other elements, molecules or compounds. Most molecules and compounds have one or more top/bottom bitron bonds.

For the Standard Model, the **valency** (or **oxidation state**) of an element (figure 28) is determined by the number of outer shell (valence) electrons, and for polyatomic ions (such as SO_4^{2-}) is the charge associated with the ion. For E2M valency relates to the total number bitron bonds formed via adsorption of external inter-layer bitron supports (as in figure 26 plus bitron bonds involving nucleons on the top and bottom nucleon layers. Atoms may display multiple valencies depending upon the degree of adsorption of outer vertical bitron supports, and upon the top and bottom nucleon layer landscape (i.e. the preferential bond sites), and to which side (i.e. top or bottom) of the atom that other elements and compounds attach.

Compound	NH_3 Ammonia	NaCN Sodium cyanide	H_2S Hydrogen sulfide	H_2SO_4 Sulfuric acid	Cl_2O_7 Dichlorine heptoxide
Diagram		$\text{Na} - \text{C} \equiv \text{N}$			
Valencies	Nitrogen: 3 Hydrogen: 1	Sodium: 1 Carbon: 4 Nitrogen: 3	Sulfur: 2 Hydrogen: 1	Sulfur: 6 Oxygen: 2 Hydrogen 1	Chlorine, 7 Oxygen, 2

+1		Valency Table										-4 or -3		0											
H	+2											+3	+4	-3	-2	-1	He								
Li	Be	<i>transition metals</i>										B	C	N	O	F	Ne								
Na	Mg	<i>Members of this group of metals display a number of valencies</i>										Al	Si	P	S	Cl	Ar								
K	Ca											Ga	Ge	As	Se	Br	Kr								
Rb	Sr											In	Sn	Sb	Te	I	Xe								
Cs	Ba											Tl	Pb	Bi	Po	At	Rn								
Fr	Ra																								
1 group		2 group												3 group		4 group		5 group		6 group		7 group		8 group	

Figure 28: Valency Groupings and Diagrams

The geometry of chemical compounds is greatly influenced by the polygonal shape of the nucleon layers. Looking at an atom from the top or bottom perspective, lateral bitron bond angles are dictated by the polygonal shape of an atom's nucleon layers. Vertical (or near vertical) bitron bonds are dictated by the top and bottom layer landscape.

All bonded groupings will adjust their position in space in accordance to the net electromagnetic field generated by all attached atoms. The net electromagnetic field acting on vertical members is always symmetrical and provides them with strength and stability. Lateral bonds, on the other hand, are usually formed by the central up quark of an 'l' form neutron, which provides for rotation in a vertical plane, and are thus able to move vertically up and down to adjust to the net electromagnetic field around them.

Using Methane (CH_4) as an example, the C-12 atom has three bitron bonds firmly securing a Hydrogen atom (i.e. a proton) above its top nucleon layer as shown in figure 29, connected to the three neutrons in the upper nucleon layer. The corresponding three neutrons in the lower nucleon layer are each attached laterally to single hydrogen atoms via single bitron bonds. Within the original intact single C-12 atom the 3 pairs of neutrons would have originally been

bitron bonded together, acting as vertical supports: however these were broken and re-assigned to bond with hydrogen atoms to form methane as shown.

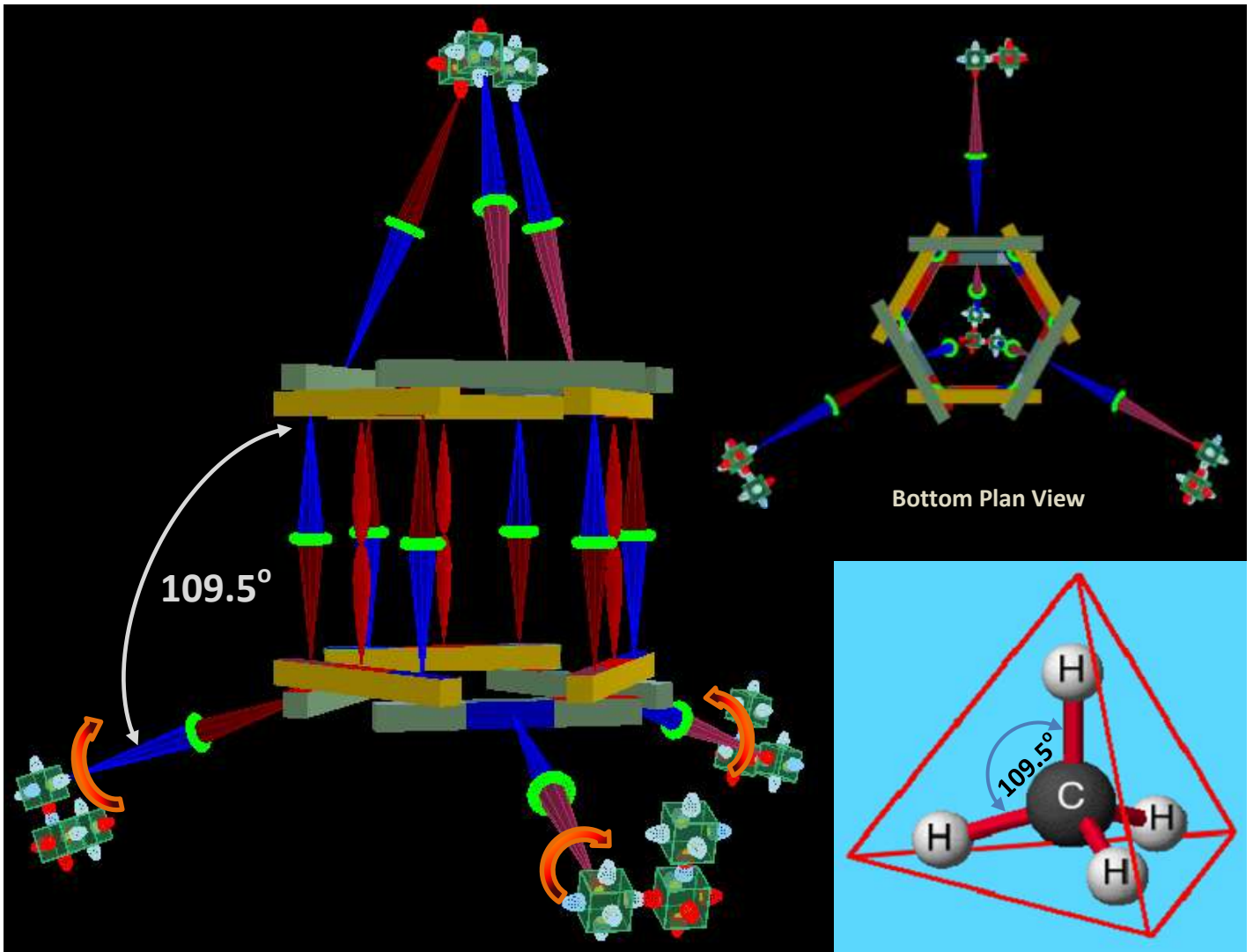


Figure 29: Bar Model for Methane (CH_4)

Within nucleon layers, down quarks within 'l' form neutrons have the ability to rotate when they are not rigidly bonded to another neutron layer within an atom's structure. In methane, the three lateral bonds with hydrogen atoms are free to swing in a vertical plane, hinged by their 'l' form neutron. The hydrogen atoms (i.e. protons) spin about their bitron bond's long axis quite rapidly to create a uniform electromagnetic field around each of the lateral hydrogen atoms. The vertical hydrogen atom, on the other hand, is rigidly held aloft by the three top nucleon layer's bitron bonds with little chance of movement or rotation.

The net result is a stable equilibrium resulting in a strong structure with the three lateral bonds being equal spaced and separated by 120° when viewed in plan projection, 109.5° to each the vertical bitron bond and to each other, forming a tetrahedral geometry.

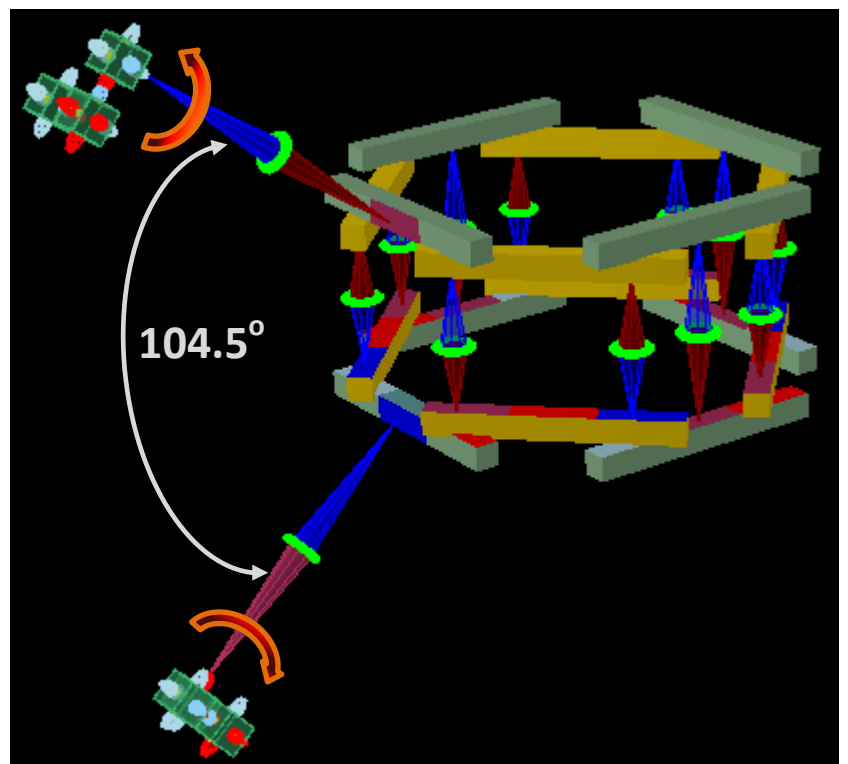


Figure 30: Bar Model for Water (H_2O)

For water (H₂O) the two hydrogen atoms are bitron bonded to a pair of neutrons in the top and bottom nucleon layers of an oxygen atom, as shown in figure 30. The free swivel action of these bitron bonds means that in the gaseous state the angle between them and the length of the bonds vary, or more accurately, vibrate around the mean angle of 104.45° and 0.9584Å (see graphs and notes of figure 31 which are from the [World of Molecules](#) web site).

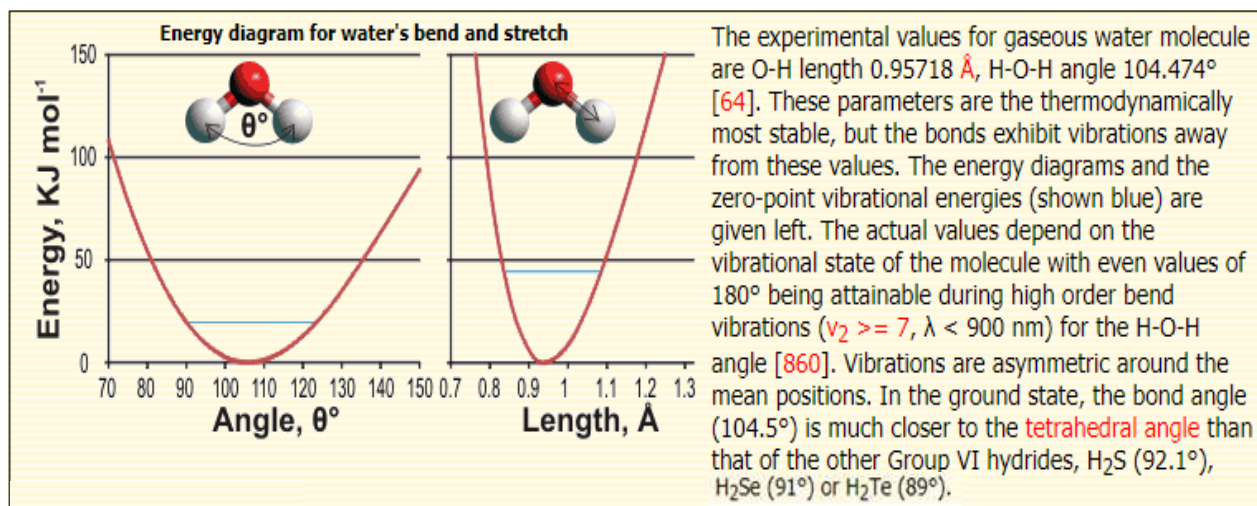


Figure 31: Bond Variation of Gaseous Water (steam)

Figure 32 shows examples of two-coordination (linear bonding) to six-coordination bonding: all examples show vertical bitron bonds (one top and/or one bottom). The trigonal planar, tetrahedral and trigonal by-pyramidal geometries (the 3 to 5-coordination examples) require hexagonal nucleon layers, whereas the others could correspond to tetragonal, hexagonal, octagonal or larger polygonal nucleon layers.

Shape	Geometry	Examples
Linear		BeCl ₂ , HgCl ₂ , CO ₂
Trigonal planar		BF ₃ , CO ₃ ²⁻ , NO ₃ ⁻ , SO ₃
Tetrahedral		CH ₄ , PO ₄ ³⁻ , SO ₄ ²⁻ , ClO ₄ ⁻ , TiCl ₄ , XeO ₄
Trigonal bipyramidal		
Octahedral		SF ₆ , WCl ₆

Figure 32: Atomic Geometry of Chemical Compounds

Whereas conventional science struggles to explain the bonding geometry of chemical compounds by complicated scenarios involving the sharing of orbital electrons, E2M provides uncomplicated and logical explanations involving the polygonal form of nucleon layers and the flexibility (albeit somewhat restricted) of neutron-based bitron bonds.

Photons, Energy Transfer and Spectral Lines

The particle versus wave debate about the nature of light has been raging since the 18th century. The wave approach has been most useful, allowing the allocation of wavelengths for the complete EMR spectrum from Gamma to long radio wave, complete with a detailed colour map of EMR by wavelength in the visible light range (see figure 2). From the colour of light, its wavelength can be determined and, using Planck's constant and the speed of light, the formula $E=hc/\lambda$ allows the corresponding energy of photons to be calculated.

The Dirac equations have sent the Standard Model and Quantum Mechanics approaches down a path of considering that the particle-wave duality of claimed from a mathematical perspective for photons also applies to all matter, including electrons, molecules and atoms. String theory, which relates to quantum gravity, supports the notion of energy transfer as vibrating strings or strands with preferred patterns of vibration, adds to the notion of wave-like photons. The Simple Universe Model's website provides excellent animations of strands.

