

Fine-structure constant from the madelung constant

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Abstract

In this short work we will study the simple and absolutely accurate expression for the fine-structure constant with the madelung constant $b_2(2)$:

$$\alpha^{-1} = -2 \cdot 3 \cdot 11 \cdot 41 \cdot 43^{-1} \cdot b_2(2)$$

This simple expression for the fine-structure constant in terms of the Archimedes constant n is:

$$\alpha^{-1} = 2 \cdot 3 \cdot 11 \cdot 41 \cdot 43^{-1} \cdot \ln 2 \cdot n$$

Also we will present the continued fraction for the fine-structure constant.

Keywords

Fine-structure constant , Dimensionless physical constants , Madelung constant , Archimedes constant

1. Introduction

One of the most important numbers in physics is the fine-structure constant α which defines the strength of the electro-magnetic field. It is a dimensionless number independent of how we define our units of mass, length, time or electric charge. A change in these units of measurement leaves the dimensionless constant unchanged. The number can be seen as the chance that an electron emits or absorbs a photon. It's a pure number that shapes the universe to an astonishing degree. Paul Dirac considered the origin of the number «the most fundamental unsolved problem of physics». The constant is everywhere because it characterizes the strength of the electromagnetic force affecting charged particles such as electrons and protons. Many eminent physicists and philosophers of science have pondered why α itself has the value that it does, because the value shows up in so many important scenarios and aspects of physics. Nobody has come up with any ideas that are even remotely convincing. A similar situation occurs with the proton-electron mass ratio μ , not because of its ubiquity, but rather how chemistry can be based on two key electrically charged particles of opposite electric charge that are opposite but of seemingly identical magnitude while their masses have a ratio that is quite large yet finite. These two questions have a huge bearing on why physics and chemistry behave the way they do. The product of the two quantities appears, at least at first glance, not to be so important.

The Madelung constant is used in determining the electrostatic potential of a single ion in a crystal by approximating the ions by point charges. It is named after Erwin Madelung, a German physicist. Because the anions and cations in an ionic solid attract each other by virtue of their opposing charges, separating the ions requires a certain amount of energy. This energy must be given to the system in order to break the anion-cation bonds. The energy required to break these bonds for one mole of an ionic solid under standard conditions is the lattice energy. The Madelung constant is also a useful quantity in describing the lattice energy of organic salts. Izgorodina and coworkers have described a generalized method (called the EUGEN method) of calculating the Madelung constant for any crystal structure.

2. Measurement of the fine-structure constant

The fine-structure constant is defined as:

$$\alpha = \frac{q_e^2}{4 \cdot n \cdot \epsilon_0 \cdot \hbar \cdot c}$$

The 2.018 CODATA recommended value of α is:

$$\alpha = \frac{q_e^2}{4 \cdot n \cdot \epsilon_0 \cdot \hbar \cdot c} = 0.0072973525693(11)$$

With standard uncertainty $0,000000011 \times 10^{-3}$ and relative standard uncertainty $1,5 \times 10^{-10}$. For reasons of convenience, historically the value of the reciprocal of the fine-structure constant is often specified. The 2.018 CODATA recommended value is given by:

$$\alpha^{-1} = 137,035999084(21)$$

With standard uncertainty $0,000000021 \times 10^{-3}$ and relative standard uncertainty $1,5 \times 10^{-10}$. There is general agreement for the value of α , as measured by these different methods. The preferred methods in 2.019 are measurements of electron anomalous magnetic moments and of photon recoil in atom interferometry. The most precise value of α obtained experimentally (as of 2.012) is based on a measurement of g using a one-electron so-called "quantum cyclotron"

apparatus, together with a calculation via the theory of QED that involved 12.672 tenth-order Feynman diagrams:

$$\alpha^{-1} = 137,035999174(35)$$

This measurement of α has a relative standard uncertainty of $2,5 \times 10^{-10}$. This value and uncertainty are about the same as the latest experimental results. Further refinement of this work were published by the end of 2.020, giving the value:

$$\alpha^{-1} = 137,035999206(11)$$

with a relative accuracy of 81 parts per trillion. We proposed in [8] the exact formula for the fine-structure constant α in terms of the golden angle, the relativity factor and the fifth power of the golden mean:

$$\alpha^{-1} = 360 \cdot \varphi^{-2} - 2 \cdot \varphi^{-3} + (3 \cdot \varphi)^{-5} \quad (1)$$

with numerical value:

$$\alpha^{-1} = 137,035999164 \dots$$

3. The search for mathematical expression for the fine-structure constant

The mystery about the fine-structure constant is actually a double mystery. The first mystery – the origin of its numerical value – has been recognized and discussed for decades. The second mystery – the range of its domain – is generally unrecognized.

– M. H. MacGregor (2007). The Power of Alpha.

When I die my first question to the Devil will be: What is the meaning of the fine structure constant?

– Wolfgang Pauli

The fine-structure constant plays an important role in modern physics. Yet it continues to be a mystery as to exactly what it represents and why it has the mystical value it has. The purpose of many sciences is to find the most accurate mathematical formula that can be found in the experimental value of fine-structure constant.