A model for EMR based upon a string, strand or a photon with particle-wave duality would sit well with E2M. However, until there is further physical proof to the validity of one particular alternative, it is more consistent to model a photon as a derived or secondary form of concentrated energy with a structure analogous to an electron. An electron (and a positron) has a rest mass of $0.511 \text{ MeV}/c^2$ whereas the rest mass of a photon is so small it cannot be directly measured; however, as inferred through satellite measurements of planetary magnetic fields, it is thought to be of the order of $3 \times 10^{-33} \text{ MeV}/c^2$. For practical purposes, a photon is usually considered to be massless.

For the purposes of this paper, a photon is represented as a torus (shown in purple as in figure 33) of concentrated energy surrounded by a swirling electromagnetic energy field. EMR is considered to be made up photons travelling that spin as they travel at the speed of light (c). The spin frequency of each photon is dependent upon its torus radius and internal energy determines the wavelength of their associated electric and magnetic fields, and hence the corresponding frequency of the EMR as described in Appendix A.

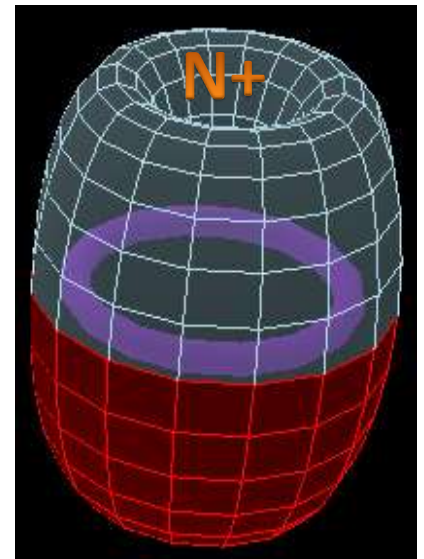


Figure 33: A Photon

The torus particle model for a photon falls firmly in the particle side of the 'particle versus wave debate'. On the surface it would seem to conflict with experimental evidence supporting the wave and particle-wave duality models, but, rather than digressing into side issues in the body of this paper, they have also been addressed in Appendix A.

Nucleons consist of three up/down quarks (U-D-U for protons and D-U-D for neutrons) and 18 CESs, of which 2 pairs of CESs form the inter-quark bonds (see figure 1(a)), leaving 8 outward facing North Pole CESs and 8 outward facing South Poles. Because the energy in and out flows of nucleons is via the outward facing poles, for free (i.e. unattached) nucleons electromagnetic energy in and out flows are balanced and unrestricted. However whenever a bitron bond forms between nucleons, it represents an energy storage mechanism in the form of the bitron, and a restriction disrupting the free 2-way flow of energy: it is analogous to a capacitor in an electric circuit and it limits flow rates.

All proton-to-neutron bonds (i.e. those converting 'L' form nucleons into 'I' form) within nucleon layers are inter-quark bonds, which are strong force bonds that act as two-way conduits for energy flow between connected nucleons. Such interconnectivity plays an energy distribution and balancing role for interconnected nucleons. However nucleon layers are connected by bitron bonds, and such connection has a buffering or capacitance effect that controls the rate at which electromagnetic energy can flow in and/or out of member nucleons.

The bulk of nucleon energy out-flows takes the form of radiant EMR energy which is generated from un-bonded CESs with outward facing South Poles. For such CESs, the narrow nature of its return or in-flow funnel restricts the return of electromagnetic energy, causing congestion that builds up into a torus of concentrated energy around the edge energy in-flow and out-flow interface. When the radius of the blockage torus grows, it expands into the CES's fast

moving out-flow energy zone to be summarily flick-ejected (left in figure 35) as EMR, forming the **base wavelength** bands within an **emission spectrum** characteristic to that particular parent atom or molecule.

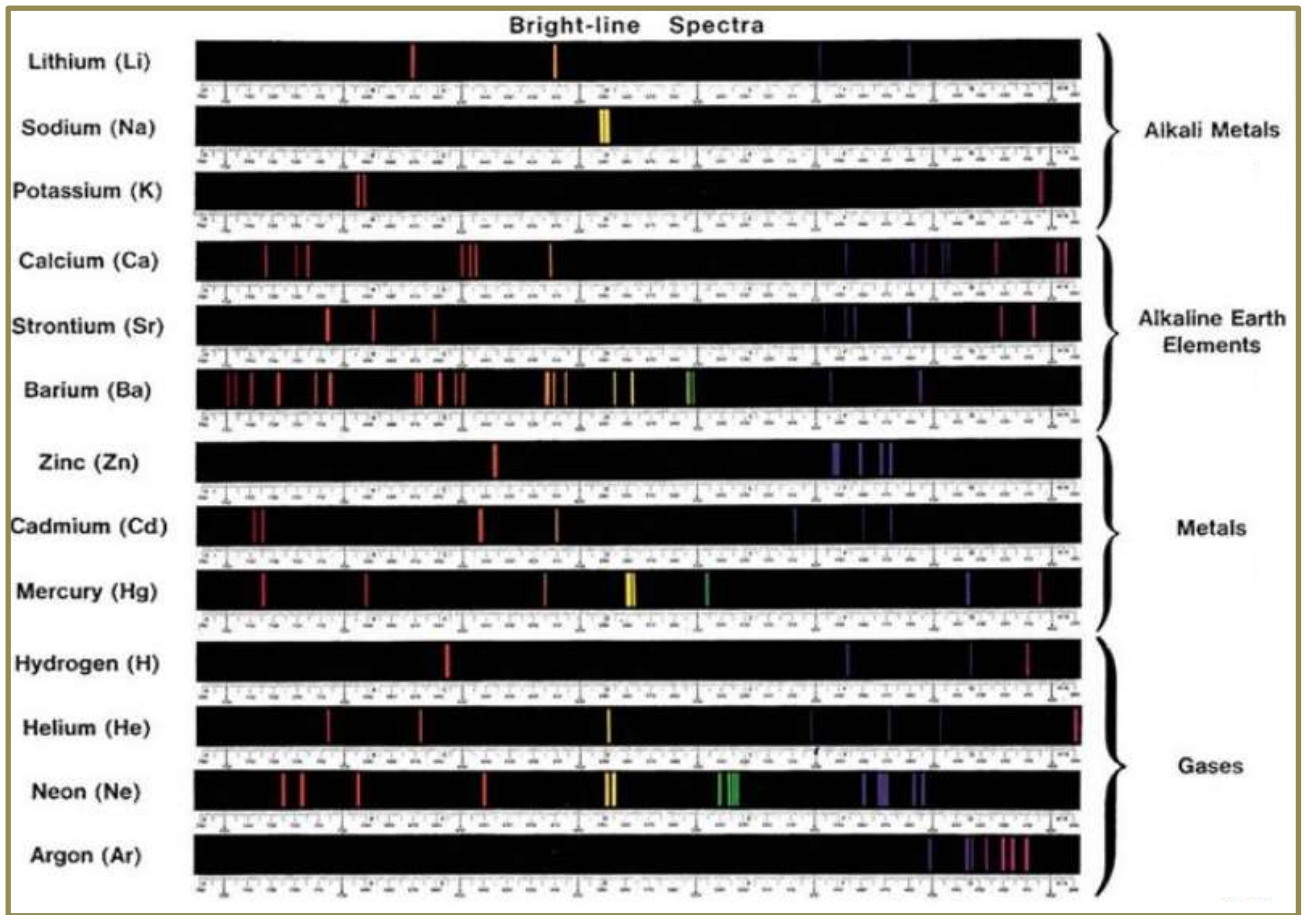


Figure 34: Emission Spectra for Common Element Groupings

Compounds and some atoms and molecules have multiple base wavelength bands within their emission spectrum. For material containing a range of different elements, each atom comes with its own nucleon layer groups and associated base wavelengths. Lone, larger atoms with polygonal embedding (as described earlier) can contain multiple different nucleon layers which can result in multiple base wavelength emissions.

Nucleons can also capture radiant energy from incoming photons, partially absorbing some of their energy, and then re-transmitting them as EMR with a reduced energy level (i.e. increased effective wavelength analogous to the **Compton Effect**, wherein there is an increase in wavelength of X-rays and other energetic electromagnetic radiations that have been elastically scattered by electrons): such EMR is referred to as **rebound EMR** (right in figure 35).

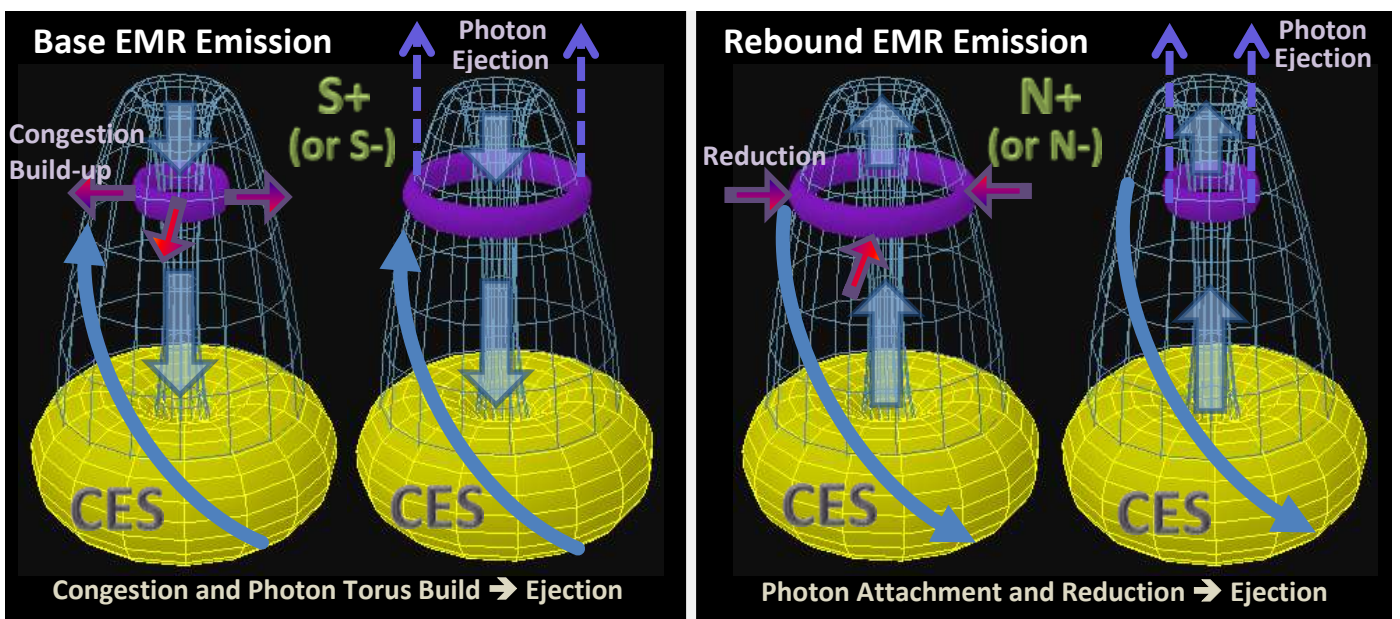


Figure 35: Base and Rebound Photon Emission

Rebound EMR is via un-bonded CESs with an outwards facing North Pole Energy. Energy from captured photons is siphoned off via the CES's outer energy return zone, causing a reduction in the photon torus size, and its inner radius in particular. The photon's reduced inner radius causes it to be caught by the fast-moving electromagnetic energy from the CES's North Pole central out-flow funnel, resulting in it being flick-ejected as EMR of a lower energy level than it had when it was captured.

The photons that can be captured by an un-bonded CES with an outward facing North Pole is limited by the strength of the CES's electromagnetic field: only photons with specific optimum size (radius) and energy level combinations are amenable to capture by a given CES field strength. The field strength of a CES is governed by its in and out energy flow rate, which is controlled by the capacitance of its nucleon layer. And because the nucleon layers of each atom are unique, each atom has its own specific group of quantum-like energy levels for photons that can be captured; and that group is unique and different to that of other atoms as evidenced by their emission spectra.

In some cases the position of a photon captured by a CES's electromagnetic field shifts so that its reducing energy level does not cause it to be ejected, and it may well stay in play while passing through several quantum energy reductions. In other cases some rebound photons are ejected with energy levels still within the marginal limits of one of the optimum quantum levels, and may thus become recaptured and be further energy reduced by another atom or molecule. Multiple energy level jumps and photon recycling thus both contribute to the generation of multiple energy bands, with secondary rebound emission lines tending to exceed the number of base bands within emission spectra.

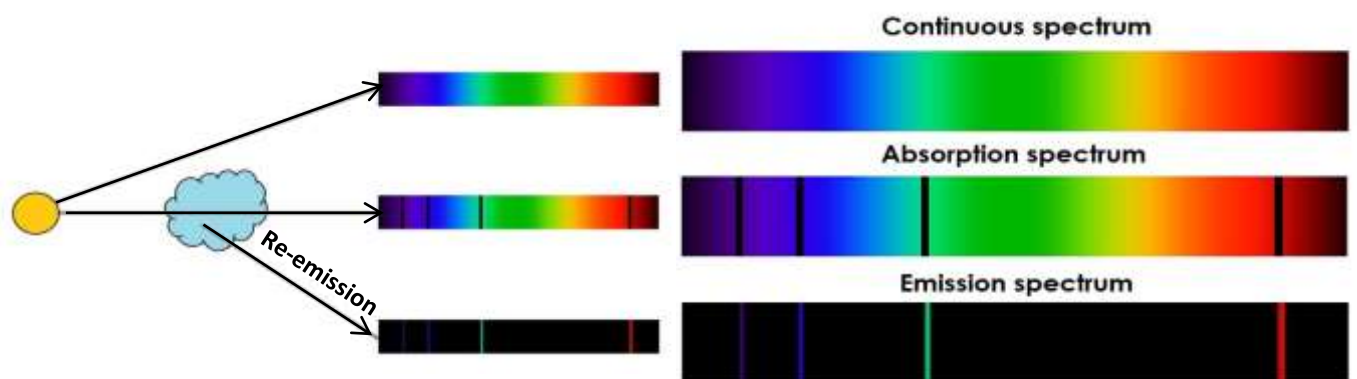


Figure 36: Absorption and Re-Emission Spectra

Another example of quantum energy level selection of photons is **spectral line absorption**, wherein photons with the optimum energy levels are captured from incident light and completely absorbed by the targeted material. Absorbed spectral energy can in turn be re-emitted with a similar pattern of spectral lines as those absorbed, as demonstrated by gaseous clouds as represented in figure 36.

EMR represents a major way in which energy (radiant) is transferred between bodies. Some photons from incident EMR are electromagnetically repelled (reflected) or diverted (refracted); others become attached to or absorbed by nucleons. The extra energy provided by the attached photons feeds back into the atom's nuclear nucleon layers, adding to the net energy of the atom.

Plasma and Cosmic Radiation

Plasma is formed by super-heating matter, which causes electrons to be stripped from the atomic structure to yield a mixture that is rich in electrons and ions. Over 99% of the matter in the visible universe is believed to be plasma.

An important source of plasma formation in space is **photoionization**, wherein photons from stellar EMR are absorbed by an existing gas mixture, causing electrons to be emitted.

Closer to Earth, lightning, naked flames, welding arcs, neon/fluorescent tubes, neon signs, plasma (some TV and computer screens), plasma lamps and globes involve the generation of plasmas; and the Earth is surrounded by a dense plasma called the **ionosphere** and impacted by the Sun's solar wind plasma.

Plasma can also be created in the laboratory by super-heating a neutral gas or subjecting it to a strong electromagnetic field to the point where it is ionised.

At lower temperatures radiant EMR energy is acquired and managed by the energy transfer and balance mechanisms as described in the previous section.

The temperature and degree of chaotic buffeting within a gaseous mix increases as the energisation levels increase, and the bitron within external bitron bonds start being released as a fast moving electron or a positron, depending upon which side of the bond the bitron exits. In such a highly energised environment, when one bitron is released another quickly forms: this bitron form-and-release and re-form process represents a much higher energy transfer rate than that of photon absorption and re-transmission. Importantly, it results in a rapid increase of electron and positron release that greatly increases the electrical conductivity of the mix, the generation of extensive long-range electromagnetic fields and production of gamma radiation (resulting from electron-positron annihilation collisions).

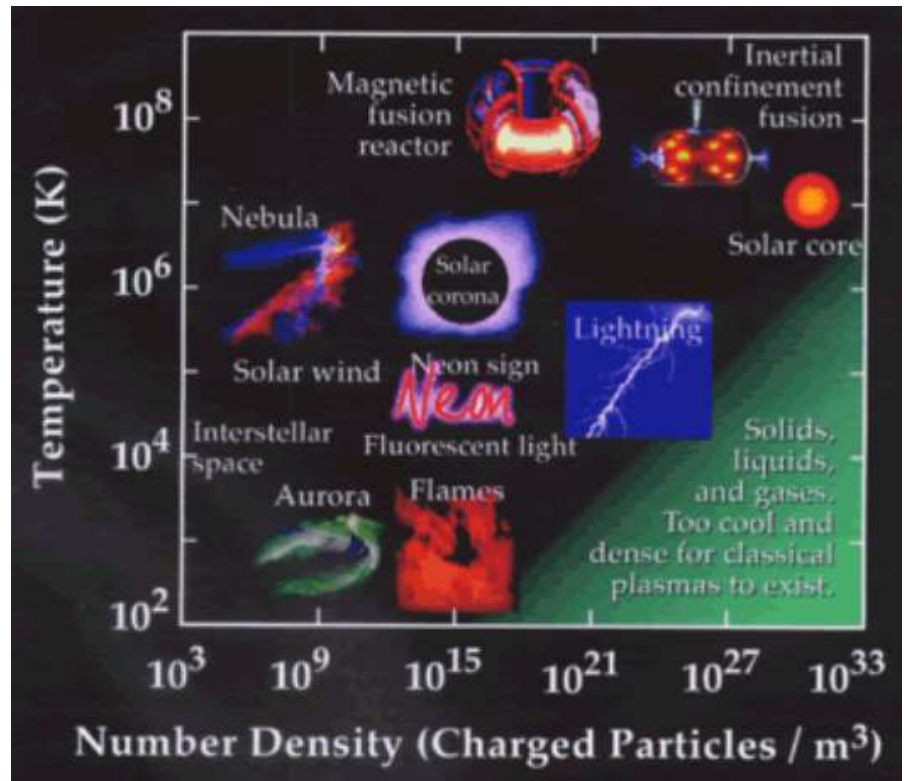


Figure 37: Plasma Temperature/Density Distribution

As the energisation continues to increase, external bitron bonds start to fail and the plasma mix becomes increasingly ionized. For Hydrogen ionization starts at about 7,000° K and by around 10,000° K it is completely ionized.

When temperatures rise to well above the 10⁴ K range, such as achieved by active stars, the bitron bonds supporting nucleon layers within atoms (i.e. atoms more complex than Hydrogen) start to fail, facilitating the separation of nested polygonal forms and the creation of atoms lower down the Periodic Table. The very fabric of the original compounds has commenced to break down, with any semblance to the original atomic structures being lost. Cooling of the plasma cloud would now result in a completely different mix of atoms and compounds.

Should energisation continue even further, nucleon layers would be further separated with the destruction of any remaining inter-layer bitron bonds, and the layers then broken down into their component nucleon and/or quark parts: such utter destruction and decomposition most likely only occurs in the crush within the bowels of a black hole.

For Hydrogen, the most elementary and widespread form of plasma in the universe, the ionization process takes place from about 7,000° K, with the bitron bonds of Hydrogen molecules being stripped away, creating a mix rich in electrons, positrons and protons (hydrogen nuclei). At this stage some interesting things start to happen:

- Firstly, because the in and out energy flow of protons is not restricted unless bitron bonded to another atom, no accumulation of excess energy occurs at outward facing South Pole CESs within the rising number of protons. As the protons no longer emit EMR (the same comments apply to neutrons as well) Hydrogen drops out of emission spectra for the plasma mix: any spectral lines are thus attributable to other contaminants.
- Secondly, because highly energised free electrons and positrons abound, protons are being toggled into neutrons via β^+ decay (see earlier) and then back again into a proton via β^- decay. This ongoing circular process generates a large number of beta rays, neutrinos and anti-neutrinos.
- Thirdly, the ongoing toggling of nucleons between proton to neutron and back again generates a mix of free protons and neutrons in close proximity, which allows them to combine and form new atoms, principally Helium, which, in such a highly energised environment, present as alpha radiation.

Cosmic radiation, the high-energy particles arriving from outer space from distant galaxies, consists mainly of protons (89%), Helium (10%) and a mix of heavier nuclei (about 1%), plus abundant levels of high-energy neutrinos, anti-neutrinos and gamma radiation: a content compatible with that of a completely ionized Hydrogen plasma mix.

Spin and the Orbital Model

Central to the E2M approach is that all concentrated energy sources (CESs) have spin, and that low level electromagnetic energy spirals around them in synch with their spin, with the circular divergent and convergent flows presenting as positive and negative charge electric fields, and the lengthwise flow presenting as a magnetic field. As electric and magnetic fields are formed by the same electromagnetic energy, they are only differentiated by their flow pattern and characteristics associated with those patterns, with E2M there is no need to balance the notional positive and negative point charges to achieve electrical neutrality: only magnetic moments need to be accounted for.

For the conventional Science approach based upon an orbital model for atomic structure, the combination of charge and spin has proven to be more problematic. In the early 1920's the Standard Model consisted of a Bohr-styled model for atomic structure consisting of negatively charged electrons orbiting a positively charged nucleus. **Orbital angular momentum** was accounted for by integer quantum numbers, but it was soon realised that the orbiting electrons possessed magnetic moments associated with their axial spin, which has been called **intrinsic angular momentum**. In the late 1920's Paul Dirac's equations, which factored in relativistic theory, allocated fermions (including electrons) quantum spin $1/2$, $3/2$ etc. to cater for intrinsic angular momentum, with anti-particles having a corresponding negative half spin number.

The **Stern–Gerlach experiment** is used to illustrate that an orbital electron possesses intrinsic angular momentum due to electron spin. The experiment involves sending a beam of silver atoms through an inhomogeneous magnetic field and observing their deflection as shown in figure 38.

Due to random thermal effects in the oven used for the experiment, the magnetic dipole moment vectors of the silver atoms are considered to be randomly oriented in space. Based on classical physics theory, a continuous spread of the atoms in the z direction corresponding to the random spread of magnetic moments could be expected, resulting in the entire lip-like atom distribution of figure 14 to be filled with atom hits: instead the experimental results had no central hits resulting in the open thin-lip distribution as shown.

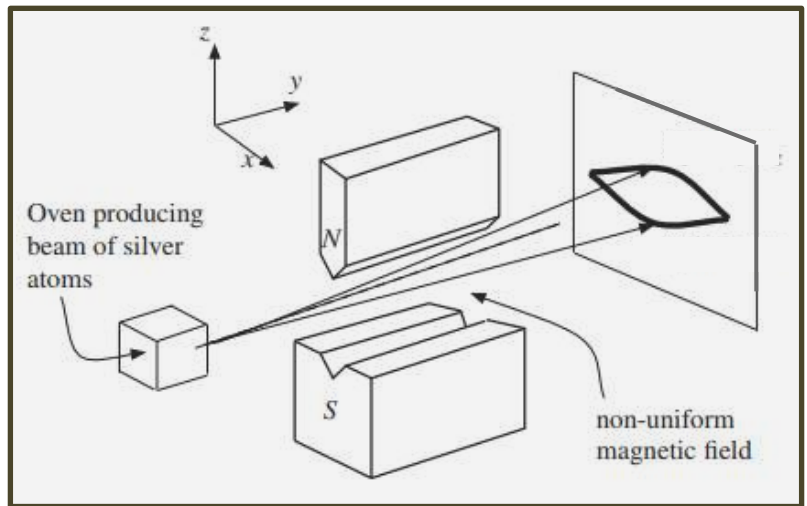


Figure 38: The Stern–Gerlach Experiment Setup

Originally the thin-lip distribution of the atom stream was interpreted that as being due to integer quantised angular momentum as per the quantum theory that then (1922 to 1925) existed. Upon the arrival of Dirac's equations in the late 1920's, the experimental results were re-evaluated and conclusions changed: the deflection pattern was now explained in terms of the intrinsic angular momentum of orbital electrons. According to the orbital model, silver atoms have one unpaired 5s electron, with all others being paired. Thus, as the 5s electron is considered to be in a zero orbital-derived angular momentum state, its quantum spin- $1/2$ results in an up or down state, resulting in an even splitting of the electron stream in the z direction.

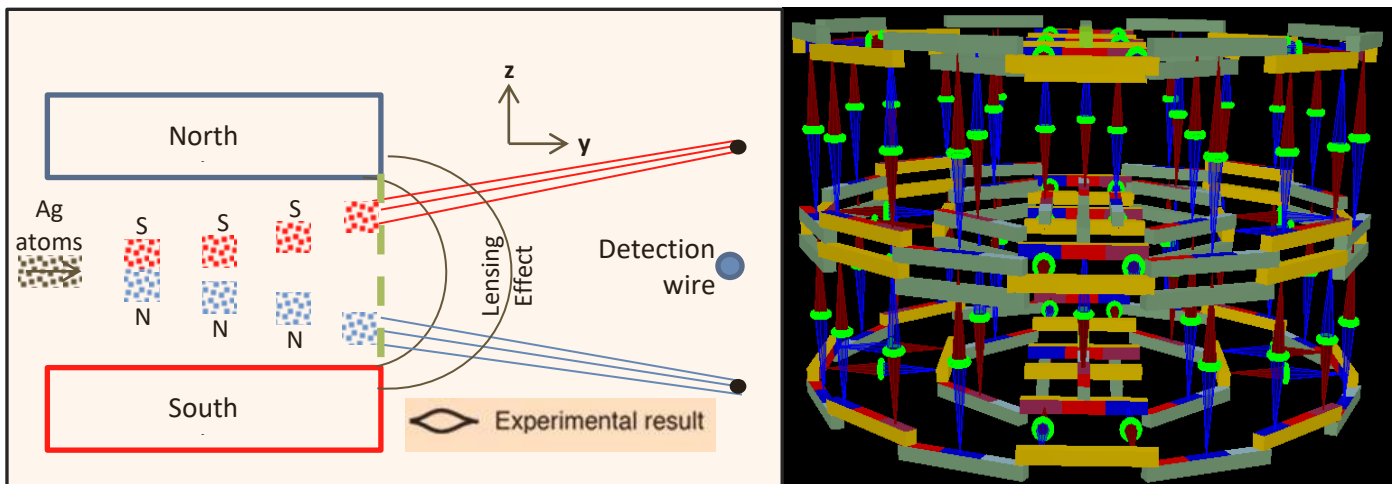


Figure 39: Silver Atom Structure and Deflection Pattern Formation

The E2M model for a silver atom (see right in figure 39) the electromagnetic field of the outer bitron bonds quickly align the atoms as they enter the applied magnetic field so their North-South axial alignment is vertical (i.e. parallel to the z direction). The outer bitron fields are magnetically balanced, but the neutron attached to the top (or bottom) nucleon layer represents unbalanced magnetic fields that cause the atom to move up or down in the z direction, separating into two groups around the central xy plane.

The silver atoms would be expected to spread in a fairly concentrated lens-like distribution band with no atoms centrally, thus producing an open lip-styled pattern. Also, in a manner not dissimilar to the optic focusing effect of a convex lens with light, the convex shape of the outer magnetic field at the atom exit-end of the magnet (dashed green line) would have a lensing effect causing the atoms to converge further (as shown in figure 39) into thinner lines.

Conclusions

Conventional Science is underpinned by the orbital model of the atom, with ongoing research placing great emphasis upon the development of mathematics to quantify and refine that model. The E2M approach is more pragmatic and energy-centric than that of conventional Science, and the resulting atomic structures, as presented in this paper, are significantly different to the orbital model, leading to explanations for many physics-related phenomena, including:

- The physical characteristics of elements
- Different allotropic forms of elements
- Bonding geometry within molecules and chemical compounds
- Beta decay and electron capture
- Electron-positron annihilation
- The ionization process during plasma formation and the Helium content of cosmic radiation
- Gravity (Appendix C)

E2M starts by providing a model for up and down quarks in terms of Preons (CESs), and shows how these build into the nucleons, with an explanation of how protons can convert into neutrons and vice versa.

Electrons are considered to consist of a toroidal core of concentrated energy enveloped by less concentrated energy responsible for their electromagnetic characteristics. The nature of the flow of concentrated core energy within the electron's torus is unknown: it could be a liquid-like flow or consist of fast-moving energy strands, vibrating strings or particles, as any of these options is compatible with the characteristics of an electron within the E2M model.

Apart from having different chirality, electrons and positrons have the same structure and electromagnetic characteristics, making it difficult to distinguish between them when they are contained within solid matter. Elaborate high energy accelerators are required to extract, collimate and axially align electrons and positrons so they can be separated by a magnetic field (e.g. the desk-top positron generator represented in figure 9). At present there is no known way to distinguish between electrons and positrons when they are contained within matter, which causes problems when considering the nature of an electric current within a wire conductor, as discussed in Appendix B.