The elementary charge of electron e was proposed by Stoney in 1.894 and discovered by Thomson in 1.896, then Planck introduced the energy quanta $h\nu$ in 1.901 and explained it as photon $E=h \cdot \nu$ by Einstein in 1.905. Planck first noticed in 1.905 that e^2/c and h have the same dimension. In 1.909, Einstein found that there are two fundamental velocities in physics: c and e^2/h requiring explanation. He said, "It seems to me that we can conclude from $h=e^2/c$ that the same modification of theory that contains the elementary quantum e as a consequence, will also contain as a consequence the quantum structure of radiation." Albert Einstein was the first to use a mathematical formula for the fine-structure constant α in 1909. This expression is:

$$\alpha = 7 \cdot \pi / 3.000$$

With numerical value $\alpha = 0.00733038286$ with an error accuracy of 0.45%. Later many scientists used other mathematical formulas for fine-structure constant but they are not at all accurate. These are Jeans 1.913, Lewis Adams 1.914, Lunn in 1.922, Peirles in 1.928 and others. Arthur Eddington was the first to focus on its inverse value and suggested that it should be an integer, that the theoretical value is $\alpha^{-1} = 136$. In his original document 1.929 he applied the value:

$$\alpha^{-1} = 16 + 1/2 \times 16 \times (16-1) = 136$$

However, the experiments themselves consistently showed that $\alpha^{-1} = 137$. This forced him to look for an error in his original theory. He soon came to the conclusion that:

$$\alpha^{-1} = 137$$

He thus argued that the extra unit was a consequence of the initial exclusion of every elementary particle pair in the universe. In the document of 1.929, Eddington considered that fine-structure constant relates in a simple way to the cosmological constants, as given by the expression:

$$\alpha = 2 \cdot \pi \cdot m \cdot c \cdot R_E / h \cdot \sqrt{N^*}$$

where: N^* the cosmic number, the number of electrons and protons in the closed universe. Eddington always kept the name and the symbol α :

$$\alpha = h \cdot c / 2 \cdot \pi \cdot q e^2$$

The first to find an exact formula for the fine-structure constant α was the Swiss mathematician Armand Wyler in 1,969. Based on the arguments concerning the congruent group, the group consists of simple Lorentz transformations such as the space-time dimensions that leave the Maxwell equations unchanged. The first form of the Wyler constant type is:

$$a_w = (9/16 \cdot \pi^3) \cdot (\pi/5!)^{1/4}$$

With numerical value $a_w = 1/137,036 = 0,00729735252$. At the time it was proposed, they agreed with the experiment to be within 1.5 ppm for the value a^{-1} .

4. Fine-structure constant with the madelung constant

We proposed in [8] the simple and absolutely accurate expression for the fine-structure constant in terms of the Archimedes constant π :

$$a^{-1} = 2 \cdot 3 \cdot 11 \cdot 41 \cdot 43^{-1} \cdot \ln 2 \cdot \pi \quad (2)$$

$$a^{-1} = (2.706/43) \cdot \ln 2 \cdot \pi \quad (3)$$

$$a^{-1} = [(14^3 - 38) \cdot 43^{-1}] \cdot \ln 2 \cdot \pi \quad (4)$$

The equivalent expressions for the fine-structure constant with the madelung constant $b_2(2)$ are:

$$a^{-1} = 2 \cdot 3 \cdot 11 \cdot 41 \cdot 43^{-1} \cdot b_2(2) \quad (5)$$

$$a^{-1} = -(2.706/43) \cdot b_2(2) \quad (6)$$

$$a^{-1} = -[(14^3 - 38) \cdot 43^{-1}] \cdot b_2(2) \quad (7)$$

with absolutely accurate numerical value:

$$a^{-1} = 137,035999078175526$$

$$a = 0,00729735256959$$

This accurate expression is the most impressive since it is simple and contains just a few prime numbers and the madelung constant. These prime numbers can be possibly connected to finite groups (Group of Lie type). The series representations for the fine-structure constant is:

$$\frac{2}{43} \times 3 \times 11 \times 41 \log(2) \pi = \frac{5412}{43} i \pi^2 \left[\frac{\arg(2-x)}{2\pi} \right] + \frac{2706}{43} \pi \log(x) - \frac{2706}{43} \pi \sum_{k=1}^{\infty} \frac{(-1)^k (2-x)^k x^{-k}}{k} \quad \text{for } x < 0$$

$$\frac{2}{43} \times 3 \times 11 \times 41 \log(2) \pi = \frac{2706}{43} \pi \left[\frac{\arg(2-z_0)}{2\pi} \right] \log\left(\frac{1}{z_0}\right) + \frac{2706}{43} \pi \log(z_0) + \frac{2706}{43} \pi \left[\frac{\arg(2-z_0)}{2\pi} \right] \log(z_0) - \frac{2706}{43} \pi \sum_{k=1}^{\infty} \frac{(-1)^k (2-z_0)^k z_0^{-k}}{k}$$

$$\frac{2}{43} \times 3 \times 11 \times 41 \log(2) \pi = \frac{5412}{43} i \pi^2 \left[\frac{\pi - \arg\left(\frac{1}{z_0}\right) - \arg(z_0)}{2\pi} \right] + \frac{2706}{43} \pi \log(z_0) - \frac{2706}{43} \pi \sum_{k=1}^{\infty} \frac{(-1)^k (2-z_0)^k z_0^{-k}}{k}$$

From this expression we pose:

$$a_b = (2.706/43) = 2 \cdot 3 \cdot 11 \cdot 41 \cdot 43^{-1}$$

This combination of prime numbers we will call the coefficient for the fine-structure constant. So the expression for the fine-structure constant can be written as:

$$a^{-1} = -a_b \cdot b_2(2)$$

The repeating decimal of the coefficient for the fine-structure constant a_b is:

$$\overline{62,930232558139534883720} \quad (\text{period } 21)$$

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