There may be subtle differences in the structure of CESs, electrons/positrons and photons that are unique to each, but at this stage they are all considered to have a toroidal structure. CESs are the larger and more energetic, and as Preons, represent the primary energy source from which other energy evolves. Electrons and positrons are thus considered to be secondary or derived forms of concentrated energy which form within bitron bonds, some of which are internal to the atom and others external between atoms. Similarly, photons are considered to be derived energy concentrations which are formed and/or captured and released by un-bonded out-facing CESs within an atom.

The most conjectural aspects of E2M are claims that of some electrons are internal to the nucleus and generated by bitron bonds, the source of EMR emissions and Gravity. It is, however, quite feasible that free electrons are captured and held by bitron bonds in the atom-building process rather than being generated from excess bitron bond energy as suggested. Similarly, photons could consist of more wave-like helical concentrations of electromagnetic energy rather than a toroidal form and/or be produced by excess energy within bitron bonds rather than by un-bonded CESs. And the explanation of Gravity may be highly speculative. However, these are all ancillary issues and, should any of these aspects require change, then E2M's basic atomic model would still remain largely intact.

Conventional Science represents the nucleus as a spherical aggregation of nucleons displaying no particular structure of note, with electrons, as negatively charged point particles or wave-like forms, orbiting at the speed of light in

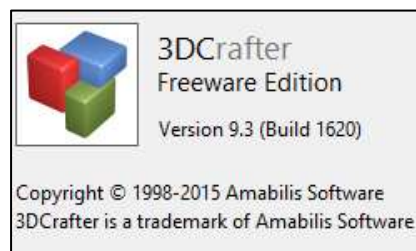
mathematically defined 'spdf' orbital patterns around the positively charged nucleus. Quantum Mechanics has diversified into several research streams, with the wave-particle duality concept, as supported by the Dirac equations, being extended to all matter (including electrons, molecules and atoms). Very little progress has been made by SM research over recent years: it now consists of a large range of elementary particles and has become a far from intuitive model. It is a very fragmented, incomplete model that seemingly provides more questions than answers.

E2M claims that the unique geometric structure an atom's nucleus contributes to its physical form and characteristics. It also provides reasonably detailed explanations across a broad range of phenomena, as summarized above. It is far removed from the orbital model, with the differences between the two being so significant that both approaches cannot be correct: either one is correct (or at least in the right ball-park), or both are incorrect, representing totally misleading, flawed false-positives. The very real dilemma that urgently needs to be resolved relates to which of the approaches represent false-positives, because should a false-positive model be adopted, used and relied upon, then the consequences to the future of Science and mankind would be significant, costly and potentially very embarrassing.

Acknowledgements and References

3D Modelling Software

3DCrafter by Amabilis Software



Technical References

Most technical detail was obtained from **Wikipedia**



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Appendix A: Electromagnetic Radiation

The 'Wave Vs Particle' Debate: Electromagnetic Radiation

E2M contends that a torus model for a photon that is similar in structure to that of an electron. Electromagnetic Radiation (EMR) is considered to consist of multiple photons, and that each photon tumbles and spins around an axis perpendicular to its direction of travel. Each photon's rotating electric and magnetic dipole fields trace as synchronous sinusoidal graphs as shown in figure 40. The peak magnetic field and electric field vector values occur at the same time whenever N- (-q) and S+ (+q) are vertical. A more accurate plot would show the negative and positive electric field plotted as mirror images, reflecting the dipole nature of a photon and the resulting zero net electric charge.

Assuming that the photon's spin frequency to be in the visible light range, figure 40 represents one of multiple photons travelling asynchronously together to form what we call a **beam of light**. The photon's spin speed is dependent upon its torus's large radius that is in turn related to the energy stored within the torus.

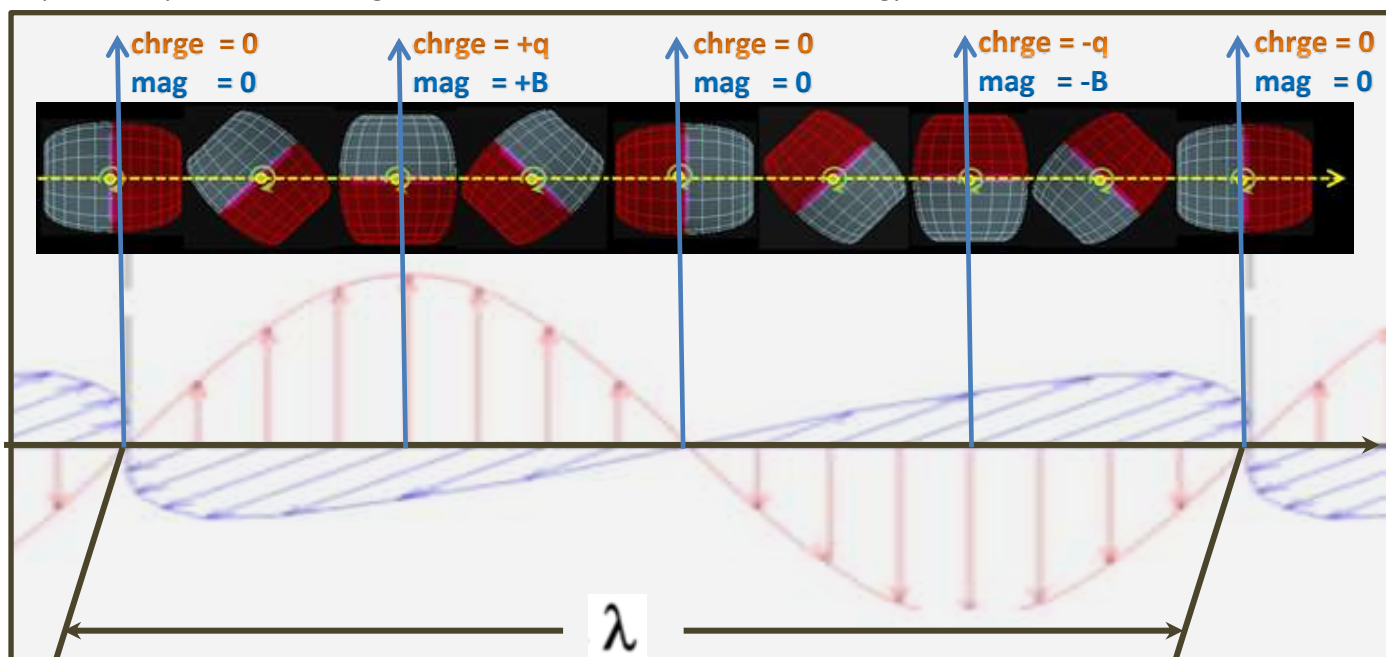


Figure 40: Photon Spin and Electromagnetic Radiation

The particle versus wave debate about the nature of light has been raging since the 18th century. The wave approach has been most useful, allowing the allocation of wavelengths for the complete EMR spectrum from Gamma to long radio wave, complete with a detailed colour map of EMR by wavelength in the visible light range (see figure 41). From the colour of light, its wavelength can be determined and, using Planck's constant and the speed of light, the formula $E=hc/\lambda$ allows the corresponding energy of photons to be calculated.

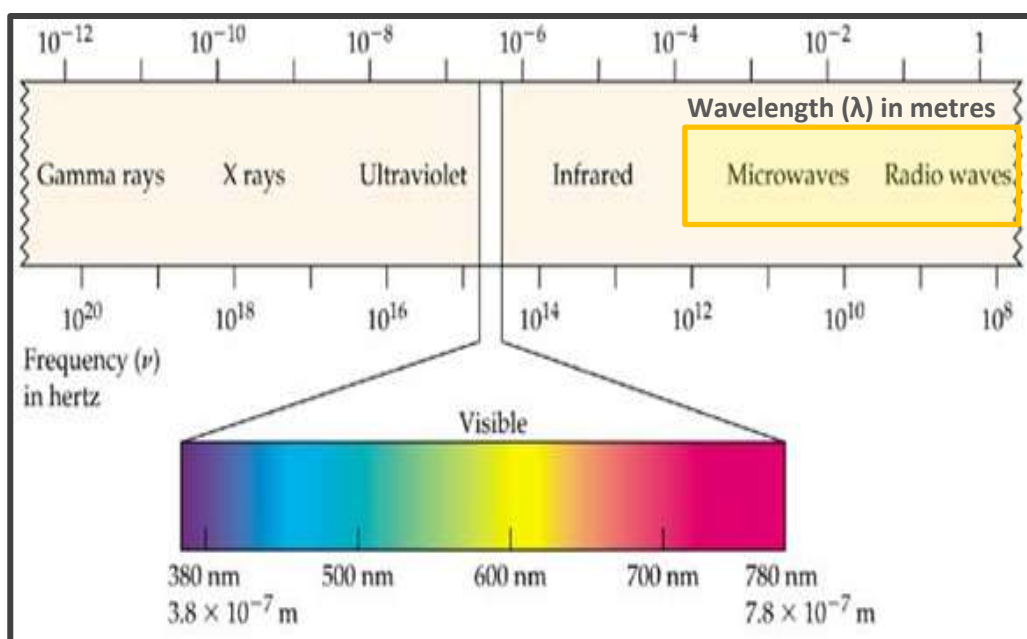


Figure 41: The EMR Spectrum

Because E2M contends that light consists of photons, it implicitly supports the particle side of the 'wave versus particle' debate.

Micro and Radio Waves

Micro and radio waves have frequencies less than 300 GHz corresponding to wavelengths from 1 millimetre upwards. Radio and micro waves can be man-made or result from natural sources. Unlike other frequencies in the EMR spectrum, radio and micro waves have the ability to pass through Earth's atmosphere intact (see figure 42): this plus the ability for data to be encoded and decoded using frequency and amplitude modulation techniques make them invaluable for communication purposes.

The ability of man-made radio waves to pass through the Earth's atmosphere intact, whereas the rest of the EMR spectrum cannot, suggests that they could possibly be different in nature to other EMR.

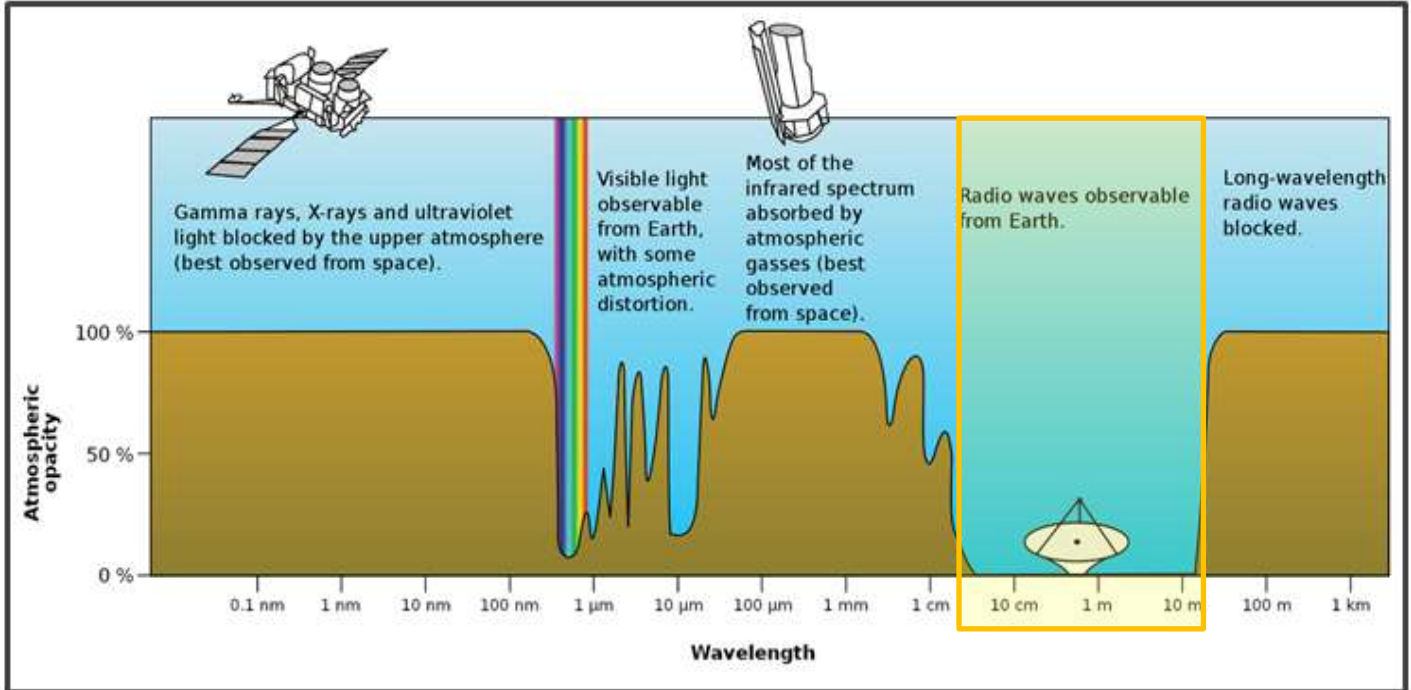


Figure 42: Electron and Positron Electromagnetic Field Patterns

Man-made micro and radio waves are generated by some type of capacitor and inductor loop circuitry that delivers an oscillating current to an aerial. It is the oscillation frequency that dictates the frequency of the emitted EMR. E2M contends that the energy-wave nature of such man-made radio/micro transmission relate to the variation in photon concentration from zero to the maximum concentration in the wave peaks, as shown in figure 43. The photon tori within each concentrated group do not spin: their electromagnetic poles are aligned and their packing density within prevents them from spinning. The result is standing-wave-like electromagnetic energy moving away from the transmission antenna. It is thus distinctly different to the 'normal' EMR such as light and X-ray radiation.

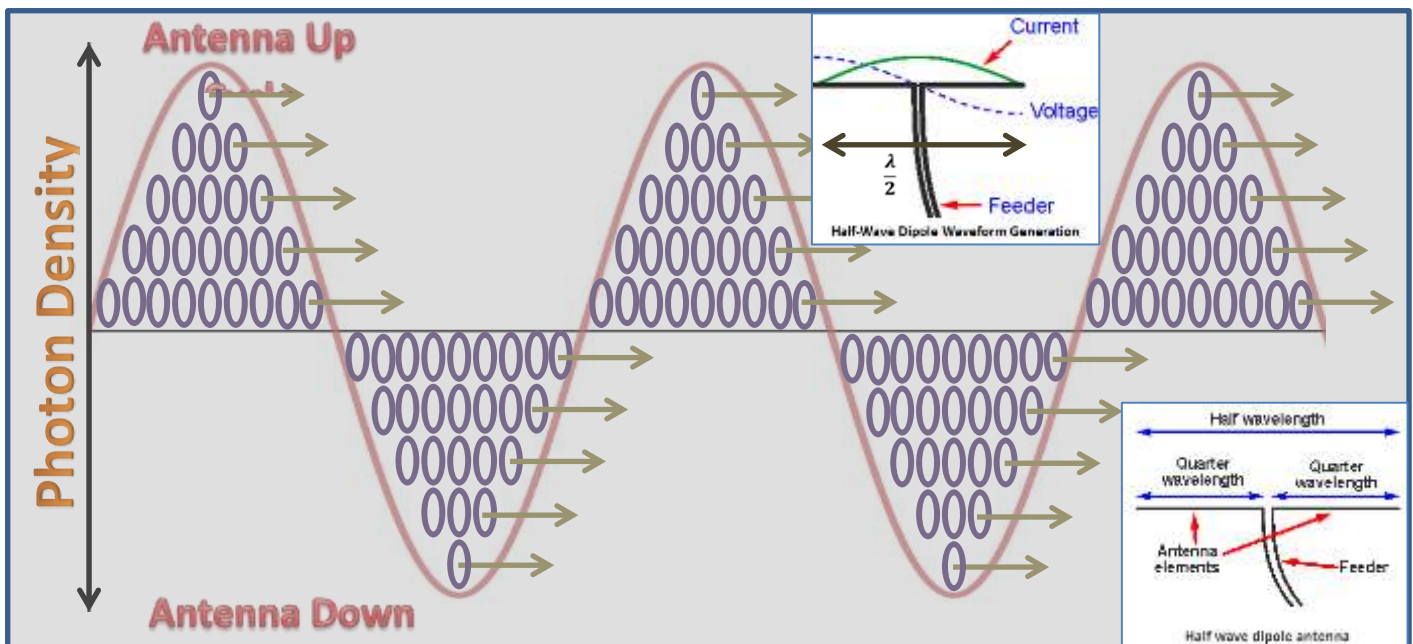


Figure 43: Man-Made Micro and Radio Waves

For a hypothetical perfect radio aerial, the transmitted wave fronts would spread as spherical sets of waves and troughs of different photon densities. Larger power transmitters are simply able to pump out more photons per cycle.

In practice, radio antennae display a range of radiation patterns. The radiating radio waves from a dipole antenna consist of electric field half-waves created by an alternating electric current applied to the antenna's two vertical metal rods. The oscillation frequency of the alternating current switches the antenna rod charge between positive (+) and negative (-), producing successive concentration bands of photons (highlighted in red and blue in figure 44). The Wikipedia animation https://en.wikipedia.org/wiki/File:Dipole_xmting_antenna_animation_4_408x318x150ms.gif shows quite clearly the outwardly radiating electric field waves corresponding to concentrated groups of photons.

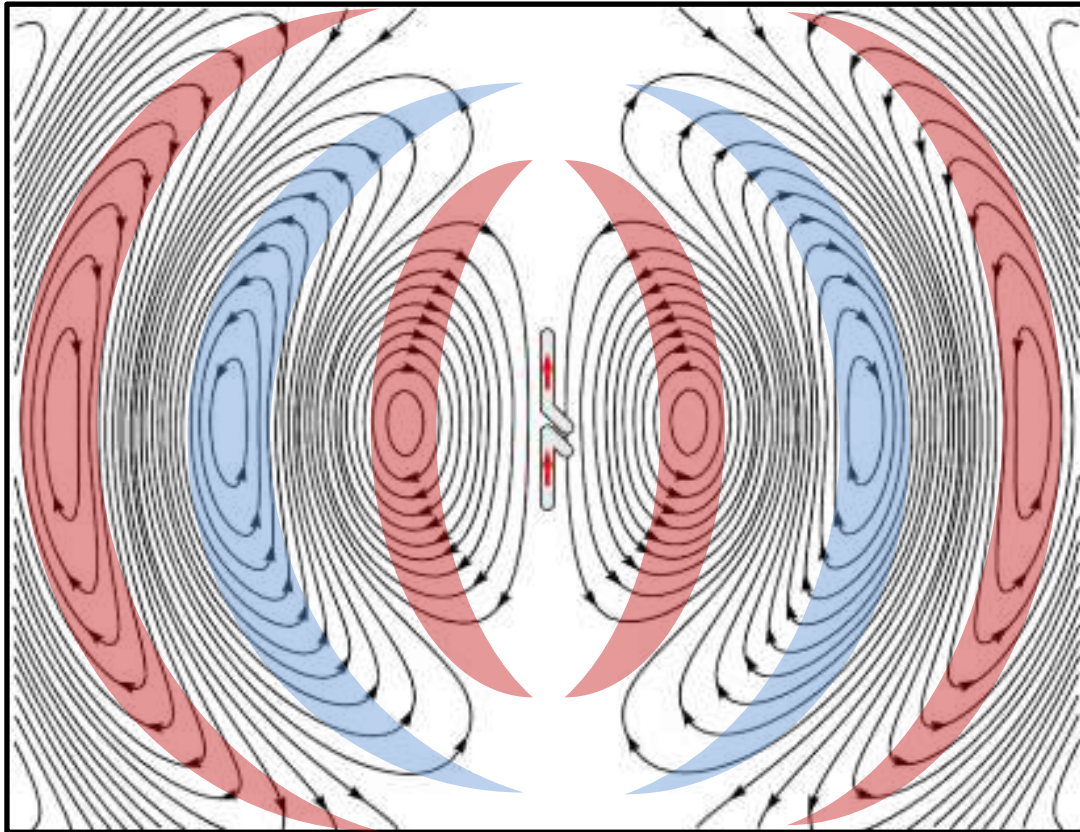


Figure 44: Electron and Positron Electromagnetic Field Patterns

As stated above, man-made micro and radio waves are considered to be different to 'normal' EMR with the corresponding frequency: they represent photons that have been concentrated into robust synthetic standing energy-waves that can be modulated to encode data, and which can be received by an appropriately tuned antenna and demodulated to decode carried data.

In terms of the particle-wave debate, E2M considers that EMR is made up of electromagnetic energy tori that can be considered to be particles: their spin (which is dependent upon the torus's radius and internal energy) determines their frequency; their spin frequency and speed (c , the speed of light) can be used to determine the wavelength of their associated electric and magnetic fields. But attaching wave-like terminology to EMR, no matter how convenient or useful it might be, does not mean that EMR travels as a wave: it can be involved in wave-like radiation but EMR essentially consists of particles. The subtle but important differences between the nature of EMR and man-made micro/radio waves are often overlooked, serving to further exacerbate the particle versus wave debate.

One of the drivers of the particle versus wave debate is that some commonly observed phenomena seemingly can only be explained by the wave approach and others only by the particle approach.

As can be seen in the tabulation right, both approaches can adequately explain reflection and refraction, but the wave approach fails to explain the photoelectric effect, and the particle approach fails to adequately explain polarisation, diffraction and interference.

Because E2M supports the particle approach, it is beholden to explain polarisation, diffraction and interference, which is the subject of the next section.

Phenomenon	Can be explained in terms of waves.	Can be explained in terms of particles.
Reflection	✓	✓
Refraction	✓	✓
Interference	✓	✗
Diffraction	✓	✗
Polarization	✓	✗
Photoelectric effect	✗	✓

The 'Wave Vs Particle' Debate: Polarisation, Diffraction and Interference Patterns

Assuming that the photon spin axis as shown in figure 1 remains stable, it represents **plane polarised EMR** because its oscillating electromagnetic fields consistently oscillate in a particular plane (vertical as shown) that contains its direction of travel.

Should the spin axis itself rotate in the plane perpendicular to its direction of travel, then it creates **circularly polarised EMR** (as shown right with the red arrows representing the rotating spin axis: the changing polarisation plane is perpendicular to the arrows). Should the spin axis not be perpendicular to the direction of travel (implying that the torus axis is not parallel to the direction of travel), then it represents **elliptically polarised EMR**.



Thus the rotating torus model of EMR allows polarisation to be explained in particle terms (*one tick*).

Ordinary white light is considered to be made up of a mixture of photons with a range of polarisation (plane polarised with a range of polarisation angles and/or circularly polarised with a range of polarisation axis rotation rates) and wavelengths (i.e. spin rates).

Constructive and destructive interference of intersecting wavefronts is well understood and documented, and provides a powerful argument in favour of a wave-like nature for EMR, and visible light in particular. There is, however, plenty of wriggle room for the particle approach.

Starting point is the interference patterns derived from a **Michelson Interferometer** setup for parallel merged beams as shown in figure 6. A single beam of **coherent light** (i.e. light with phase synchronised photon spin) is split into two identical beams by a partially reflecting mirror beam splitter. Each of the split beams travels a different path to be re-combined before arriving at a detector. The path length difference of each beam creates a phase difference between them, producing a characteristic concentric interference pattern (insert figure 45).

Whether or not the coherent light as used in the Michelson Interferometer experiments is polarised or not, it will be plane polarised when reflected by the beam-splitting mirror (M1 light at point C, and C' for M2 light). Beams of light, made up of plane polarised photon tori with a wavelength defined by one complete rotation around their spin axis, would have exactly the same phase difference as would waves of electromagnetic energy under the wave model.

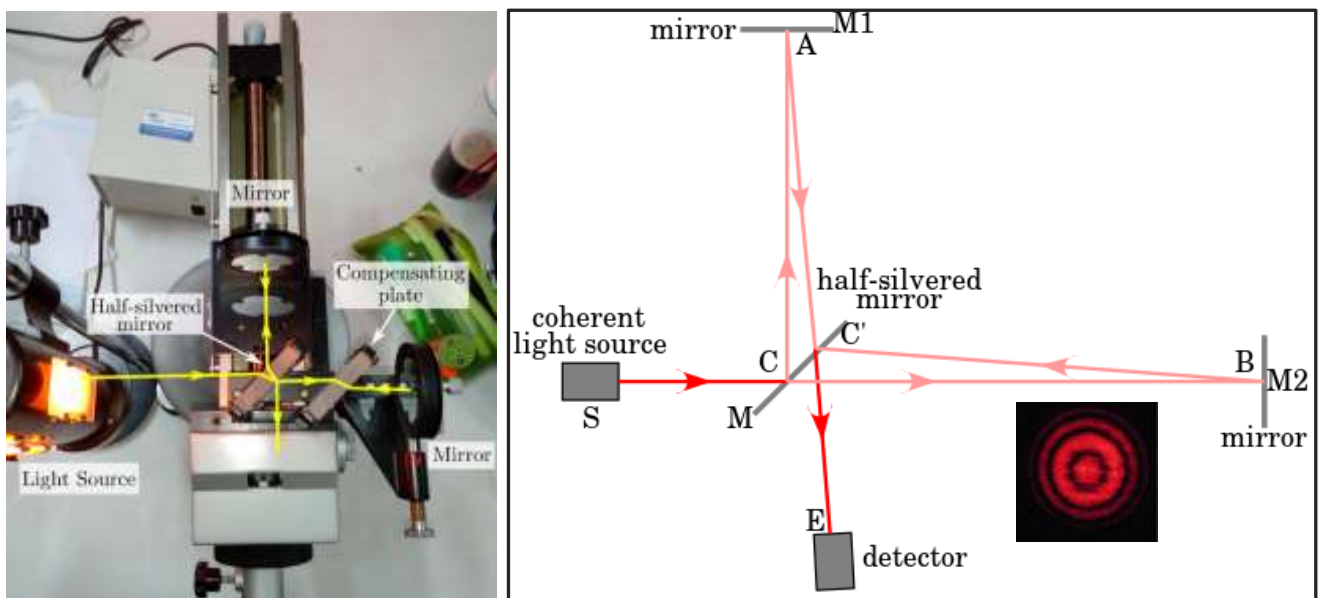


Figure 45: Michelson Interferometer setup: parallel merged beams

When photon tori are very close or superimposed **destructive interference** occurs when the polarisation planes are aligned and both the electric field flow (clockwise and anti-clockwise) and magnetic field flow directions (North to South) oppose each other: viz. S+ & N- (see figure 46), or S- & N+ for their anti-particles. The magnetic fields thus cancel each other, thus and the net electric charge is zero.

When photon tori are very close or superimposed **constructive interference** occurs when the polarisation planes are aligned and the electromagnetic flow directions are in the same direction: viz. N- & N- (see figure 46) or S+ & S+ (or N+ & N+ or S- & S- for their anti-particles): the result is a larger energy density and a brighter point of light.

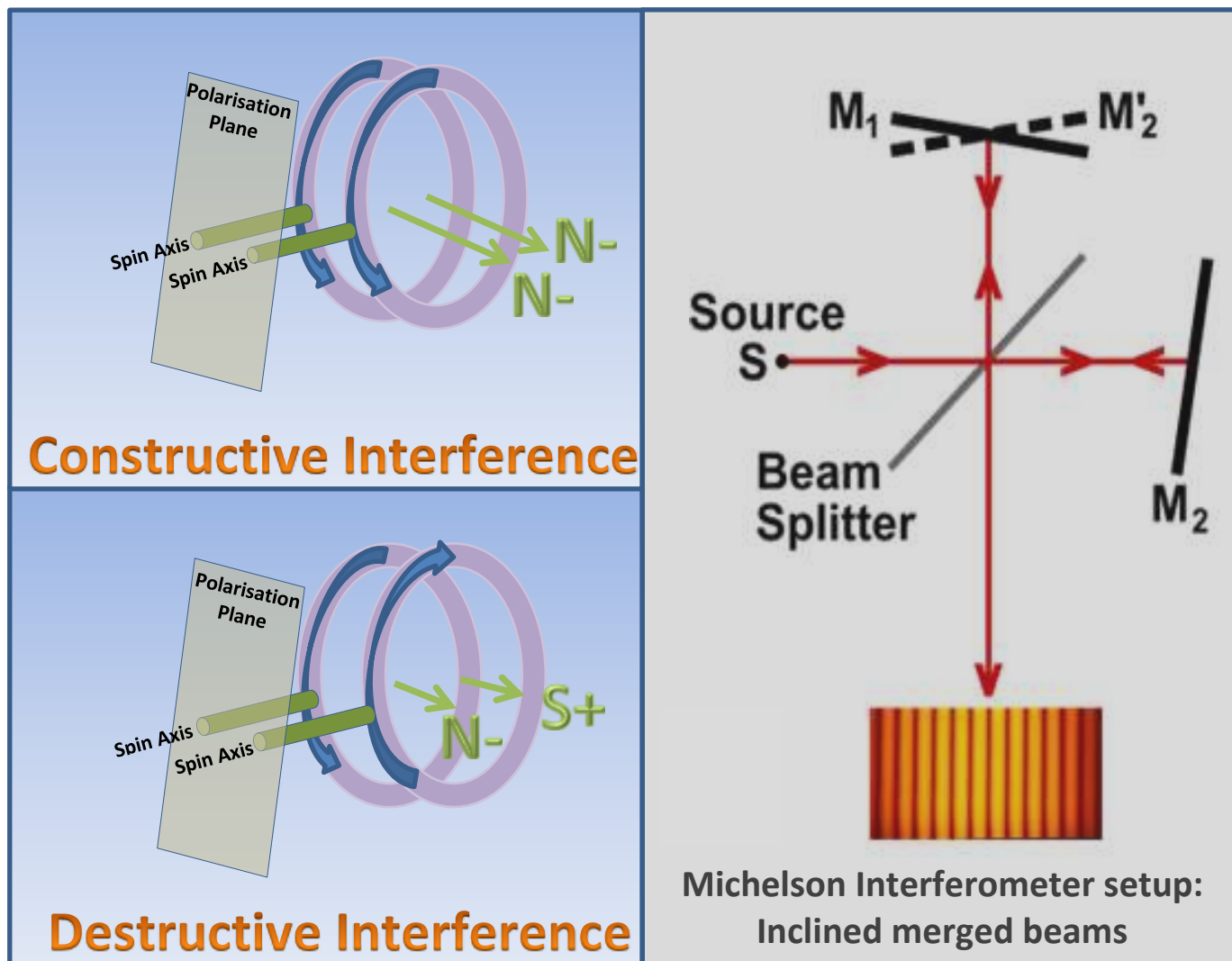


Figure 46: Constructive versus Destructive Interference and Inclined Michelson Interferometer

The reasons for the pattern of the interference patterns for the torus particle approach are exactly the same as for the wavefront model: for the parallel merged beams the virtual images of the light source caused by the beam splitting are one behind the other for the parallel setup, and are side-by-side for the inclined setup. What is different is the explanation for the destructive and constructive interference.

The constructive and destructive interference can thus be explained in torus-particle terms, which provides **another tick** for the particle approach. Now onto diffraction: the vehicle for this explanation is the 2-slit electron experiment.

One of the better documented single and 2-slit experiments is the paper titled ‘Controlled double-slit electron diffraction’, by Roger Bach, Damian Pope, Sy-Hwang Liou and Herman Batelaan (New Journal of Physics Volume 15, March 2013; downloadable at <http://iopscience.iop.org/article/10.1088/1367-2630/15/3/033018>). The referenced paper provides an excellent historical overview of the most significant experimental evidence on the subject since Richard Feynman’s thought experiment concept and is one of the few single/double slit experiments that provide full details (although the backstop distance is missing) of the setup for electron streams and for single electron-by-electron accumulation, together with good clear presentation of the results.

The wall and mask (bottom right of figure 47) were constructed from 100 nm thin silicon-nitride membrane coated with approximately 2 nm of gold, and the slits are 62nm wide separated by 272nm. Whenever a metal (such as gold) and a dielectric (such as air) interface a surface effect is an alternating series of micro positive and negative charges called **Surface Plasmons** (see figure 48). Should the frequency of these coherent delocalized electric oscillations correspond with the spin of an electron torus (which is travelling at about one hundredth the speed of light) then significant deflection can result in opposite directions depending upon how in or out of synchronisation are the two frequencies. The resultant deflection distributions from the top slit are shown in blue, and those from the top slit in red; the theoretical net distribution, a sum of the two, are shown in green in figure 47.

Have a close look at the single electron diffraction distribution (bottom left figure 47): it clearly presents as a banded continuum of dots that is more compatible with overlapping deflection distributions (i.e. the theoretical net distribution graph). It has no bands of total blackout as would be expected from destructive interference.

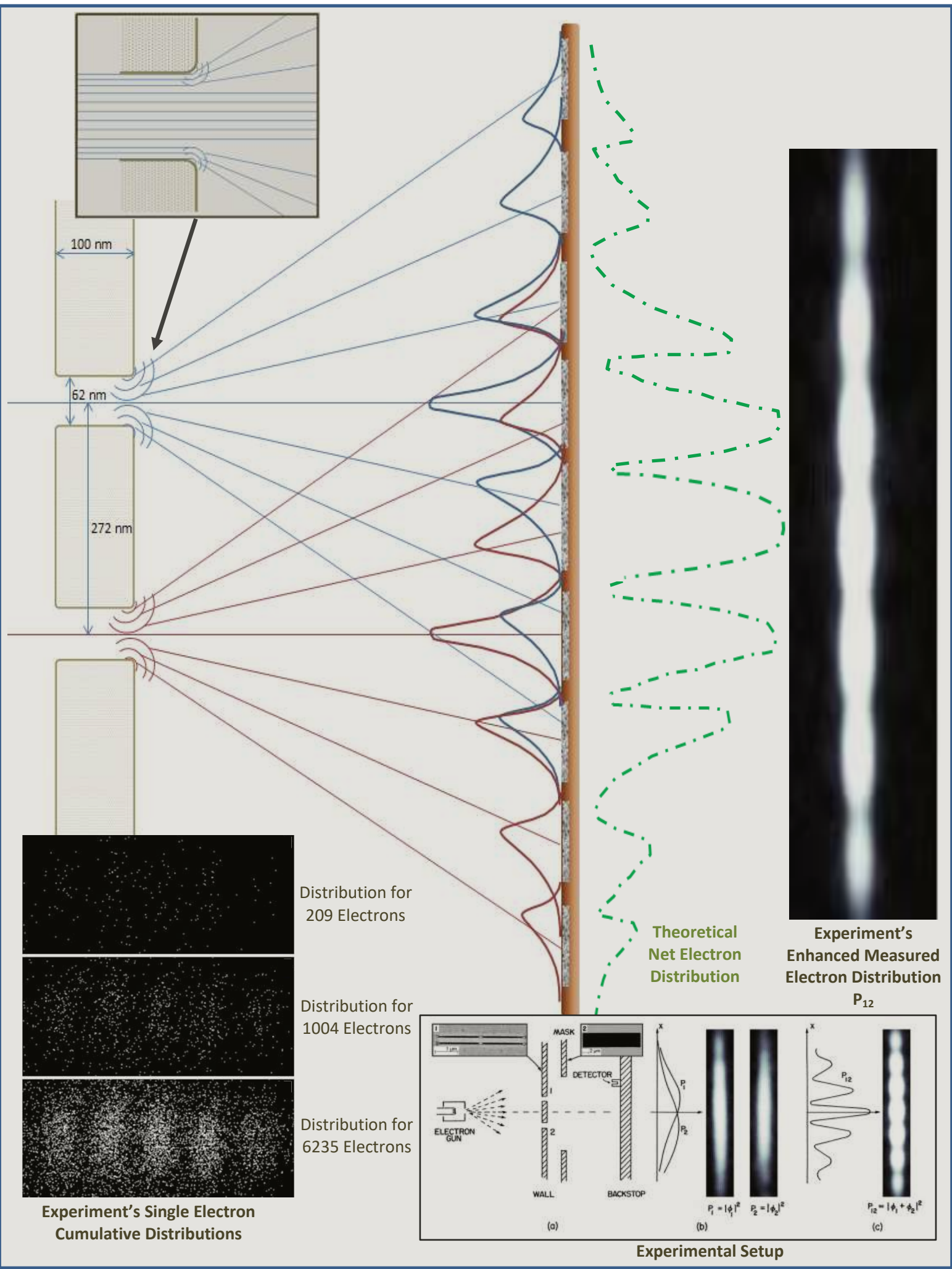


Figure 47: 2-Slit Electron Electromagnetic Deflection and Distribution Pattern

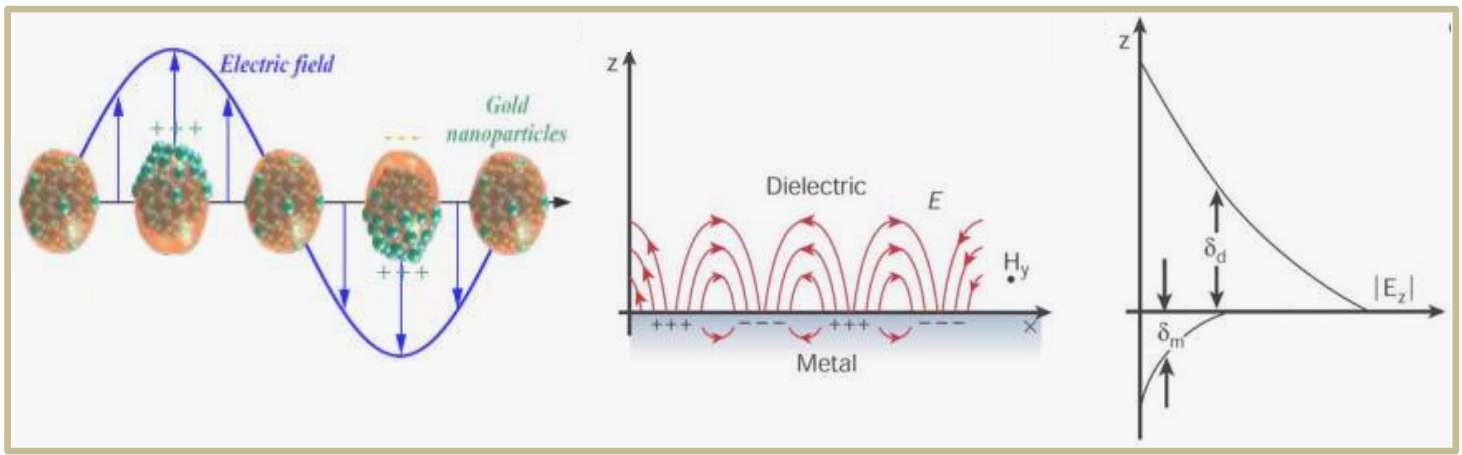


Figure 48: Surface Plasmons

The banding thus results from a **diffraction effect** which is caused by the relative frequency of the slit's surface plasmon charges and the spin speed of the electrons: some are pushed up, some down, some sideways and some show no net change in direction. And if the slits are wider, the effect is greatly reduced. Should destructive and constructive interference be responsible then the dark bands would be quite black rather than the mottled ones of the experiment.

When a beam of electrons (as opposed to single electrons) is used, destructive and constructive interference can take place should collision electron pair orientations correspond to those shown in figure 46. Such interference may slightly enhance the interference pattern, but, as can be seen in the experimental results P_{12} to the right of figure 8, is not sufficient to create distinct black no-go banding due to destructive interference.

When light is used the photon density is greater, the photon particles have less energy and are more fragile, and the polarising effect becomes more prominent: thus for light the destructive and constructive interference effect is increased significantly. Similar rationale can be applied to explain the single slit experimental results.

With a feasible particle-based explanation of the diffraction phenomenon (the **third tick**, as shown yellow highlighted group in the tabulation right), the particle-alone approach can explain all EMR phenomena without having to resort to a particle-wave duality explanation.

Phenomenon	Can be explained in terms of waves.	Can be explained in terms of particles.
Reflection	✓	✓
Refraction	✓	✓
Interference	✓	✓
Diffraction	✓	✓
Polarization	✓	✓
Photoelectric effect	✗	✓

Appendix B: Electricity

Conventional Physics theory defines electricity as the movement of negatively charged electrons within a wire conductor. When electricity flows, magnetic fields and related forces are generated; and electricity can be induced to flow by a magnetic field: this is often referred to as the duality of electromagnetism. The conventional view struggles with the concept of a negative charge moving from a positive to a negative because the accumulation of negative charge at one end of a circuit (or at a capacitor plate or a resistor) should repel approaching negatively charged electrons. The compromise offered in Science texts is to consider the electron to move from negative to positive, or to consider it to be a positive charge moving from positive to negative, or to simply ignore the problem completely.

The only difference between electrons and positrons is their chirality, and it is most difficult, if not impossible, to distinguish between them unless they are moving in free space and their deflection by an electromagnetic field noted (see figure 9). E2M contends that electric current is a measure of the net electric field strength of axially aligned electrons and positrons: electrons with their N- poles pointing in one direction and positrons with their N+ poles pointing in the opposite direction as shown in figure 49.

Movement of electrons and positrons within the conducting medium (e.g. copper wire) is consequential to their alignment. Because they are free electrons and positrons, it is the screw-thread like thrust from their swirling electromagnetic energy fields (b) that causes them to shuffle past each other in opposite directions (see figure 4) at quite low speeds (less than one metre per hour). Electrons, with their N- poles pointing in their direction of travel, form the negative side of the circuit whereas positrons, with their N+ poles pointing in their opposite direction of travel, form the positive. Electric current is a consequence of electron/positron alignment rather than movement.

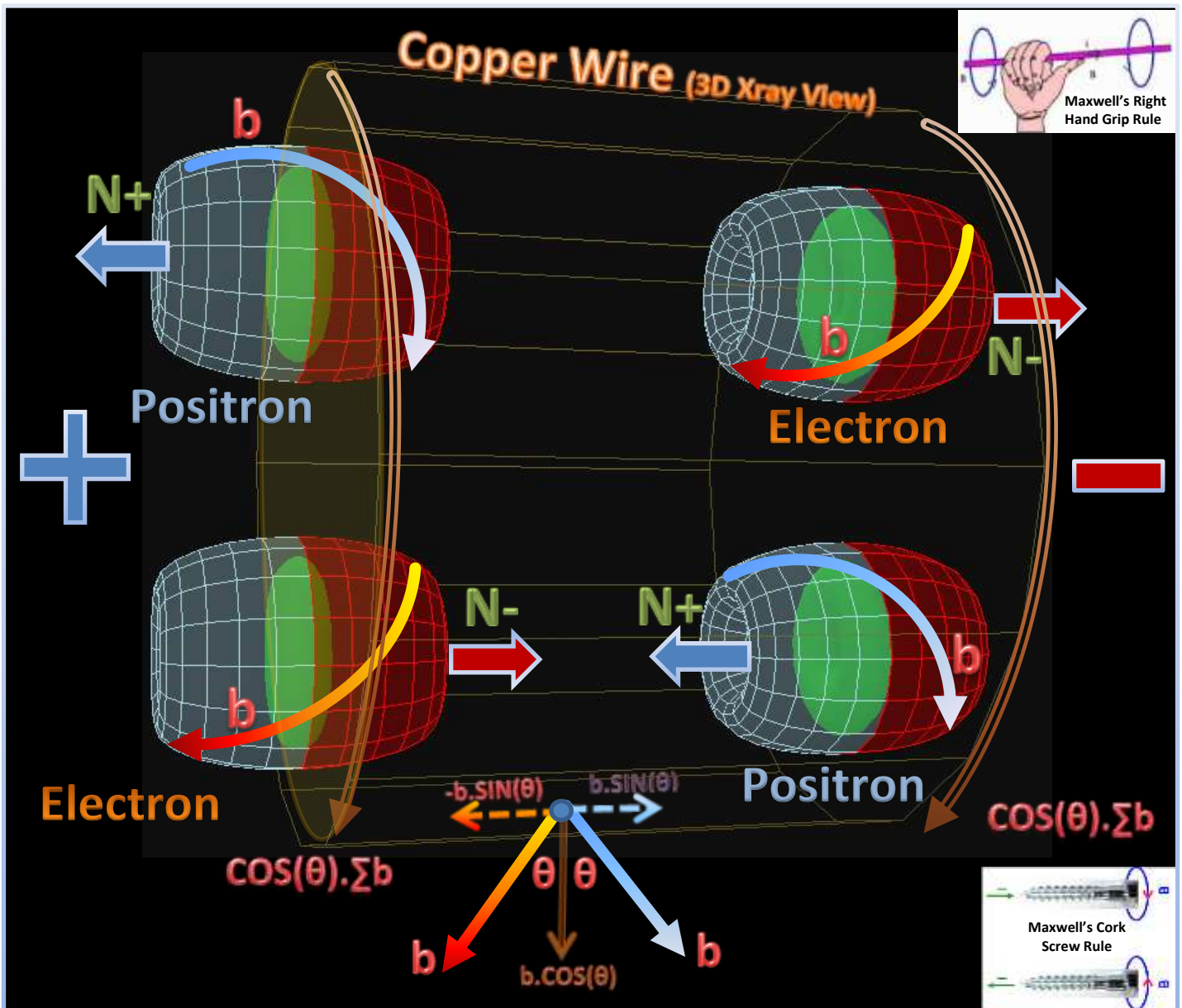


Figure 49: Electric Current: Electron and Positron Movement

Electric **current** (amperes) is a measure of the net electric field strength of the aligned electrons and positrons, and is not related to their speed within the conducting medium, which is consequential to their alignment. That does not mean that electron and positron movement is not important.

The alignment of free electrons and positrons results in a circular magnetic field in the order of $\cos(\theta) \cdot \Sigma b$ around the wire in a direction that conforms with Maxwell's Right-Hand Grip (or Cork Screw) Rule, as shown in figure 4 inserts. Because the magnetic field around the wire is circular, in E2M terms it is more compatible with an electric field energy pattern: however, within a loop or coil circuit it does create a magnetic field as will be covered shortly in the discussion about induced electric currents. The along-wire magnetic field components ($b \cdot \sin(\theta)$) from the aligned electrons and positrons are in opposite directions so as to cancel each other out. Thus the electric potential difference created by electron and positron alignment is the only electromagnetic energy field internal to a wire conductor.

Whenever an obstacle is encountered within an electric circuit, such as a resistor or a capacitor, because of their shuffle movement, electrons build up on one side of the obstruction to create a negative charge; positrons, moving in the opposite direction, create a positive charge on the other side. Thus an electric potential difference is created across the obstacle. Using the resistor example, it limits the number of aligned electrons and positrons that can move through the resistor per unit of time. It is the net electric field strength as defined by the steady state number of aligned electrons and positrons shuffling through the resistor that determines the current. By Ohm's Law, a resistor of R ohms (Ω) and an electric potential difference of V volts results in a current of V/R amperes when the circuit is active.

Dielectrics and Electrical Capacitors

The **dielectric material** used in an electric capacitor is an electric insulator whose free electrons are constrained, unable to shuffle move when aligned by an active electric circuit.

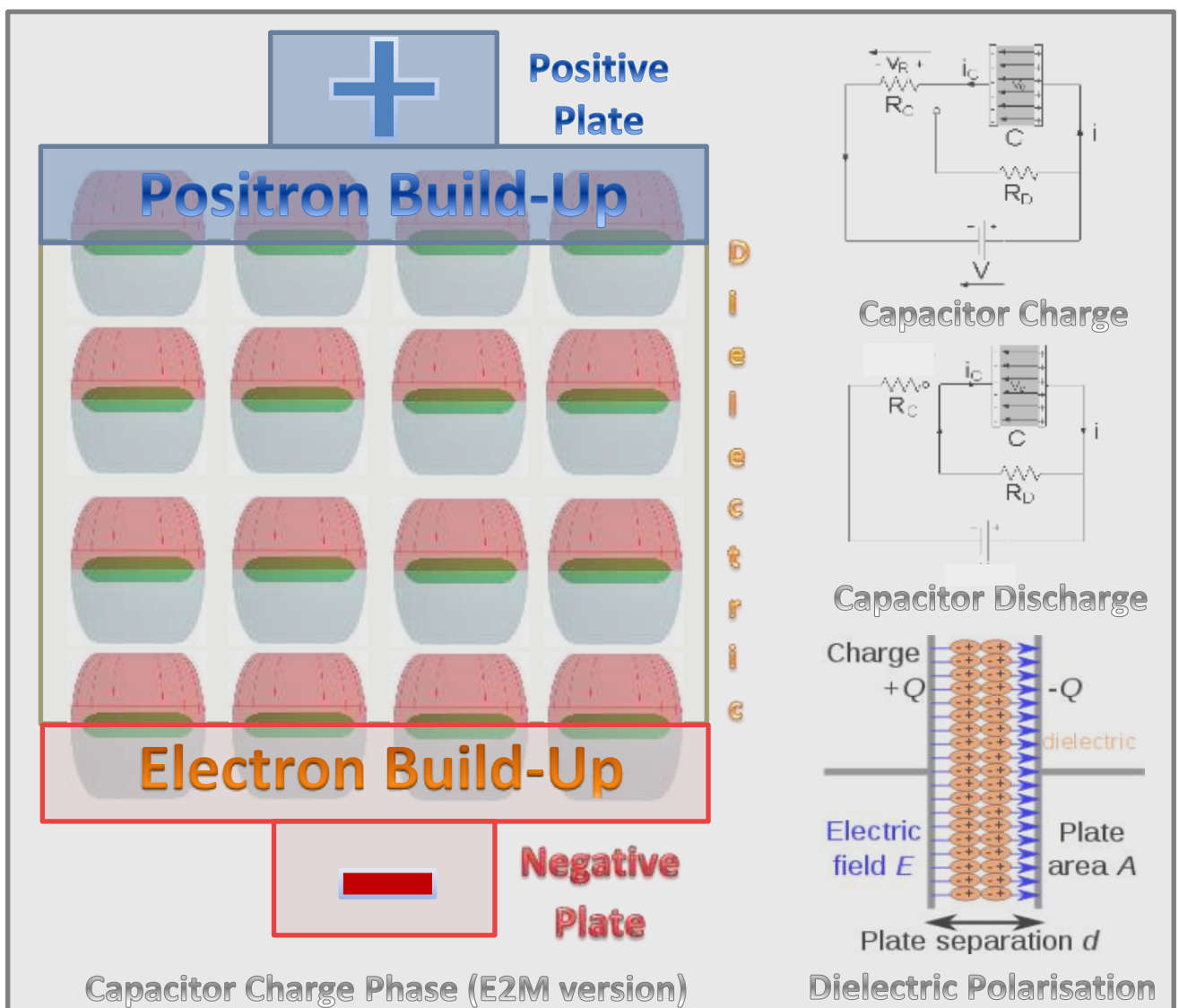


Figure 50: Electric Capacitor Charge and Discharge Phases

During the **charge phase** (figure 50), electrons build up when the dielectric resistor is encountered, forming the negative plate: positrons moving in the opposite direction accumulate forming the positive plate. The voltage generated between the plates causes the alignment of electrons and positrons within the dielectric; which continues to build until their combined voltage equals that of the DC energy source. The capacitor is then full.

For the **discharge phase** the aligned positrons and electrons in the dielectric become the energy source: the dielectric's electrons' aligned positive poles now present as a positive terminal (i.e. the charging phase's negative plate becomes the discharge phase's positive plate). Similarly the other plate presents as the negative terminal, causing the in discharge current to flow in the opposite direction to that of the charging current.

Interestingly, [the Wikipedia definition for a dielectric](#) is 'an electrical insulator that can be polarised by an applied electric field' and their diagram for 'dielectric polarisation' (duplicated bottom right in figure 50) uses electric dipoles, and is remarkably similar to the E2M model. This represents an unintentional endorsement of the E2M explanation.

Chemical DC Electricity Generation and Recharge

The **Galvanic Cell** of figure 6(a) shows Direct Current (DC) electricity generation by chemical reaction (i.e. a chemical DC battery). The electrolyte releases a mixture of free electrons and positrons attached to cadmium atoms in the anode to create cadmium cations in solution.

The newly freed electrons and positrons create an emf across the wire connecting to the copper cathode, causing the shuffle migration of free positrons towards the positive cathode. In parallel with the chemical reactions at the anode, at the positive cathode free positrons (and some electrons) attach to copper cations in solution to deposit copper metal at the cathode.

Keep in mind that the only difference between an electron and a positron is their chirality as defined by the swirl direction of their electromagnetic fields (i.e. clockwise or anti-clockwise): both can attach to and be released from atoms via chemical reactions. Neither electrons nor positrons are point charges, albeit convenient to mathematically treat them so, particularly in regards to the mathematical models for atom structure based upon the assumption of a positively charged nucleus balanced electrically by negative orbital electrons. E2M contends that both electrons and positrons consist of a central torus core of concentrated energy that generates an external electromagnetic dipole field as represented in the figures in this article. They are very, very similar to each other: so much so that it is quite difficult to distinguish between them except by the manner in which they inter-react.

For the **Electrolytic Cell** of figure 51(b), when a DC current (shown as 0.74volts from an AC-to-DC converter) is applied electrons and positrons are axially orientated along the wire conductor; electrons start to shuffle towards the negative cadmium cathode; and positrons start to shuffle towards the positive copper anode. The concentration of free electrons at the cathode end attracts cadmium ions in solution, which readily accept available, free electrons (and some positrons), depositing cadmium metal on the cathode. The reverse occurs at the anode, with the concentration of positrons causes the release of electrons (and some positrons) from the metallic copper atoms to create aqueous copper cations, with the extra electrons shuffling towards the cathode to continue cadmium side of the equation.

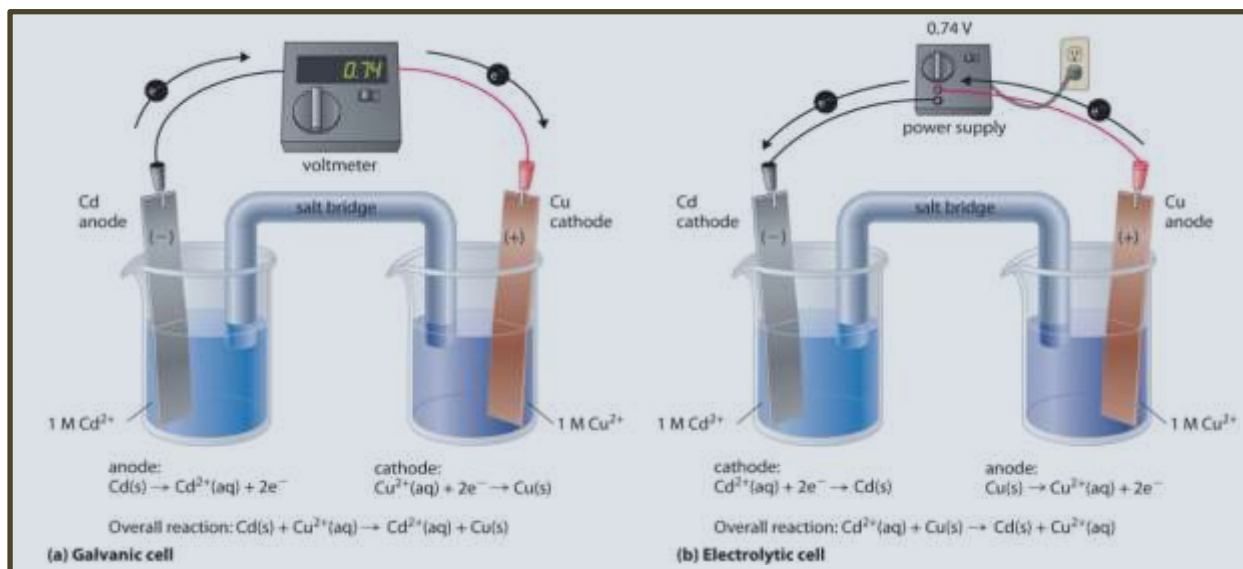


Figure 51: Galvanic and Electrolytic Cells

Faraday's Law of electromagnetic induction states that whenever a conductor is forcefully moved in an electromagnetic field, an electromagnetic force is induced which causes a current to flow. And the E2M opinion is that the current flow represents the synchronous movement of aligned free electrons and positrons within the conductor.

In figure 7 the movement (at speed v) of conductor rod (PQ) through an external magnetic field (B) generates an anti-clockwise electric current (I) in the circuit.

The middle diagram (of the lower 3 in figure 52) shows that when the rod is stationary within a magnetic field, the flux deflects around the rod; the electrons and positrons are in effect shielded from the external magnetic field and thus no current results.

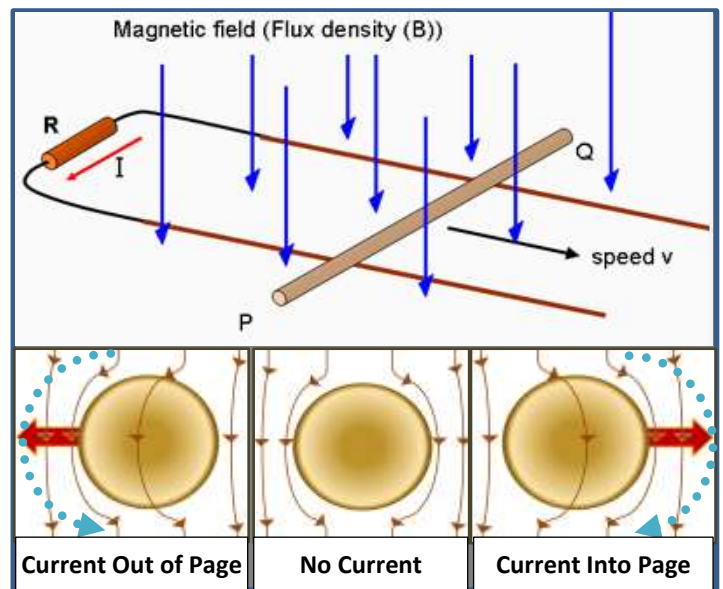


Figure 52: Magnetic Field Induced Current

When the rod moves to the right as shown in the right-most diagram, the magnetic flux is forced to pass through the rod, causing the free electrons and positrons to align, generating an electric current in the rod and thus the circuit. The alignment causes the electrons to shuffle in a direction into the page, creating a current direction that corresponds to Maxwell's Right-hand rule plus a clockwise circular magnetic field (shown as a blue dashed arc).

When the rod moves to the left (left-most diagram), the free electrons and positrons align in the opposite direction, reversing the electric current and circular magnetic field directions.

When the direction of the external magnetic field is reversed, the direction of the induced current and circular magnetic field is also reversed for each rod movement direction.

The E2M model provides a simple, consistent causative explanation for

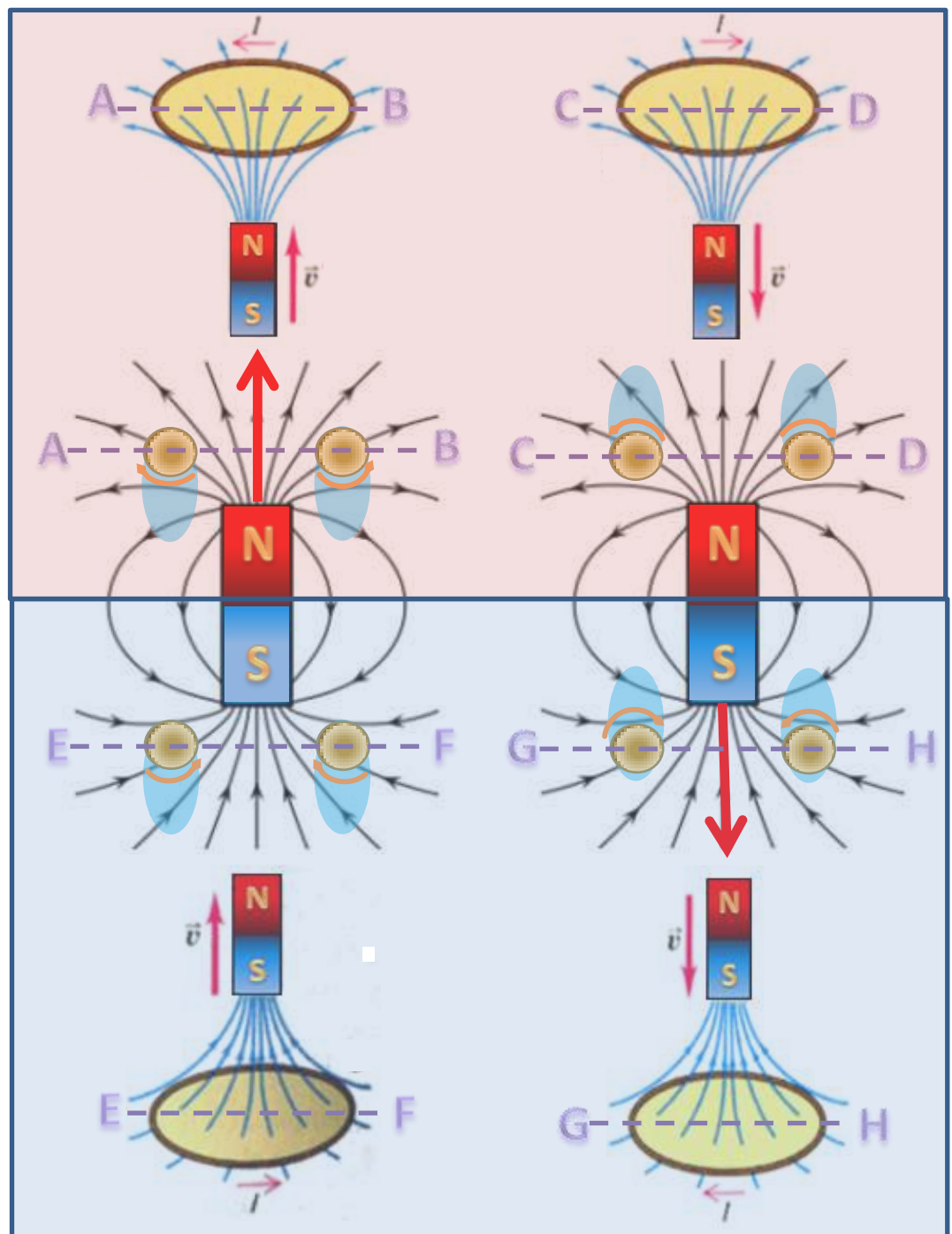


Figure 53: Magnet-Motion Induced Current within a Loop

induced current, its direction and the associated circular magnetic field. Current Science texts and associated articles provide descriptive explanation rather than a causative explanation of these phenomena. They provide is a series of inter-related rules and conventions but little real explanation, which adds mystery and confusion (just ask any high school student) around electromagnetic induction and electricity.

The electromagnetic induction of a loop circuit caused by moving a magnet towards and away from it in different N-S polar orientations is shown in figure 53.

The key to understanding the direction of the loop current and the associated circular magnetic field is to look at the cross-sectional view of the loops (AB, CD, EF and GH).

The blue oval annotation attached to each loop cross-section indicates the direction from which the magnetic flux approaches: the flux passes through the loop, aligning the free electrons and positrons therein and generating a circular magnetic as shown by the orange arc arrows.

The direction of the induced loop current can be confirmed for each of the 4 scenarios by using Maxwell's right-hand grip rule. There is no mystery here – just simple Physics.

As mentioned earlier, the induced circular magnetic field that forms around a wire conductor is considered by E2M to represent an electric field pattern: thus electricity in a wire conductor is not considered to have a magnetic component. However, the electric field convergence inside a loop current a loop current, as shown in figure 54, generates a magnetic field pattern that is similar to that of a dipole magnet or to an electron or positron (except that it does not have their swirling spiral pattern). This serves to highlight the subtlety of electric and magnetic field duality and the transition from one to the other.

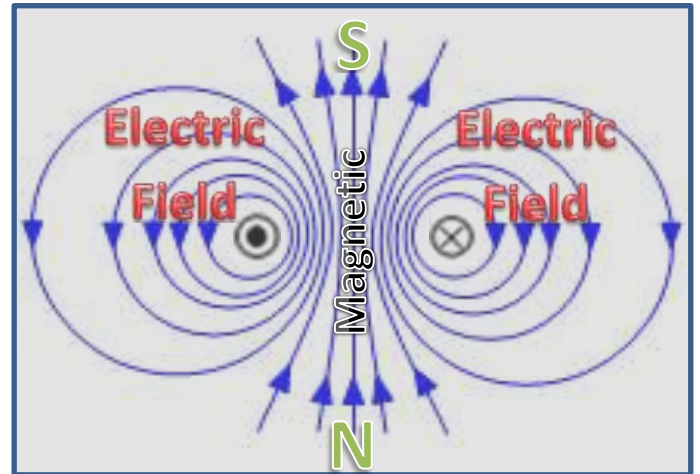


Figure 54: Electromagnetic fields around a Loop

The main difference between the conventional view and that of E2M is that E2M considers an electric current to involve the two-way shuffle movement of electric dipoles (electrons and positrons).

For alternating current (AC), created by switching current direction at a specific frequency, the E2M view considers that the aligned free electron and positron tori are simply flipped 180° to reverse the electric current for each cycle. In other respects E2M and the conventional view are fully compatible, and as the techniques for creating AC electricity from induced currents (e.g. by rotating a U-shaped loop through a magnetic field) are well documented, as are the ways of creating and managing DC electricity: consequently they will not be addressed in this series of articles.

Appendix C: The Pull of Gravity

Not all electromagnetic energy leaving CESs, electrons, positrons and photons via a North Pole returns to the corresponding South Pole: very small quantities of the normally connected swirling electromagnetic energy is leaked and 'lost'. Although much of the 'lost' energy is absorbed by other adjacent electromagnetic fields, a small amount still manages to escape and accumulate as a very weak pool of 'stagnant' energy around the object from which it escaped. E2M contends that low-level leakage energy accumulates **atmosphere**-like around all normal matter: thus it has been called an **Enersphere**.

For nucleons within an object's atomic structure, electromagnetic energy returning via the converging funnel-like South Pole CESs provides a positive pull on the enersphere immediately surrounding the parent object and sucks in a miniscule amount of leakage energy. E2M contends that the suction-like pull of the out-facing South Pole CESs results in a minute inwardly-directed force, the nett sum of which totalled over the billions of nucleons (1 billion = 10^9 , and one litre of water contains approximately 2×10^{27} or 2 octillion nucleons) within an object equates to the pull of **Gravity**.

The more nucleons that an object contains, then the larger its enersphere would be and the more out-facing South Pole CESs it would have that would contribute to its gravitational pull. Thus it is the total number of nucleons within an object that dictates its **mass** (implicit in $E=mc^2$), and, in relation to Earth, it is the size of an object's enersphere and its location within Earth's enveloping enersphere that dictates its **weight**.

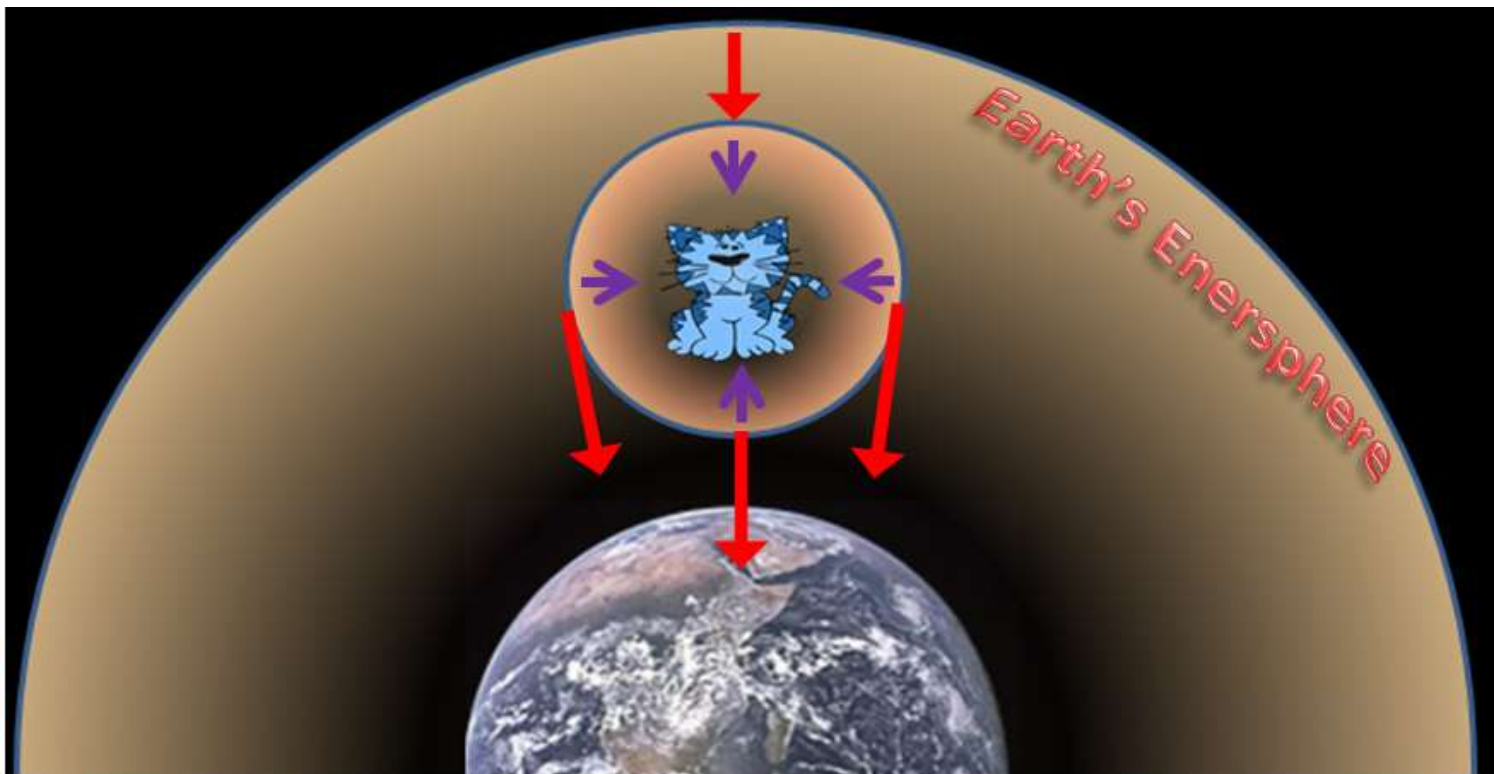


Figure 55: Model for Earth's Gravity

Figure 55 represents a cat sitting with its enersphere fully within the Earth's enersphere. The purple arrows show the direction of the forces pulling the cat's enersphere towards the cat, and the red arrows the considerably greater forces pulling the cat's enersphere towards the surface of the Earth. The cat is heading earthwards, free falling under the influence of Gravity.

The depiction of figure 55 is idealised because the shown bubble boundary surface does not exist - when one object's enersphere overlaps another's they simply merge into one, with the energy levels of each enersphere adjusting appropriately to the other. The suction-like pull of the out-facing South Pole CESs within each object (here the cat and Earth) would continue to pull energy from the resource-in-common, their merged enerspheres. The forces acting in the zone of the cat, however, would be similar to those shown, as would be their nett effect.

At the macro level, enerspheres of large bodies extend well into space (e.g. the Moon, Earth, the Sun, the stars and galaxies) intersecting and acting together to create Gravity pull between systems. The huge, far reaching enerspheres

of large objects in space suggests that their enerspheres could possibly be in expansion mode, albeit slowly. It is difficult to imagine any region of space without enersphere energy, even if it is only from the vapour trails of photons.

Enerspheres are a remarkable bi-product of the way energy combines to make atoms and of atoms to make matter. E2M contends that the pull of Gravity is due to the retrieval of enersphere energy rather than being a mysterious external force or a result from the warping of **space-time**.

Historical Note. René Descartes, famous for his philosophical assertion *cogito ergo sum* ("I think, therefore I am"), in 1644 proposed that **aether**, the medium then considered to separate objects and matter (cf. space), is filled with vortices whose **inward pressure** is '*nothing else than gravity*'.



Place Descartes's vortices within matter (i.e. within nucleons) and then consider the aether immediately surrounding matter to be enersphere energy, then philosophically speaking Descartes' 374 year old explanation for gravity was not far removed from that of E2M.

It should also be noted that, at this stage, E2M does subscribe to the notion that aether is the substance or medium in space that supports the propagation of EMR in a wave form.